

## Supplementary Material

### Reactions of quinine with 2-chloro-4,6-dimethoxy-1,3,5-triazine

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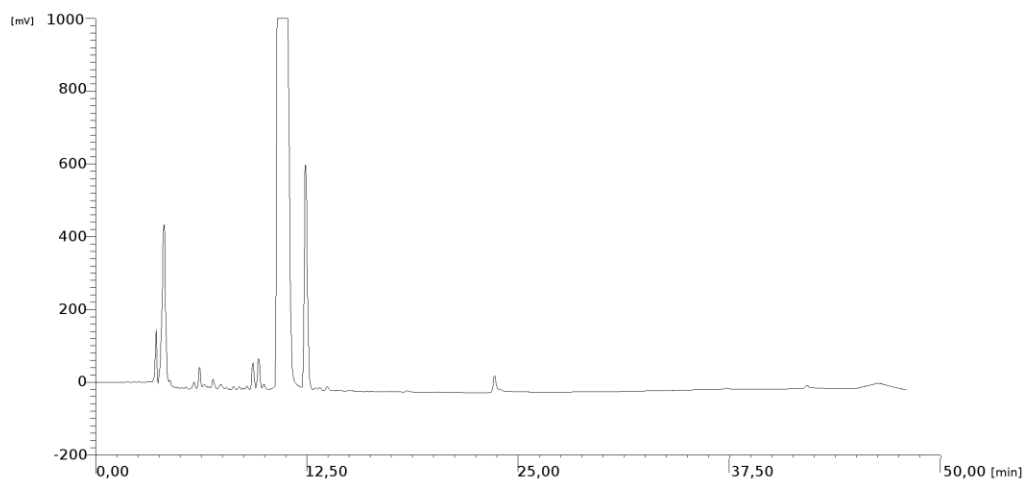
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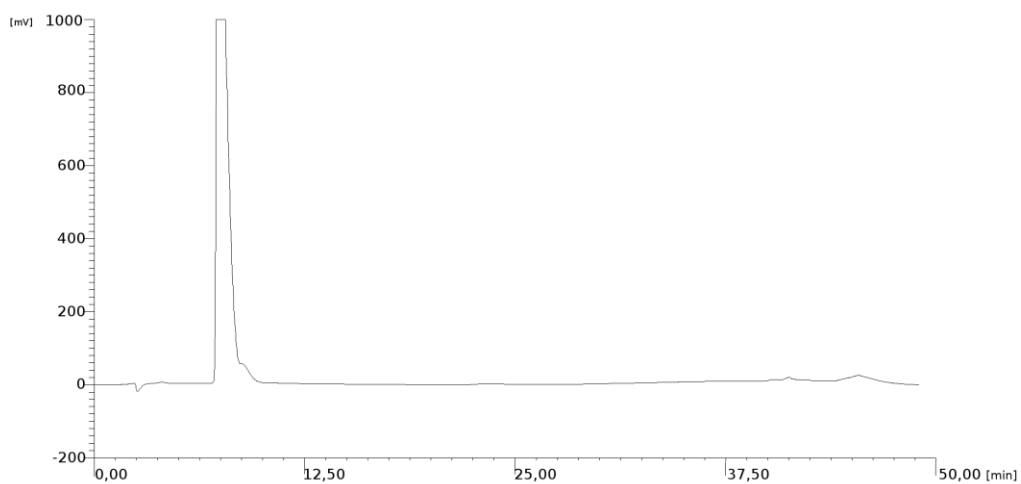
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a)



b)



**Figure S1.** Isolation of the main component **5** from the product mixture obtained by the reaction of quinine (**1**) and CDMT (**2**). Panel a) after preparative chromatography; panel b) after recrystallization. Column Vydac C18 (250x4.6). Programme: 1) 0-5 min. isocratic acetonitrile / water 50 / 50%; 2) 5-35 min. gradient acetonitrile / water 50-97%; 3) 35-40 min, isocratic acetonitrile/water 97/3%; 4) 40-50 min. isocratic acetonitrile / water 50 / 50. Note shortening of  $R_t$  after recrystallization.

## 2. Crystal structure data

The intensity data was collected on the Bruker SMART APEX II CCD diffractometer equipped with a Cu K $\alpha$  (1.54178 Å) radiation source. Crystal structure refinement was carried out with SHELX.<sup>7,8</sup> Structure refinement details are summarized in **Table S1**. The molecular structure of compound **5** is shown in **Figure 1**. Program MERCURY<sup>9</sup> was used for molecular graphics. CCDC: 2027177 contains the supplementary crystallographic data for this paper. The data is provided free of charge by The Cambridge Crystallographic Data Centre via [www.ccdc.cam.ac.uk/structures](http://www.ccdc.cam.ac.uk/structures).

