

## Supplementary Material

### Acid-catalyzed reaction of 1-(2,2-dimethoxyethyl)ureas with phenols as an effective approach to diarylethanes and dibenzoxanthenes

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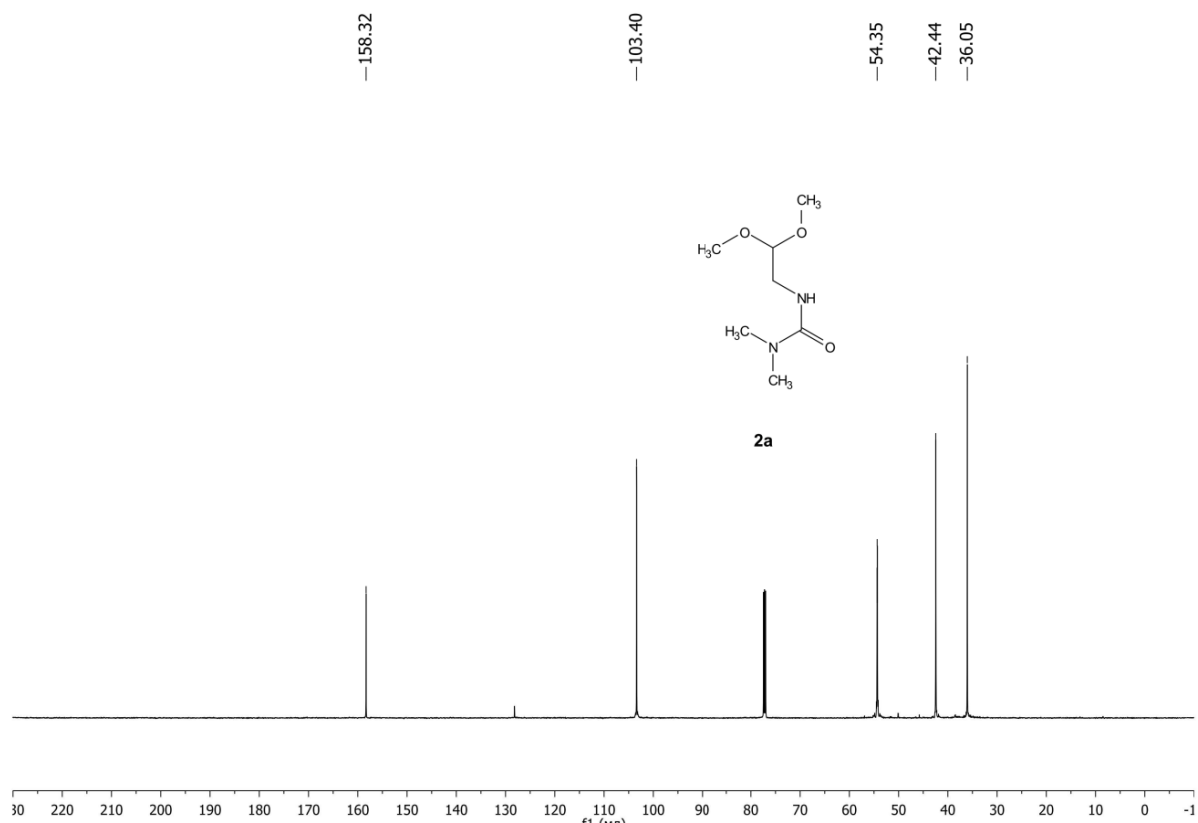
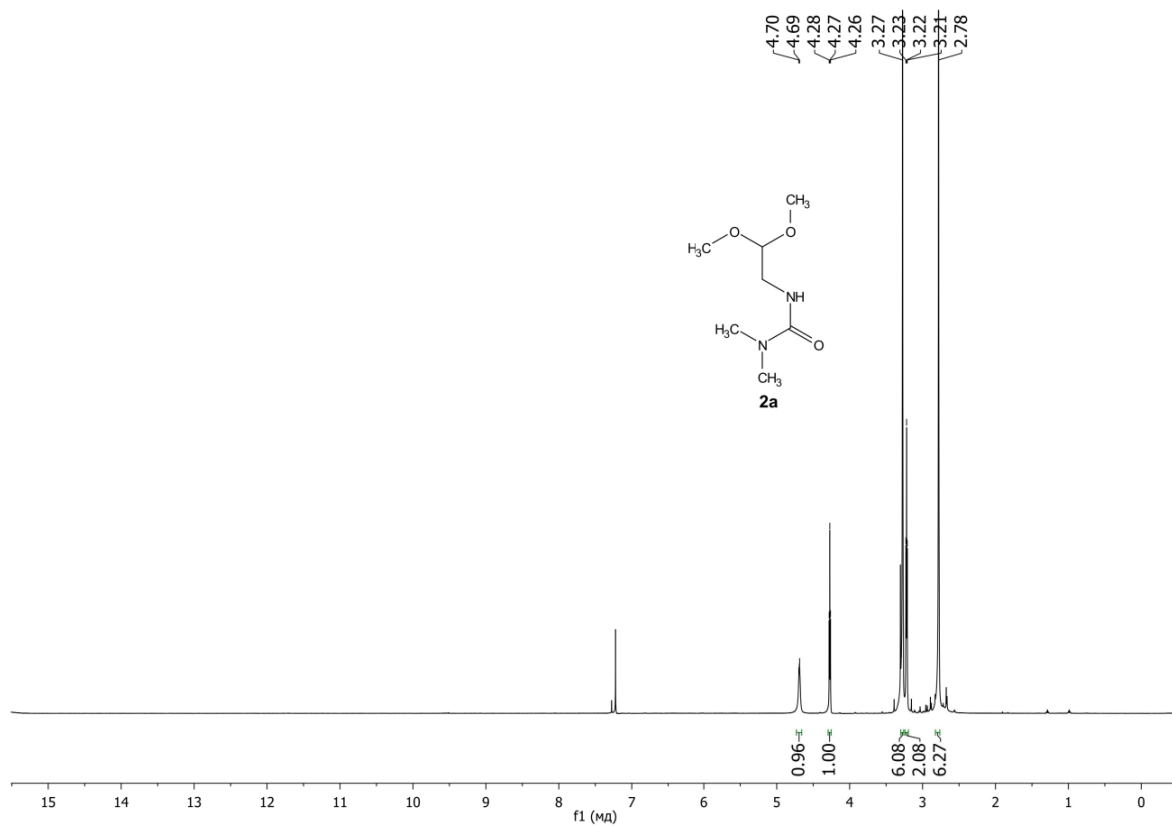
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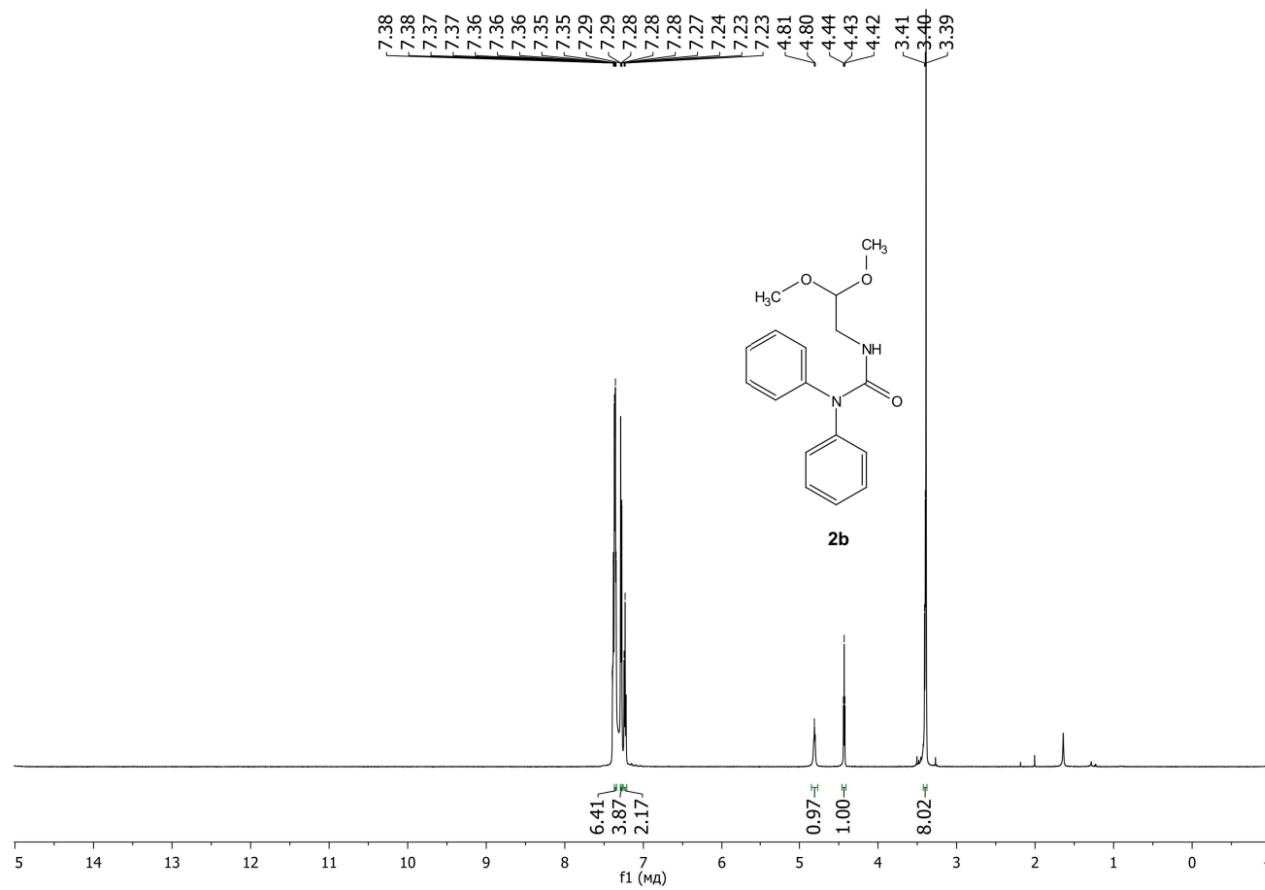
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#### Table of Contents

|                                                      |    |
|------------------------------------------------------|----|
| <sup>1</sup> H and <sup>13</sup> C NMR spectra ..... | S2 |
|------------------------------------------------------|----|



