

Recent advances on organocatalysed asymmetric Mannich reactions

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Abstract

The asymmetric Mannich reaction is one of the most useful carbon-carbon bond forming reactions for the synthesis of chiral molecules containing nitrogen. The resulting β -amino carbonyl compounds are valuable synthons in the preparation of many natural products with useful biological properties. This review provides an overview of asymmetric Mannich reactions in recent years under different organocatalytic systems, including: chiral amines, chiral bifunctional thiourea, chiral Brønsted acids and other chiral organocatalytic systems.

Keywords: Asymmetric Mannich reaction, organocatalysis, enantioselective, diastereoselectivity

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Introduction

The asymmetric Mannich reaction¹⁻⁷ is one of the most powerful carbon-carbon bond-forming reactions for the construction of nitrogen-containing compounds. The utilization of this reaction allows for the preparation of optically enriched β -amino carbonyl compounds and their

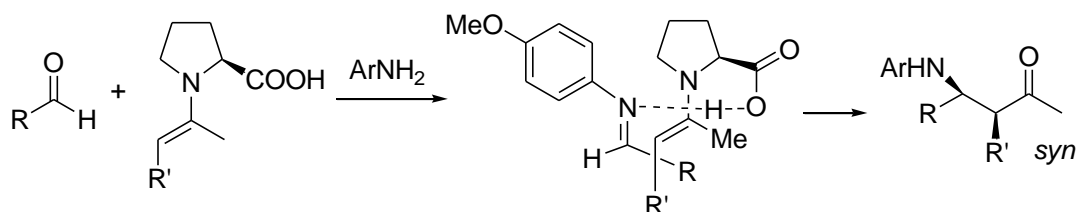
derivatives. In some instances these reactions have proven effective for the generation of biologically significant and synthetically useful β -amino acids that contain a quaternary stereocenter substituted with a nitrogen atom adjacent to the carbonyl group.⁸⁻¹⁰ Traditionally, asymmetric Mannich reactions are catalyzed by chiral transition-metal complexes.¹¹⁻¹⁷ In 2000, List described firstly the *L*-proline catalysed Mannich reaction.¹⁸⁻¹⁹ This landmark discovery stimulated the rapid development of many asymmetric organocatalytic Mannich reactions. The typical organocatalytic approach to asymmetric Mannich reaction is based on enamine activation of carbonyl compounds using secondary amine organocatalysts.²⁰ Other types of organocatalysts have also been successfully used for Mannich-type reactions. This review provides an overview of asymmetric Mannich reactions covering from 2007 to now under different organocatalytic systems in recent years. Several organocatalytic approaches will be reviewed, which can be divided in catalysis by (i) chiral amines, (ii) chiral bifunctional thioureas (iii) chiral Brønsted acids, and (iv) other chiral organocatalysis.

1. Catalysed by Chiral Amines

Organocatalytic Mannich reactions can be carried out either as three-component, one-pot reactions or as reactions of preformed imines with aldol donors. Chiral amines resulting in chiral enamines can attack a Mannich acceptor, usually a prochiral aldimine, thereby introducing one or two stereocenters in the Mannich product. The catalytic cycle is completed by regeneration of the amine catalyst through hydrolysis. The products are β -aminoaldehydes or β -aminoketones, which are optionally substituted at the α -position.⁵

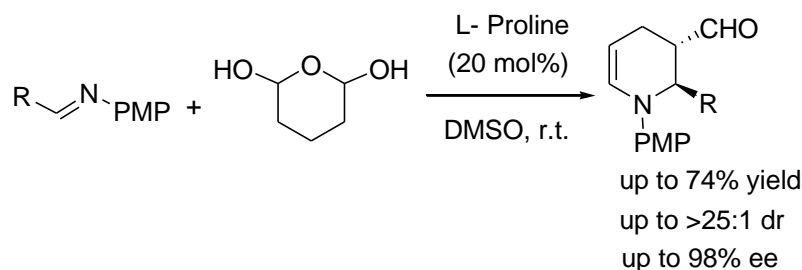
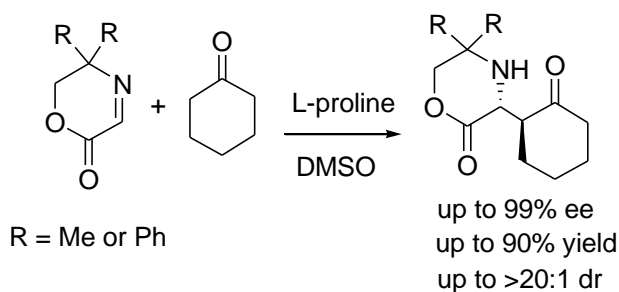
1.1 *L*-Proline and its derivatives

Among a wide variety of organocatalysts that have been used in the asymmetric Mannich reaction, the most widely used is proline. *L*-proline-catalysed Mannich reactions gives easy access to syn-products. Mechanistically, the stereochemical outcome of all of the reactions can be explained by invoking a transition state as depicted in Scheme 1. The stereochemical repulsion between the PMP-group and the proline moiety, in combination with protonation of the imine by the acid-functionality of proline, accounts for a *si*-face attack of the (*E*)-aldimine (from *p*-anisidine and acceptor aldehyde) by the *si*-face of the (*E*)-enamine formed by the ketone and proline.²¹ This model explains the stereochemical outcome of many similar reactions that have appeared in literature.

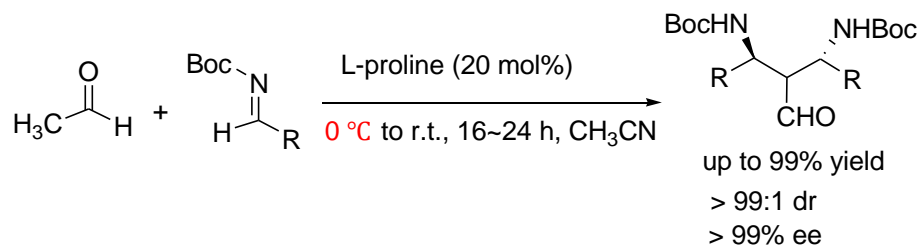


Scheme 1

In 2008, Xu and coworkers²² reported an enantioselective synthetic method for substituted tetrahydropyridines via a proline-mediated cascade Mannich-type/intramolecular cyclization (Scheme 2). The advantage of the organocatalyst was that the Mannich-type reaction proceeded efficiently with excellent diastereo- and enantioselectivity in the presence of water. This strategy would easily provide access to structurally diverse N-PMP piperidines. In the same year, Glorius *et al.*²³ developed the proline-catalyzed Mannich reaction of unactivated ketones, and demonstrated that the use of cyclic acceptors enabled the highly stereoselective synthesis of chiral 3-substituted morpholin-2-ones (Scheme 3). These products corresponded to α -D-amino acids that were protected at the N and O terminus by the diphenylethylene group. This protecting group for α -amino acids could be cleaved readily by hydrogenolysis in aqueous ethanol to furnish the free amino acid.

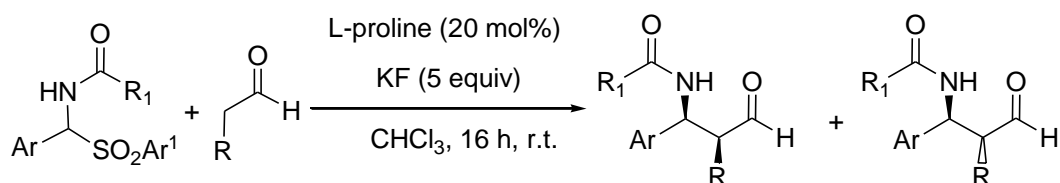
**Scheme 2****Scheme 3**

List¹⁹ introduced the one-pot catalytic asymmetric synthesis of pseudo- C_2 -symmetric β,β' -diaminoaldehydes with extremely high stereoselectivities, starting from acetaldehyde and either aromatic or aliphatic N-Boc imines (Scheme 4). The method was effectively extended to cross-Mannich reactions, furnishing β,β' -diamino aldehydes containing three adjacent stereogenic centers.

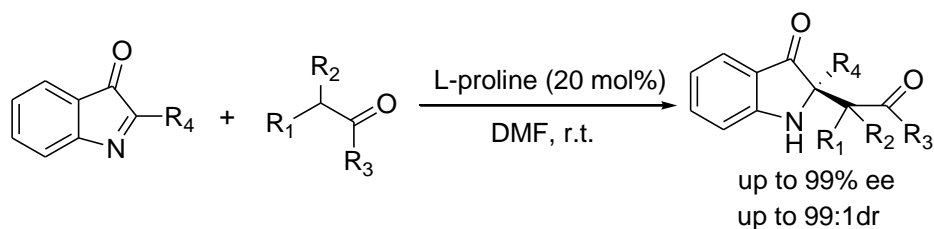


Scheme 4

In 2010, Zhao and coworkers²⁴ reported the one-pot organocatalytic reactions between α -amido sulfones and unmodified aldehydes proceeded with high chemo- and enantioselectivities to furnish β -amino aldehydes in high yields with up to 95:5 dr and up to 99% ee (Scheme 5). In the same year, Li *et al.*²⁵ developed that 2-Aryl-3*H*-indol-3-ones reacted with aldehydes or ketones to afford the corresponding aza-quaternary carbon addition product in good yield with moderate to excellent regioselectivity and enantioselectivity, showed *L*-proline was an effective catalyst in the reaction. The system was applied to the reaction of 2-(2-bromo-phenyl)-3*H*-indol-3-one and acetaldehyde to produce 2-[2-(2-bromophenyl)-3-oxoindolin-2-yl] acetaldehyde, which was a precursor for the synthesis of some alkaloids such as hinckdentine A (Scheme 6).



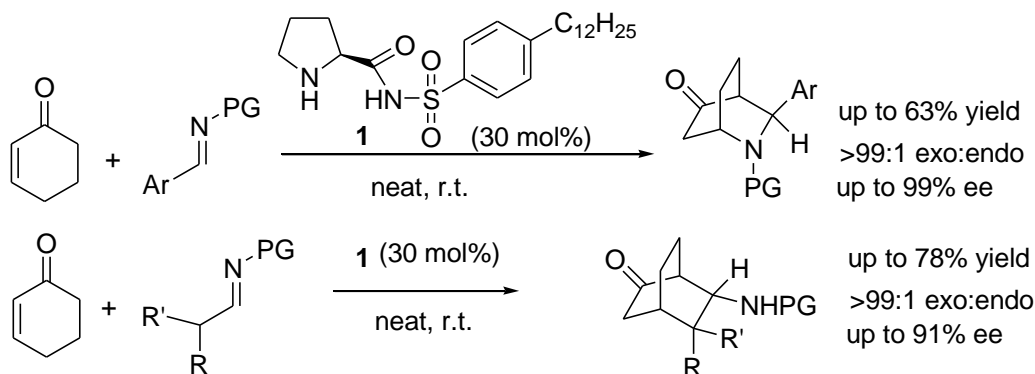
Scheme 5



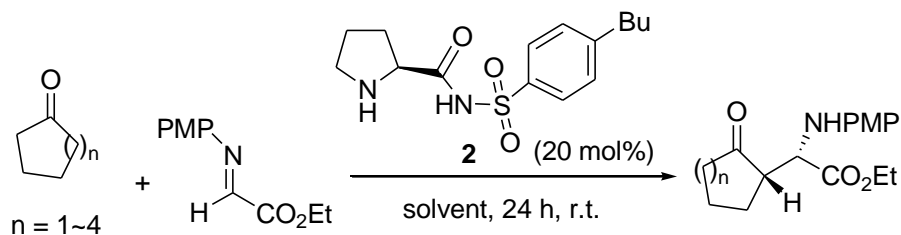
Scheme 6

In 2009, Carter and Yang²⁶ developed an organocatalyzed method for accessing nitrogen-containing [2.2.2]-bicyclic scaffolds in a highly enantioselective and diastereoselective manner (Scheme 7). The *p*-dodecylphenylsulfonamide catalyst (**1**) allowed for the scope of this formal aza-Diels-Alder process to be expanded to include aryl imines. Additionally, alkyl imines proceeded with a divergent and novel reaction pathway, further demonstrating the utility of this technology. Next year, Sebesta *et al.*²⁷ showed *L*-proline-derived sulfonamides (**2**) was effective

catalysts in the Mannich reaction of cyclohexanone with N-PMP-protected α -imino ethylglyoxylate with practical advantages in comparison with *L*-proline in different solvents and ionic liquids (Scheme 8). Ionic liquids could be used as solvents as well, although in this case, proline sulfonamides were less diastereo- and enantioselective than common organic solvents. Owing to larger differences in ionic liquids than in molecular solvents, a broader range of ionic liquids seemed to be necessary to gain deeper insight into the reactivity of these catalysts in the Mannich reaction.