**Table S1.** Structure refinement and experimental details

Crystal data	
Chemical formula	C <sub>30</sub> H <sub>35</sub> ClN <sub>8</sub> O <sub>6</sub>
$M_r$	639.11
Crystal system, space group	Orthorhombic, $P2_12_12_1$
Temperature (K)	100
$a, b, c$ (Å)	7.7702 (2), 10.6624 (3), 38.4161 (10)
$V$ (Å <sup>3</sup> )	3182.73 (15)
$Z$	4
Radiation type	Cu $K\alpha$
$\mu$ (mm <sup>-1</sup> )	1.53
Crystal size (mm)	0.60 × 0.07 × 0.04
Data collection	
Diffractometer	Bruker SMART APEX CCD
$T_{\min}, T_{\max}$	0.643, 0.754
No. of measured, independent and observed [ $I > 2\sigma(I)$ ] reflections	36943, 6330, 6186
$R_{\text{int}}$	0.025
$(\sin \theta/\lambda)_{\text{max}}$ (Å <sup>-1</sup> )	0.619
Refinement	
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.024, 0.065, 1.04
No. of reflections	6330
No. of parameters	411
$\Delta\rho_{\text{max}}, \Delta\rho_{\text{min}}$ (e Å <sup>-3</sup> )	0.21, -0.19

**Table S2.** Parameters of  $^1\text{H}$  MNR,  $^{13}\text{C}$  NMR and DEPT-135 spectra

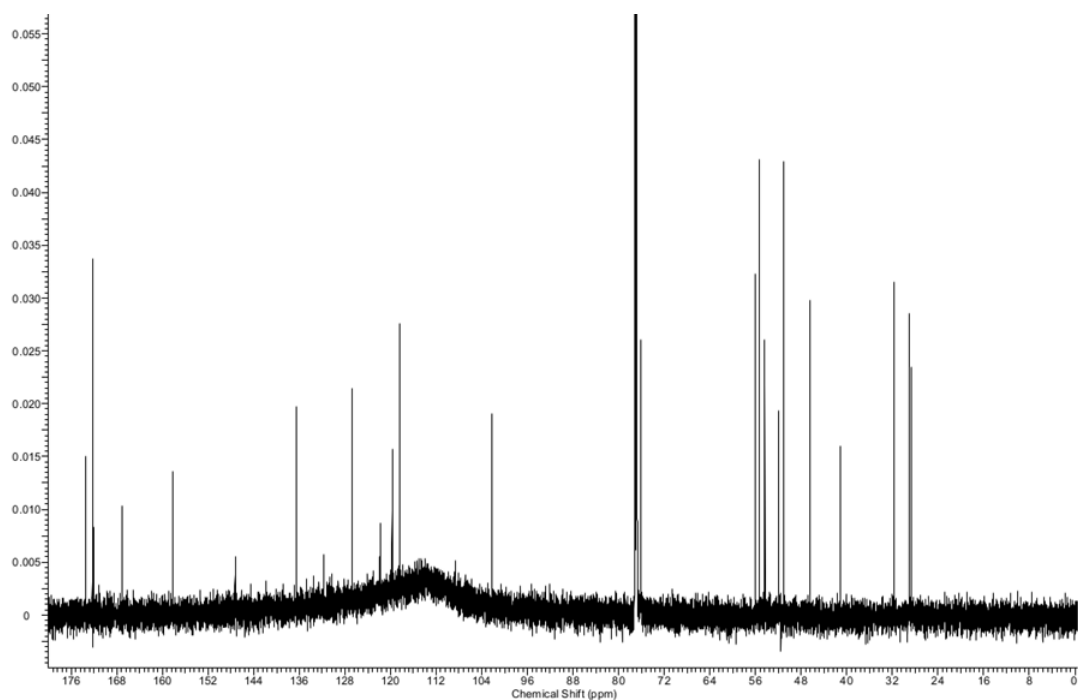
parameters	$^1\text{H}$	$^{13}\text{C}$	DEPT-135
Acquisition Time (sec)	2.2719	0.7864	0.7864
Frequency (MHz)	700.08	176.04	176.04
Nucleus	$^1\text{H}$	$^{13}\text{C}$	$^{13}\text{C}$
Number of Transients	128	12288	8192
Original Points Count	32768	32768	32768
Points Count	32768	32768	32768
Receiver Gain	2050.00	2050.00	2050.00
SW(cyclical) (Hz)	14423.08	41666.67	41666.67
Solvent	$\text{CDCl}_3$	$\text{CDCl}_3$	$\text{CDCl}_3$
Spectrum Offset (Hz)	4308.0820	17607.8320	17600.0859
Sweep Width (Hz)	14422.64	41665.39	41665.39
Temperature (degree C)	27.000	27.100	27.000