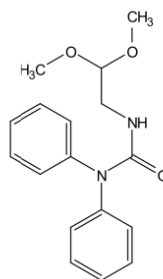


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—142.78  
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—127.36  
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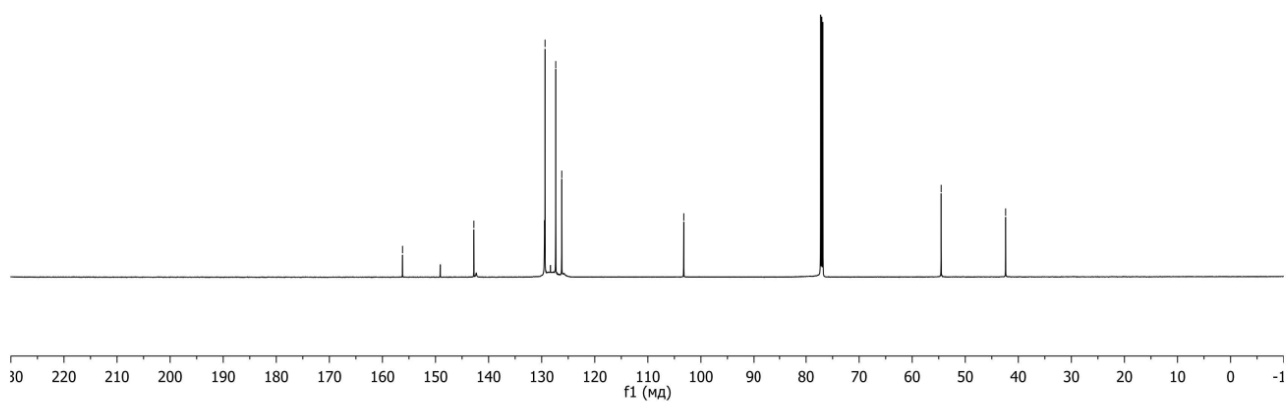
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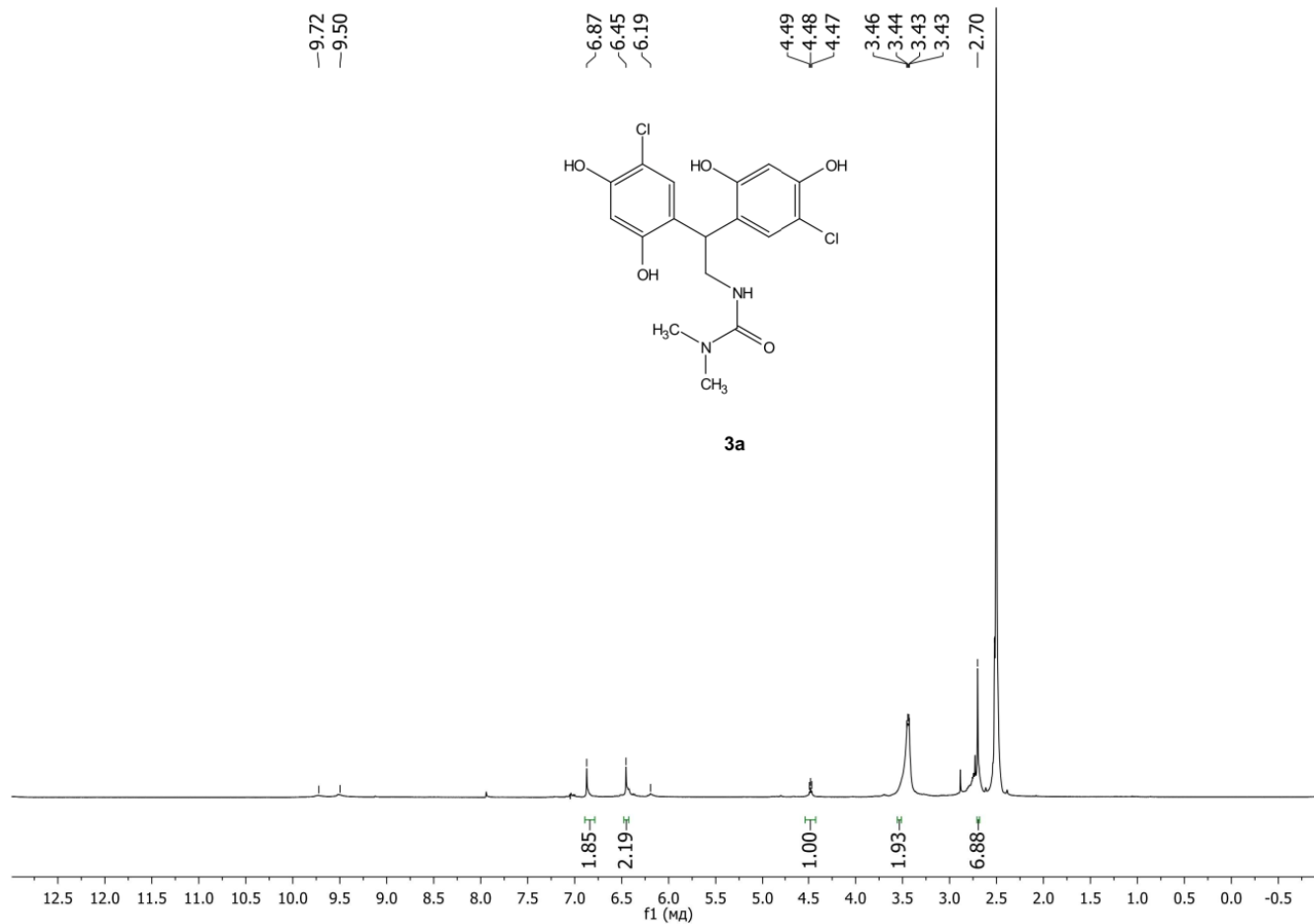
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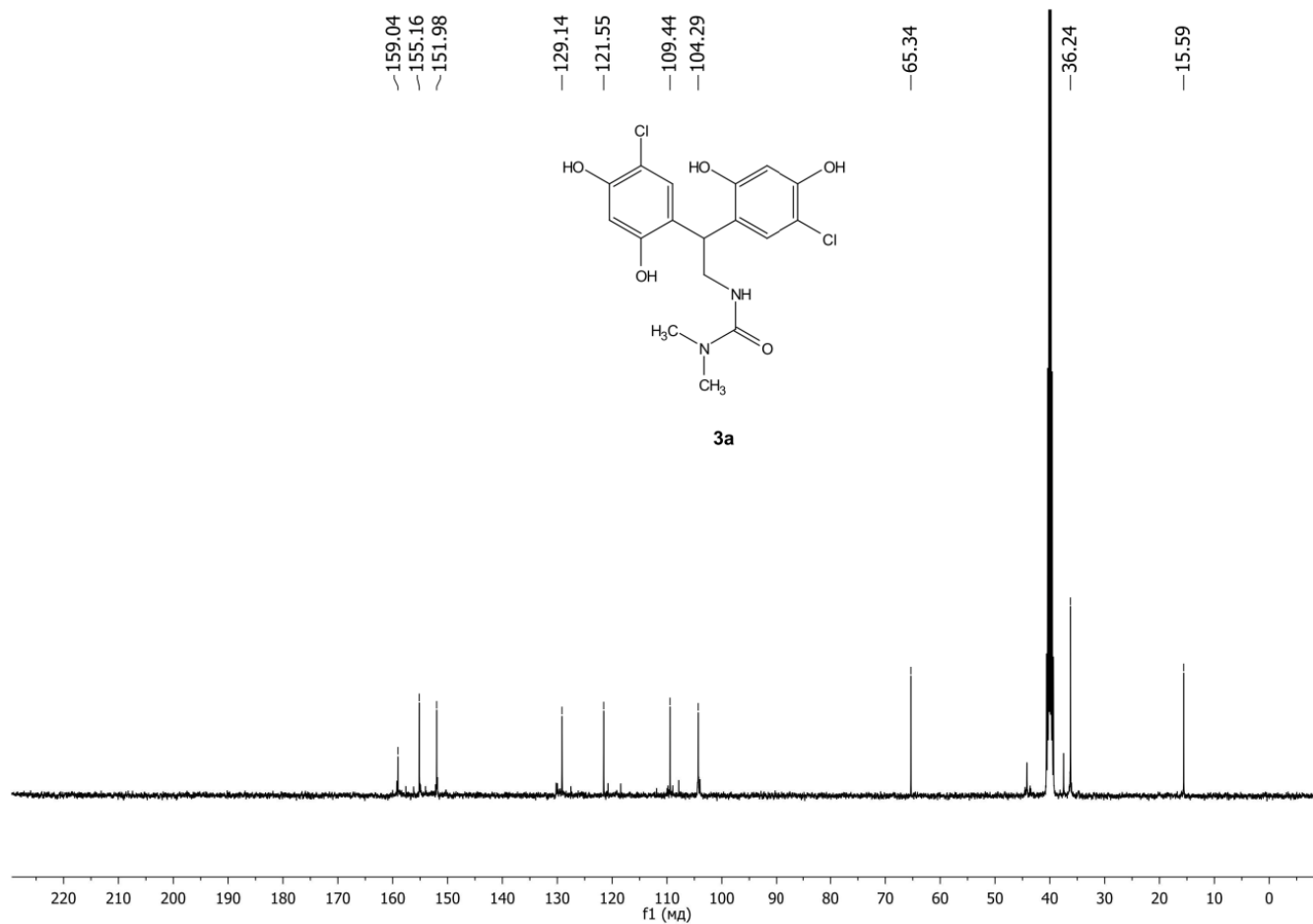
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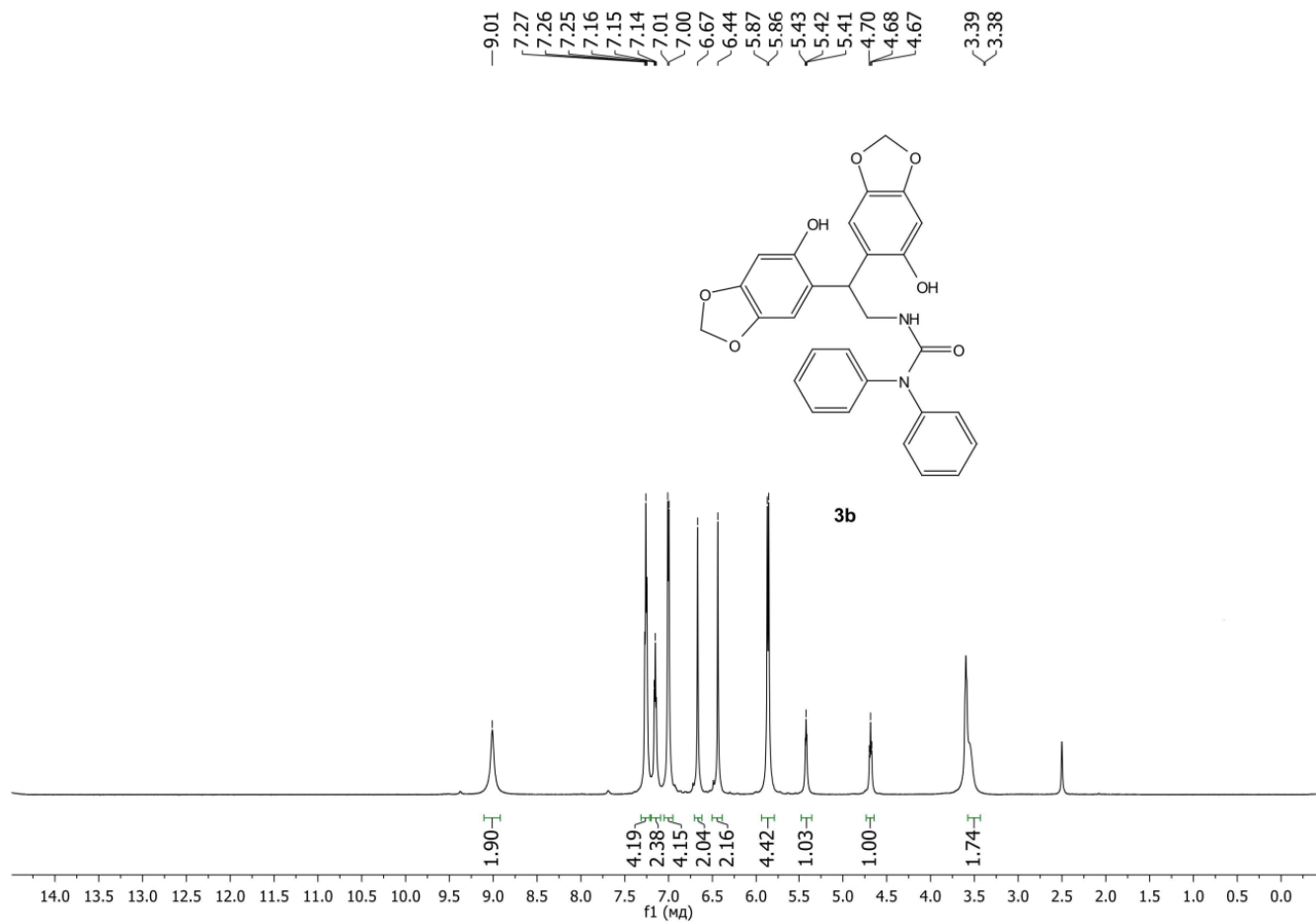


