

Short and efficient method for the preparation of furo[3,2-*f*] quinoline

Mercedesz Törincki, Pál Kolonits, Endre Pálosi, Melinda Fekete, and Lajos Novák*

Department of Organic Chemistry and Technology, Budapest University of Technology and Economics, Research Group for Alkaloid Chemistry, Hungarian Academy of Sciences, Gellért tér 4, 1111 Budapest, Hungary
E-mail: lnovak@mail.bme.hu

Dedicated to Professor Csaba Szántay on his 80th birthday

Abstract

Furo[3,2-*f*]quinolines **5** were prepared by reacting the sodium salt of quinolin-6-ol and allyl bromides **2**. The allyl aryl ethers **3** formed were then thermally rearranged under standard or microwave conditions. Acid-catalyzed cyclization of the products **4** afforded the title compounds.

Keywords: Furo[3,2-*f*]quinoline, allyl aryl ethers, Claisen rearrangement, ring closure, sonochemical conditions, pyrano[3,2-*f*]quinoline

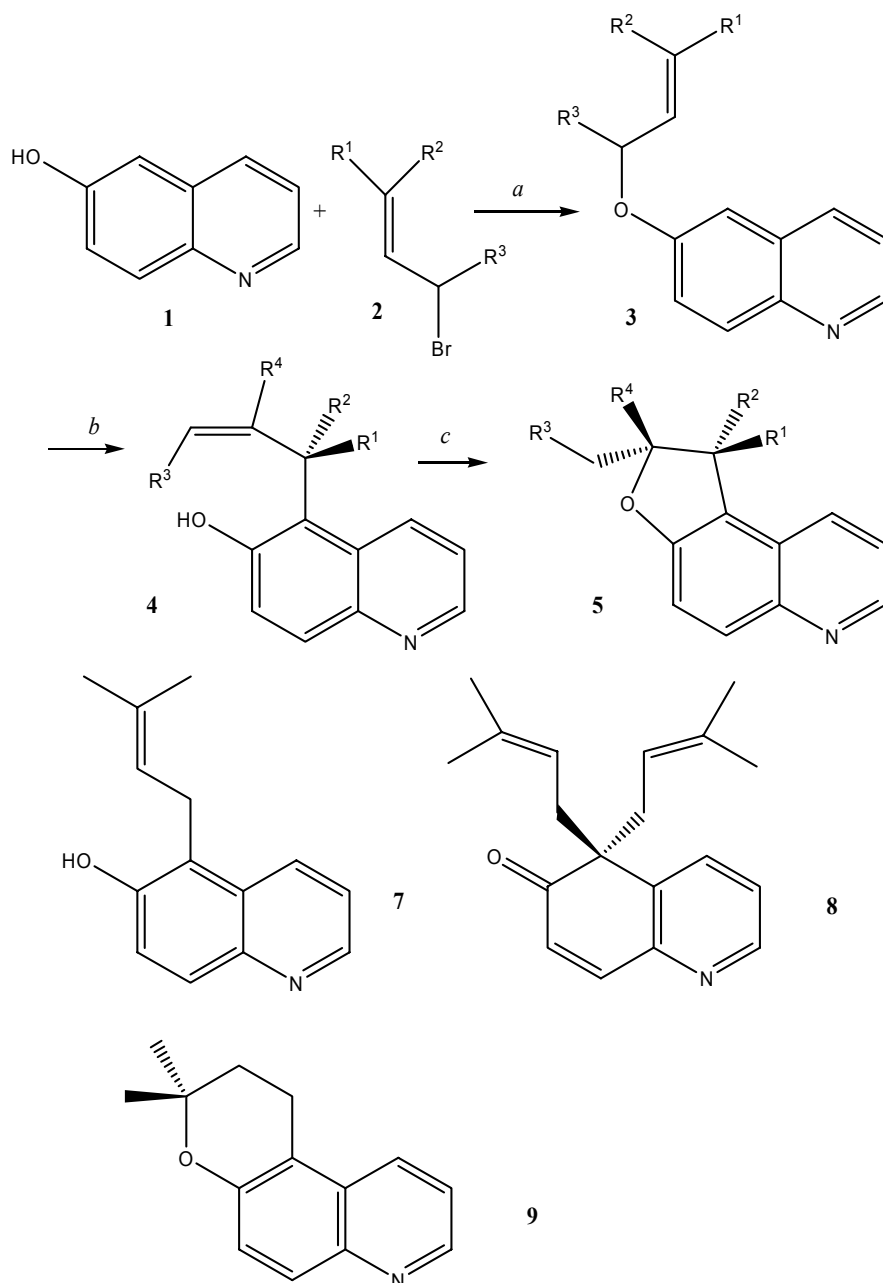
Introduction

Furo[3,2-*f*]quinolines have not received much attention. Only three papers have so far been published dealing with the the preparation of this ring system. B. B. Dey and T. R. Seshadri prepared the furoquinoline skeleton by the thermal decomposition of a quinolinopyrone derivative.¹ R. Royer *et al.*² described the synthesis of substituted furo[3,2-*f*]quinolines by the rearrangement of allylaryl ether followed by cyclization. Later on, M. Natsume *et al.*³ isolated the furoquinoline as by-product of their duocarmycin SA's synthesis.

Recently we have developed an effective method for the preparation of furo[2,3-*f*]isoquinolines by aromatic Claisen rearrangement and subsequent cyclization.⁴ Following these synthetic efforts toward the preparation of novel heterocyclic compounds which might be useful intermediates for the development of molecules of pharmaceutical or biological interest, we planned the elaboration of new synthesis generally applicable for the preparation of furo[3,2-*f*]quinolines. We present here a method for the preparation of furoquinolines **5a-c** and their cycloalkano analogues **5d,e**.

