Supplementary Material

Novel conversions of a multifunctional, bio-sourced lactone carboxylic acid

Trinadh Kaicharla, Sangjun Lee, Ruiqin Wang, Ashok D. Pehere, Shu Xu, and Thomas R. Hoye*

Department of Chemistry, 207 Pleasant St. SE, University of Minnesota, Minneapolis, MN 55455 USA Email: <u>hoye@umn.edu</u>

Table of Contents

General Experimental Protocols	S2
Experimental Procedures for Synthesis and Characterization of All New Compounds	
Supplementary Material References	\$24
Copies of ¹ H and ¹³ C NMR Spectra	S25

General Experimental Protocols:

Nuclear magnetic resonance (NMR) spectra (¹H and ¹³C) were recorded on a Bruker HD-500 spectrometer. ¹H chemical shifts in CDCl₃ samples are referenced to TMS (δ 0.00), in acetone-*d*₆ samples to residual solvent protons at 2.05, DMSO-*d*₆ to residual solvent protons at 2.50, and in CD3CN to residual solvent protons at 1.94. Data are reported using the following format: chemical shift (ppm) [multiplicity, coupling constant(s) (in Hz), integral (to the nearest whole integer), and assignment of the proton]. Coupling constant values have been deduced using protocols previously described.^{1,2} Non-first order multiplets in a ¹H NMR spectrum are designated by 'nfom'. Non-first order doublets in a ¹H NMR spectrum (e.g., present in a 1,4-disubstitutedbenzene ring) are designated by 'nfod' and the apparent doublet coupling constant (actually, e.g., *J*_{2,3} + *J*_{2,3}') is indicated as *J*_{app}. ¹³C{¹H} NMR chemical shifts are referenced to the carbon atom in CDCl₃ to 77.16 ppm, in acetone-*d*₆ to 29.8 ppm, and in DMSO-*d*₆ to 39.5 ppm.

Infrared (IR) spectral data were collected on a Bruker spectrometer (model Alpha II). Samples were deposited as films on a diamond window (solids by evaporation from DCM; liquids by direct application) in the mode of attenuated total reflectance (ATR). Peaks are reported in cm⁻¹.

High-resolution mass spectrometry (HRMS) measurements were made using ESI ionization with a Thermo instrument (model Orbitrap Velos, which has a mass accuracy of \leq 3). An external calibrant (PierceTM LTQ) was used. Samples were injected directly into the ion source.

Medium pressure liquid chromatography (MPLC) was used to purify products. Silica gel (normal-phase, 20-40 µm, 60 Å pore size, Teledyne RediSep Rf Gold[®]) columns were hand-packed into Michel-Miller[®] glass columns. The equipment used consisted of a Waters HPLC pump (model 510) and a Waters (R401) differential refractive index detector. Preparative flash chromatography was done on silica gel (230-400 mesh) columns. Thin layer chromatography (TLC) was performed on silica-gel-coated, plastic-backed plates (Machery-Nagel) that were visualized by staining with KMnO₄ solution.

Heating of reactions performed above ambient temperature was done in silicone oil baths that had been preequilibrated to the desired temperature prior to immersion of the reaction vessel.

(±)-3a,7a-(Methanooxymethano)benzofuran-2,10(3*H*)-dione (5) and 3-Isochromanone (6)



The lactone acid 1 (200 mg, 0.95 mmol) was suspended in chloroform (10 mL) in a 20 mL Schlenk flask. Trifluoromethanesulfonic acid (TfOH, 40 μ L, 0.45 mmol) was added to the mixture. A brown color was immediately observed and this mixture was stirred at 80 °C for 30 min. The color of the solution turned dark brown and formation of a black precipitate was observed. The chloroform supernatant was filtered through short pad of silica that was eluted with 20 mL of additional chloroform. The obtained chloroform solution was concentrated in vacuo and the residue was purified by MPLC (7:3 hexanes:EtOAc elution) to give the rearranged dilactone **5** (68 mg, 0.38 mmol, 37%) as a white crystalline solid along with 3-isochromanone **6** (34 mg, 0.38 mmol, 24%), also as a white crystalline solid.

Data for the dilactone (5)

¹**H NMR** (500 MHz, Chloroform-*d*) δ 6.23–6.07 (nfom, 2*H*), 6.03–5.87 (nfom, 2*H*), 4.75 (dd, *J* = 10.8, 0.7 Hz, 1H, C8*H*_aH_b), 4.25 (d, *J* = 10.9 Hz, 1H, C8H_aH_b), 3.16 (dd, *J* = 18.1, 0.7 Hz, 1H, C3H_aH_b), and 2.75 (d, *J* = 18.1 Hz, 1H, C3H_aH_b).

¹³C{¹H} NMR (126 MHz, Chloroform-*d*) δ 176.5, 171.7, 125.2, 123.9, 123.3, 122.9, 87.5, 76.1, 51.0, and 38.6.

IR (neat): 1777, 1250, and 736 cm⁻¹.

HRMS (ESI) m/z: $[M + Na]^+$ Calcd for $C_{10}H_8NaO_4^+$ for 215.0320; Found 215.0308.

mp: 105–107 °C

3-Isochromanone (6) (NMR data matches well with literature values³):

¹**H NMR** (500 MHz, Chloroform-*d*) δ 7.36–7.29 (m, 2H), 7.26–7.21 (m, 2H), 5.31 (s, 2H, CH₂O), and 3.71 (s, 2H, CH₂CO).

¹³C{¹H} NMR (126 MHz, Chloroform-*d*) δ 170.8, 131.7, 131.1, 128.9, 127.5, 127.2, 124.8, 70.2 and 36.3.

IR (neat): 1746, 1408, 1250, 1222, 1030, and 775 cm⁻¹.

mp: 80–82 °C (lit.⁴ 80–81 °C).

(±)-2-((3a*R*,6*S*)-1-Oxotetrahydro-3*H*-3a,6-epoxyisobenzofuran-7a(1*H*)-yl)acetic acid (9)



The lactone acid **1** (4.9 g, 23.3 mmol) and Pd/C (10%, 350 mg) were placed in a 250 mL Fisher-Porter tube with a stirring bar. Under a nitrogen atmosphere, THF (100 mL) was added and the reaction vessel was purged with H₂. The reactor headspace was pressurized to ca. 10 psi with H₂ and then the pressure was slowly released. This cycle was repeated twice more. The reactor was then pressurized to 30 psi of H₂ and the mixture was stirred for 3 h at room temperature. The crude reaction mixture was filtered through a small plug of silica (eluant: 10 mL THF). The volatiles were removed under reduced pressure to provide a white solid, which was dried under vacuum to give the lactone acid **9** (4.8 g, 97%) as a white crystalline solid.

¹**H NMR** (500 MHz, Acetone- d_6) δ 4.59 (dd, J = 10.3 Hz, 1H, C₃HaHb), 4.51 (dd, J = 5.3, 5.3 Hz, 1H, H6), 4.41 (d, J = 10.3 Hz, 1H, C3H_a H_b), 2.85 (d, J = 15.6 Hz, 1H, H_a H_b), 2.60 (d, J = 15.6 Hz, 1H, C8H_a H_b), 2.22 (ddd, J = 12.4, 5.0, 2.4 Hz, 1H, C7H_{endo} H_{exo}), 2.09–2.01 (m, 1H), 1.89 (dddd, J = 12.3, 9.2, 5.6, 2.4 Hz, 1H), 1.82 (d, J = 12.4 Hz, 1H, C7 H_{endo} Hexo), and 1.77–1.66 (m, 2H).

¹³C{¹H} NMR (126 MHz, Acetone-*d*₆) δ 179.7, 172.1, 92.8, 76.9, 69.5, 53.8, 45.6, 39.9, 29.5, and 25.1.

IR (neat): 2978, 1767, 1732, 1134, and 1005 cm⁻¹.

HRMS (ESI) m/z: $[M + Na]^+$ Calcd for $C_{10}H_{12}NaO_5^+$ for 235.0582; Found 235.0566.

mp: 163–164 °C.