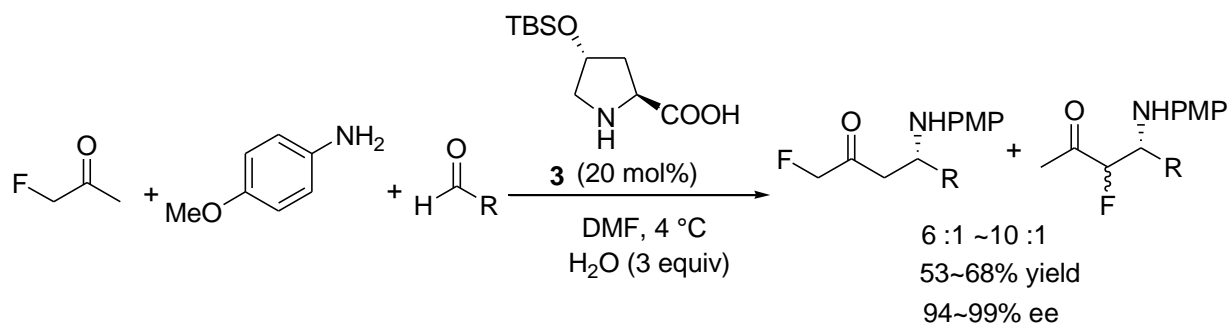


Scheme 7

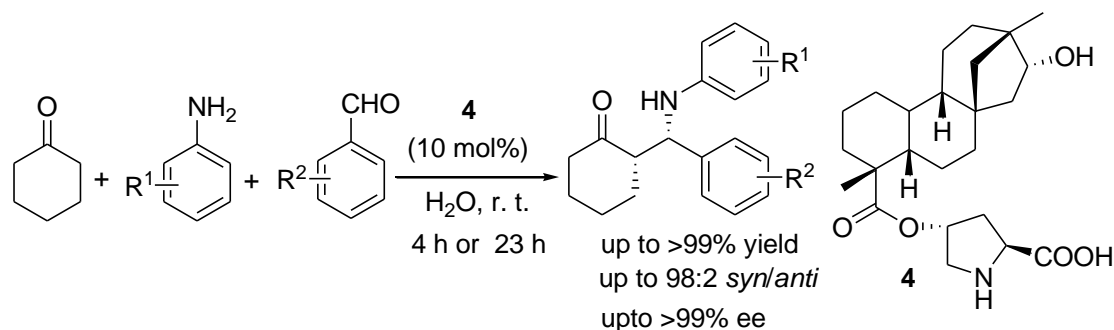


Scheme 8

In 2011, Lu *et al.*²⁸ found the direct Mannich protocol with highly enantioselectivity employing fluoroacetone, *p*-anisidine, and aldehydes catalyzed by 4-siloxypyrrolidine (**3**), the approach allowed efficient access for pharmaceutically important fluorinated β -amino ketones (Scheme 9). Recently, An *et al.*²⁹ developed the asymmetric three-component Mannich reactions of cyclohexanone and anilines with aromatic aldehydes in the presence of H₂O mediated by Isosteviol–proline (**4**) as highly efficient amphiphilic organocatalysts, and afforded *syn*-Mannich products with excellent diastereoselectivities (*syn/anti* up to 98 :2) and enantioselectivities (up to >99% ee) (Scheme 10).



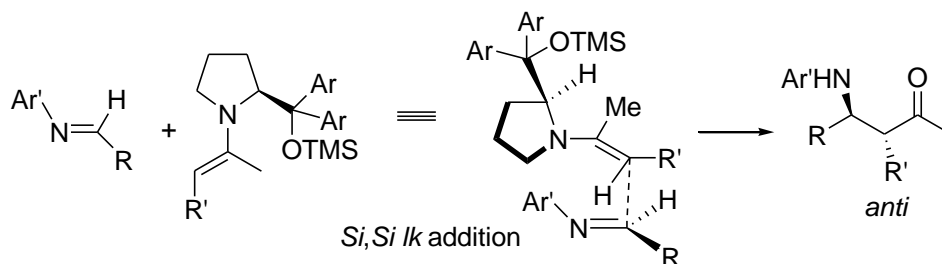
Scheme 9



Scheme 10

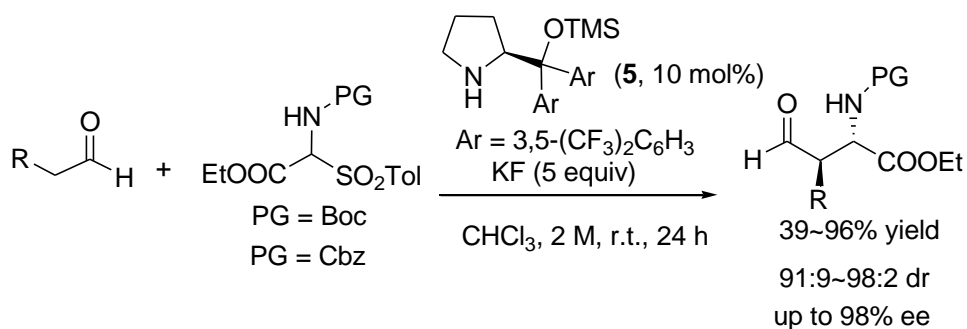
1.2 Pyrrolidine derivatives

Nowadays, a series of enamine forming amines are available that give rise to the anti-products in good selectivity. The stereoselectivity observed when catalysts are used can be rationalized through the proposed transition state in the well-established enamine catalysis mechanism (Scheme 11).³⁰ Thus, nucleophilic attack on the imine preferentially occurs from the (*Si,Si*) *lk* face to afford the corresponding anti- β -amino aldehydes as major products. The asymmetric induction caused by the bulky pyrrolidine substituent is the opposite of that induced by hydrogen-bonding when proline is used as catalyst.

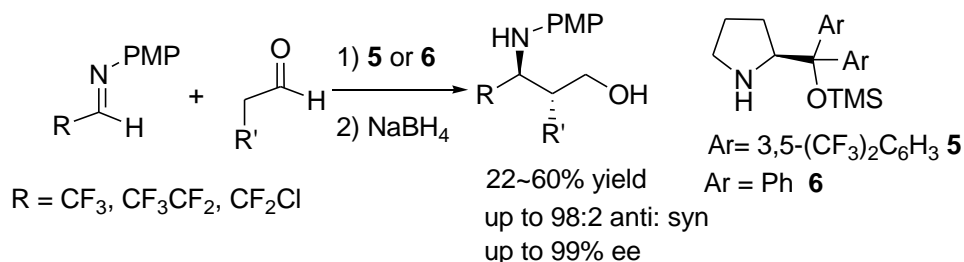


Scheme 11

In 2008, Melchiorre group³¹ described the the Hayashi-Jørgensen catalyst (**5**) catalyzed *anti*-selective Mannich reaction of aldehydes with N-Cbz- and N-Boc-protected imines generated *in situ* from stable amido sulfones (Scheme 12). Besides the high level of efficiency and stereocontrol achieved, this approach introduced important synthetic advantages by avoiding the requirement to preform the *N*-carbamoyl imines. The following year, Fustero *et al.*³² reported that α,α -diarylprolinol trimethylsilyl ether (**5** or **6**) catalyzed the asymmetric Mannich reaction between fluoroalkyl aldimines and aldehydes. The corresponding Mannich adducts were reduced and afforded *anti*- β -alkyl- γ -fluoroalkyl- γ -amino alcohols in highly diastereo- and enantioselectivities (Scheme 13).



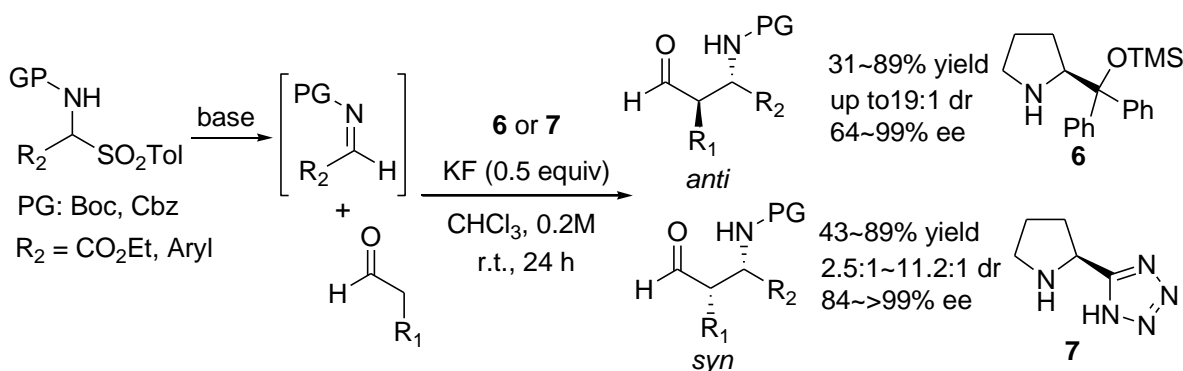
Scheme 12



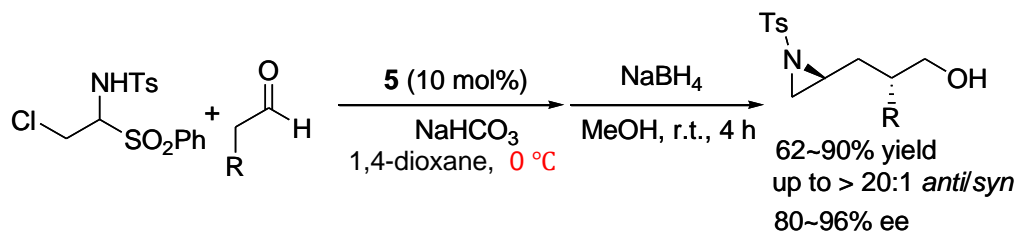
Scheme 13

Melchiorre *et al.*³³ developed a highly efficient system for the asymmetric amino catalytic Mannich reaction of unmodified aldehydes with *in situ* generated N-carbamoyl imines in 2010, (Scheme 14). The main feature of this method lay in the operational simplicity: the highly reactive N-carbamate-protected imines were generated *in situ* from stable, easily handled α -amido sulfones. The judicious selection of commercially available chiral amine catalysts **6** and **7** allowed full control of the stereochemistry of the Mannich process; either the *syn*- or *anti*- β -amino aldehydes were accessible with very high stereocontrol. In 2011, Hayashi group³⁴ also showed a highly diastereo- and enantioselective asymmetric Mannich reaction of imines derived from aliphatic and aromatic aldehydes catalyzed by diarylprolinol silyl ether **5** (Scheme 15). Later, They still developed a one-pot synthesis of chiral aziridine derivatives with excellent

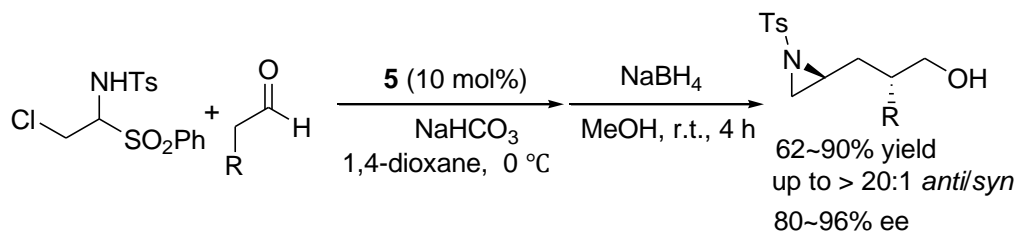
diastereo- and enantioselectivities through uninterrupted sequential reactions, including desulfonylative formation of the N-Ts imine derived from chloroacetaldehyde, a diarylprolinol silyl ether mediated asymmetric Mannich reaction, reduction, and aziridine formation (Scheme 16).³⁵ Because the generated product possesses several functional groups, with excellent diastereoselectivity and enantioselectivity and the synthetic procedure is simple, this method offers an efficient route for the preparation of chiral aziridine derivatives.