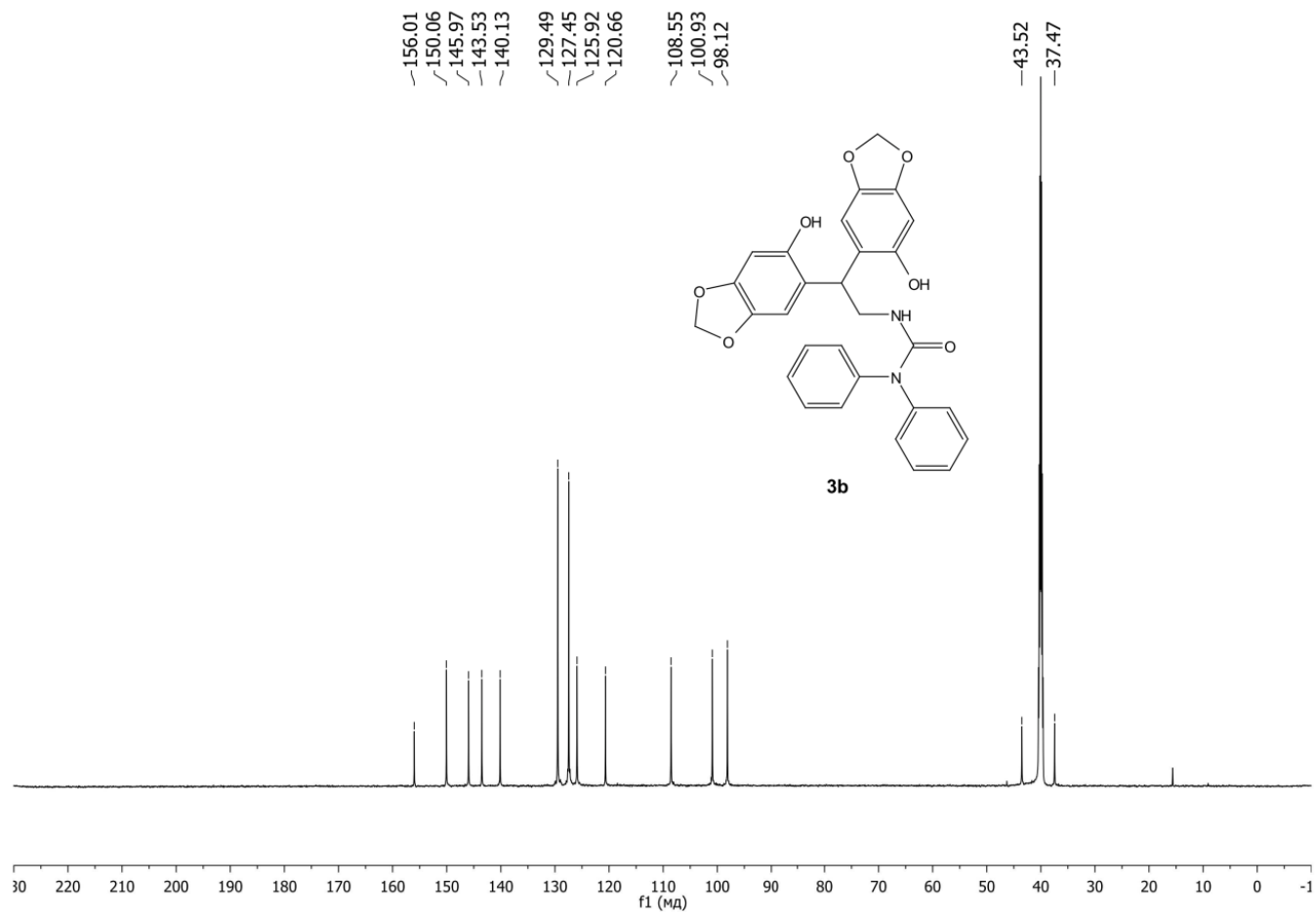
**2b**



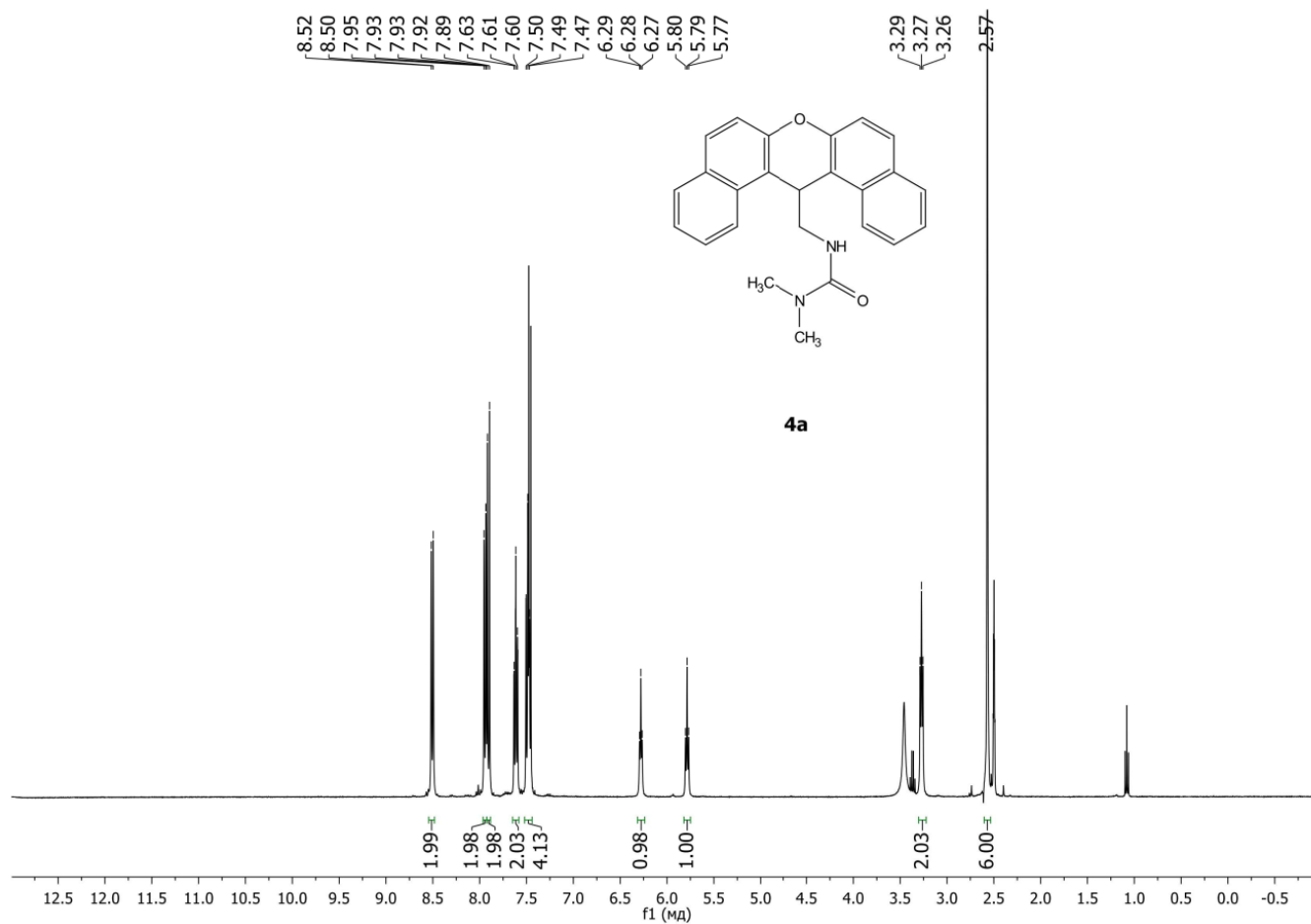


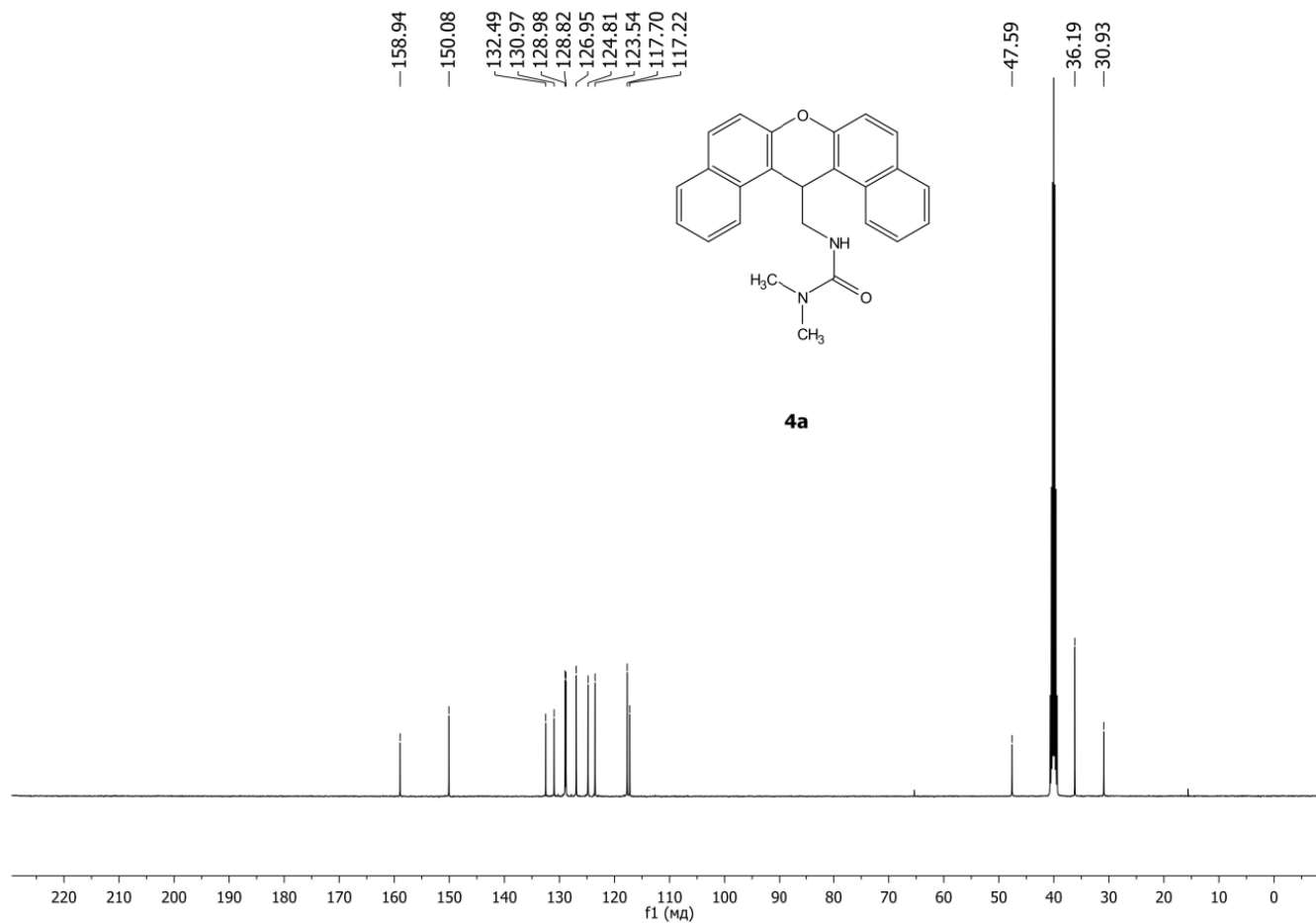