(±)-4,7-Dihydro-3a,7a-(methanooxymethano)benzofuran-2,10(3*H*)-dione (10b) and (±)-6,7-Dihydro-3a,7a-(methanooxymethano)benzofuran-2,10(3*H*)-dione (10a):



The lactone acid **9** (42 mg, 0.2 mmol) was suspended in chloroform (2 mL) in a 5 mL Schlenk flask. Trifluoromethanesulfonic acid (TfOH, 6 μ L, 30 mol%) was added. A pale brown color was immediately observed. This solution was stirred for 2 h at 80 °C. The chloroform was concentrated in vacuo; a ¹H NMR spectrum indicated a nearly 1:1 mixture of dilactone products. The residue was purified and partially separated by flash column chromatography on SiO₂ (2:1 hexanes:EtOAc elution) to give the dilactones **10b** (7 mg) as the faster eluting and **10a** (17 mg) as the slower eluting components (24 mg, 62% total yield), each as a colorless oil.

NMR Data for 10a

¹**H NMR** (500 MHz, Chloroform-*d*) δ 6.06 (ddd, J = 9.8, 5.7, 2.2 Hz, 1H, *H5*), 5.80 (ddd, J = 9.9, 2.8, 1.4 Hz, 1H, *H4*), 4.53 (d, J = 11.2, 1H, C8*H*_aH_b), 4.32 (d, J = 11.1 Hz, 1H, C8*H*_aH_b), 3.21 (d, J = 18.2, 1H, C3*H*_aH_b), 2.68 (d, J = 18.3 Hz, 1H, C3H_aH_b), 2.40 (ddddd, J = 18.7, 5.9, 5.9, 2.0, 1.5 Hz, 1H, C6H_aH_b), 2.27 (ddd, J = 14.0, 5.8, 1.9, 1H, C7*H*_aH_b), 2.19-2.11 (ddddd, J = 18.7, 11.8, 5.8, 2.4, 2.4 Hz 1H, C6H_aH_b), and 1.79 (ddd, J = 13.8, 11.9, 6.2 Hz, 1H, C7H_aH_b).

¹³C{¹H} NMR (126 MHz, Chloroform-*d*) δ 176.0, 172.2, 130.2, 122.5, 87.6, 71.4, 49.9, 38.4, 26.2, and 22.8.

NMR Data for 10b

¹**H** NMR (500 MHz, Chloroform-*d*) δ 5.97–5.91 (nfom, 1H, *H5* or *H6*), 5.91–5.87 (nfom, 1H, *H5* or *H6*), 4.59 (d, *J* = 10.9 Hz, 1H, C8*H*_aH_b), 4.20 (d, *J* = 10.8 Hz, 1H, C8H_aH_b), 3.10 (d, *J* = 18.3 Hz, 1H, C3*H*_aH_b), 2.74 (d, *J* = 18.4 Hz, 1H, C3H_aH_b), 2.68–2.56 (m, 2H), 2.50 (m, 2H).

¹³C{¹H} NMR (126 MHz, Chloroform-*d*) δ 178.7, 172.4, 125.6, 124.8, 88.0, 75.0, 48.1, 38.2, 31.0, and 29.8.

Data from the mixture:

IR (neat): 2924, 2853, 1770, 1191,1041, and 983 cm⁻¹.

HRMS (ESI) m/z: $[M + H]^+$ Calcd for $C_{10}H_{11}O_4^+$ for 195.0657; Found 195.0645.

(±)-2-((1a*R*,2*R*,6a*R*,6b*R*)-4-Oxotetrahydro-6*H*-2,6a-epoxyoxireno[2,3-*e*]isobenzofuran-3a(4*H*)-yl)acetic acid (11)



The lactone acid **1** (2.28 g) was added to a 100 mL round-bottom flask. To this was added 30% H_2O_2 (10 mL) and formic acid (36 mL). The resulting homogenous mixture was place into a preheated (60 °C) oil bath for 2 h. The resulting colorless solution was evaporated under vacuum to obtain a white, sticky, solid material. This solid was recrystallized from ethanol (10 mL) to provide the epoxide of **11** (2.1 g, 86%) as a white crystalline solid.

Caution: On one occasion, a large exotherm was observed following this similar procedure.⁵ We do not recommend using this procedure involving the *in situ* generation of the unstable performic acid. Therefore, we developed the following protocol.

A preferred procedure for this epoxidation reaction using peroxyacetic acid.

Lactone acid **1** (210 mg, 1 mmol) was added to a 20 mL glass vial. To this was added CH₃CO₃H in AcOH (2 mL, 32% peracid, ca. 8 mmol). The resulting homogenous mixture was stirred for 4 days at ambient temperature. To this colorless solution was added Et₂O (~4 mL) followed by hexanes (~8 mL). A white precipitate appeared. The supernatant liquid phase was decanted the precipitate was washed with a small amount of Et₂O. This solid was recrystallized from ethanol (2 mL) to provide the epoxide of **11** (185 mg, 82%) as a white crystalline solid.

¹**H NMR** (500 MHz, DMSO-*d*6): δ 12.64 (s, 1H, -CO₂*H*), 4.80 (d, *J* = 10.8 Hz, 1H, *H*6), 4.54 (d, *J* = 5.2 Hz, 1H, *H*2), 4.48 (d, *J* = 10.8 Hz, 1H, *H*6'), 3.78 (d, *J* = 3.3 Hz, 1H, *H1* or *H7*), 3.67 (d, *J* = 3.3 Hz, 1H, *H1* or *H7*), 2.68 (d, *J* = 15.3 Hz, 1H, CH2CO₂H), 2.63 (d, *J* = 15.3 Hz, 1H, *H3*_{endo}).

¹³C{¹H} NMR (126 MHz, DMSO-*d*₆) δ 177.1, 171.0, 88.4, 74.6, 67.4, 54.6, 48.6, 46.1, 38.5, and 36.0.

IR (neat): 3083 (br), 1769, 1730, 1004, and 862 cm⁻¹.

HRMS (ESI) m/z: $[M + H]^+$ Calcd for $C_{10}H_{11}O_6^+$ for 227.0556; Found 227.0543.

mp: 158–159 °C.

 $(\pm) \textbf{-3-Bromodihydro-} 2H, 7H, 9H-2, 6a-methanodifuro [3, 2-b: 3', 4'-c] pyran-5, 7(6H)-dione\ (12a) and \\$

 $(\pm) - 10 - Bromodihydro-7H, 9H-2, 6a: 3, 9a - dimethanofuro [3, 4-e][1, 4] dioxocine-5, 7(6H) - dione (12b)$



The lactone acid **1** (210 mg, 1.0 mmol) was dissolved in acetone (5 mL) in a 20 mL Schlenk flask. *N*-Bromosuccinimide (196 mg, 1.1 mmol) was added to the mixture stirred at room temperature for overnight. The reaction mixture was concentrated in vacuo; a proton NMR spectrum of this crude product mixture indicated the presence of an ca. 6:1 mixture of **12a** to **12b**. The residue was purified by MPLC (6:4 hexanes:EtOAc elution) to give a mixture of these largely coeluting bromolactones **12a and 12b** (210 mg, 73%) as a broad melting white crystalline solid.

NMR Data for major bromolactone 12a (extracted from the spectrum of the mixture).

¹**H NMR** (500 MHz, Chloroform-*d*) δ 5.14 (dd, *J* = 1.4, 0.8 Hz, 1H, *H3a*), 4.80 (d, *J* = 11.4 Hz, 1H, *H9*), 4.70 (dd, *J* = 5.5, 0.8 Hz, 1H, *H2*), 4.64 (d, *J* = 11.6 Hz, 1H, *H9'*), 4.03 (d, *J* = 1.6 Hz, 1H, *H3*), 3.18 (d, *J* = 18.8 Hz, 1H, *H6*), 2.83 (dd, *J* = 13.9, 5.7 Hz, 1H, *H10_{exo}*), 2.73 (d, *J* = 18.7 Hz, 1H, *H6'*), and 1.96 (d, *J* = 13.9 Hz, 1H, *H10_{endo}*).

¹³C{¹H} NMR (126 MHz, Chloroform-*d*) δ 175.0 (C7), 163.7 (C5), 85.9 (C3a), 84.9 (9 a), 83.7 (C2), 65.7 (C9), 52.2 (C3), 48.6 (C6a), 40.3 (C10), and 32.6 (C6).