Scheme 14



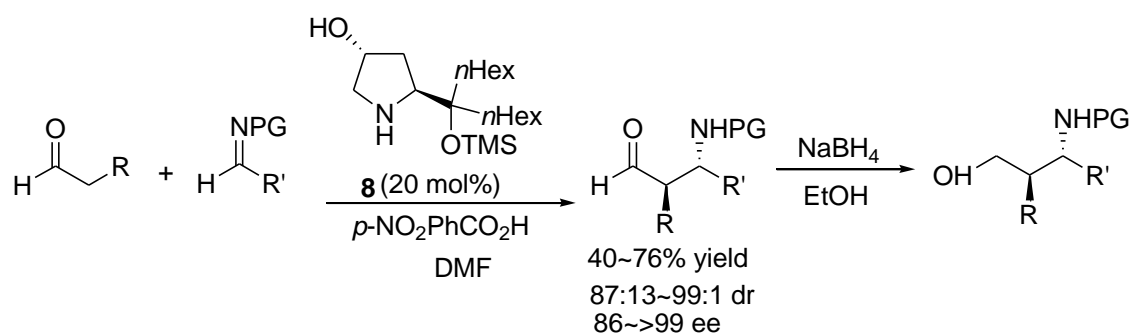
Scheme 15



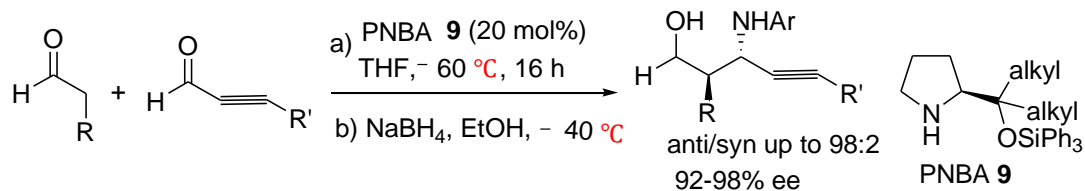
Scheme 16

In 2010, Palomo group³⁶ reported an anti-selective Mannich reaction of aldehydes with N-sulfonyl imines had been developed by using a 4-hydroxypyrrolidine in combination with an external Brønsted acid (Scheme 17). This catalyst system combined an amino group to activate the aldehyde donor substrate, and a hydroxy group together with an external Bronsted acid to

activate the imine acceptor component, while controlling the stereochemistry of the process. The Mannich adducts could be easily reduced or oxidized and then deprotected to give the corresponding β -amino acids and β -amino alcohols with good yields. The reaction results also showed that this ternary catalytic system was practical in other enamine-based reactions. In 2012, The group³⁷ still reported anti-selective and highly enantioselective Mannich reaction of aldehydes and unactivated imines mediated by the combined use of a Brønsted acid with an α,α -dialkylprolinol ether catalyst (**9**) with good yields (typically 70~75%), anti : syn ratios greater than 90 : 10, and ee values usually above 95% (Scheme 18). The method worked particularly well with propargylic imines and, unlike previous catalytic routes to optically active propargylamines, provided adducts featuring two contiguous stereocenters and a functionalized side chain amenable for ulterior synthetic applications.



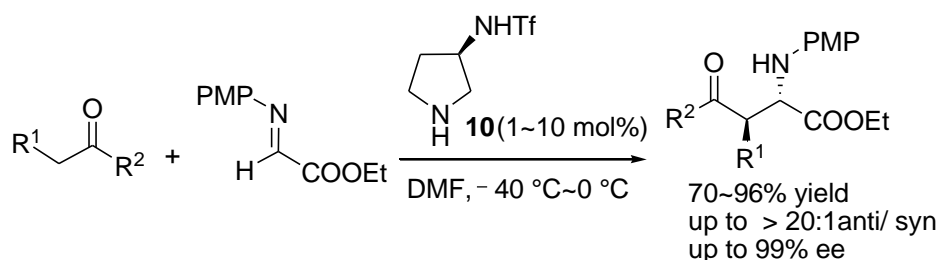
Scheme 17



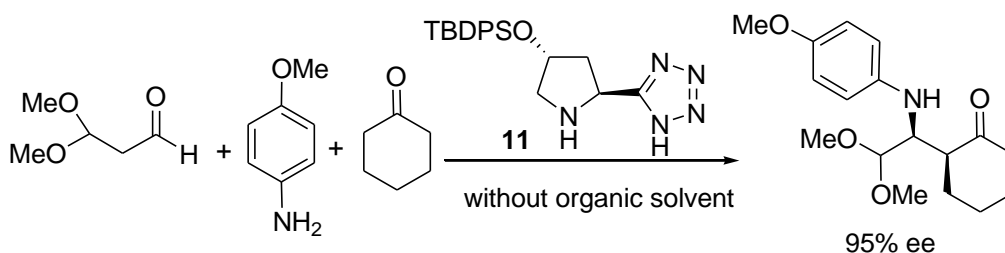
Scheme 18

Blanchet and coworkers³⁸ reported an *anti*-selective direct and three-component Mannich reaction catalysed by 3-trifluoromethanesulfonamido-pyrrolidine (**10**) which achieved high yields and selectivities for various substrates ranging from linear and branched aldehydes to ketones (Scheme 19). The research disclosed that the acidity of the trifluoromethylsulfonamide group was critical to achieve high stereoselectivity, and C-2 symmetry of catalyst was not a key structural feature for a high stereoselectivity. Similar work for the enantioselective *anti*-selective Mannich-type reactions of aldehydes and ketones with imines catalyzed by 3-pyrrolidinecarboxylic acid and related pyrrolidine derivatives was also reported by Tanaka *et al.*³⁹ in 2008. Hayashi group⁴⁰ developed an asymmetric Mannich reaction mediated by the

siloxytetrazole hybrid organocatalyst (**11**) in the presence of water without using organic solvents (Scheme 20).



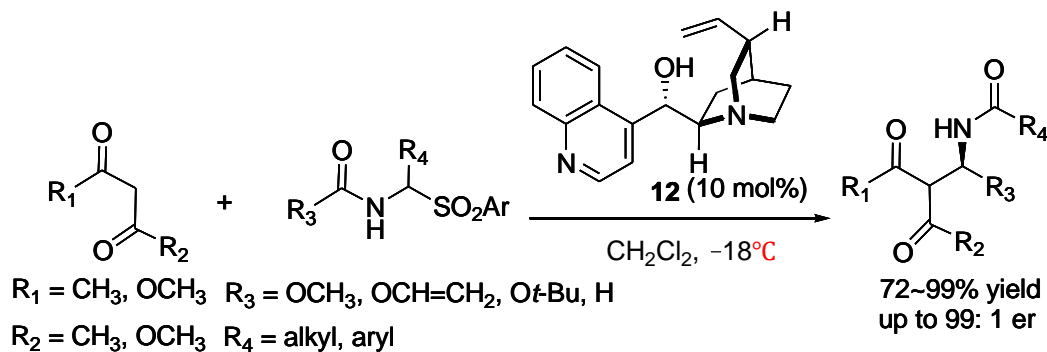
Scheme 19



Scheme 20

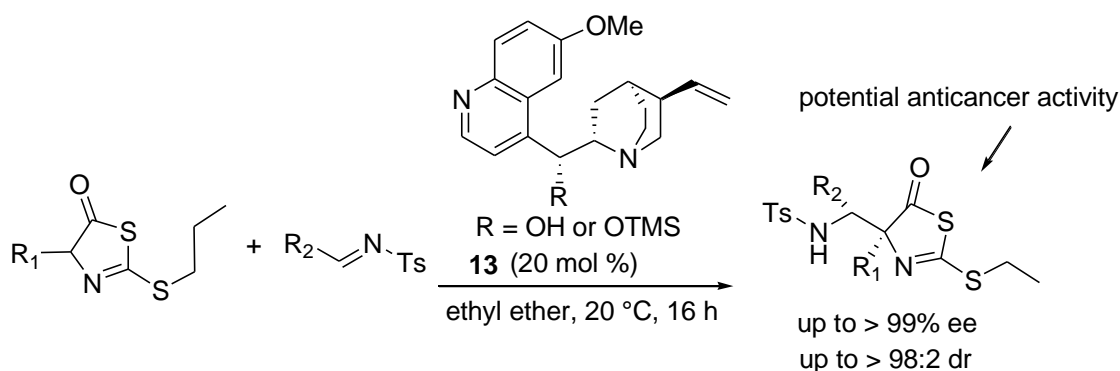
1.3 Cinchona alkaloids or theirs derivatives

Schaus *et al.*⁴¹ were first to describe the cinchona alkaloid-catalyzed Mannich reaction of dicarbonyl compounds with α -amido sulfones as acyl imine precursors in good yields and high enantioselectivities, and in diastereoselectivities that range from 1:1 to > 95:5 (Scheme 21). The reaction required 10 mol % of the cinchona alkaloid catalyst (**12**), which served as a general base to generate acyl imines *in situ*, and aqueous Na_2CO_3 to maintain the concentration of free alkaloid catalyst. The cinchonine catalyzed reactions gave practical access to highly functionalized building blocks which had been employed in the synthesis of chiral dihydropyrimidones, a class of compounds rich in diverse biological activity.



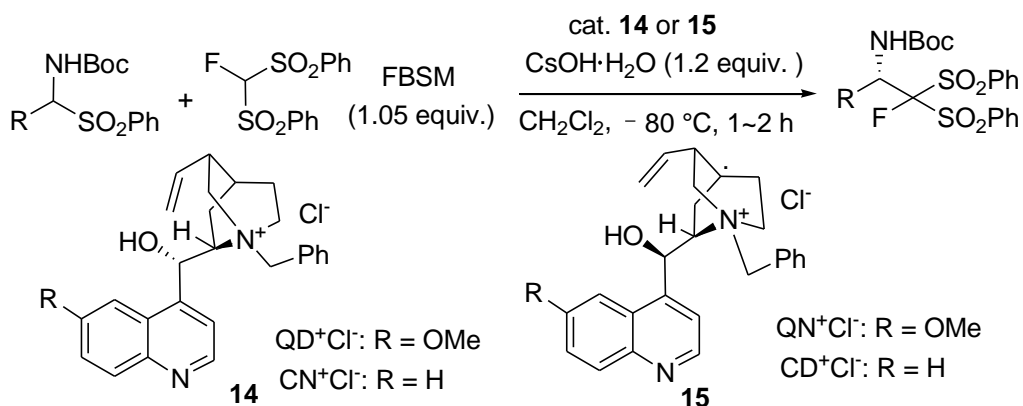
Scheme 21

Last year, Wang *et al.*⁴² showed the aza-Mannich addition of 2-(ethylthio)-thiazolones and *N*-tosyl aldimines were catalysed by cinchona alkaloid catalyst system (**13**) (Scheme 22). A series of masked chiral 2-(ethylthio)-thiazolone derivatives by establishing a carbon- and nitrogen-substituted quaternary carbon stereocenter, were synthesized with high levels of diastereo- (up to >98:2) and enantioselectivities (up to >99%). Several new potential anticancer active derivatives were obtained.



Scheme 22

Shibata and Toru *et al.*⁴³ described the catalytic enantioselective fluorobisphenylsulfonyl methylation of *in situ* generated imines from α -amido sulfones under the combination of Mannich type conditions with fluorobis(phenylsulfonyl)methane chemistry (Scheme 23). The α -fluorobisphenylsulfonylmethylated amines could be converted to α -mono-fluoromethyl amines by reductive desulfonylation.