Results and Discussion

Ethers of quinolin-6-ol **3a-d** were synthesized by reaction of the sodium salt of quinolinol (**1**) with the appropriate alkyl bromide **2a-d** in good to acceptable yields (Scheme 1, Table 1, entries 1-4).



Scheme 1. Reagent and conditions: (a) NaH, DME, r.t.; (b) microwave oven, 175°C; or chlorobenzene, reflux; (c) H₂SO₄, 100°C.

The allyl ether **3a** was subjected to thermal [3,3] rearrangement in a microwave oven to afford **4a**.⁵ Acid-catalyzed intramolecular cyclization of the latter afforded the known furo[3,2]quinoline **5a**² (entries 5 and 10).

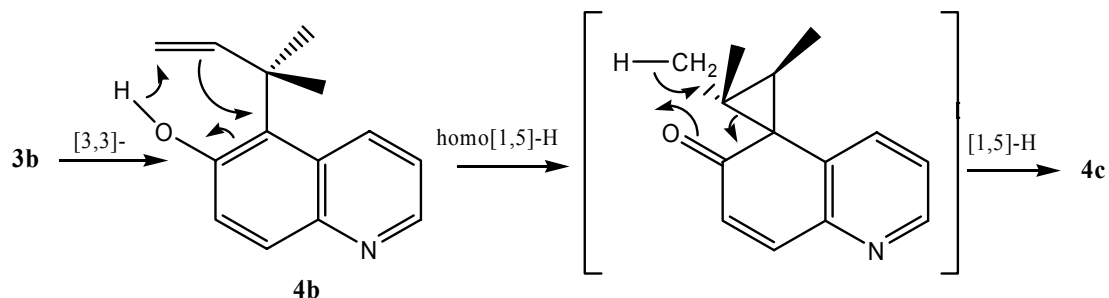
Starting with compound **3b**, the microwave assisted rearrangement gave two products **4b** and **4c** (entry 6). Compound **4b** was formed in the normal Claisen rearrangement.⁶ The unexpected product took its origin from consecutive rearrangement reactions (Scheme 2). Namely, [3,3]-sigmatropic rearrangement of ether **3b** yielded intermediate **4c** which then underwent a homo[1,5]-H shift to afford compound **6**. Further [1,5]-H migration on this intermediate led to the formation of **4c**.⁷

Table 1. Compounds prepared by the rearrangement of **3a-d** and subsequent cyclization

Entry	Substrate	Product	R ¹	R ²	R ³	R ⁴	Temp (°C)	Time (h)	Yield ^a (%)
1	2a	3a	H	H	H	-	r.t.	24	69
2	2b	3b	Me	Me	H	-	r.t.	24	83
3	2c	3c	H	(CH ₂) ₃		-	r.t.	24	46
4	2d	3d	H	(CH ₂) ₄		-	r.t.	24	38
5	3a	4a	H	H	H	H	175 ^b	8	49.5
6	3b	4b	Me	Me	H	H	175 ^b	10	44
		4c	Me	H	H	Me			12
7	2b	3b	Me	Me	H	H	r.t.	16	16.8
		7	H	H	Me	Me			26.2
		8	-	-	-	-			32.4
8	3c	4e	H	(CH ₂) ₃		H	132 ^c	100	70
9	3d	4f	H	(CH ₂) ₄		H	132 ^c	100	38
10	4a	5a	H	H	H	Me	100	1.5	35
11	4b	5b	Me	Me	H	H	100	1	42.2
12	4c	5c	Me	H	Me	H	100	1	40
13	7	9	-	-	-	-	100	1	54
14	4d	5d	H	(CH ₂) ₃		H	100	1.5	75
15	4e	5e	H	(CH ₂) ₄		H	100	1.5	74

^aIsolated and unoptimized yields; ^bin microwave oven; ^cin chlorobenzene.