HMBC HSQC

NMR Data for minor bromolactone 12b (extracted from the spectrum of the mixture).

¹**H** NMR (500 MHz, Chloroform-*d*) δ 5.05 (nfom, 1H, *H3*), 4.64 (d, *J* = 11.9 Hz, 1H, H9), 4.55 (s, 1H, H10), 4.47 (d, *J* = 11.9 Hz, 1H, H9'), 3.41 (d, *J* = 18.7 Hz, 1H, *H6*), 3.03 (ddd, *J* = 18.6, 2.3, 1.0 Hz, 1H, *H6'*), 2.45 (dddd, *J* = 12.7, 2.4, 2.4, 1.5 Hz, 1H, *H11_{exo}*), and 2.34 (dd, *J* = 12.7, 1.1 Hz, 1H, *H11_{endo}*). The resonance for H2 could not be definitively identified.

Data for the mixture:

IR (neat): 3010, 1778, 1750, 1707, 1375, 1190, 1062, 1009, and 585 cm⁻¹.

HRMS (ESI) m/z: $[M + H]^+$ Calcd for $C_{10}H_{10}BrO_5^+$ for 288.9712; Found 288.9697.

mp: 188–194 °C.

(±)-3-Iododihydro-2*H*,7*H*,9*H*-2,6a-methanodifuro[3,2-b:3',4'-c]pyran-5,7(6*H*)-dione (13a)



The lactone acid **1** (42 mg, 0.20 mmol) *N*-iodosuccinimide (44 mg, 0.22 mmol) were combined in a NMR tube and acetone- d_6 (0.6 mL) was added. The pale brown homogenous mixture was allowed to stand at room temperature. After 4 days NMR analysis indicated ~40% of unreacted starting acid along with unreacted NIS. The reaction mixture was concentrated in vacuo and the residue was purified by MPLC (6:4 hexanes:EtOAc elution) to give a the iodolactone **13a** (35 mg, 52%) as a white crystalline solid. This sample contained ca. 4% of a contaminant with many analogous resonances in the ¹H NMR spectrum, although its structure could not be definitively identified as that of an isomeric iodolactone (cf. **12b** vs. **12a**).

NMR Data for iodolactone 13a

¹**H NMR** (500 MHz, Chloroform-*d*) δ 5.25 (dd, *J* = 2.0, 0.6 Hz, 1H, *H3a*), 4.79 (d, *J* = 11.5 Hz, 1H, *H9*), 4.75 (dd, *J* = 5.2, 0.6 Hz, 1H, *H2*), 4.64 (d, *J* = 11.5 Hz, 1H, *H9'*), 3.99 (d, *J* = 2.0 Hz, 1H, *H3*), 3.18 (d, *J* = 18.7 Hz, 1H, *H6*), 2.734 (dd, *J* = 13.8, 5.4 Hz, 1H, *H10_{exo}*), 2.733 (d, *J* = 18.7 Hz, 1H, *H6'*), and 1.99 (d, *J* = 13.9 Hz, 1H, *H10_{endo}*).

¹³C{¹H} NMR (126 MHz, Chloroform-*d*) δ 175.2, 163.9, 87.6, 85.3, 85.2, 65.6, 48.5, 41.5, 32.7, and 24.6.

IR (neat): 3003, 1777, 1749, 1221, 1193, and 1121.

HRMS (ESI) *m/z*: [M + H]⁺ Calcd for C₁₀H₁₀IO₅⁺ for 336.9573; Found 336.9558. **mp**: 184–186 °C. (±)-(3a*R*,5*R*,7a*R*)-5-Hydroxy-4,5-dihydro-3a,7a-(methanooxymethano)benzofuran-2,10 (3*H*)-dione (7b)



The lactone acid **1** (210 mg, 1.0 mmol) and $Sc(OTf)_3$ (49 mg, 0.1 mmol) were combined in a screw-capped culture tube. Acetonitrile (5.0 mL) was added. This mixture was placed in a preheated (70 °C) oil bath and kept there for 6 h. The brownish homogenous mixture was evaporated under vacuum and the crude material was purified by flash column chromatography (hexanes:EtOAc 1:1) to provide a sample of **7b** as a white crystalline solid, which contained an impurity that appeared to be itaconic acid (ca. 7 mol%). This material was recrystallized from ethanol to give (102 mg, 49%) of **7b**.

¹**H NMR** (500 MHz, Acetone-*d*₆): δ = 6.15 (ddd, *J* = 10.3, 2.2, 1.1 Hz, 1H, *H*6), 5.91 (dd, *J* = 10.3, 2.1 Hz, 1H, *H7*), 4.65 (dd, *J* = 11.1, 0.6 Hz, 1H, *H8*), 4.64 (dddd, *J* = 9.6, 5.0, 2.1, 2.1 Hz, 1H, *H5*), 4.49 (d, *J* = 11.2 Hz, 1H, *H8'*), 3.07 (d, *J* = 18.4 Hz, 1H, *H3*), 2.82 (d, *J* = 18.4 Hz, 1H, *H3'*), 2.36 (ddd, *J* = 13.6, 5.1, 1.2 Hz, 1H, *H4_{eq}*), and 2.21 (dd, *J* = 13.5, 9.4 Hz, 1H, *H4_{ax}*).

¹³C{¹H} NMR (126 MHz, Acetone-*d*₆): δ 178.2, 172.7, 138.2, 123.1, 86.6, 74.7, 62.8, 49.8, 35.5 and 34.6.

IR (neat): 3439, 1776, 1251, 1061, and 936 cm⁻¹.

HRMS (ESI) m/z: $[M + Na]^+$ Calcd for $C_{10}H_9NaO_5^+$ for 233.0426; Found 233.0411.

mp: 144–145 °C.

(±)-(3a*R*,5*S*,7a*R*)-5-Hydroxytetrahydro-3a,7a-(methanooxymethano)benzofuran-2,10(3*H*)-dione (14):



A mixture of $Sc(OTf)_3$ (5 mg, 0.01mmol) and the lactone acid **9** (21 mg, 0.1 mmol) were taken into a screw-capped culture tube. Acetonitrile-d₃ (0.6 mL) was added. The resulting colorless homogenous solution was heated at 80 °C for 5 h. The solvent was evaporated under vacuum from the resulting colorless solution and the crude material was eluted through a silica gel plug with the aid of EtOAc. The eluent was concentrated and gave **14** (17 mg, 80%) as a white crystalline solid. The proton NMR spectrum of this reaction mixture showed a very clean conversion to the set of resonances for the product **14**.

¹**H NMR** (500 MHz, Acetone-*d*₆): δ 4.64 (d, *J* = 11.3 Hz, 1H, *H*8), 4.45 (dd, *J* = 11.3, 0.6 Hz, 1H, *H*8'), 3.97 (dddd, *J* = 10.5, 10.5, 4.0, 4.0 Hz, 1H, *H*5), 3.09 (d, *J* = 18.0 Hz, 1H, *H*3), 2.79 (dd, *J* = 18.0, 0.7 Hz, 1H, *H*3'), 2.9 (br s, 1H, OH), 2.29 (ddd, *J* = 14.2, 4.4, 3.6 Hz, 1H, *H*7_{eq}), 2.23 (ddd, *J* = 13.8, 4.0, 2.2, Hz, 1H, *H*4_{eq}), 1.98 (ddddd, *J* = 13.5, 4.6 3.8, 3.8, 2.2 Hz, 1H, *H*6_{eq}), 1.90 (ddd, *J* = 14.5, 13.4, 4.7 Hz, 1H, *H*7_{ax}), 1.62 (ddd, *J* = 13.8, 10.6, 0.4 Hz, 1H, *H*4_{ax}), and 1.50 (dddd, *J* = 13.5, 13.5, 10.2, 4.2 Hz, 1H, *H*6_{ax}).