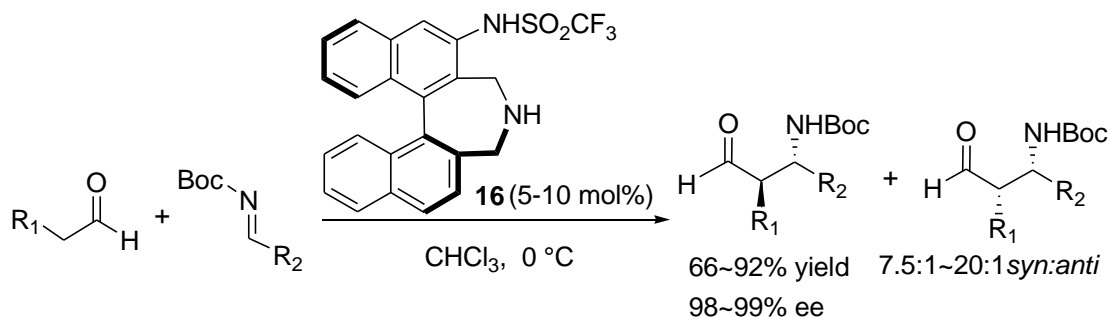


Scheme 23

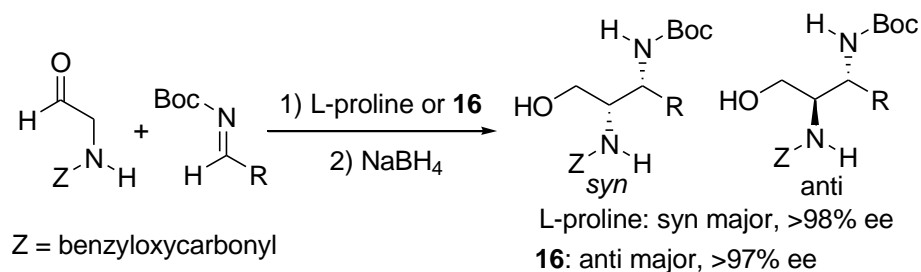
1.4 Other chiral amines

In 2009, the Maruoka group⁴⁴ developed a highly *anti*- and enantioselective direct asymmetric Mannich reaction catalysed by axially chiral amino sulfonamide (**16**) (Scheme 24). For instance,

in the presence of a catalytic amount of **16**, the reactions between aldehydes and α -imino esters proceeded smoothly to produce a higher *anti/syn* ratio as well as higher enantioselectivity Mannich products than previously possible for Mannich products. The axially chiral amino sulfonamide **16** was also successfully applied to asymmetric direct cross-aldol reaction between two different aldehydes. The advantage of catalyst **16** was giving mainly *syn* products, whereas proline showed the opposite *anti* selectivity. The same year, they still reported that a highly stereoselective direct asymmetric Mannich reaction between acetaldehyde and *N*-Boc-protected imines, as well as an anti-selective direct asymmetric Mannich reaction of *N*-Boc-protected imines by using the less nucleophilic chiral amino sulfonamide **16** to suppress the undesired side reactions.⁴⁵ Recently, this group⁴⁶ introduced both *syn*- and *anti*-selective asymmetric direct Mannich reactions of *N*-protected aminoacetaldehydes with *N*-Boc-protected imines catalyzed by proline and the axially chiral amino sulfonamide **16** (Scheme 25). This organocatalytic process represented the first example of a Mannich reaction using *Z*- or Boc-protected aminoacetaldehyde as a new entry of α -nitrogen functionalized aldehyde nucleophile in enamine catalysis. The obtained optically active vicinal diamines were useful chiral synthons as exemplified by the formal synthesis of (-)-agelastatin A.



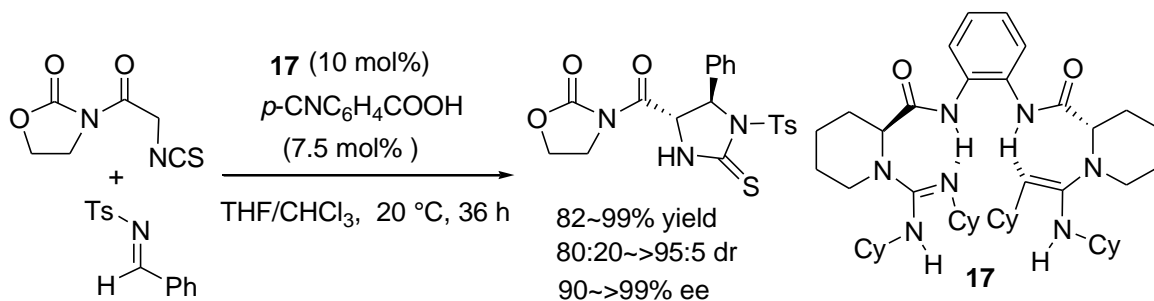
Scheme 24



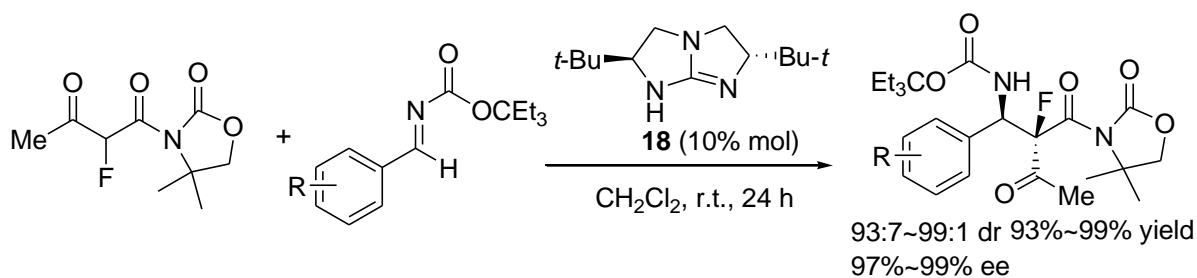
Scheme 25

In 2011, a highly efficient bisguanidine organocatalyst (**17**) for the Mannich-type reaction of isothiocyanato imide with *N*-Ts-protected imines was developed by Feng group (Scheme 26).⁴⁷ Significant progress had been made with an extremely broad substrate scope, giving optically

active α,β -diamino acid derivatives in excellent yields with high diastereoselectivities (up to > 95:5 d.r.) and excellent enantioselectivities (up to 99% ee) under mild conditions. Tan *et al.*⁴⁸ reported the highly enantio- and diastereoselective Mannich reaction catalyzed by guanidine (**18**) with α -fluoro- β -keto acyloxazolidinone as the fluorocarbon nucleophile. Fluoro- β -amino acid derivatives with chiral fluorinated carbon were obtained through selective deacylation or decarboxylation reactions. Besides, the employment of *N*-tosylimines could result in remarkably efficient enantioselective *anti*-Mannich reactions (Scheme 27).⁴⁹ The involvement of both oxygen atoms of sulfone in hydrogen bonding network to stabilize the transition state was unprecedented, and might have implications for the design of novel organocatalytic systems.

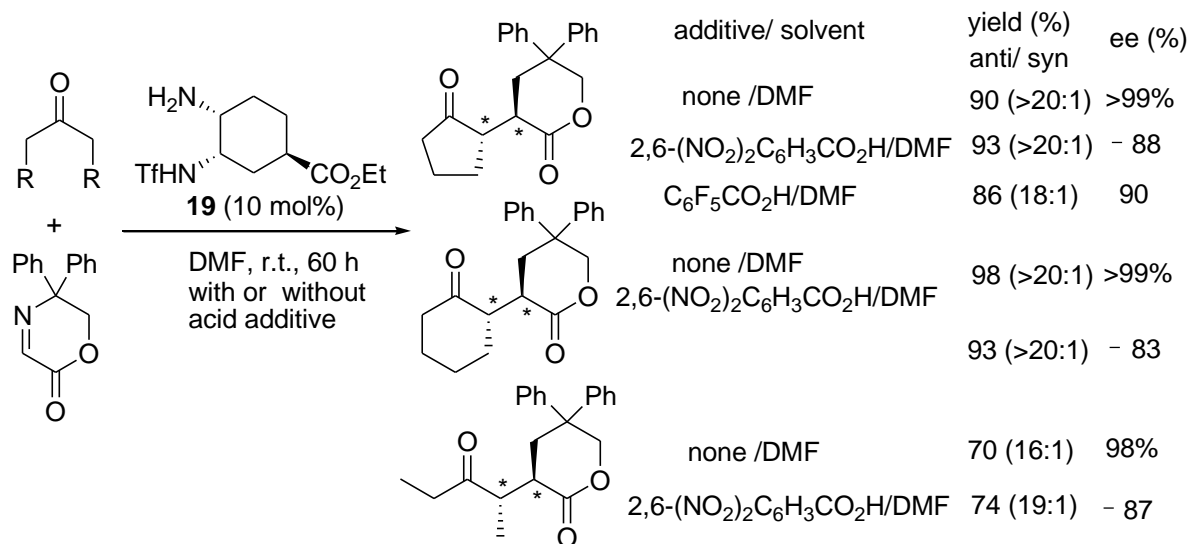


Scheme 26



Scheme 27

In 2012, Moteki *et al.*⁵⁰ introduced to a practical synthesis of both enantiomeric Mannich products in asymmetric Mannich reactions catalyzed by catalyst (**19**) with or without an acid additive, and the reaction gave *anti*-selective Mannich reaction of a cyclic imino ester (Scheme 28).

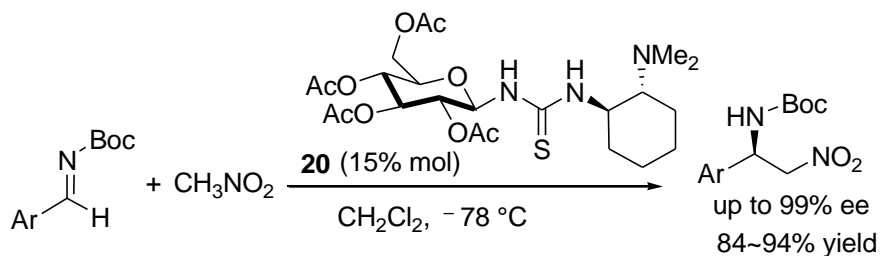


Scheme 28

2. Catalysed by Chiral Bifunctional Thioureas

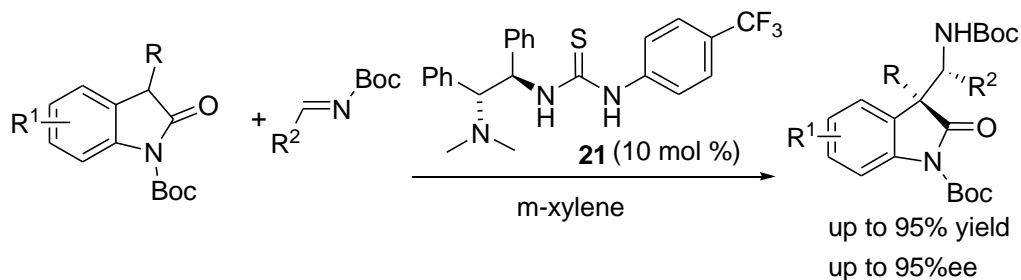
Thiourea-catalysed reactions were already earlier reported by Jacobsen group in 2002. The group firstly developed the thiourea-catalysed Mannich reaction that an efficient route to *N*-Boc-protected β -amino acids *via* the enantioselective addition of silyl ketene acetals to *N*-Boc aldimines in 2002.⁵¹ Later, this group reported again a highly enantioselective thiourea-catalyzed nitro-Mannich reactions in 2005.⁵²

In 2008, a novel bifunctional chiral thiourea organocatalyst (**20**) bearing a glycosyl scaffold and a tertiary amino group starting from readily available α -*D*-glucose was synthesized by the Zhou group (Scheme 29).⁵³ This thiourea was an effective organocatalyst for the asymmetric aza-Henry reaction between *N*-Boc imines and nitromethane. The corresponding adducts were obtained in good to excellent yields with highly *anti*-selective (93:7~99:1) and enantioselective (96~99% ee).