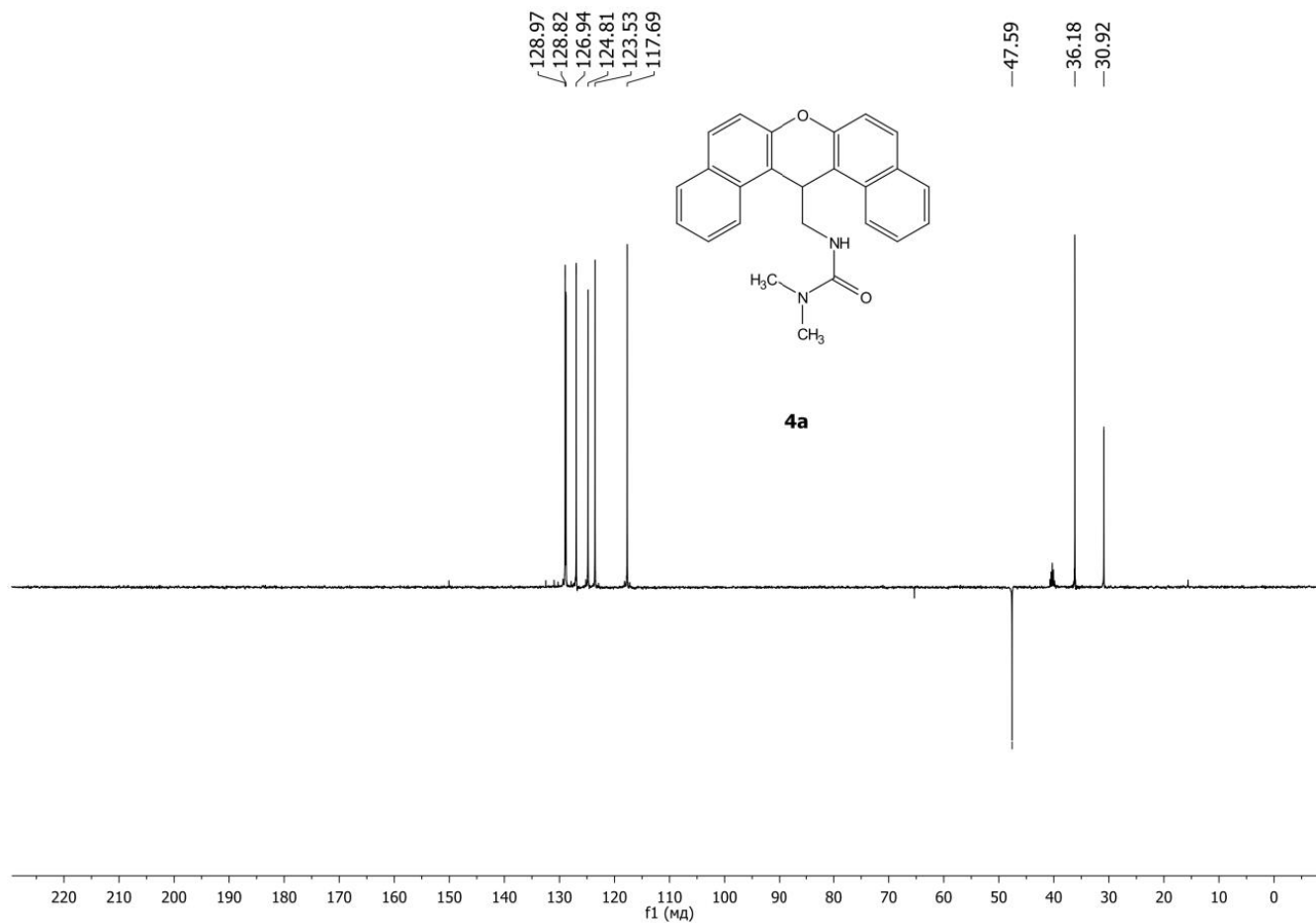


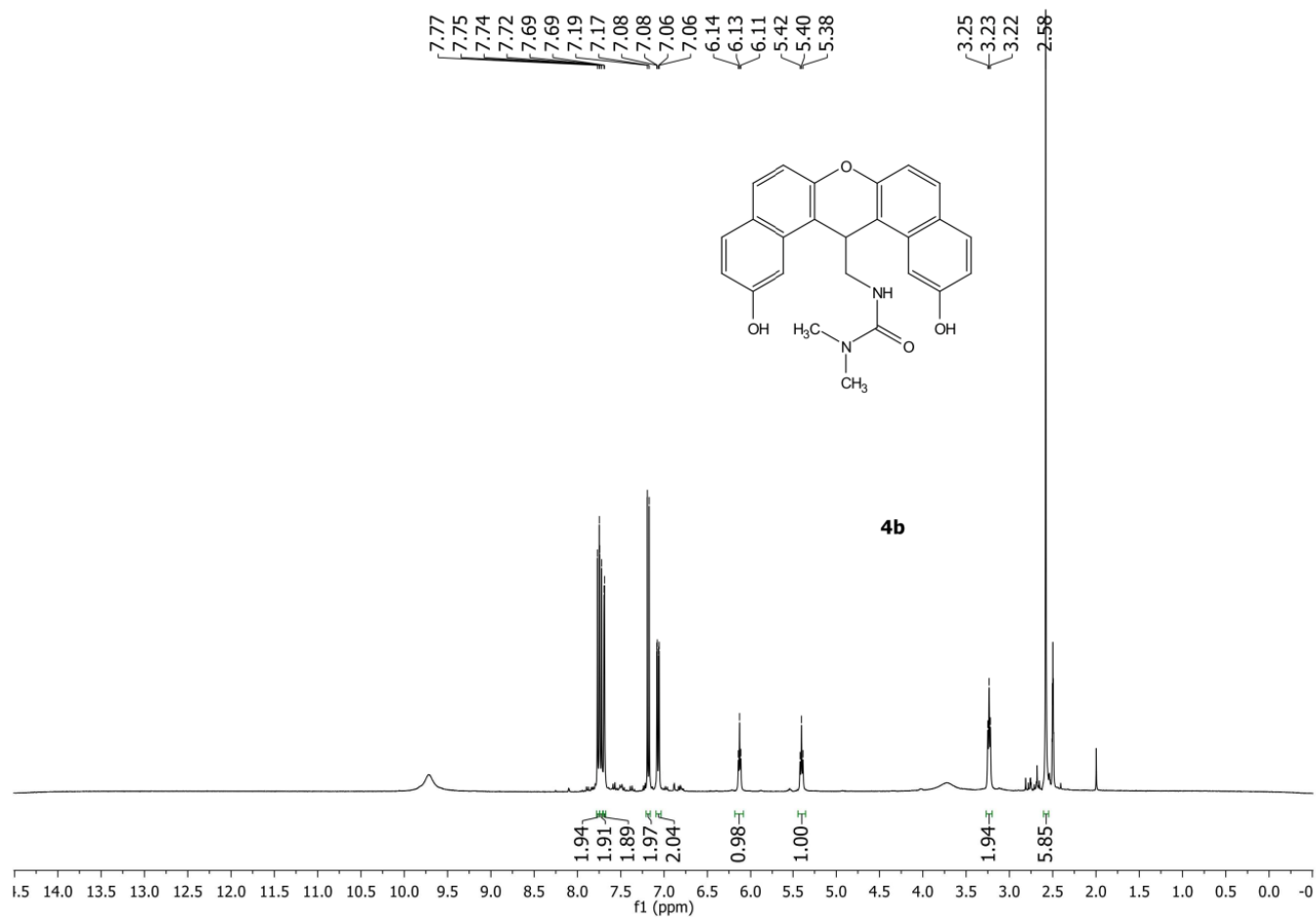


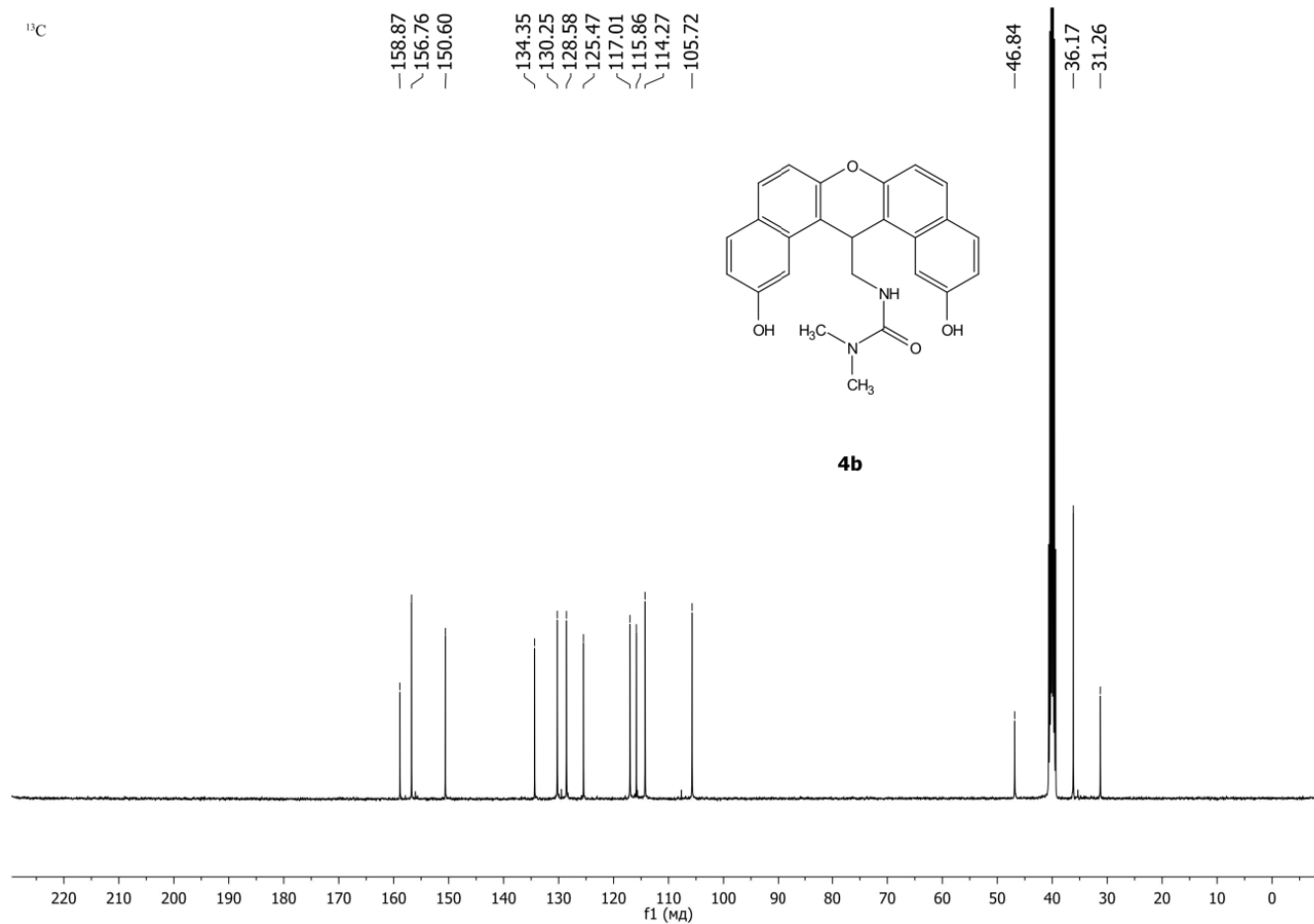