¹**H NMR** (500 MHz, MeCN-*d*₃): δ 4.49 (d, *J* = 11.2 Hz, 1H, *H*8), 4.42 (dd, *J* = 11.3, 0.5 Hz, *H*8'), 3.80 (dddd, *J* = 10.3, 10.3, 3.8, 3.8 Hz, 1H, *H*5), 2.92 (d, *J* = 18.1 Hz, 1H, *H*3), 2.80 (dd, *J* = 18.1, 0.6 Hz, 1H, *H*3'), 2.22 (ddd, *J* = 14.5, 4.3, 3.8 Hz, 1H, *H*7_{eq}), 2.15 (ddd, *J* = 13.8, 3.9, 2.2, Hz, 1H, *H*4_{eq}), 1.91 (ddddd, *J* = 13.5, 4.8 3.7, 3.7, 2.2 Hz, 1H, *H*6_{eq}), 1.79 (ddd, *J* = 14.4, 13.3, 4.7 Hz, 1H, *H*7_{ax}), 1.55 (ddd, *J* = 13.9, 10.5, 0.4 Hz, 1H, *H*4_{ax}), and 1.36 (dddd, *J* = 13.5, 13.5, 10.1, 4.1 Hz, 1H, *H*6_{ax}).

13C{1H} NMR (126 MHz, Acetone-d6): δ 178.2, 172.0, 87.3, 70.9, 63.7, 49.2, 35.4, 35.2, 30.4, and 27.6.

IR (neat): 3052, 3001, 2952, 2899, 2835, 2228, 1702, 1156, 1113, 1020, 999, 883, and 764 cm⁻¹. **HRMS** (ESI) *m/z*: [M + Na]⁺ Calcd for C₁₀H₁₂NaO₅⁺ for 235.0582; Found 235.0568. **mp**: 116–117 °C.

ARKIVOC 2024, *ii*, S1-S86

(±)-3-Hydroxydihydro-2*H*,7*H*,9*H*-2,6a-methanodifuro[3,2-*b*:3',4'-*c*]pyran-5,7(6*H*)-dione (15) Method A:



The epoxy acid **11** (1.59 g, 6.2 mmol) was added to a 250 mL round bottom flask. THF (70 mL) was added, followed by BF₃.OEt₂ (0.44 mL, 3.5 mmol). This mixture was placed in a preheated (70 °C) oil bath and kept there for 12 h. The resulting colorless solution was evaporated under vacuum to obtain a white sticky solid. This solid was triturated with Et₂O, aided by sonication. The white solid was filtered under vacuum and washed several times with additional fresh Et₂O. The solid was finally recrystallized from ethanol to provide the sample of **15** (1.35 g, 85%) as a white crystalline solid.

Method B:



The lactone epoxide **11** (22 mg, 0.1 mmol) and Sc(OTf)₃ (4.9 mg, 0.01 mmol) were combined in a screw-capped culture tube. Acetonitrile- d_3 (0.6 mL) was added to the NMR tube. The colorless homogenous mixture was monitored at room temperature. After full consumption of the starting epoxide (36 h), the crude material was passed through a short pad of silica and eluted with 5 mL of EtOAc. The solution was evaporated to provide the white solid material, which was then washed with several mL of Et₂O to give the sample of **15** (19 mg, 84%) as a white crystalline solid.

¹**H** NMR (500 MHz, Acetone- d_6) δ 4.92 (dd, J = 11.4, 0.5 Hz, 1H, $H9\beta$), 4.66 (d, J = 11.4 Hz, 1H, $H9\alpha$), 4.65 (dd, J = 1.2, 1.2 Hz, 1H, H3a), 4.42 (dd, J = 5.8, 1.3 Hz, 1H, H2), 4.13 (dd, J = 1.2 Hz, 1H, H3), 3.05 (d, J = 18.7 Hz, 1H, $H6\alpha$), 2.94 (dd, J = 18.6, 0.6 Hz, 1H, $H6\beta$), 2.47 (dd, J = 13.6, 5.9 Hz, 1H, $H1\alpha$), and 2.06 (d, J = 13.7 Hz, 1H, $H1\beta$).

¹³C{1H} NMR (126 MHz, Acetone- d_6) δ 177.3 (C7), 165.4 (C5), 85.9 (3a), 84.6 (9a), 84.0 (C2), 81.2 (C3), 66.9 (C9), 49.6 (6a), 38.0 (C1), and 32.5 (C6).

Assignments based on: HSQC, HMBC, NOESY correlations.

IR (neat): 3453, 1776, 1745, 1378, 1223, 1065, and 1008 cm⁻¹.

HRMS (ESI) m/z: $[M + Na]^+$ Calcd for $C_{10}H_{10}NaO_6^+$ for 249.0375; Found 249.0357.

mp: 196–198 °C.

ARKIVOC 2024, *ii*, S1-S86

(±)-(3a*R*,5*R*,7a*R*)-2,10-Dioxo-2,3,4,5-tetrahydro-3a,7a-(methanooxymethano)benzofuran-5-yl acrylate (16a)



A mixture of the lactone acid **1** (210 mg, 1.0 mmol) and $Sc(OTf)_3$ (49 mg, 0.1 mmol) were combined in a screw-capped culture tube. Acetonitrile (5.0 mL) and acrylic anhydride (230 mL, 1.95 mmol) were added. The homogenous mixture was stirred at room temperature for 20 h. The solvent was evaporated under vacuum and the crude material was purified by MPLC (hexanes:EtOAc 4:6) to provide the sample of **16a** (176 mg, 67%) as a white crystalline solid.

¹**H NMR** (500 MHz, Chloroform-*d*): $\delta = 6.42$ (dd, J = 17.3, 1.5 Hz, 1H, $H_ZH_EC=C$), 6.20 (ddd, J = 10.3, 3.2, 0.8 Hz, 1H, *H6*), 6.10 (dd, J = 10.3, 1.6 Hz, 1H, *H7*), 6.09 (dd, J = 17.3, 10.5 Hz, 1H, =CHC=O), 5.92 (dd, J = 10.5, 1.3 Hz, 1H, H_Z $H_EC=C$), 5.56 (dddd, J = 8.1, 4.7, 3.2, 1.6 Hz, 1H, *H5*), 4.69 (d, J = 11.0 Hz, 1H, *H8*), 4.38 (d, J = 11.0 Hz, 1H, *H8'*), 3.09 (d, J = 18.6 Hz, 1H, *H3*), 2.84 (d, J = 18.6 Hz, 1H, *H3'*), 2.40 (ddd, J = 13.9, 4.7, 0.8 Hz, 1H, *H4_{eq}*) and 2.21 (dd, J = 13.9, 8.1 Hz, 1H, *H4_{ax}*).

¹³C{¹H} NMR (126 MHz, Chloroform-d): δ 176.5, 171.4, 165.1, 132.6, 131.7, 127.5, 126.4, 84.7, 74.9, 64.8, 47.8, 36.2 and 31.5.

IR (neat): 2989, 1781, 1721, 1634, 1189, 1051, and 941 cm⁻¹.

HRMS (ESI) m/z: $[M + H]^+$ Calcd for $C_{13}H_{13}O_6^+$ for 265.0712; Found 265.0696.

mp: 122–123 °C.

(±)-(3a*R*,5*R*,7a*R*)-2,10-Dioxo-2,3,4,5-tetrahydro-3a,7a-(methanooxymethano)benzofuran-5yl methacrylate (16b)



A mixture of the lactone acid **1** (2.2 g, 10.5 mmol) and $Sc(OTf)_3$ (515 mg, 1.05 mmol) were combined in a screw-capped culture tube. Acetonitrile (30 mL) and methacrylic anhydride (2.2 mL, 15.7 mmol) were added. The homogenous mixture was stirred at room temperature for 16 h. The solvent was evaporated under vacuum and the crude material was purified by flash column chromatography (hexanes:EtOAc 4:6) to provide the sample of **16b** (1.9 gm, 65%) as a white crystalline solid.

¹**H NMR** (500 MHz, Chloroform-*d*): δ 6.24 (ddd, J = 10.2, 3.4, 0.6 Hz, 1H, *H*6), 6.12 (dd, J = 10.3, 1.4 Hz, 1H, *H7*), 6.09 (dq, J = 1.1, 1.1 Hz, 1H, *H*_ZH_EC=C), 5.65 (dq, J = 1.5, 1.5 Hz, 1H, H_ZH_EC=C), 5.53 (dddd, J = 7.5, 4.8, 3.4, 1.4 Hz, 1H, *H5*), 4.69 (d, J = 10.9 Hz, 1H, *H8*), 4.38 (d, J = 11.0 Hz, 1H, *H8'*), 3.11 (d, J = 18.6 Hz, 1H, *H3*), 2.85 (d, J = 18.6 Hz, 1H, *H3'*), 2.35 (ddd, J = 1.7, 1.0 Hz, 3H, -CH₃).