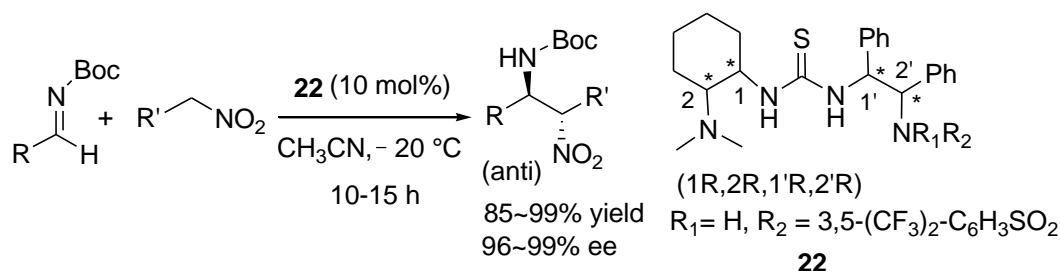
Scheme 29

The asymmetric Mannich reaction of 3-substituted oxindoles and N-Boc imines employing bifunctional thiourea-tertiary amine organocatalysts (**21**) based on diphenylethylene-diamine (DPEN) scaffold exhibited high diastereoselectivities.⁵⁴ The corresponding Mannich adducts bearing adjacent quaternary and tertiary chiral centers were generally obtained in good to excellent enantioselectivities (up to 95% ee) (Scheme 30).



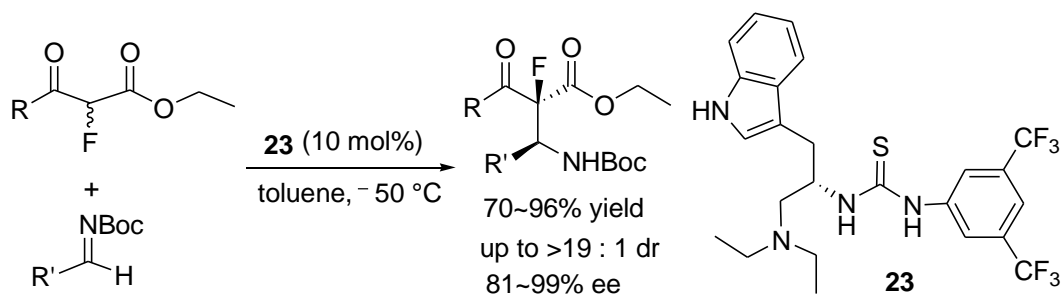
Scheme 30

Wang *et al.*⁵⁵ developed the highly *anti*-selective (93:7-99:1) and excellent enantioselective (96-99% ee) nitro-Mannich reactions catalyzed by chiral bifunctional thiourea catalyst (**22**) bearing multiple hydrogen-bonding donors that perform well over a broad scope of substrates (Scheme 31). This methodology was a nice complement the highly *syn*-selective version using a heterobimetallic Cu-Sm-Shiff base complex.



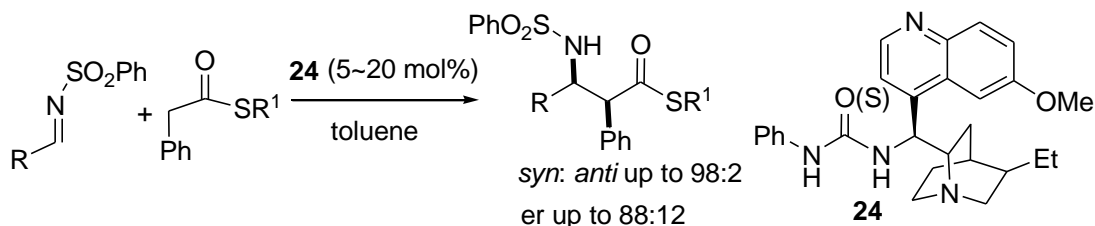
Scheme 31

Huang group⁵⁶ considered a novel tryptophan based bifunctional thiourea catalyst (**23**) that was remarkably effective in promoting the asymmetric Mannich reaction of α -fluoro- β -ketoesters (Scheme 32). The resulting compounds with fluorinated quaternary and tertiary stereocenters could be converted readily into α -fluoro- β -amino acids and α -fluoro- β -lactams. Preliminary computational studies suggested that the indole moiety of the catalyst played a crucial role in substrate binding. They disclosed that tertiary amine-thiourea bifunctional catalysts could be derived readily from natural amino acids, a strategy which may eventually lead to the discovery of various novel multifunctional organic catalysts.



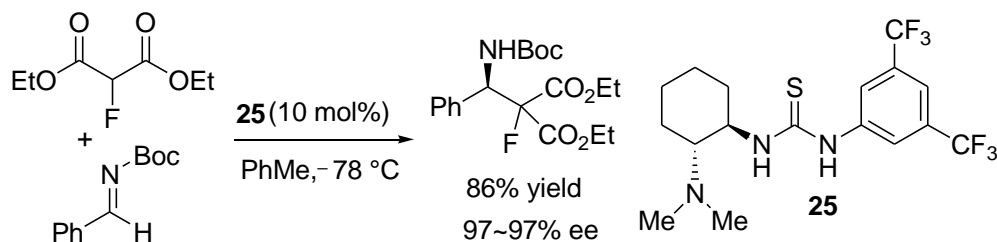
Scheme 32

In 2010, Coltart *et al.*⁵⁷ developed the organocatalytic Mannich reaction based on proximity-assisted intracomplex soft enolization of thioesters using simple derivatives of cinchona alkaloid-based catalysts (**24**) (Scheme 33). This approach to enolization was based on the cooperative action of a carbonyl activating hydrogen bonding (thio)urea moiety and an amine base contained within a single catalytic entity to facilitate intracomplex deprotonation. Significantly, this allowed thioesters over a range of acidity to react efficiently, thereby opening the door to the development of a general mode of enolization-based organocatalysis of monocarboxylic acid derivatives.



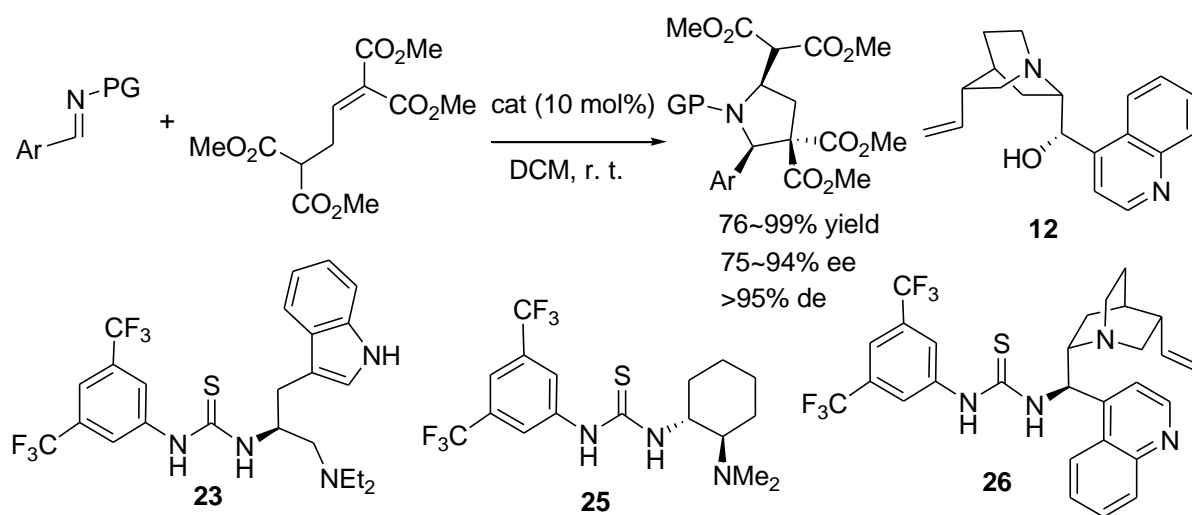
Scheme 33

Lee *et al.*⁵⁸ described the highly enantioselective Mannich reaction of diethyl fluoromalonate with *N*-Boc-aldimines promoted by chiral bifunctional organocatalysts (**25**), and the corresponding products β -amino- β -fluoromalonates were obtained with excellent enantioselectivity (93–97% ee) under mild reaction conditions (Scheme 34).



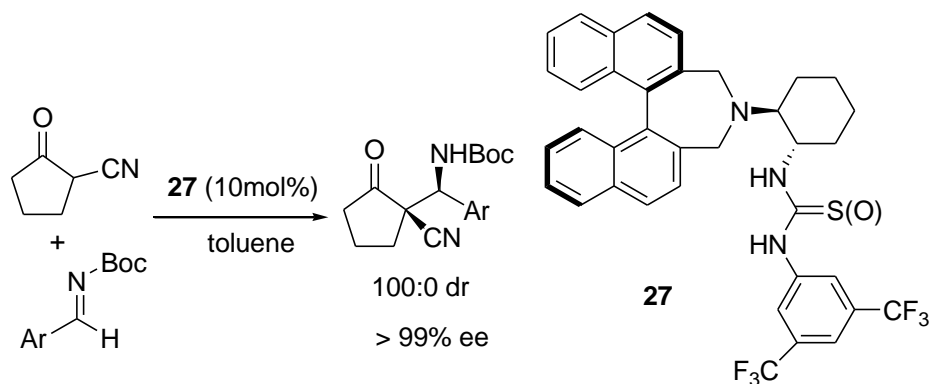
Scheme 34

In 2010, Enders *et al.*⁵⁹ researched an efficient domino Mannich/aza-Michael reaction between carbamate-protected aryl aldimines and γ -malonate-substituted α,β -unsaturated methyl esters promoted using the cinchona alkaloid catalyst **12** and the bifunctional thiourea catalysts (**23**, **25**, **26**) (Scheme 35). The reaction furnished 2,5-*cis*-configured polysubstituted pyrrolidines in good to excellent yields (76~99%), enantioselectivities (75~94%) and excellent diastereoselectivities (de >95%).



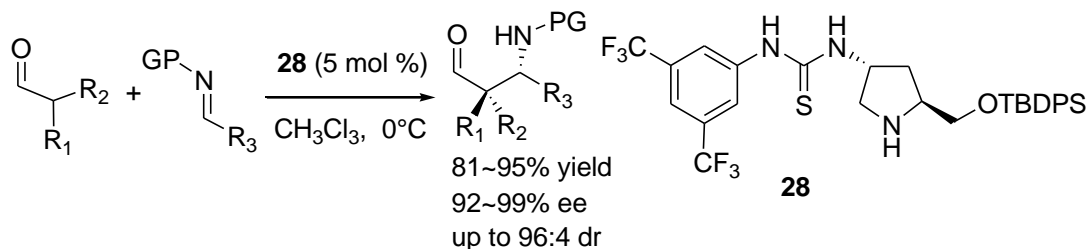
Scheme 35

Lee and Kim⁶⁰ described the catalytic enantioselective electrophilic Mannich-type reaction of α -cyano ketones with *N*-Boc-aldimines promoted by chiral bifunctional organocatalysts (**27**), and afforded the corresponding β -amino- α -cyano ketones with excellent diastereoselectivities (up to *syn/anti* = 100/0), and excellent enantioselectivities (up to 99% ee) under mild reaction conditions (Scheme 36).