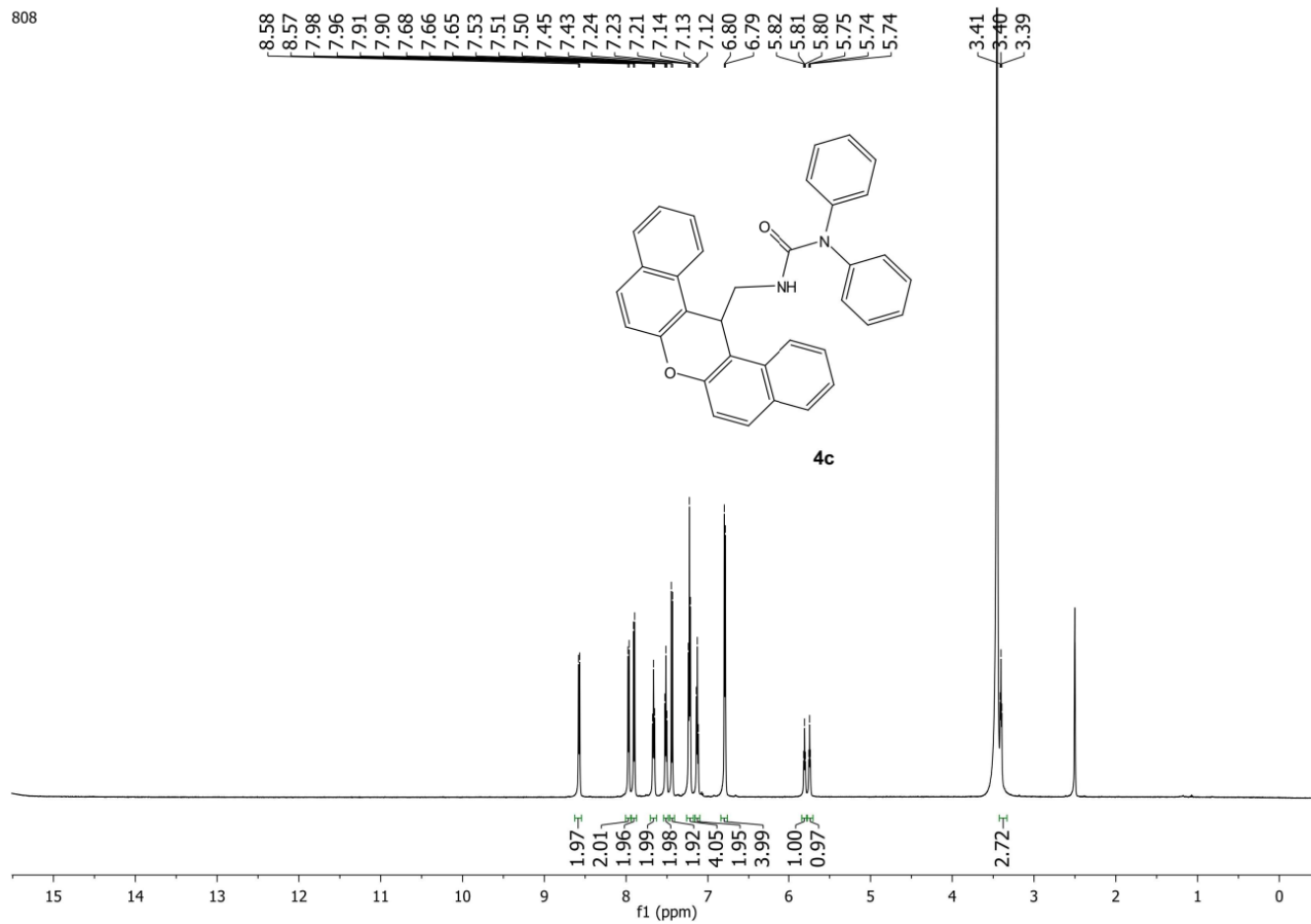


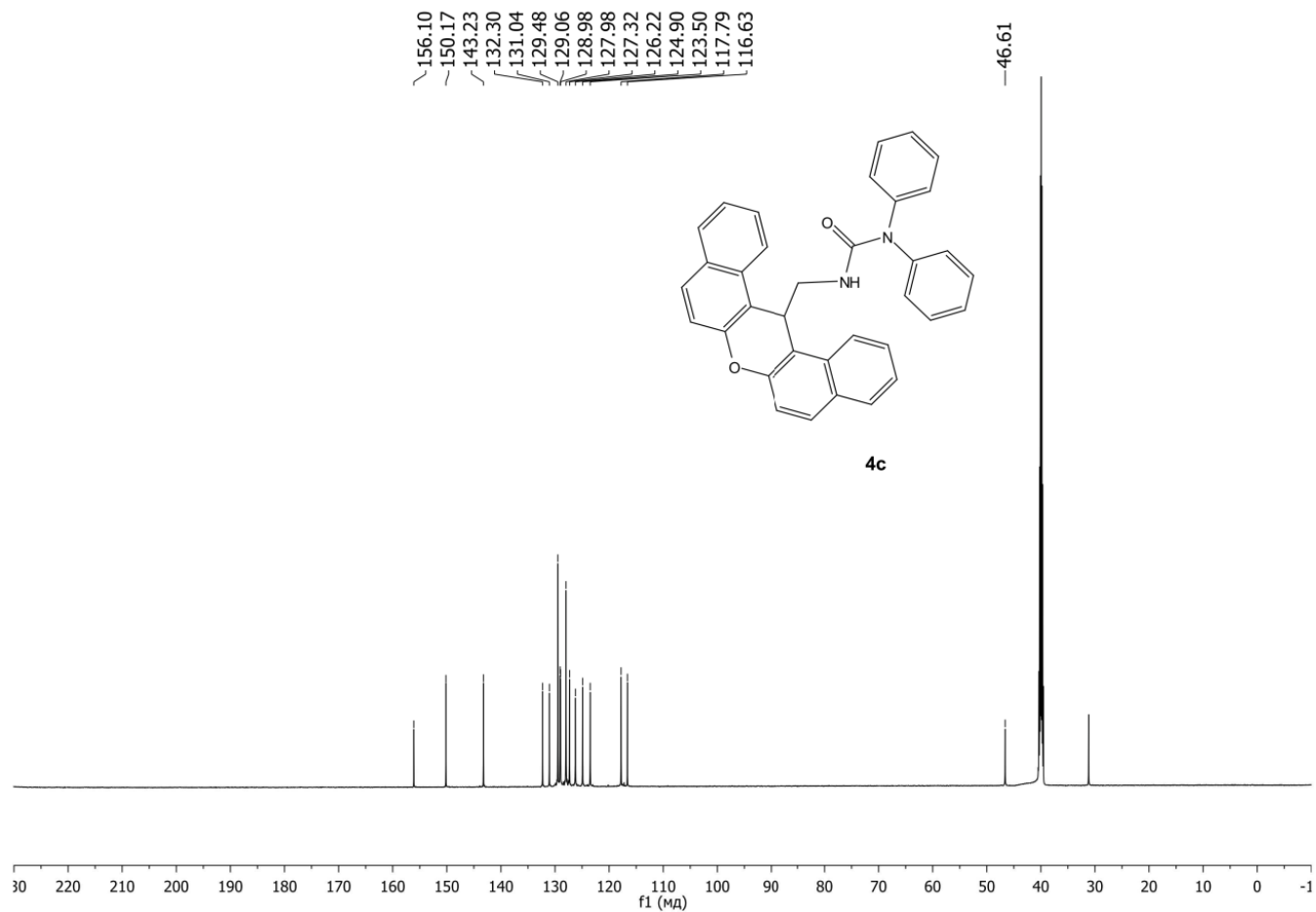


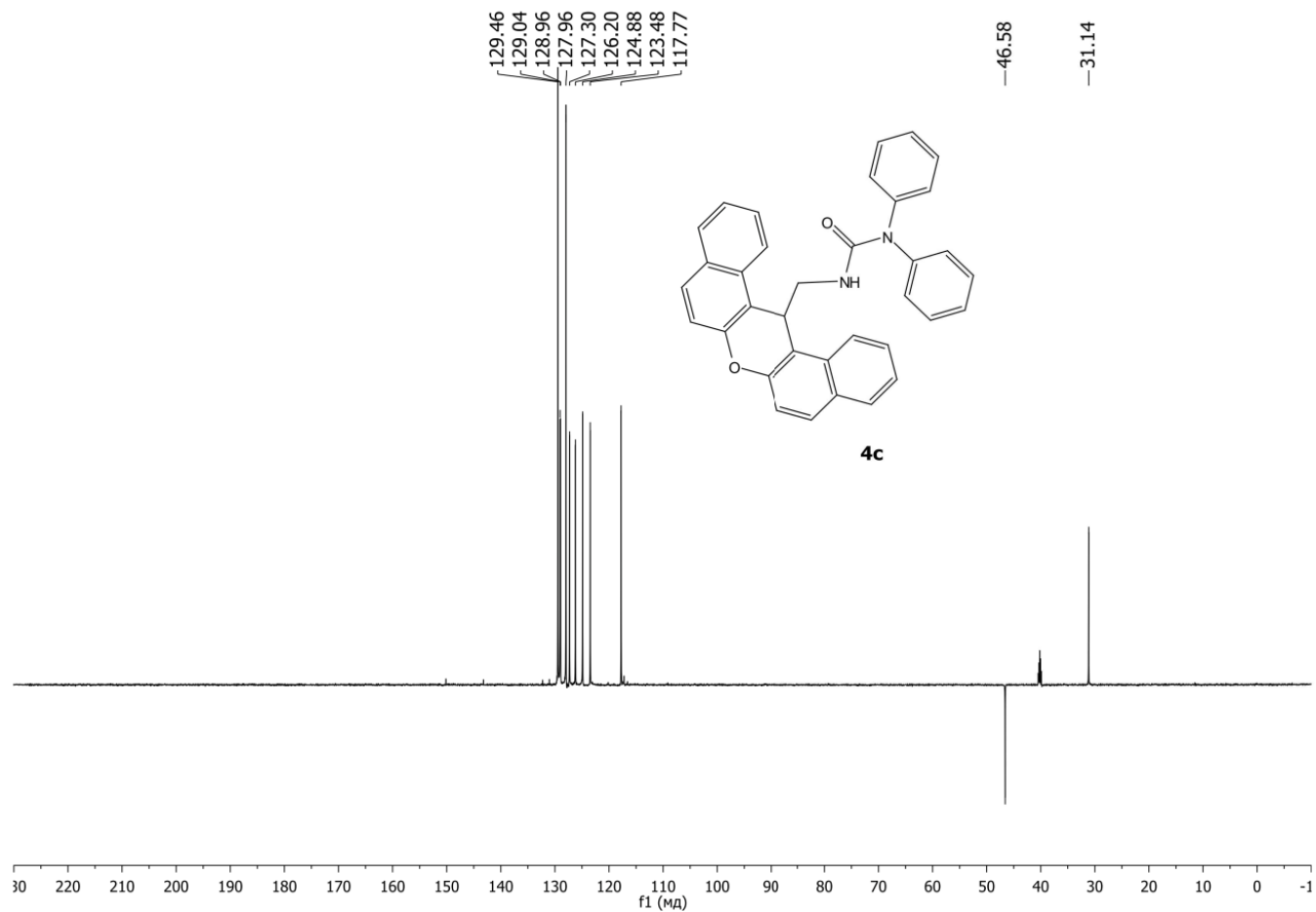