¹³C{¹H} NMR (126 MHz, Chloroform-d): δ 176.6, 171.5, 166.4, 135.5, 131.6, 127.2, 126.7, 84.7, 75.3, 64.8, 47.6, 36.7, 31.8, and 18.3.

IR (neat): 1781, 1713, 1635, and 1168 cm⁻¹.

HRMS (ESI) m/z: $[M + H]^+$ Calcd for $C_{14}H_{15}O_6^+$ for 279.0869; Found 279.0851.

mp: 105–106 °C.

(±)-(3aR,5R,7aR)-2,10-Dioxo-2,3,4,5-tetrahydro-3a,7a-(methanooxymethano)benzofuran-5yl Acetate (16c):



A mixture of $Sc(OTf)_3$ (49 mg, 0.1mmol) and the lactone acid **1** (210 mg, 1 mmol) were taken into a screw-capped culture tube. Acetonitrile (5 mL) and acetic anhydride (190 µL, 2.0 mmol) were added. The dark green-colored homogenous mixture was stirred at room temperature for 16 h. The solvent was evaporated under vacuum and the crude material was purified by flash column chromatography (hexanes:EtOAc, 1:1) to provide the sample of **16c** (130 mg, 51%) as a sticky oil.

¹**H NMR** (500 MHz, **CDCl**₃): 6.13 (ddd, *J* = 10.4, 3.0, 0.8 Hz, 1H, *H***6**), 6.07 (ddd, *J* = 10.3, 1.6 Hz, 1H, *H***7**), 5.49 (dddd, *J* = 8.1, 4.7, 3.0, 1.6 Hz, 1H, *H*5), 4.67 (dd, *J* = 11.0, 0.5 Hz, 1H, *H*8), 4.37 (d, *J* = 11.0 Hz, 1H, *H*8'), 3.06 (d, *J* = 18.6 Hz, 1H, *H*3), 2.82 (d, *J* = 18.5 Hz, 1H, *H*3'), 2.38 (ddd, *J* = 14.0, 4.8, 0.8 Hz, 1H, *H*4), 2.12 (dd, *J* = 14.0, 8.3 Hz, 1H, *H*4'), and 2.07 (s, 3H).

Assignments based on HMBC HSQC correlations.

¹³C{1H} NMR (126 MHz, CDCl₃) δ 176.5 (C10), 171.4 (C2), 170.1 (CH₃CO₂), 131.8 (C6), 126.1 (C7), 84.7 (C7a), 74.7 (C8), 64.6 (C5), 47.9 (C3a), 36.0 (C3), 31.2 (C4), and 20.9 (CH₃CO₂).

IR (neat): 3023, 2937, 1775, 1732, 1373, 1224, 1009, and 938 cm⁻¹.

HRMS (ESI) m/z: $[M + Na]^+$ Calcd for $C_{12}H_{12}NaO_6^+$ for 275.0532; Found 275.0515.

(3a*R*,5*S*,7a*R*)-2,10-Dioxohexahydro-3a,7a-(methanooxymethano)benzofuran-5-yl acrylate (17a)



A mixture of Sc(OTf)₃ (38 mg, 0.094 mmol) and the lactone acid **9** (200 mg, 0.94 mmol) was taken into a screw-capped culture tube. Acetonitrile (2 mL) was added followed by acrylic anhydride (178 μ L 1.9 mmol). The colorless homogeneous mixture was stirred at room temperature for 16 h. The solvent was evaporated under vacuum and the crude material was purified by flash column chromatography (hexanes:EtOAc, 2:3) to provide the sample of **17a** (186 mg, 0.70 mmol 75%) as a white crystalline solid.

¹**H NMR** (500 MHz, Acetone-*d*₆): $\delta = 6.35$ (dd, *J*= 17.2, 1.5 Hz, 1H, *H*_ZH_EC=C), 6.11 (dd, *J*= 17.2, 10.5 Hz, 1H, CHCOO), 5.90 (dd, *J*= 10.5, 1.5 Hz, 1H, *H*_ZH_EC=C), 5.11 (dddd, *J*= 9.9, 9.9, 3.9, 3.9, 1H, H5), 4.70 (d, *J* = 11.3 Hz, 1H, *H8*),, 4.50 (d, *J* = 11.3, 0.5 Hz, 1H, *H8*'), 3.20 (d, *J* = 18.0 Hz, 1H, *H3*), 2.89 (dd, *J*=18.0, 0.6 Hz, 1H, *H3*'), 2.39–2.34 (nfom, 1H, *H6^{exo}*), 2.13–2.06 (two nfom, 2H, *H6^{endo}* and *H7*), 2.34 (ddd, *J* = 13.8, 4.0, 1.8 Hz, 1H, *H4^{exo}*), 1.95 (ddd, *J* = 13.9, 9.9, 0.4 Hz, 1H, *H4^{endo}*), and 1.84–1.70 (nfom, 1H, *H7*').

¹³**C NMR** (126 MHz, Acetone-*d*₆): δ = 178.3, 172.6, 165.5, 131.6, 129.3, 87.5, 72.2, 68.2, 49.3, 36.4, 32.5, 27.8, and 27.2.

IR (neat): 2949, 2874, 1779, 1719, 1409, 1190, and 764 cm⁻¹.

HRMS (ESI) m/z: $[M + H]^+$ Calcd for $C_{13}H_{15}O_6^+$ for 267.0869; Found 267.0851.

mp: 138–140 °C.

(±)-(3a*R*,5*S*,7a*R*)-2,10-Dioxohexahydro-3a,7a-(methanooxymethano)benzofuran-5-yl acetate (17c)



A mixture of $Sc(OTf)_3$ (215 mg, 0.47 mmol) and the lactone acid **9** (1 g, 4.7 mmol) were taken into a screw-capped culture tube. Acetonitrile (8 mL) followed by acetic anhydride (1.1 mL, 9.4 mmol) was added. The colorless homogenous mixture was stirred at room temperature for 16 h. The solvent was evaporated under vacuum and the crude material was purified by flash column chromatography (hexanes:EtOAc, 4:6) to provide the sample of **17c** (980 mg, 82%) as a white crystalline solid.

¹**H NMR** (500 MHz, **CDCl**₃): 4.91 (dddd, *J* = 9.3, 8.9, 3.9, 3.9 Hz, 1H, *H5*), 4.48 (dd, *J* = 11.1, 0.7 Hz, 1H, *H8*), 4.38 (d, *J* = 11.0 Hz, 1H, *H8'*), 3.03 (dd, *J* = 18.3, 0.7 Hz, 1H, *H3*), 2.78 (d, *J* = 18.3 Hz, 1H, *H3'*), 2.28 (ddd, *J* = 14.9, 4.7, 4.4 Hz, 1H, *H7*), 2.21 (ddd, *J* = 14.3, 3.9, 1.9 Hz, 1H, *H4*), 2.08 (ddddd, *J* = 14.0, 4.9, 4.9, 3.9, 1.9 Hz, 1H, *H6*), 2.03 (s, 3H, COC*H*₃), 1.90 (dd, *J* = 14.1, 9.2 Hz, 1H, *H4'*), 1.89 (dd, *J* = 14.9, 12.1, 5.0 Hz, 1H, *H7'*), and 1.54 (dddd, *J* = 13.9, 11.9, 8.8, 4.5 Hz, 1H, *H6'*).

¹³C{1H} NMR (126 MHz, CDCl₃) δ 176.8 (C10), 171.6 (C2), 170.1 (CH₃CO), 86.2 (C7a), 71.9 (C8), 66.5 (C5), 48.0 (C3a), 36.4 (C3), 32.5 (C4), 27.7 (C7), 26.6 (C6), and 21.1 (CH₃CO).

Assignments based on HMBC HSQC correlations.