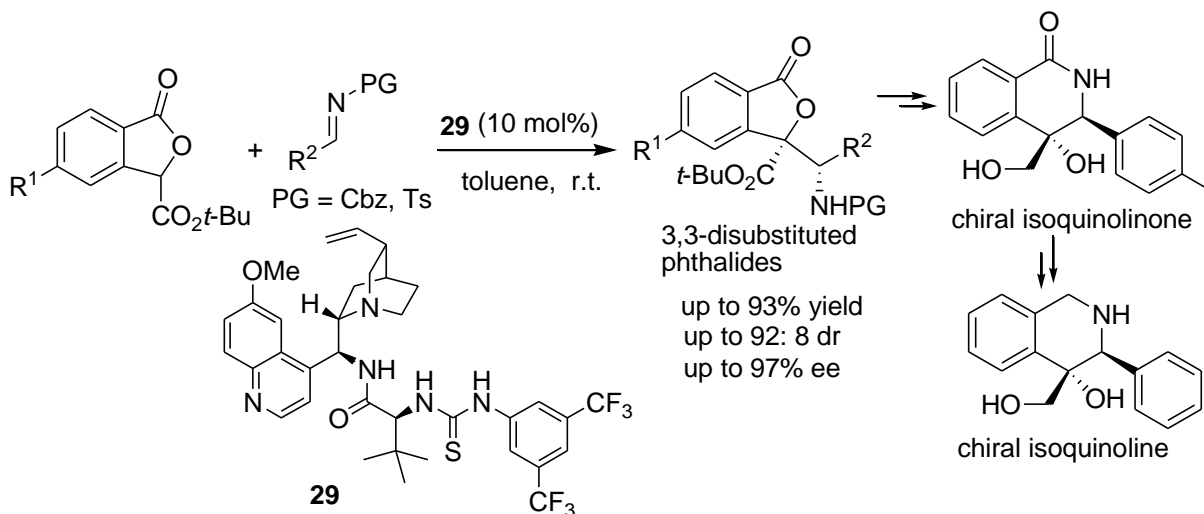
Scheme 36

In 2011, Peng *et al.*⁶¹ identified an efficient catalytic system for the direct *anti*-Mannich reaction of simple aldehydes with preformed N-Boc and N-Cbz imines (Scheme 37). Only 5 mol% catalyst (**28**) loading was needed to give the corresponding products in excellent yields (up to 95%), diastereoselectivities (up to 96:4 dr) and enantioselectivities (up to >99% ee).



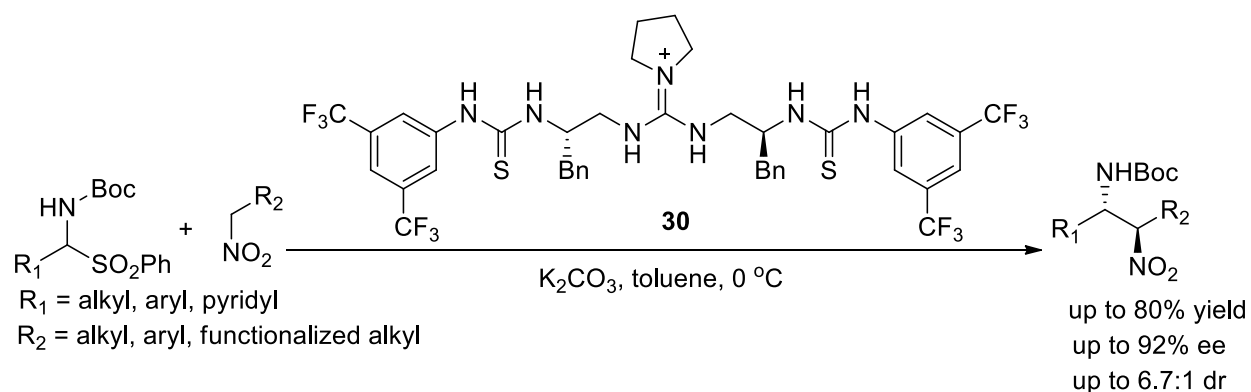
Scheme 37

Recently, Luo *et al.*⁶² reported the synthesis of 3,3-disubstituted phthalide derivatives using a quinidine-based multifunctional catalyst (**29**) in excellent yields, with good diastereo- and enantioselectivities (Scheme 38). The method led to convenient synthesis of chiral isoquinolinones and isoquinolines had also been demonstrated.



Scheme 38

In 2011, Peng *et al.*⁶³ achieved the asymmetric nitro-Mannich reactions of nitroalkanes and in situ generated N-Boc-imines using a new type of thiourea-guanidine bifunctional organocatalyst (**30**). The novel transformations exhibited good diastereoselectivities, and the adducts bearing adjacent chiral centers were generally obtained in moderate to high enantioselectivities (up to 94% ee). This reaction provided a concise and alternative route converting readily accessible and stable N-carbamate amido sulfones into optically active 1,2-diamino compounds (Scheme 39).

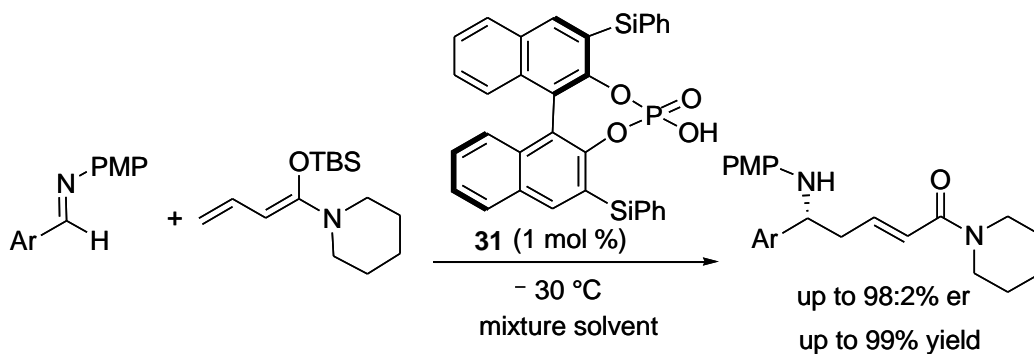


Scheme 39

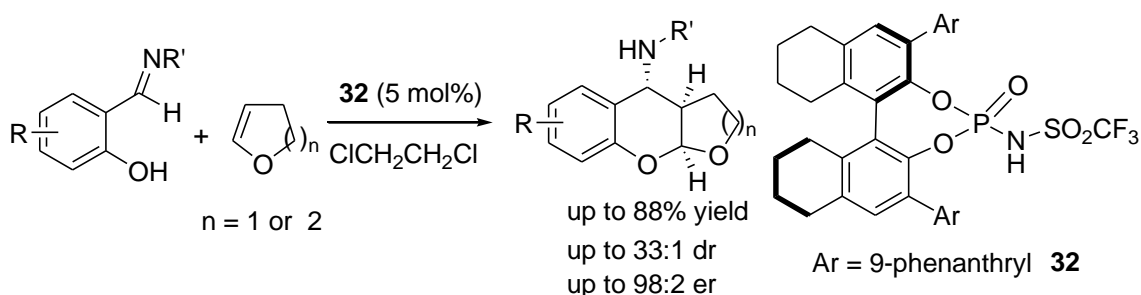
3. Catalysed by Chiral Brønsted Acids

It was an important pathway for enantioselective organocatalysed Mannich reactions proceeds *via* enantiopure Brønsted acids. As an early example, Akiyama *et al.*^{64,65} reported enantioselective Mannich-type reaction of aldimines with silyl enolates, and β -aminoesters catalysed by a series of chiral phosphate catalysts, of which phosphoric acid proved to give the best results.

Schneider and co-workers⁶⁶ developed that vinylketene silyl *N,O*-acetals readily participate vinylogous Mukaiyama-Mannich reactions catalyzed by Brønsted acid (**31**) with aromatic and heteroaromatic aldimines, and afforded δ -amino- α,β -unsaturated amides in good yields and enantioselectivities (Scheme 40). Direct three-component vinylogous Mannich reactions produced the products with almost identical yield and enantioselectivity, thus avoiding the synthesis of the imines in a separate step. The utility of the vinylogous Mannich products was demonstrated through conversion into various functional building blocks including a short synthesis of the enantiomerically highly enriched 2-phenylpiperidine. In 2010, Magnus and Lin⁶⁷ developed the enantioselective domino Mannich-ketalization reaction of *o*-hydroxy benzaldehydes with electron-rich alkenes promoted by Brønsted acid (**32**) afforded an effective and direct access to optically pure 4-aminobenzopyrans in good yields with excellent enantiomeric ratios (up to 98:2 er) (Scheme 41).

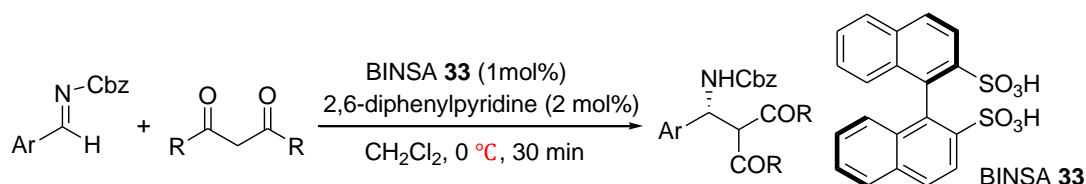


Scheme 40



Scheme 41

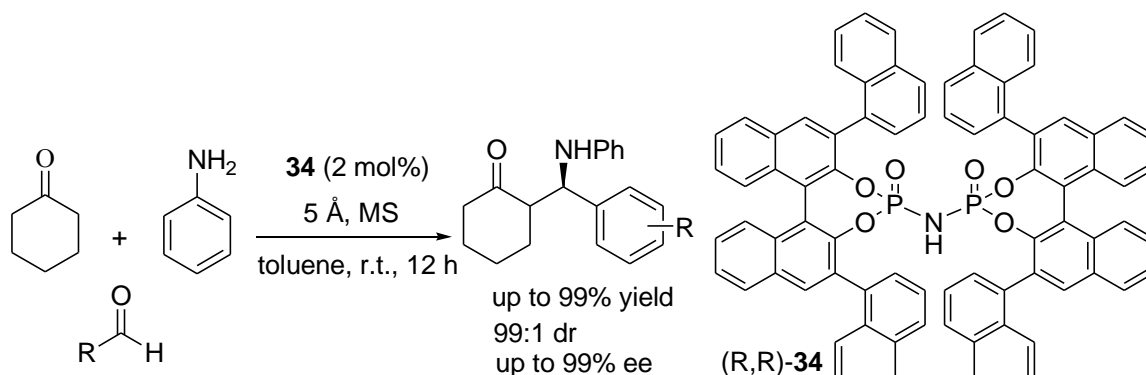
BINSA (**33**) were a highly effective chiral Brønsted acid that could be combined with an achiral Brønsted base. In 2008, Ishihara *et al.*⁶⁸ reported a direct Mannich-type reactions of a variety of 1,3-diketones and a 1,3-ketoester equivalent with arylaldimines proceeded smoothly with high enantioselectivities in the presence of 1 mol % of BINSA and 2 mol % of 2,6-biarylpyridine (Scheme 42). They think that BINSA should be a powerful chiral auxiliary like BINOL, BINAP (2,2'-bis(diphenylphosphino)-1,1'-binaphthalene), BINAM (2,2'-diamino-1,1'-binaphthalene), etc., and could trigger a new frontier in acid-base chemistry in asymmetric catalyses.



Scheme 42

In 2012, Chen *et al.*⁶⁹ reported an efficient and highly sterically hindered Brønsted acid catalyst (**34**) in asymmetric three-component Mannich reactions, and optically active *syn*- β -

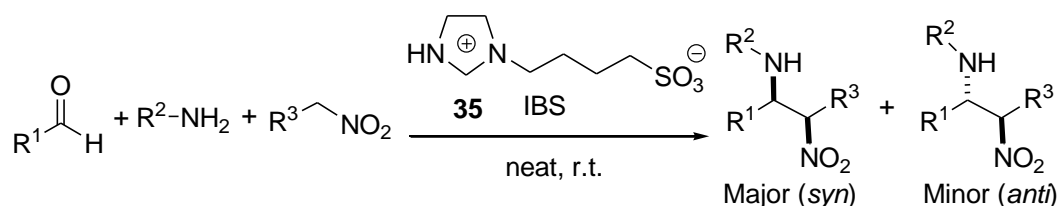
amino ketones were obtained in high yields (up to 99%) with excellent diastereoselectivity (99:1) and enantioselectivity (up to 99% ee) (Scheme 43). A gram-scale reaction was also performed to prove the synthetic application value of this reaction.