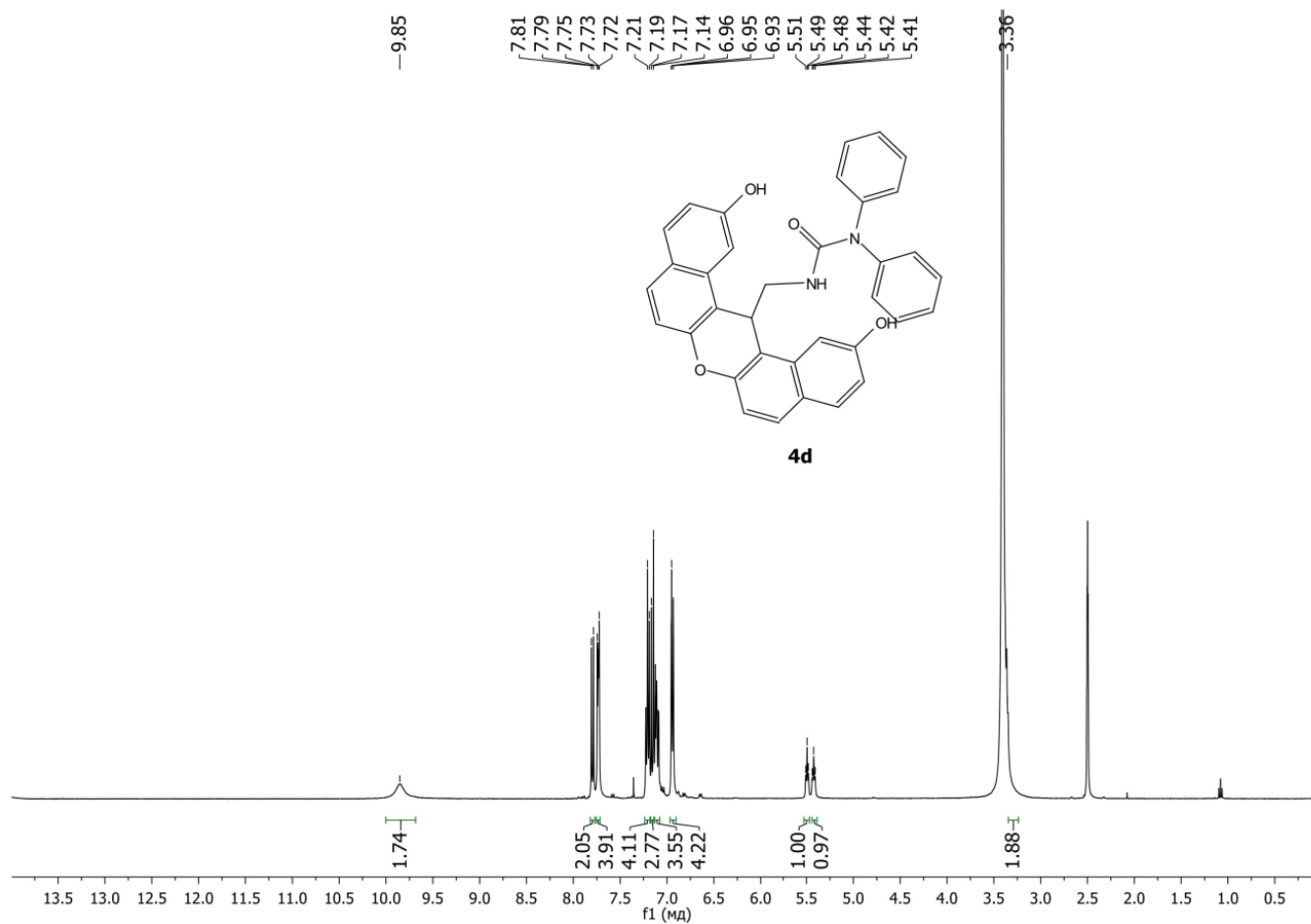
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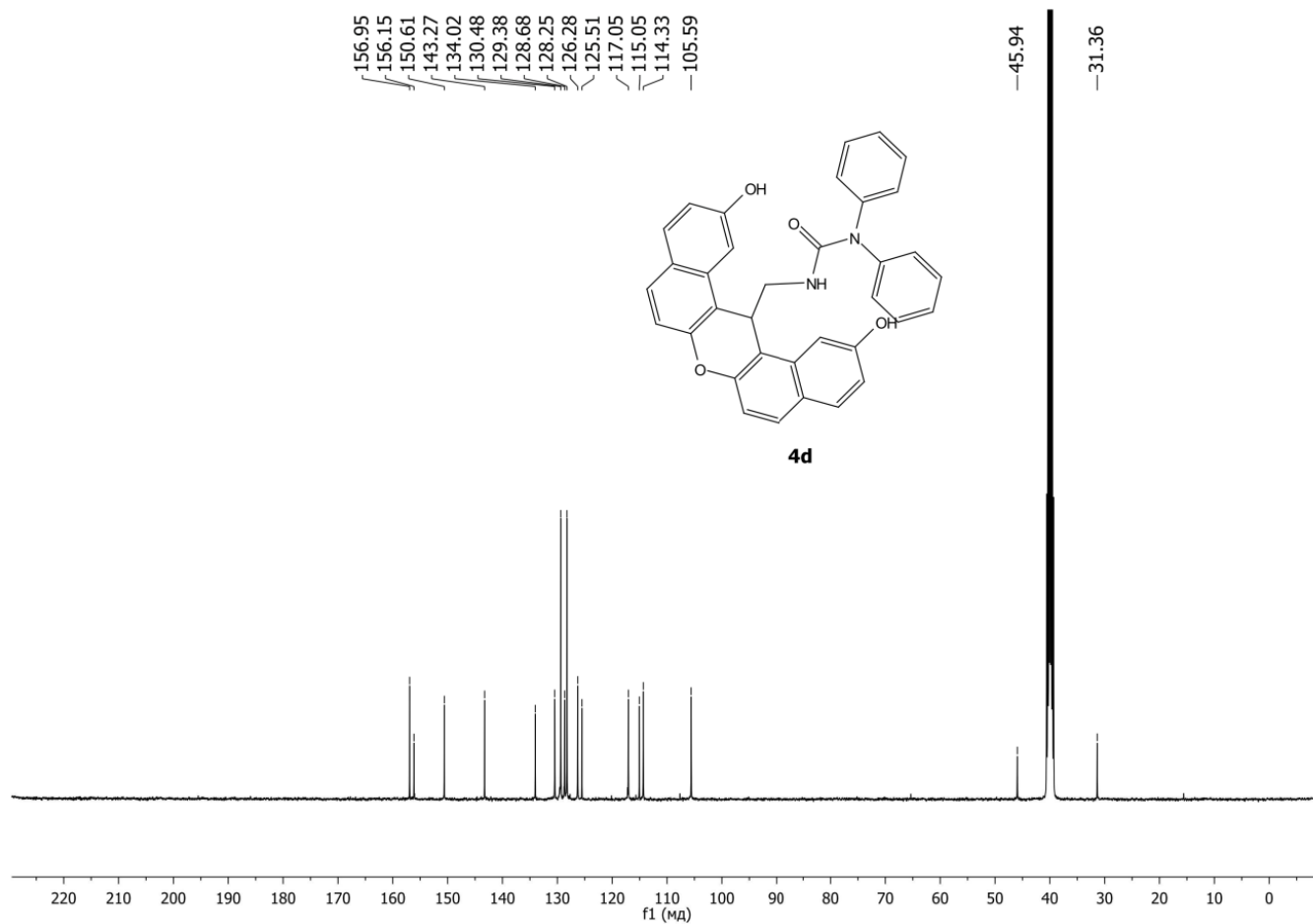