**Table S3.** Parameters of COSY spectra

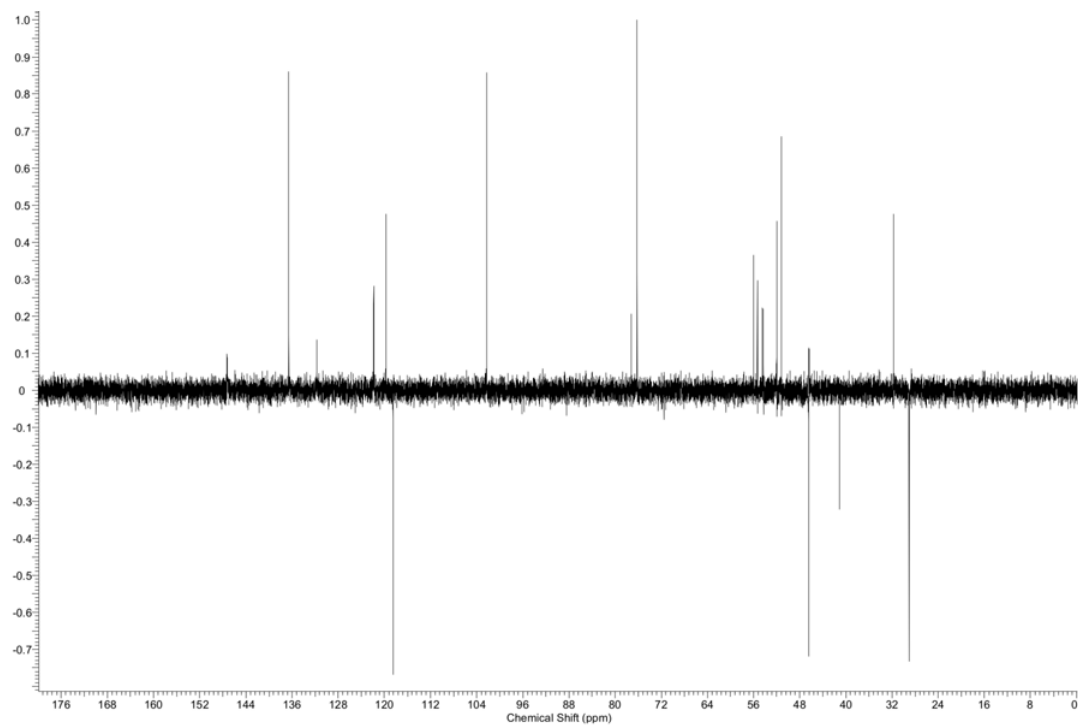
parameters	COSY
Acquisition Time (sec)	(0.2433, 0.0609)
Frequency (MHz)	(700.08, 700.08)
Nucleus	( $^1\text{H}$ , $^1\text{H}$ )
Number of Transients	24
Original Points Count	(2048, 512)
Points Count	(2048, 1024)
Pulse Sequence	cosygpqf
Solvent	$\text{CDCl}_3$
Spectrum Offset (Hz)	COSY
Sweep Width (Hz)	(8413.40, 8392.81)
Temperature (degree C)	27.000

**Table S3.** Parameters of HSQC spectra

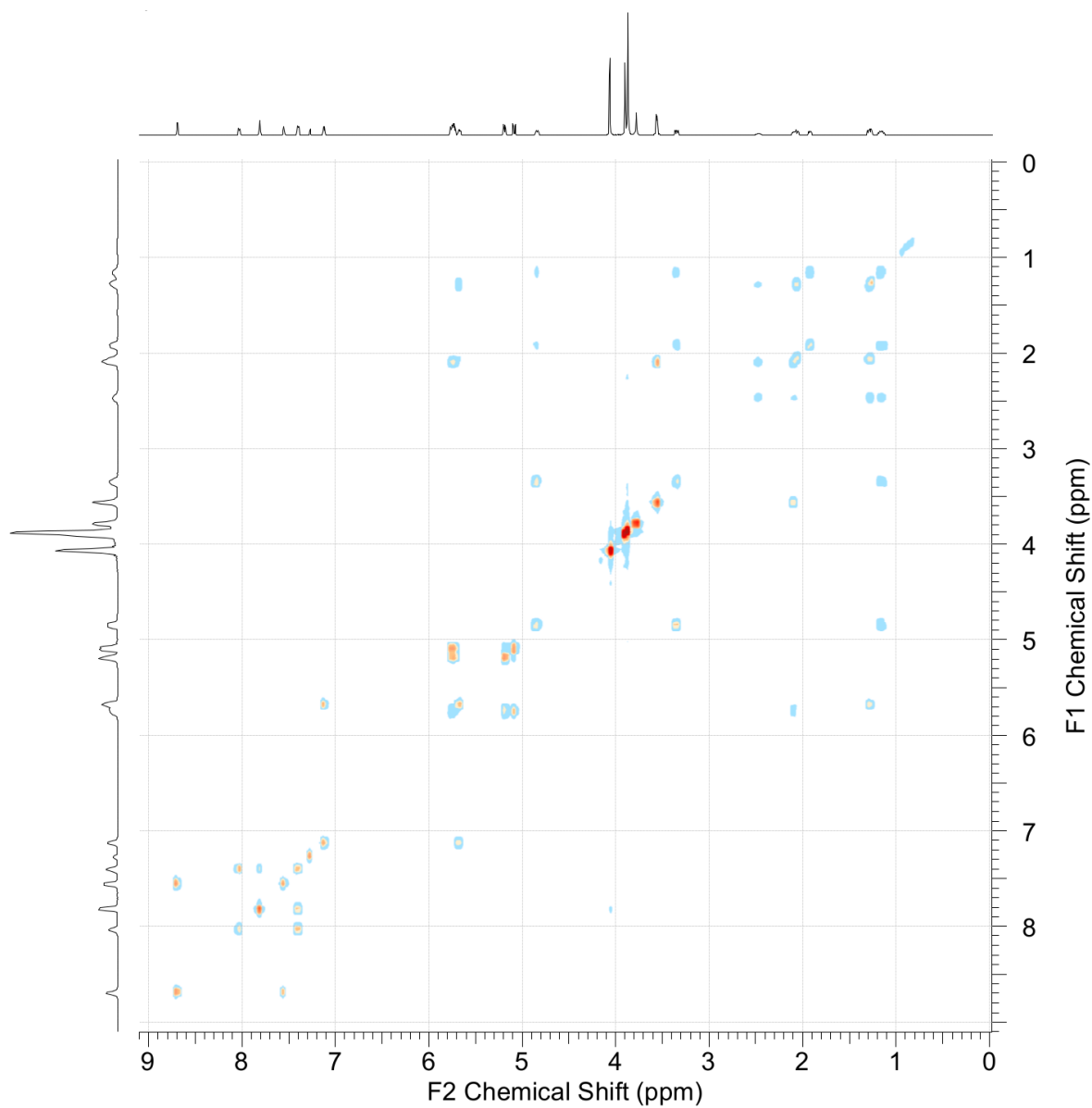
parametry	HSQC
Acquisition Time (sec)	(0.1217, 0.0171)
Frequency (MHz)	(700.08, 176.05)
Nucleus	( $^1\text{H}$ , $^{13}\text{C}$ )
Number of Transients	32
Original Points Count	(1024, 512)
Points Count	(1024, 2048)
Pulse Sequence	hsqcedetgpsisp2.2.pp
Solvent	CHLOROFORM-d
Spectrum Offset (Hz)	HSQC
Sweep Width (Hz)	(8409.29, 29913.78)
Temperature (degree C)	27.200