Scheme 43

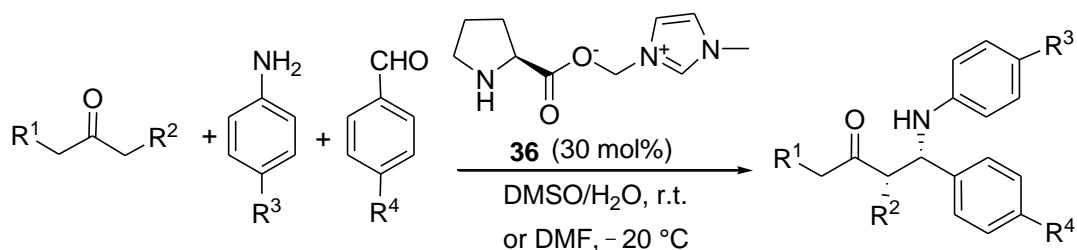
4. Catalysed by Other Chiral Organocatalytic Systems

Hajra *et al.*⁷⁰ found that imidazole-based zwitterionic-type molten salts (**35**) were new class of catalysts for the aza-Henry reaction in excellent yields and with highly selectivity (Scheme 44). Most significantly, the *syn*- β -nitroamine was obtained predominantly under the present reaction conditions.



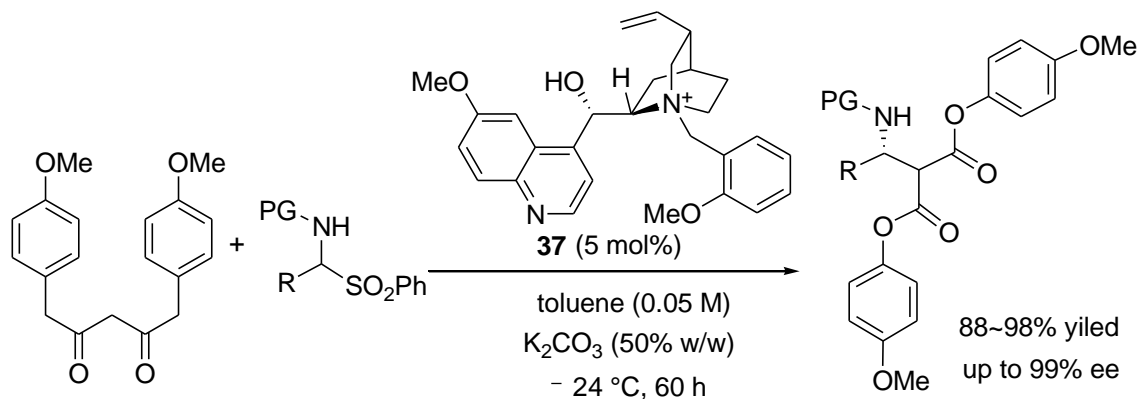
Scheme 44

In 2010, Wang *et al.*⁷¹ reported the employment of CIL [EMIm][Pro] (**36**) as a catalyst for the one-pot three-component asymmetric Mannich reaction with excellent chemo-, regio-, and enantioselectivities either under mild conditions or at a low temperature (Scheme 45). The desired products were isolated in up to 99% yield and with up to > 99 dr and > 99% ee. Additionally, this catalyst was readily prepared from rather inexpensive starting materials and the reactions could be conducted in wet solvent without an inert atmosphere. The proposed mechanism and transition state had been discussed on the basis of the stereochemistry of the corresponding Mannich products.



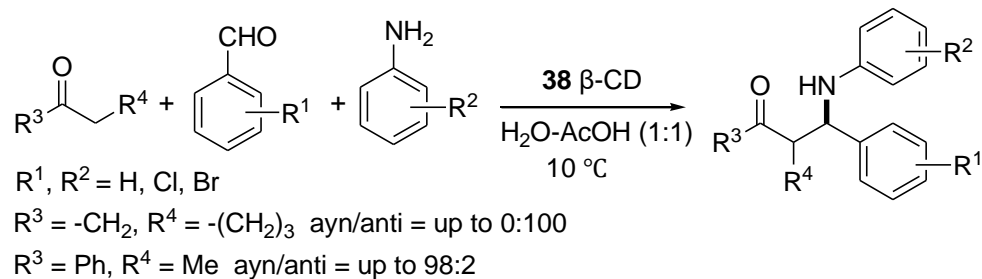
Scheme 45

Bernardi and Ricci group⁷² reported the asymmetric Mannich-type reaction of different malonates and β -ketoesters could react with *N*-tert-butoxycarbonyl (*N*-Boc) and *N*-benzyloxycarbonyl (*N*-Cbz) protected α -amido sulfones was promoted using phase-transfer catalyst (**37**) under very mild and user-friendly conditions (Scheme 46). The optimised protocol avoided the preparation and the isolation of the relatively unstable *N*-Boc and *N*-Cbz imines that were generated *in situ* from the bench-stable α -amido sulfones. The corresponding Mannich bases were generally obtained in good yields and enantioselectivities, and could be readily transformed into key compounds, such as optically active β -amino acids in one simple step.



Scheme 46

Recently, Pitchumani *et al.*⁷³ developed a highly efficient diastereoselective Mannich reaction has been carried out in water using a catalytic amount of β -cyclodextrin (**38**) as a chiral host in the presence of acetic acid to give the corresponding β -aminoketones (Mannich bases) with good yield (up to 98%) and excellent diastereomeric excess (up to >99%) (Scheme 47). This Brønsted acid-chiral cyclodextrin composite catalyzed reaction proceeds in a *syn*-selective manner with 98:2 *syn/anti* selectivity when propiophenone was used as the ketone moiety and in an *anti*-selective manner with 100:0 (*anti/syn*) selectivity when cyclohexanone was used.



Scheme 47

Discussion

This review demonstrates clearly the diversity and power of organocatalysed asymmetric Mannich reactions. The use of proline and derivatives as the catalyst affords easy access to *syn*-products in good yields with high regio-, diastereo- and enantioselectivity, but pyrrolidine derivatives as the catalyst mainly gives to the *anti*-products. Chiral cinchona alkaloids, bifunctional thioureas and their derivatives have also been successfully employed in combination with electron-poor imines and active methylene compounds with high diastereo- and enantioselectivity. Chiral Brønsted acids (mostly phosphoric acids) have been employed to include the iminium ion in a chiral ion pair, which also results in enantioselective addition onto the iminium species. Besides, other chiral organocatalytic systems have been used and exhibited the exciting results in organocatalysed asymmetric Mannich reactions in recent years. For example, the using of chiral ion and chiral phase-transfer catalyst consulting in good yields as well as highly regio-, and enantioselectivities. Transition-metal-catalysed enantioselective reactions will certainly continue to play a central role in the future; however, metal-free catalysts appear to be an emerging trend over the past few years in asymmetric Mannich reactions. In past few years, enantioselective organocatalytic asymmetric Mannich reactions have obtained rapid development, regarding the applications of this type of reactions, but efforts must be directed to research highly enantioselective and new type of chiral organocatalysts, in particular, and develop some procedures could be valuable to practical application in asymmetric Mannich reactions.

Acknowledgements

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References

1. Cordova, A. *Acc. Chem. Res.* **2004**, *37*, 102.
<http://dx.doi.org/10.1021/ar030231l>
PMid:14967057
2. Kobayashi, S.; Ishitani, H.; Kobayashi, S.; Ishitani, H. *Chem. Rev.* **1999**, *99*, 1069.
<http://dx.doi.org/10.1021/cr980414z>
PMid:11749440
3. Arend, M.; Westermann, B.; Risch, N. *Angew. Chem. Int. Ed.* **1998**, *37*, 1044.
[http://dx.doi.org/10.1002/\(SICI\)1521-3773\(19980504\)37:8<1044::AID-ANIE1044>3.0.CO;2-E](http://dx.doi.org/10.1002/(SICI)1521-3773(19980504)37:8<1044::AID-ANIE1044>3.0.CO;2-E)
4. Arrayás, R. G. and Carretero, J. C. *Chem. Soc. Rev.* **2009**, *38*, 1940.
<http://dx.doi.org/10.1039/b820303b>
PMid:19551174
5. Verkade, J. M. M.; van Hemert, L. J. C.; Quaedflieg, P. J. L. M.; Rutjes, F. P. J. T. *Chem. Soc. Rev.* **2008**, *37*, 29.
<http://dx.doi.org/10.1039/b713885g>
PMid:18197331
6. Pellissier, H. *Tetrahedron* **2007**, *63*, 9267.
<http://dx.doi.org/10.1016/j.tet.2007.06.024>
7. Graaff, C. D.; Ruijter, E. and Orru, R. V. A. *Chem. Soc. Rev.* **2012**, *41*, 3969.
<http://dx.doi.org/10.1039/c2cs15361k>
PMid:22546840
8. Ma, J. A. *Angew. Chem. Int. Ed.* **2003**, *42*, 4290.
<http://dx.doi.org/10.1002/anie.200301600>
PMid:14502699
9. Liu, M.; Sibi, M. P. *Tetrahedron* **2002**, *58*, 7991.
[http://dx.doi.org/10.1016/S0040-4020\(02\)00991-2](http://dx.doi.org/10.1016/S0040-4020(02)00991-2)
10. Müller, R.; Goesmann, H.; Waldmann, H. *Angew. Chem. Int. Ed.* **1999**, *38*, 184.
[http://dx.doi.org/10.1002/\(SICI\)1521-3773\(19990115\)38:1/2<184::AID-ANIE184>3.0.CO;2-E](http://dx.doi.org/10.1002/(SICI)1521-3773(19990115)38:1/2<184::AID-ANIE184>3.0.CO;2-E)
11. Shibasaki, M.; Matsunaga, S. *J. Organomet. Chem.* **2006**, *691*, 2089.
<http://dx.doi.org/10.1016/j.jorganchem.2005.10.025>
12. Friestad, G. K.; Mathies, A. K. *Tetrahedron* **2007**, *63*, 2541.
<http://dx.doi.org/10.1016/j.tet.2006.11.076>
13. Charette, A. B.; Boezio, A. A.; Cote, A.; Moreau, E.; Pytkowicz, J.; Desrosiers, J. N.; Legault, C. *Pure Appl. Chem.* **2005**, *77*, 1259.
<http://dx.doi.org/10.1351/pac200577071259>
14. Chen, Z. H.; Yakura, K.; Matsunaga, S. and Shibasaki M. *Org. Lett.* **2008**, *10*, 3239.
<http://dx.doi.org/10.1021/ol800965t>
PMid:18610973

15. Wang, J.; Shi, T.; Deng, G. H.; Jiang, H. L. and Liu, H. *J. Org. Chem.* **2008**, *73*, 8563.
<http://dx.doi.org/10.1021/jo8019169>
PMid:18844412
16. Shepherd, N. E.; Tanabe, H.; Xu, Y. J.; Matsunaga, S. and Shibasaki, M. *J. Am. Chem. Soc.* **2010**, *132*, 3666.
<http://dx.doi.org/10.1021/ja1002636>
PMid:20192184
17. Kang, Y. K.; Kim, D. Y. *Tetrahedron Lett.* **2011**, *52*, 2356.
<http://dx.doi.org/10.1016/j.tetlet.2011.02.087>
18. List, B.; Pojarliev, P.; Biller, W. T.; Martin, H. J. *J. Am. Chem. Soc.* **2002**, *124*, 827.
<http://dx.doi.org/10.1021/ja0174231>
19. Chandler, C.; Galzerano, P.; Michrowska, A. and List, B. *Angew. Chem. Int. Ed.* **2009**, *48*, 1978.
<http://dx.doi.org/10.1002/anie.200806049>
PMid:19199308
20. Mukherjee, S.; Yang, J. W.; Hoffmann, S.; List, B. *Chem. Rev.* **2007**, *107*, 5471.
<http://dx.doi.org/10.1021/cr0684016>
PMid:18072803
21. List, B.; Pojarliev, P.; Biller, W. T. and Martin, H. J. *J. Am. Chem. Soc.* **2002**, *124*, 827.
<http://dx.doi.org/10.1021/ja0174231>
22. Han, R. -G.; Wang, Y.; Li, Y. -Y. and Xu P. F. *Adv. Synth. Cat.* **2008**, *350*, 1474
<http://dx.doi.org/10.1002/adsc.200800253>
23. Hahn, B. T.; Fröhlich, R.; Harms, K. and Glorius, F. *Angew. Chem. Int. Ed.* **2008**, *47*, 9985.
<http://dx.doi.org/10.1002/anie.200803515>
PMid:19006133
24. Deiana, L.; Zhao, G. -L.; Dziedzic, P.; Rios, R.; Vesely, J.; Ekstr, J.; Córdova, A. *Tetrahedron Lett.* **2010**, *51*, 234.
<http://dx.doi.org/10.1016/j.tetlet.2009.10.130>
25. Li, L. Q.; Han, M. Y.; Xiao, M. X.; Xie, Z. X. *Synlett* **2011**, 1727.
26. Yang, H. Carter, R. G. *J. Org. Chem.* **2009**, *74*, 5151.
<http://dx.doi.org/10.1021/jo9009062>
PMid:19743541
27. Veverkova, E.; Strasserova, J.; Sebesta, R.; Toma, S. *Tetrahedron: Asymmetry* **2010**, *21*, 58.
<http://dx.doi.org/10.1016/j.tetasy.2009.12.013>
28. Lu, M.; Lu, Y. P.; Tan, P. K. A.; Lau, Q. Y.; Zhong, G. F. *Synlett* **2011**, 477.
29. An, Y. -J.; Wang, C. -C.; Liu, Z. -P. and Tao, J. -C. *Helv. Chim. Acta* **2012**, *95*, 43.
<http://dx.doi.org/10.1002/hlca.201100265>
30. Franzén, J.; Marigo, M.; Fielenbach, D.; Wabnitz, T. C.; Kjærsgaard, A.; Jørgensen, K. A. *J. Am. Chem. Soc.* **2005**, *127*, 18296.