IR (neat): 1780, 1750, 1727, 1185, and 1063 cm⁻¹.

HRMS (ESI) m/z: $[M + Na]^+$ Calcd for $C_{12}H_{14}NaO_6^+$ for 277.0688; Found 277.0672.

mp: 125–127 °C.

In a separate experiment, this transformation was monitored over time by in situ ¹H NMR spectroscopy in CD₃CN at ambient temperature. $Sc(OTf)_3$ (10 mol%), Ac_2O (18 uL), and acid 1 were used. The reaction was very selective for formation of the acetate **17c**. A copy of the spectrum of the reaction mixture sample is included at the end of this SI. The following spectral data were taken from a NMR sample of purified **17c** in CD₃CN.

¹**H** NMR (500 MHz, **CD₃CN**): 4.91 (dddd, *J* = 9.8, 9.8, 4.0, 4.0 Hz, 1H, *H5*), 4.51 (d, *J* = 11.2 Hz, 1H, *H8*), 4.38 (dd, *J* = 11.4, 0.8 Hz, 1H, *H8*'), 2.96 (d, *J* = 18.2 Hz, 1H, *H3*), 2.78 (dd, *J* = 18.3, 0.8 Hz, 1H, *H3'*), 2.26 (ddd, *J* = 14.6, 4.2, 4.2 Hz, 1H, *H7*), 2.20 (ddd, *J* = 14.0, 4.0, 2.0 Hz, 1H, *H4*), 2.00 (ddddd, *J* = 13.5, 4.9, 4.2, 4.2 1.9 Hz, 1H, *H6*), 1.97 (s, 3H, COC*H*₃), 1.91 (ddd, *J* = 14.7, 12.6, 4.9 Hz, 1H, *H7'*), 1.83 (dd, *J* = 14.0, 10.0 Hz, 1H, *H4'*), and 1.54 (dddd, *J* = 13.7, 12.7, 9.5, 4.3 Hz, 1H, *H6'*).

ARKIVOC 2024, ii, S1-S86

(±)-(3a*R*,5*S*,7a*R*)-2,10-Dioxohexahydro-3a,7a-(methanooxymethano)benzofuran-5-yl methacrylate (17b)



A mixture of the lactone acid **9** (5.0 g, 23.6 mmol) and $Sc(OTf)_3$ (580 mg, 1.18 mmol) were combined in a screw-capped culture tube. Acetonitrile (50 mL) and methacrylic anhydride (5.3 mL, 35.3 mmol) were added. The homogenous mixture was stirred at room temperature for 16 h. The solvent was evaporated under vacuum and the crude material was purified by flash column chromatography (hexanes:EtOAc 4:6) to provide the sample of **17b** (5.1 gm, 76%) as a white crystalline solid.

¹**H NMR** (500 MHz, Chloroform-d): 6.08 (dq, J = 1.5, 1.0 Hz, 1H, $H_ZH_EC=C$), 5.61 (dq, J = 1.5, 1.5 Hz, 1H, $H_ZH_EC=C$), 5.01 (dddd, J = 8.7, 8.7, 3.9, 3.9 Hz, 1H, H5), 4.49 (dd, J = 11.0, 0.4 Hz, 1H, H8), 4.40 (d, J = 11.0 Hz, 1H, H8'), 3.06 (dd, J = 18.3, 0.6 Hz, 1H, H3), 2.81 (d, J = 18.3 Hz, 1H, H3'), 2.29 (ddd, J = 14.8, 5.5, 4.5 Hz, 1H, $H7_{eq}$), 2.24 (ddd, J = 14.4, 3.8, 1.6 Hz, 1H, $H4_{eq}$), 2.13 (ddddd, J = 14.1, 5.2, 5.1, 4.1, 1.6 Hz, 1H, $H6_{eq}$), 2.03 (dd, J = 14.3, 8.8 Hz, 1H, $H4_{ax}$), 1.96 (ddd, J = 14.9, 11.5, 4.8 Hz, 1H, $H7_{ax}$), 1.92 (dd, J = 1.0, 1.5 Hz, 3H, -CH₃), and 1.63 (dddd, J = 14.2, 11.5, 8.5, 4.5 Hz, 1H, H6').

¹³C{¹H} NMR (126 MHz, Chloroform-d): δ 176.7 (C10), 171.7 (C2), 166.4 (=CC=O), 135.8 (=CC=O), 126.7 (=CH₂), 86.2 (C7a), 72.1 (C8), 66.8 (C5), 47.9 (C3a), 36.7 (C3), 32.7 (C4), 27.6 (C7), 26.4 (C6), and 18.3 (CH₃).

Assignments based on HMBC HSQC correlations.

IR (neat): 2957, 1781, 1714, 1635, and 1168 cm⁻¹.

HRMS (ESI) m/z: $[M + H]^+$ Calcd for $C_{14}H_{17}O_6^+$ for 281.1025; Found 281.1006.

mp: 155–157 °C.

(±)-(3aR,5S,7aR)-2,10-Dioxohexahydro-3a,7a-(methanooxymethano)benzofuran-5-yl Benzoate (17d):



A mixture of Sc(OTf)₃ (25 mg, 0.05 mmol) and the lactone acid **9** (106 mg, 0.5 mmol) were taken into a screw-capped culture tube. Acetonitrile (3 mL) and benzoyl chloride (116 μ L, 1.0 mmol) were added. The colorless homogenous mixture was stirred at room temperature for 16 h. The solvent was evaporated under vacuum and the crude material was purified by flash column chromatography (hexanes:EtOAc, 1:1) to provide the **17d** (128 mg, 81%) as a white crystalline solid.

¹**H NMR** (500 MHz, **CDCl**₃): 7.98 (nfom, 2H, $Ph_{o+o'}$), 7.63 (tt, J = 7.5, 1.4 Hz, 1H, Ph_p), 7.42 (nfom, 2H, $Ph_{m+m'}$), 5.19 (dddd, J = 9.0, 9.0, 3.9, 3.9 Hz, 1H, H5), 4.52 (dd, J = 11.0, 0.6 Hz, 1H, H8), 4.44 (d, J = 11.0 Hz, 1H, H8'), 3.09 (dd, J = 18.2, 0.6 Hz, 1H, H3), 2.78 (d, J = 18.3 Hz, 1H, H3'), 2.36 (ddd, J = 14.3, 4.0, 1.8 Hz, 1H, $H4_{eq}$), 2.33 (ddd, J = 14.9, 5.1, 4.4 Hz, 1H, $H7_{eq}$), 2.23 (dddddd, J = 14.0, 4.9, 4.9, 3.8, 1.8 Hz, 1H, $H6_{eq}$), 2.10 (dd, J = 14.3, 9.1 Hz, 1H, $H4_{ax}$), 2.0 (ddd, J = 14.9, 11.9, 4.9 Hz, 1H, $H7_{ax}$), and 1.72 (dddd, J = 14.1, 11.8, 8.7, 4.4 Hz, 1H, $H6_{ax}$).

¹³C{1H} NMR (126 MHz, CDCl₃) δ 176.8, 171.7, 165.7, 133.6, 129.8, 129.5, 128.7, 86.3, 72.0, 67.2, 48.0, 36.6, 32.8, 27.7 and 26.7.

IR (neat): 1780, 1714, 1274, 1113, and 714 cm⁻¹.

HRMS (ESI) m/z: $[M + Na]^+$ Calcd for $C_{17}H_{16}NaO_6^+$ for 339.0845; Found 339.0823.

mp: 143–144 °C.

(±)-(3aR,5S,7aR)-2,10-Dioxohexahydro-3a,7a-(methanooxymethano)benzofuran-5-yl Pivalate (17e):



A mixture of Sc(OTf)₃ (25 mg, 0.05 mmol) and the lactone acid **9** (106 mg, 0.5 mmol) were taken into a screw-capped culture tube. Acetonitrile (3 mL) and pivaloyl chloride (123 μ L, 1.0 mmol) were added. The colorless homogenous mixture was stirred at room temperature for 20 h. The solvent was evaporated under vacuum and the crude material was purified by flash column chromatography (hexanes:EtOAc, 1:1) to provide the pivalate **17e** (105 mg, 71 %) as a white crystalline solid.