**Figure S2.**  $^{13}\text{C}$  NMR spectrum of the main product **5** formed in reaction of quinine with CDMT.

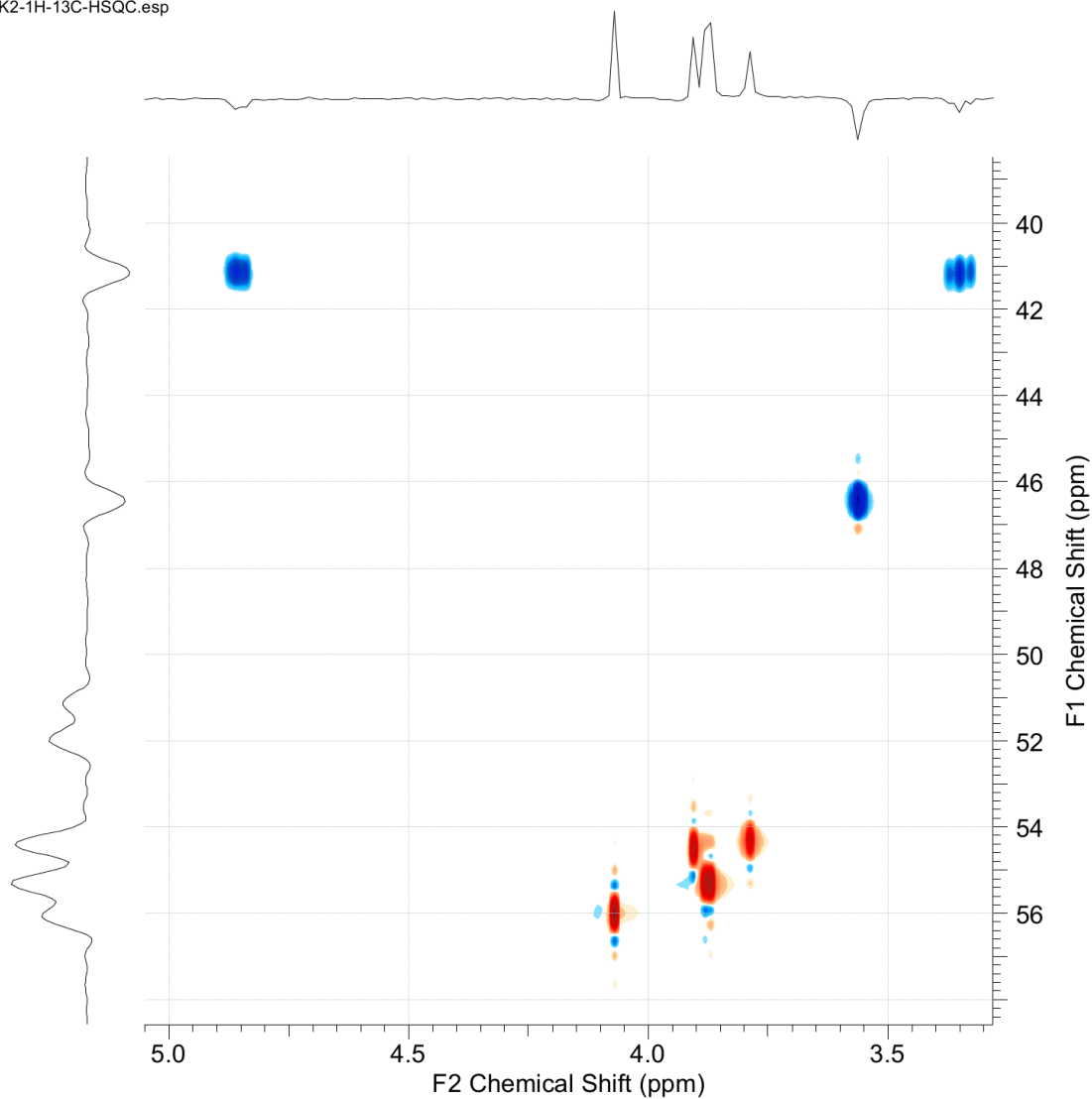


**Figure S3,** DEPT-135 spectrum in  $\text{CDCl}_3$  of the main product **5** formed in reaction of quinine with CDMT.

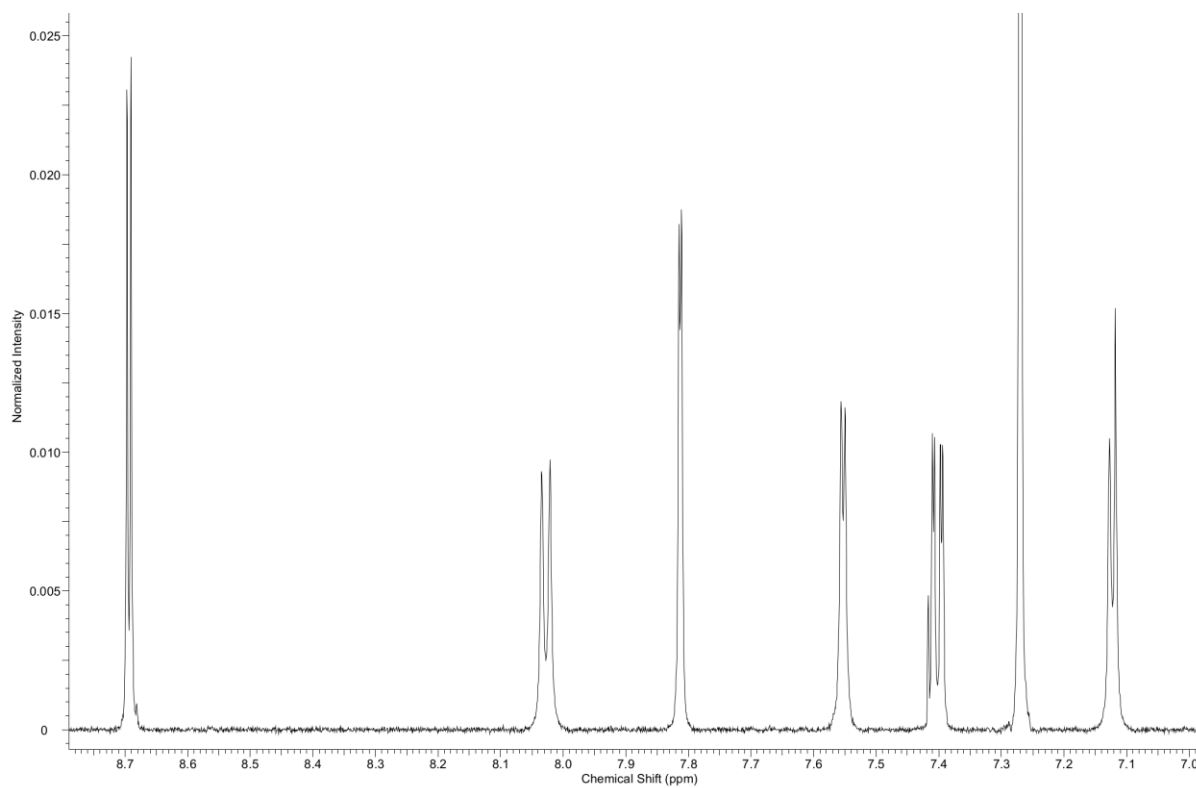


**Figure S4.** COSY spectrum of the main product **5** formed in reaction of quinine with CDMT.

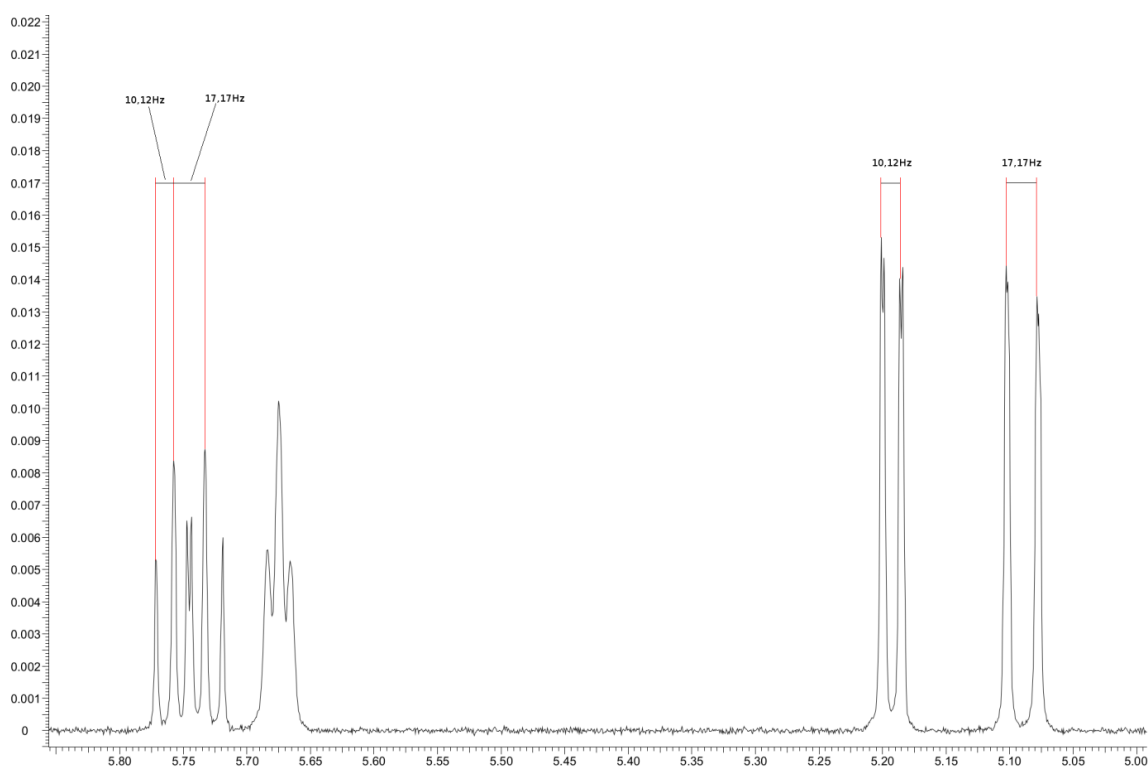
K2-1H-13C-HSQC.esp