<http://dx.doi.org/10.1021/ja056120u>

PMid:16366584

31. Gianelli, C.; Sambri, L.; Carlone, A.; Bartoli, G. and Melchiorre, P. *Angew. Chem. Int. Ed.* **2008**, *47*, 8700.

<http://dx.doi.org/10.1002/anie.200803819>

PMid:18846535

32. Fustero, S.; Mojarrad, F.; Carrión, M. D. P.; Sanz-Cervera, J. F. and Ace-a, J. L. *Eur. J. Org. Chem.* **2009**, 5208.

<http://dx.doi.org/10.1002/ejoc.200900509>

33. Galzerano, P.; Agostino, D.; Bencivenni, G.; Sambri, L.; Bartoli, G.; Melchiorre P. *Chem. Eur. J.* **2010**, *16*, 6069.

PMid:20397160

34. Urushima, T.; Ishikawa, H. and Hayashi, Y. *Chem. Eur. J.* **2011**, *17*, 8273.

<http://dx.doi.org/10.1002/chem.201101077>

PMid:21656864

35. Hayashi, Y.; Urushima, T.; Sakamoto, D.; Torii, K.; Ishikawa, Hayato. *Chem. Eur. J.* **2011**, *17*, 11715.

<http://dx.doi.org/10.1002/chem.201101668>

PMid:21887838

36. Gómez-Bengo, E.; Maestro, M.; Mielgo, A.; Otazo, I.; Palomo, C. and Velill, I. *Chem. Eur. J.* **2010**, *16*, 5333.

37. Gómez-Bengo, E.; Jiménez, J.; Lapuerta, I.; Mielgo, A.; Oiarbide, M.; Otazo, I.; Velilla, I.; Vera, S. and Palomo, C. *Chem. Sci.* **2012**, *3*, 2949.

<http://dx.doi.org/10.1039/c2sc20590d>

38. Pouliquen, M.; Blanchet, J.; Lasne, M. -C. and Rouden, J. *Org. Lett.* **2008**, *10*, 1029.

<http://dx.doi.org/10.1021/ol8000975>

PMid:18225912

39. Zhang, H. L.; Mitsumori, S.; Utsumi, N.; Imai, M.; Garcia-Delgado, N.; Mifsud, M.; Albertshofer, K.; Cheong, P. H. -Y.; Houk, K. N.; Tanaka, F. J. and Barbas, C. F. C. *J. Am. Chem. Soc.* **2008**, *130*, 875.

<http://dx.doi.org/10.1021/ja074907+>

PMid:18163619

40. Hayashi, Y.; Urushima, T.; Aratake, S.; Okano, T. and Obi, K. *Org. Lett.* **2008**, *10*, 21.

<http://dx.doi.org/10.1021/ol702489k>

PMid:18052181

41. Lou, S.; Dai, P. and Schaus, S. E. *J. Org. Chem.* **2007**, *72*, 9998.

<http://dx.doi.org/10.1021/jo701777g>

PMid:18047372

42. Liu, X. D.; Deng, L. J.; Song, H. J.; Jia, H. Z. and Wang, R. *Org. Lett.* **2011**, *13*, 1494.
<http://dx.doi.org/10.1021/ol200185h>
PMid:21344918
43. Mizuta, S.; Shibata, N.; Goto, Y.; Furukawa, T.; Nakamura, S. and Toru, T. *J. Am. Chem. Soc.* **2007**, *129*, 6394.
<http://dx.doi.org/10.1021/ja071509y>
PMid:17461589
44. Kano, T.; Yamaguchi, Y. and Maruoka, K. *Chem. Eur. J.* **2009**, *15*, 6678.
<http://dx.doi.org/10.1002/chem.200900267>
PMid:19479934
45. Kano, T.; Yamaguchi, Y. and Maruoka, K. *Angew. Chem. Int. Ed.* **2009**, *48*, 1838.
<http://dx.doi.org/10.1002/anie.200805628>
PMid:19173354
46. Kano, T.; Sakamoto, R.; Akakura, M. and Maruoka, K. *J. Am. Chem. Soc.* **2012**, *134*, 7516.
<http://dx.doi.org/10.1021/ja301120z>
PMid:22486203
47. Chen, X. H.; Dong, S. X.; Qiao, Z.; Zhu, Y.; Xie, M. S.; Lin, L. L.; Liu, X. H. and Feng, X. M. *Chem. Eur. J.* **2011**, *17*, 2583.
<http://dx.doi.org/10.1002/chem.201002571>
PMid:21271617
48. Pan, Y. H.; Zhao, Y. J.; Ma, T.; Yang, Y. Y.; Liu, H. J.; Jiang, Z. Y. and Tan, C. -H. *Chem. Eur. J.* **2010**, *16*, 779.
PMid:19943289
49. Cheng, L. L.; Han, X.; Huang, H. M.; Wong, W. and Lu, Y. X. *Chem. Commun.* **2007**, 4143.
<http://dx.doi.org/10.1039/b706793c>
PMid:17925956
50. Moteki, S. A.; Han, J. W.; Arimitsu, S.; Akakura, M.; Nakayama, K.; Maruoka, K. *Angew. Chem. Int. Ed.* **2012**, *51*, 1187.
<http://dx.doi.org/10.1002/anie.201107239>
PMid:22190385
51. Wenzel, A. G. and Jacobsen, E. N. *J. Am. Chem. Soc.* **2002**, *124*, 12964.
<http://dx.doi.org/10.1021/ja028353g>
52. Yoon, T. P. and Jacobsen, E. N. *Angew. Chem. Int. Ed.* **2005**, *44*, 466.
<http://dx.doi.org/10.1002/anie.200461814>
PMid:15624148
53. Wang, C. G.; Zhou, Z. H. and Tang, C. C. *Org. Lett.* **2008**, *10*, 1707.
<http://dx.doi.org/10.1021/ol8003035>
PMid:18396893

54. Tian, X.; Jiang, K.; Peng, J.; Du, W. and Chen, Y. -C. *Org. Lett.* **2008**, *10*, 3583.
<http://dx.doi.org/10.1021/ol801351j>
PMid:18642826
55. Wang, C. -J.; Dong, X. -Q.; Zhang, Z. -H.; Xue, Z. -Y. and Teng, H. -L. *J. Am. Chem. Soc.* **2008**, *130*, 8606.
<http://dx.doi.org/10.1021/ja803538x>
PMid:18549213
56. Han, X.; Kwiatkowski, J.; Xue, F.; Huang, K. -W. and Lu, Y. X. *Angew. Chem. Int. Ed.* **2009**, *48*, 7604.
57. Kohler, M. C.; Yost, J. M.; Garnsey, M. R. and Coltart, D. M. *Org. Lett.* **2010**, *12*, 3376.
<http://dx.doi.org/10.1021/ol101152b>
PMid:20608684
58. Lee, L. J. and Young, K. D. *Synthesis* 2010, 1860.
59. Enders, D.; Göddertz, D. P.; Bece-e, C. and Raabe, G. *Adv. Synth. Catal.* **2010**, *352*, 2863.
<http://dx.doi.org/10.1002/adsc.201000658>
60. Lee, J. H. and Kim, D. Y. *Adv. Synth. Catal.* **2009**, *351*, 1779.
<http://dx.doi.org/10.1002/adsc.200900268>
61. Chuan, Y. -M.; Chen, G. -H.; Gao, J. -Z.; Zhang, H. and Peng, Y. -G. *Chem. Commun.* **2011**, *47*, 3260.
<http://dx.doi.org/10.1039/c0cc05249c>
PMid:21283837
62. Luo, J.; Wang, H. F.; Zhong, F. R.; Kwiatkowski, J.; Xu, L.W. and Lu, Y. X. *Chem. Commun.* **2012**, *48*, 4707.
<http://dx.doi.org/10.1039/c2cc31439h>
PMid:22488217
63. Huang, W., Peng, C.; Guo, L.; Hu, R.; Han, B. *Synlett* **2011**, *20*, 2981.
64. Takahiko, A.; Itoh, J.; Yokota, K. and Fuchibe, K. *Angew. Chem. Int. Ed.* **2004**, *43*, 1566.
<http://dx.doi.org/10.1002/anie.200353240>
PMid:15022235
65. Yamanaka, M.; Itoh, J.; Fuchibe, K.; Akiyama, T. *J. Am. Chem. Soc.* **2007**, *129*, 6756.
<http://dx.doi.org/10.1021/ja0684803>
PMid:17477527
66. Giera, D.S.; Sickert, M. and Schneider, C. *Org. Lett.* **2008**, *10*, 4259.
<http://dx.doi.org/10.1021/ol8017374>
PMid:18754625
67. Rueping, M. and Lin, M. -Y. *Chem. Eur. J.* **2010**, *16*, 4169.
<http://dx.doi.org/10.1002/chem.201000203>
PMid:20309979
68. Hatano, M.; Maki, T.; Moriyama, K.; Arinobe, M.; Ishihara, K. *J. Am. Chem. Soc.* **2008**, *130*, 16858.

<http://dx.doi.org/10.1021/ja806875c>

PMid:19053478

69. Chen, Y. -Y.; Jiang, Y. -J.; Fan, Y. -S.; Sha, D.; Wang, Q. F.; Zhang, G. L.; Zheng, L. Y.; Zhang, S. Q. *Tetrahedron: Asymmetry* **2012**, *23*, 904.

<http://dx.doi.org/10.1016/j.tetasy.2012.06.008>

70. Kundu, D.; Debnath, R. K.; Majee, A.; Hajra, A. *Tetrahedron Lett.* **2009**, *50*, 6998.

<http://dx.doi.org/10.1016/j.tetlet.2009.09.153>

71. Zheng, X.; Qian, Y.-B. and Wang, Y. M. *Eur. J. Org. Chem.* **2010**, 515.

<http://dx.doi.org/10.1002/ejoc.200901088>

72. Marianacci, O.; Micheletti, G.; Bernardi, L.; Fini, F.; Fochi, M.; Pettersen, D.; Sgarzani, V. and Ricci, A. *Chem. Eur. J.* **2007**, *13*, 8338.

<http://dx.doi.org/10.1002/chem.200700908>

PMid:17705329

73. Sukumari, S.; Azath, I. A.; Pitchumani, K. *Synlett* **2012**, *23*, 2328.

<http://dx.doi.org/10.1055/s-0032-1317156>

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