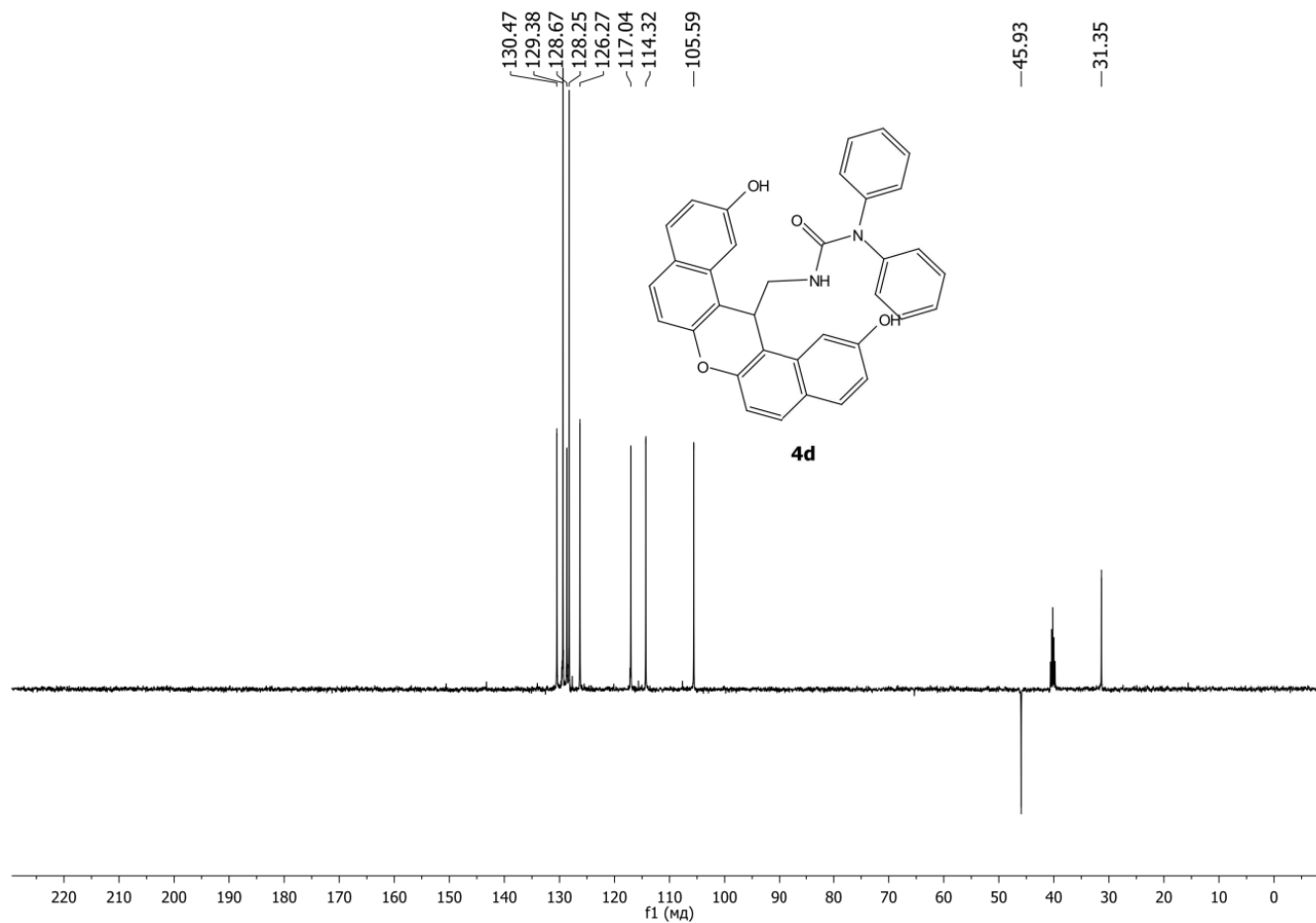








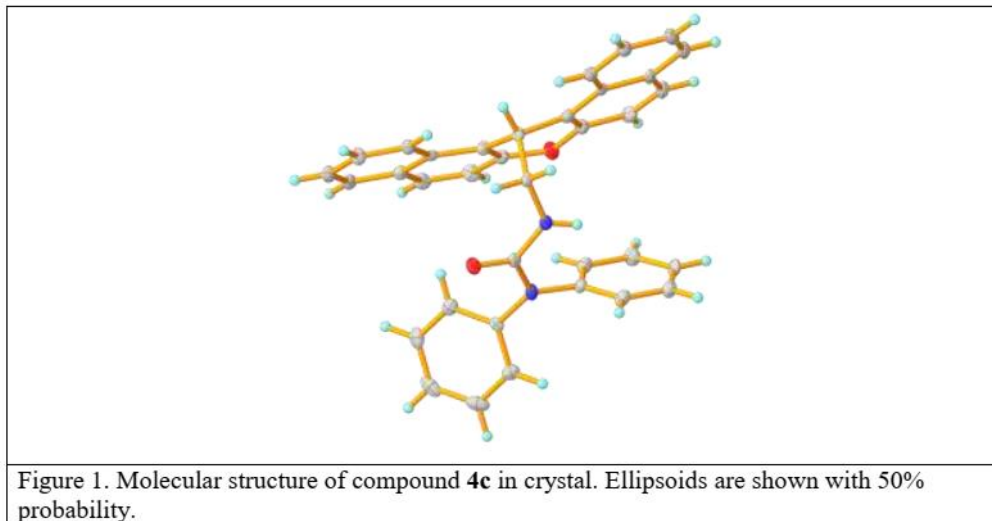


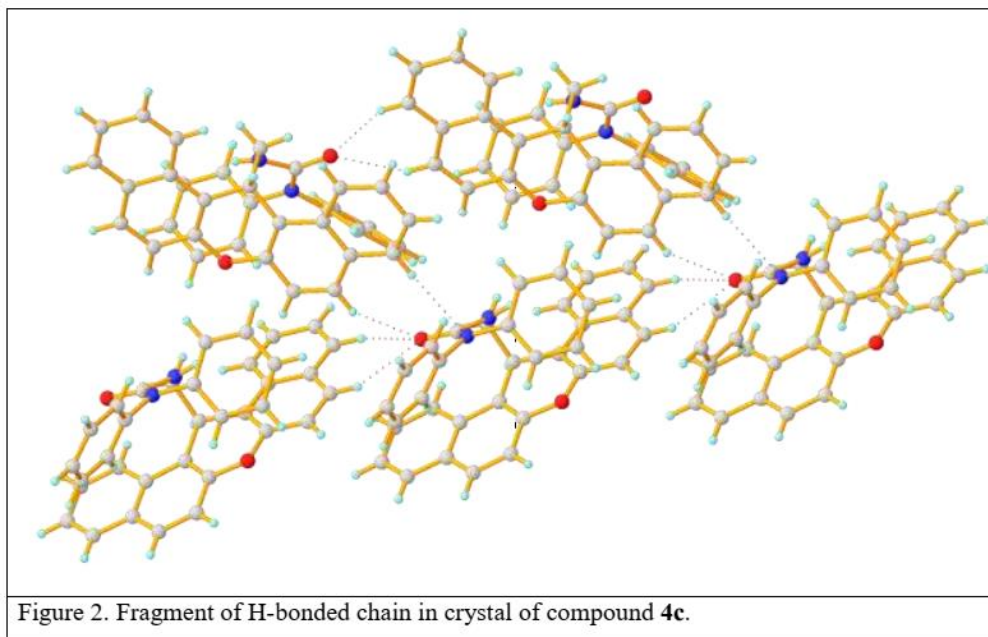


**Experimental.** The X-ray diffraction data for the crystal of **4c** were collected on a Smart Apex II automatic diffractometer using graphite monochromated radiation. The structures were solved by direct methods and refined by full-matrix least-squares using the SHELXL97<sup>1</sup> program. All the non-hydrogen atoms were refined with anisotropic atomic displacement parameters. All figures were made using the program OLEX2<sup>2</sup>. Crystallographic data for the structure reported in this paper have been deposited with the Cambridge Crystallographic Data Center (deposit number is 1950456).

**Crystal data for 4c:** C<sub>35</sub>H<sub>26</sub>N<sub>2</sub>O<sub>2</sub>, *M* = 506.58, colorless crystal, monoclinic, space group *P*2<sub>1</sub>/*c*, *Z* = 4, *a* = 10.1977(9), *b* = 12.3675(13), *c* = 20.1934(18)Å, β = 103.246(7)°, *V* = 2479.0(4) Å<sup>3</sup>, ρ<sub>calc</sub> = 1.357 g/cm<sup>3</sup>, μ = 0.084 mm<sup>-1</sup>, 19605 reflections collected (±*h*, ±*k*, ±*l*), 5950 independent (*R*<sub>int</sub> = 0.1283) and 2945 observed reflections [*I* ≥ 2 σ(*I*)], 353 refined parameters, *R* = 0.0780, *wR*<sup>2</sup> = 0.2132, residual electron density 0.327(-0.379)eÅ<sup>-3</sup>.

The bond lengths, valence and torsion angles in the molecules of **4c** in crystal are in the ranges typical for every bond type. The conformation of molecule **4c** is folded and stabilized by CH(Ph)...π and π...π interactions.





Crystal packing of investigated compound is formed by the large number of non-covalent interactions. Infinite chains are formed by the hydrogen bonds with additional stabilization of Lp... $\pi$  interactions. Folded layers are formed perpendicular to them due to stacking and CH... $\pi$  interactions.

Table 1. H-bonds in crystal of compound **4c**.

| H-bond               | D – H, Å | H...A, Å | D...A, Å | D - H...A, ° |
|----------------------|----------|----------|----------|--------------|
| C(7)-H(7)...O(15)    | 0.93     | 2.54     | 2.862(4) | 101          |
| C(17)-H(17B)...O(15) | 0.97     | 2.37     | 2.775(4) | 104          |
| C(22)-H(22)...O(15)  | 0.93     | 2.53     | 3.383(4) | 153          |