**Figure S5.** HSQC spectrum of fragment 3.30 – 5.00 ppm.

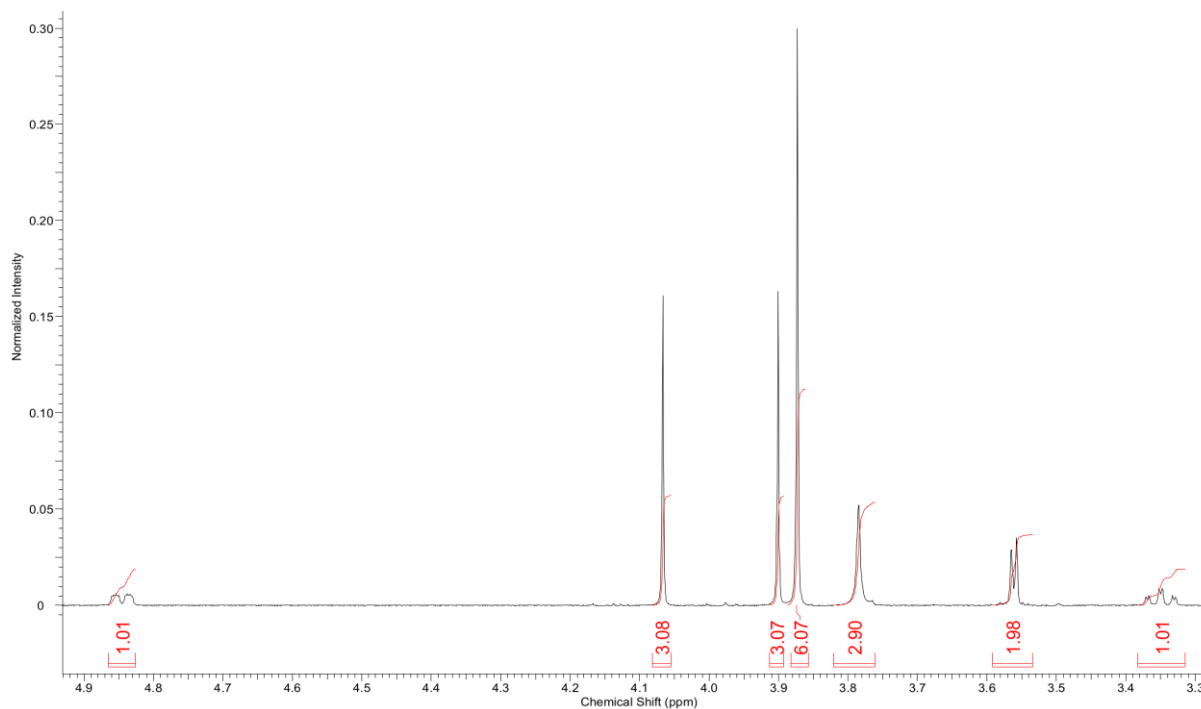


**Figure S6.** Fragment 7.00-8.75 ppm of  $^1\text{H}$  NMR spectrum of product **5** formed in reaction of quinine with CDMT.