¹**H NMR** (500 MHz, **CDCl**₃): δ 5.0 (dddd, J = 9.3, 9.3, 4.0, 4.0 Hz, 1H, H5), 4.68 (dd, J = 11.2, 0.5 Hz, 1H, H8), 4.50 (d, J = 11.2 Hz, 1H, H8'), 3.13 (dd, J = 18.1, 0.5 Hz, 1H, H3), 2.90 (d, J = 18.1 Hz, 1H, H3'), 2.32 (ddd, J = 14.2, 4.5, 1.9 Hz, 1H, $H4_{eq}$), 2.25 (ddd, J = 14.0, 4.0, 1.8 Hz, 1H, $H7_{eq}$), 2.08 (dd, J = 14.4, 12.1, 4.7 Hz, 1H, $H7_{ax}$), 2.03 (dddd, J = 13.7, 4.9, 3.9, 1.9 Hz, 1H, $H6_{eq}$), 1.95 (dd, J = 14.1, 9.5 Hz, 1H, $H4_{ax}$), and 1.70 (dddd, J = 13.7, 11.9, 9.1, 5.3 Hz, 1H, $H6_{ax}$).

¹³C{1H} NMR (126 MHz, CDCl₃) δ 177.3, 176.7, 171.8, 86.5, 71.5, 66.7, 48.1, 38.3, 35.9, 31.7, 26.8, 26.4, and 26.1.

IR (neat): 2971, 1780, 1721, 1281, and 1154 cm⁻¹.

HRMS (ESI) m/z: $[M + Na]^+$ Calcd for $C_{15}H_{20}NaO_6^+$ for 319.1158; Found 319.1133.

mp: 86–88 °C.

(±)-5,7-Dioxotetrahydro-2*H*,7*H*,9*H*-2,6a-methanodifuro[3,2-b:3',4'-c]pyran-3-yl acrylate (18a)



A mixture of the lactone epoxide **11** (252 mg, 2 mmol) and Sc(OTf)₃ (98 mg, 0.2 mmol) was placed into a screw-capped culture tube. Acetonitrile (8 mL) and acrylic anhydride (400 μ L, 3.4 mmol) were added. The homogenous mixture was stirred at room temperature for 12 h. The solvent was evaporated under vacuum and the crude material was purified by flash column chromatography (hexanes:EtOAc 4:6) to provide **18a** (354 mg, 63%) as a white crystalline solid.

¹**H NMR** (500 MHz, Acetone-*d*₆) $\delta = 6.43$ (dd, *J*= 17.3, 1.5 Hz, 1H, H_ZH_EC=C), 6.21 (dd, *J*= 17.3, 10.4 Hz, 1H, CHCOO), 5.98 (dd, *J*= 10.5, 1.5 Hz, 1H, *H*_ZH_EC=C), 5.12 (d, *J* = 1.4 Hz, 1H, *H3a*), 5.00 (d, *J* = 11.6 Hz, 1H, *H9*), 4.98 (dd, *J* = 1.3, 1.3 Hz, 1H, *H3*), 4.72 (d, *J* = 11.5 Hz, 1H, *H9'*), 4.67 (dd, *J* = 5.8, 1.2 Hz, 1H, *H2*), 3.17 (d, *J* = 18.8 Hz, 1H, *H6*), 3.01 (d, *J* = 18.8 Hz, 1H, *H10*_{exo}), and 2.34 (d, *J* = 13.8 Hz, 1H, *H10*_{endo}).

¹³C{¹H} NMR (126 MHz, Acetone-*d*₆) δ 177.0, 165.6, 165.0, 132.6, 128.6, 85.0, 82.7, 82.6, 81.7, 66.7, 49.6, 37.8, and 32.3.

IR (neat): 1779, 1749, 1725, 1633, 1184, and 1063 cm⁻¹.

HRMS (ESI) m/z: $[M + H]^+$ Calcd for $C_{13}H_{13}O_7^+$ for 281.0661; Found 281.0645.

mp: 197–199 °C.

(±)-(5,7-Dioxotetrahydro-2*H*,7*H*,9*H*-2,6a-methanodifuro[3,2-*b*:3',4'-*c*]pyran-3-yl methacrylate (18b)



Smaller scale (with chromatography): A mixture of lactone epoxide **11** (226 mg, 1 mmol) and $Sc(OTf)_3$ (49 mg, 0.1 mmol) was placed into a screw-capped culture tube. Acetonitrile (6 mL) and methacrylic anhydride (220 µL, 1.5 mmol) were added. The homogenous mixture was stirred at room temperature for 12 h. The solvent was evaporated under vacuum and the crude material was purified by flash column chromatography (hexanes:EtOAc 4:6) to provide the sample of **18b** (220 mg, 75%) as a white crystalline solid.

Larger scale (with recrystallization): A mixture of the lactone epoxide **11** (4.0 g, 17.7 mmol) and $Sc(OTf)_3$ (712 mg, 1.8 mmol) was placed into a screw-capped culture tube. Acetonitrile (120 mL) and methacrylic anhydride (4.0 mL, 35.4 mmol) were added. The homogenous mixture was stirred at room temperature for 12 h. The solvent was evaporated under vacuum to obtain a white solid. This solid was washed several times with Et₂O (these washing contained largely methacrylic acid and anhydride; ¹H NMR). The solid was recrystallized from hexane and methanol to provide the sample of **18b** (4.3 g, 82%) as a white crystalline solid.

¹**H NMR** (500 MHz, Acetone- d_6) δ 6.14 (dq, J = 1.5, 1.1 Hz, 1H, $H_ZH_EC=C$), 5.72 (dq, J = 1.6, 1.6 Hz, 1H, $H_ZH_EC=C$), 5.09 (d, J = 1.4 Hz, 1H, H4), 5.00 (d, J = 11.5 Hz, 1H, H9), 4.98 (dd, J = 1.3, 1.3 Hz, 1H, H3), 4.72 (d, J = 11.5 Hz, 1H, H9'), 4.67 (dd, J = 5.9, 1.3 Hz, 1H, H2), 3.18 (d, J = 18.8 Hz, 1H, H6), 3.01 (d, J = 18.8 Hz, 1H, H6'), 2.59 (dd, J = 13.8, 5.9 Hz, 1H, $H1_{exo}$), 2.34 (d, J = 13.7 Hz, 1H, $H1_{endo}$), and 1.93 (dd, J = 1.6, 1.0 Hz, 3H, -CH₃).

¹³C{¹H} NMR (126 MHz, Acetone-*d*₆) δ 177.0, 166.8, 165.0, 136.7, 126.8, 85.0, 82.8, 82.6, 81.7, 66.7, 49.6, 37.9, 32.3, and 18.2.

IR (neat): 3052, 3001, 2952, 2899, 2835, 1783, 1719, 1702, 1156, 1113, 1020, 999, 883, and 764 cm⁻¹.

HRMS (ESI) m/z: $[M + H]^+$ Calcd for $C_{14}H_{15}O_7^+$ for 295.0818; Found 295.0800.

mp: 169–170 °C.

(±)-5,7-Dioxotetrahydro-2*H*,7*H*,9*H*-2,6a-methanodifuro[3,2-*b*:3',4'-*c*]pyran-3-yl acetate (18c)



A mixture of Sc(OTf)₃ (36 mg, 0.07 mmol) and epoxy lactone acid **11** (163 mg, 0.73 mmol) were combined in a screw-capped culture tube. Acetonitrile (1.5 mL) and acetic anhydride (140 μ L, 1.48 mmol) were added. The colorless homogenous mixture was stirred at room temperature for 16 h. The solvent was evaporated under vacuum and the crude material was purified by flash column chromatography (hexanes:EtOAc, 4:6) to provide the sample of **18c** (165 mg, 84%) as a white crystalline solid.

¹**H** NMR (500 MHz, Acetone- d_6) δ 5.01 (d, J = 1.4 Hz, 1H, H_3), 4.98 (d, J = 11.5 Hz, 1H, $H_9\beta$), 4.89 (dd, J = 1.3, 1.3 Hz, 1H, H_3a), 4.70 (d, J = 11.6 Hz, 1H, $H_9\alpha$), 4.60 (dd, J = 5.9, 1.3 Hz, 1H, H_2), 3.15 (d, J = 18.8 Hz, 1H, $H_6\alpha$), 3.00 (d, J = 18.8 Hz, 1H, $H_6\beta$), 2.56 (dd, J = 13.7, 5.9 Hz, 1H, $H_1\alpha$), 2.30 (d, J = 13.7 Hz, 1H, $H_1\beta$), and 2.08 (s, 3H).