Table 2. Lp... $\pi$  interactions in crystals of **4c**.

| Lp... $\pi$          | X..Cg*   | X-Perp | Gamma | Y-X..Cg  |
|----------------------|----------|--------|-------|----------|
| C(14)-O(15)... Cg(6) | 3.800(3) | 3.340  | 28.49 | 75.4(2)  |
| C(14)-O(15)... Cg(7) | 3.873(3) | 3.378  | 29.29 | 111.4(2) |

\*Cg – centroid of the ring.

Table 3. C-H... $\pi$  interactions in crystals of **4c**.

| C-H... $\pi$        | H..Cg, Å | H-Perp, Å | Gamma, ° | X-H..Cg, ° | X..Cg, Å | C-H... $\pi$ |
|---------------------|----------|-----------|----------|------------|----------|--------------|
| C(7)-H(7)...Cg(6)   | 2.96     | 2.79      | 19.37    | 153        | 3.811(4) | 63           |
| C(25)-H(25)...Cg(2) | 2.84     | -2.74     | 15.83    | 125        | 3.468(4) | 51           |

Table 4.  $\pi \dots \pi$  interactions in crystals of **4c**.

| $\pi \dots \pi$ | Cg-Cg, Å | Alpha, °  | Beta, ° | Gamma, ° | CgI_Perp, Å | CgJ_Perp, Å |
|-----------------|----------|-----------|---------|----------|-------------|-------------|
| Cg(2)...Cg(7)   | 5.391(2) | 68.02(17) | 42.6    | 61.2     | 2.5974(15)  | 3.9682(14)  |
| Cg(3)...Cg(6)   | 5.589(2) | 39.72(16) | 59.8    | 48.3     | 3.7215(14)  | 2.8086(14)  |
| Cg(4)...Cg(3)   | 4.849(2) | 47.81(16) | 10.0    | 38.0     | 3.8213(14)  | -4.7751(14) |
| Cg(4)...Cg(5)   | 5.065(2) | 1.44(17)  | 49.0    | 50.3     | 3.2354(15)  | 3.3254(15)  |
| Cg(4)...Cg(7)   | 4.641(2) | 10.56(17) | 47.7    | 39.0     | 3.6043(14)  | 3.1224(14)  |
| Cg(5)...Cg(4)   | 5.065(2) | 1.44(17)  | 50.3    | 49.0     | 3.3254(15)  | 3.2355(15)  |
| Cg(5)...Cg(5)   | 4.410(2) | 0.03(17)  | 42.4    | 42.4     | 3.2580(15)  | 3.2580(15)  |
| Cg(5)...Cg(7)   | 4.949(2) | 10.99(17) | 50.2    | 48.5     | 3.2778(15)  | 3.1707(14)  |
| Cg(6)...Cg(3)   | 5.589(2) | 39.72(16) | 48.3    | 59.8     | 2.8085(14)  | 3.7215(14)  |
| Cg(7)...Cg(4)   | 4.641(2) | 10.56(17) | 39.0    | 47.7     | 3.1224(14)  | 3.6044(14)  |
| Cg(7)...Cg(5)   | 4.949(2) | 10.99(17) | 48.5    | 50.2     | 3.1707(14)  | 3.2778(15)  |

1. Sheldrick, G. M. SHELXTL v.6.12, Structure Determination Software Suite, Bruker AXS, Madison, WI, USA, 2000.
2. Dolomanov, O. V.; Bourhis, L. J.; Gildea, R. J.; Howard, J. A. K.; Puschmann, H. J. Appl. Crystallogr. 2009, 42, 339-341.