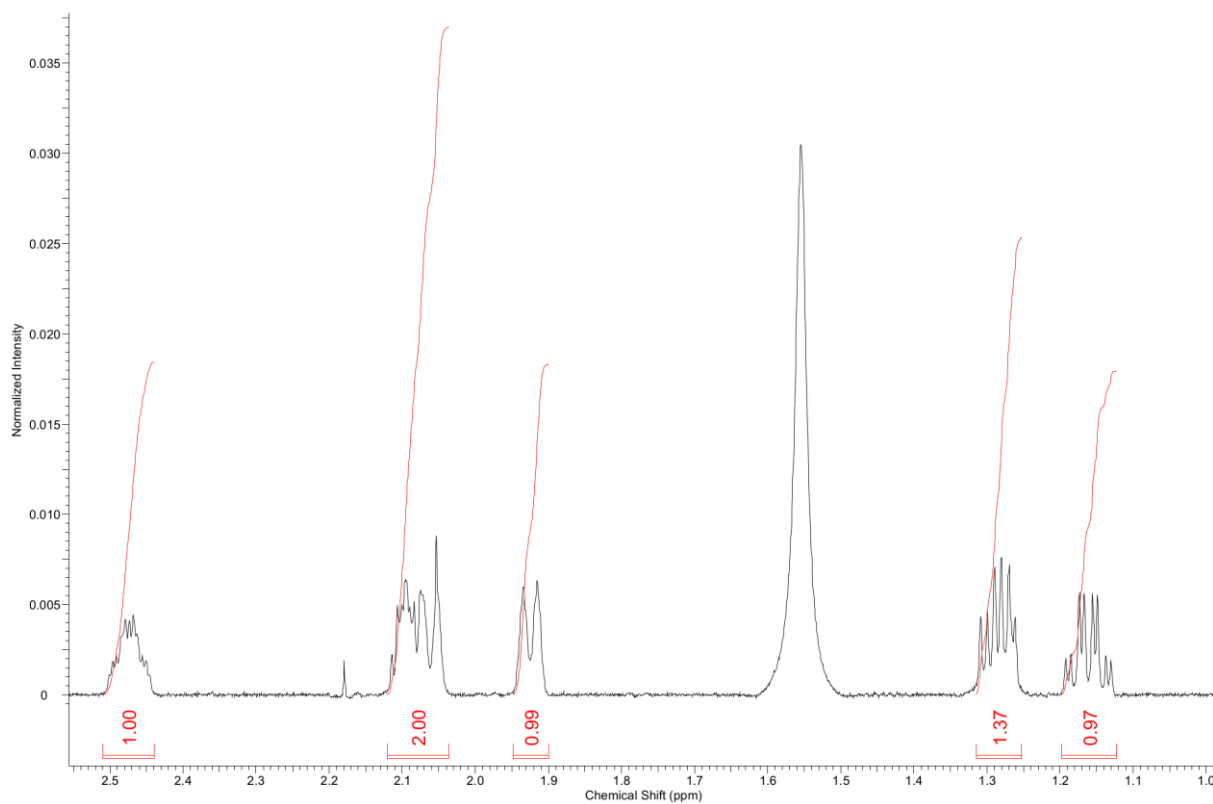


**Figure S7.** Fragment 5.00-5.80 ppm of  $^1\text{H}$  NMR spectrum of **5** formed in reaction of quinine with CDMT.

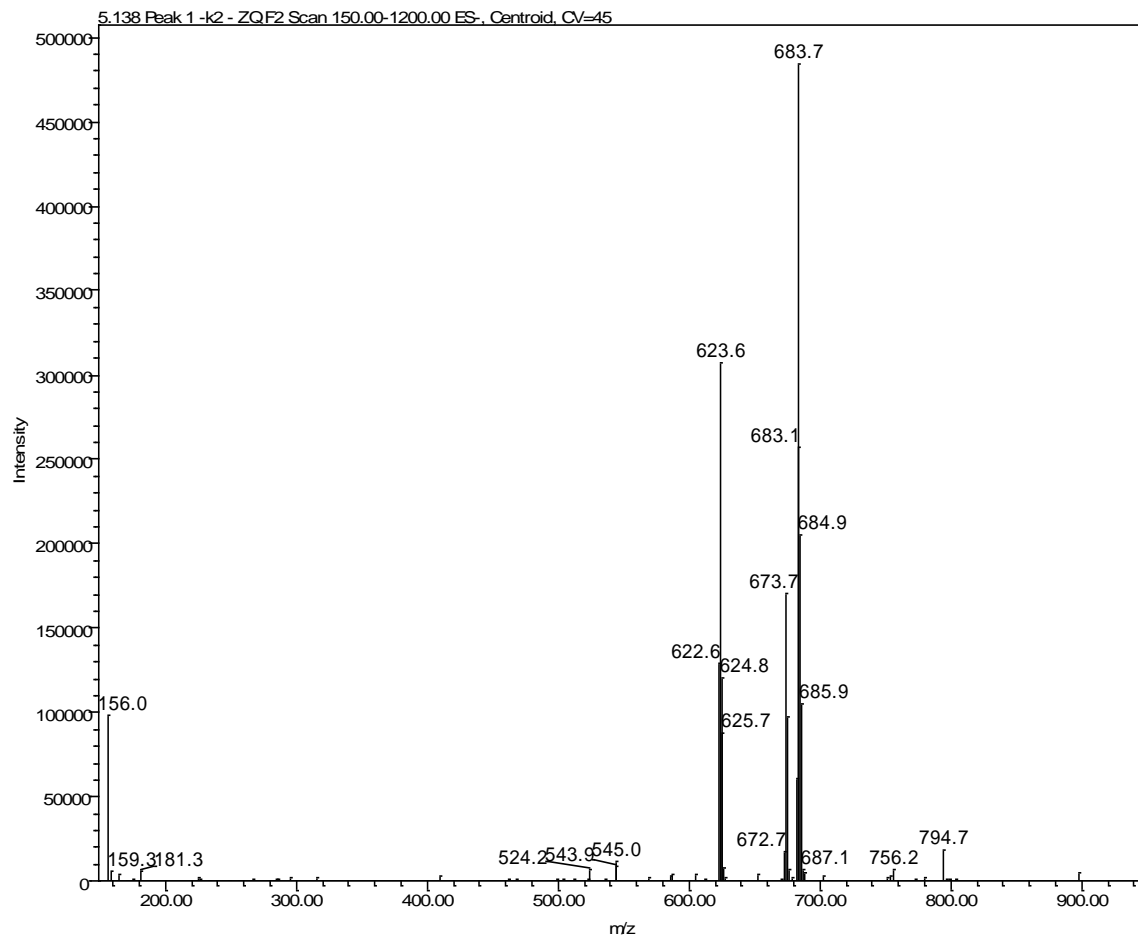




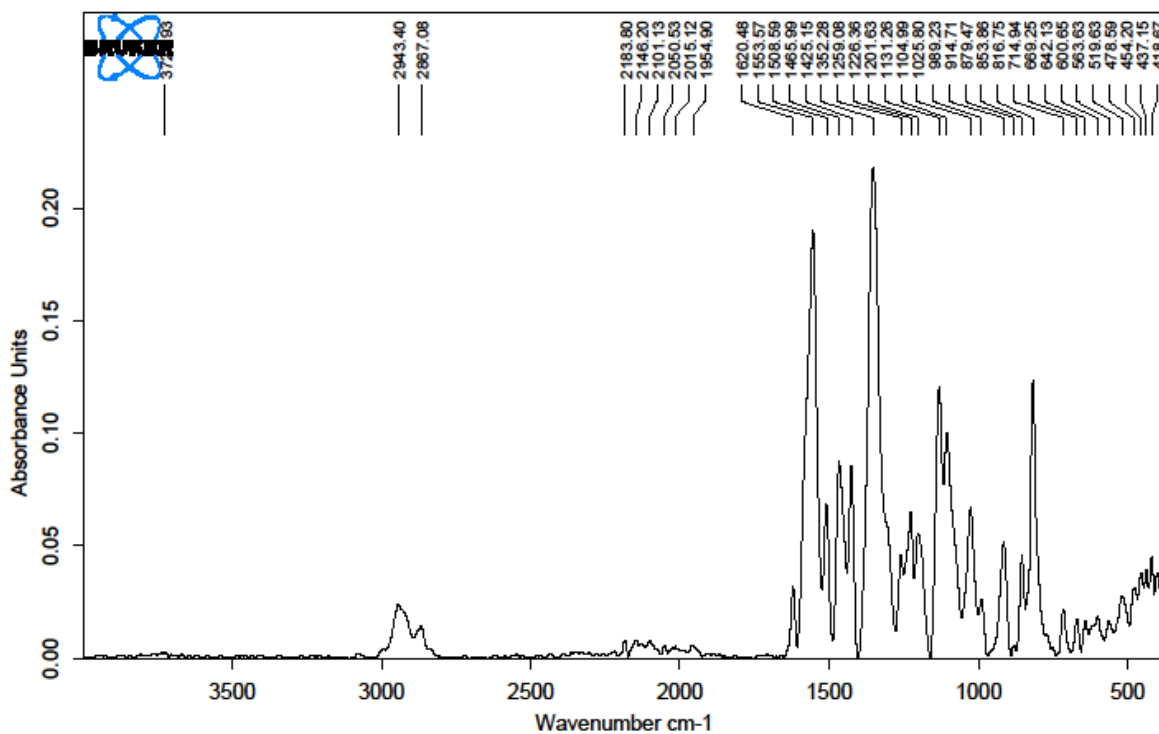
**Figure S8.** Fragment 3,30 to 4,90 ppm of <sup>1</sup>H NMR spectrum of product **5** formed in reaction of quinine with CDMT.



**Figure S9.** Fragment 1,20 to 2,50 ppm of <sup>1</sup>H NMR spectrum of product formed in reaction of quinine with CDMT.



**Figure S10.** Mass spectrum (ESI) of the main product **5** formed in reaction of quinine with CDMT isolated after crystallization from ethyl acetate / n-heptane



**Figure S11.** IR spectrum of the main product **5** formed in reaction of quinine with CDMT isolated after crystallization from ethyl acetate / n-heptane.