¹³C{1H} NMR (126 MHz, Acetone- d_6) δ 177.0 (C7), 170.4 (MeCO), 165.0(C5), 84.9 (9a), 82.7(C3a), 82.4 (C3), 81.8 (C2), 66.7 (C9), 49.6 (6a), 37.9 (C1), 32.3 (C6), and 20.7 (CH₃).

Assignments based on: HSQC, HMBC, NOESY correlations.

IR (neat): 2990, 1780, 1750, 1374, and 1237 cm⁻¹.

HRMS (ESI) *m/z*: [M + Na]⁺ Calcd for C₁₂H₁₂NaO₇⁺ for 291.0481; Found 291.0462.

mp: 197–198 °C.

Summary of selected HMBC and NOESY data as well as side-by-side listing of ¹³C and ¹H NMR chemical shifts and proton-proton coupling constants (Table S1).



Table S1.	1D ¹³ C and	¹ H NMR	data for	the alcohol	15 and	the acetate	ester 18c
-----------	------------------------	--------------------	----------	-------------	--------	-------------	-----------

Atom #	δ13C(OH)	δ13C(OAc)	δ1H(OH)	Js(OH) (Hz)	δ1H(OAc)	JS(OAc) (Hz)
1α	<mark>38.0</mark>	<mark>37.9</mark>	<mark>2.47</mark>	13.6, 5.9	<mark>2.56</mark>	13.7, 5.9
1β	"	"	<mark>2.06</mark>	13.7	<mark>2.30</mark>	13.7
2	<mark>84.0</mark>	<mark>81.8</mark>	<mark>4.42</mark>	5.8, 1.3	<mark>4.60</mark>	5.9 <i>,</i> 1.3
3	<mark>81.2</mark>	<mark>82.4</mark>	4.13	1.2	<mark>5.01</mark>	1.4
3a	<mark>85.9</mark>	<mark>82.7</mark>	<mark>4.65</mark>	1.2, 1.2	<mark>4.89</mark>	1.3, 1.3
5	<mark>165.4</mark>	<mark>165.0</mark>	-	-	-	-
6α	<mark>32.5</mark>	<mark>32.3</mark>	<mark>3.05</mark>	18.7	<mark>3.15</mark>	18.8
6β			<mark>2.94</mark>	18.6, 0.6	<mark>3.00</mark>	18.8, 0.5
6a	<mark>49.6</mark>	<mark>49.6</mark>	-	-	-	-
7	<mark>177.3</mark>	<mark>177.0</mark>	-	-	-	-
9α	<mark>66.9</mark>	<mark>66.7</mark>	<mark>4.66</mark>	11.4	4.70	11.6
9β	"		<mark>4.92</mark>	11.4, 0.5	<mark>4.98</mark>	11.5, 0.5
9a	<mark>84.6</mark>	<mark>84.9</mark>	-	_	-	_
H3 <u>C</u> CO2	-	<mark>20.7</mark>	-	-	2.08	-
H ₃ C <u>C</u> O ₂	-	<mark>170.4</mark>	-	-	-	-

Carbon δs from HSQC, HMBC; Proton δs assignments supported by NOESY with protons in addition to their geminal partners

Supplementary Material References

- ¹ Hoye, T. R.; Hanson, P. R.; Vyvyan, J. R. A practical guide to first-order multiplet analysis in ¹H NMR spectroscopy. *J. Org. Chem.* **1994**, *59*, 4096–4103.
- ² Hoye, T. R.; Zhao, H. A method for easily determining coupling constant values: an addendum to "a practical guide to first-order multiplet analysis in ¹H NMR spectroscopy". J. Org. Chem. 2002, 67, 4014–4016.
- ³ Sim, J.; Ryou, B.; Choi, M.; Lee, C.; Park, C.-M. Electro-chemical C(sp3)-H functionalization of γ-lactams based on hydrogen atom transfer. *Org. Lett.* **2022**, *24*, 4264–4269.
- ⁴ Bertolini, V.; Appiani, R.; Pallavicini, M.; Bolchi, C. Green oxidation of ketones to lactones with oxone in water. *J. Org. Chem.* **2021**, *86*, 15712–15716.
- ⁵ Santacesaria, E.; Russo, V.; Tesser, R.; Turco, R.; Di Serio, M. Kinetics of performic acid synthesis and decomposition. *Ind. Eng. Chem. Res.* **2017**, *56*, 12940–12952.





©AUTHOR(S)





©AUTHOR(S)







©AUTHOR(S)



210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 f1 $_{\rm (ppm)}$




















































©AUTHOR(S)









-178.20 -172.72	-138.22	 -86.63	-74.70	-62.76	-49.78	∕35.51 ∕34.61
$ \begin{array}{c c} HO & & O \\ HO & 7a & O \\ \hline 7 & 5 & 7a & O \\ \hline 6 & O & 4 & 3 \\ \hline 6 & O & 4 & 3 \\ \hline 7 & 2 & 0 & 0 \end{array} $						
$ \begin{array}{c} 2 & \bullet \\ 7b \\ 1^{3}C \text{ NMR} \\ acetone-d_{6} \\ 126 \text{ MHz} \\ \end{array} $						

210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 f1 (ppm)







Page S47



210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 fl $_{(\rm ppm)}$







⊦20





HO^{3a}

Ò 15

NOESY

acetone-d₆

500 MHz

p 8

1)

©AUTHOR(S)

98

-2.5

-3.0

-3.5 (-3.5 -

-4.0

-4.5

-5.0

11





©AUTHOR(S)

	$^{176.52}$ $^{171.44}$ $^{165.09}$	1		127 EQ	<pre>/ 131.67 / 131.67 / 127.49</pre>	¹ 126.35					-74.91	-64.80	-47.82	-36 24	-31.50	
	0 7 6 13 C N CDC 126 M	⁸ 0 10 7a 3a 0 4 3 2 0 a MR Cl ₃ MHz														
180	170	160	150	140	130	120	110	100	90	80	70	60	50	40	30	20
							f	l ₍ ppm ₎)							





<pre>~176.60 -171.50 </pre> /166.35	ノ135.50 ノ131.61 ノ127.19 人126.67		64.81 47.61		-18.28
O 7 6 0 7 6 0 4 3 3 0 0 4 3 0 0 10 0 0 10 1					
180 170 160 150 2	140 130 120 110 100 f1 ₍ pp	90 80 m	70 60 50	40 30	20 10













©AUTHOR(S)

~176.48 ~171.44 ~170.09		84.74	77.16 76.91 74.70 -64.64	-47.87	35.96 31.23 20.86	
$ \begin{array}{c} 0 \\ 0 \\ 7 \\ 7 \\ 6 \\ 0 \\ 7 \\ 7 \\ 7 \\ 7 \\ 7 \\ 7 \\ 7 \\ 7 \\ 7 \\ 7$						
						~-4~
180 170 160 150 14	130 120	110 100 90 f1 ₍ ppm ₎	80 70 60	50 40	30 20 10	0











f1(ppm)































©AUTHOR(S)














©AUTHOR(S)







210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 $f1_{(ppm)}$







-176.96 $\sqrt{165.62}$ $\sqrt{164.98}$	~132.56 ~128.61	84.99 82.66 81.73 -66.67		
$ \begin{array}{c} $				
210 200 190 180 170 160 15	50 140 130 120 1 f1 ₍ p	10 100 90 80 70 60 pm	50 40	30 20 10

Page S79



©AUTHOR(S)

-176.10	16.cot 164.13	-135.86	-125.98	84.08 81.98 81.75 80.87	-65,79	-48.69	-36.99 -31.39	-17.33
$ \begin{array}{c} $								
210 200 190 180 170) 160 150 140	0 13	0 120 110 100 f1 ₍ ppm ₎	90 80 7	0 60	50 4	0 30	20 10