Having the above result, we tried to synthesize compounds **3b** and **4b** with a phase transfer catalyzed reaction. Interestingly, the quaternary ammonium salt catalyzed reaction of quinolin-6-ol **1** with prenyl bromide (**2b**) afforded three compounds (entry 7). Besides the expected ether **3b**, compounds **7** and **8** were isolated as major products. The formation of these new compounds is probably the result of direct aromatic electrophilic substitutions.



Scheme 2

Acid-catalyzed intramolecular cyclization of **4b** gave furoquinoline **5b** in moderate yield. Likewise, compound **4c** yielded **5c** by acid (H_2SO_4 or HClO_4) promoted ring closure (entries 11 and 12).

Treatment of the dimethylallyl derivative **7** with sulfuric acid provided pyrano [3,2-*f*]quinoline derivative **9** (entry 13).

Due to their sensitivity at the temperature of the microwave oven, compounds **3c** and **3d** were submitted to thermal rearrangement in boiling chlorobenzene. This protocol was especially effective for the preparation of **4d**, but a moderate result was obtained for the synthesis of **4e** (entries 8 and 9). Acid-catalyzed cyclization of **4d** and **4e** yielded **5d** and **5e** (respectively), as an approximately 4:1 mixture of *trans* and *cis* stereoisomers (entries 14 and 15). Small amounts of these mixtures were separated by preparative HPLC and the stereochemistry of isomers was established by ^1H and ^{13}C NMR studies. For example, in the *cis*-fused isomer of **5d** we saw a NOE interaction between the 7a proton and 11a proton (4.82 and 3.45, respectively). This interaction was absent in the spectrum of the corresponding *trans*-isomer. For **5e**, on the bases of γ -effect in chemical shift between C_8 and C_{12} in the spectrum the *cis*-isomer was identified. This correlation is in concord with our earlier findings in the series of furo[2,3-*f*]isoquinolines.⁴

In summary, we have developed a general and efficient method for the preparation of [3,2-*f*]quinoline derivatives **5a-e** from quinolin-6-ol (**1**) and allyl bromides **2a-d**. This process involves the thermal rearrangement of ethers **3a-d**, followed by acid-catalyzed intramolecular cyclization of the products **4a-e**. This synthesis using the readily available starting compounds seems to provide a powerful methodology for the construction of furo-condensed quinoline derivatives.

Experimental Section

General Procedures. Solvents were used as received from commercial vendors and no further attempts were made to purify or dry them. Mps were determined on a Büchi apparatus and are uncorrected. ^1H and ^{13}C NMR spectra were obtained on a Bruker DRX-500 spectrometer. All NMR spectra are reported in ppm relative to TMS. IR spectra were measured on a Specord 2000

spectrometer. Merck precoated silica gel 60 F₂₅₄ plates were used for TLC and Kieselgel 60 for column chromatography. Solvent were mixed on a v/v basis. HPLC chromatographic analyses and separation were performed with a Waters 600 equipped with a photodiode array detector 990. Stationary phase for compound **5e** was SupelcosilTM SPLC-18-DB, 250x10 mm, eluents MeOH/H₂O/H₃PO₄ 4:6:0.05. For compound **5d** stationary phase was Waters Symmetry C₁₈, 150x3.9 mm. For the separation of isomers of **5f** we worked on SupelcosilTM PLC-18 column, 250x21.2 mm, eluents MeOH/3% tartaric acid solution/H₃PO₄ 4:6:0.02.

Microwave accelerated reactions were conducted in an CEM Focused MicrowaveTM Synthesis System (CEM Corporation, Matthews, NC, USA).

3-Bromocyclohex-1-ene (**2c**)⁸ and 3-bromocyclohept-1-ene (**2d**)⁹ were prepared by literature procedures.

Preparation of ethers 3. General procedure

To a cold stirred suspension of NaH (24 mmol, 63.7% in mineral oil) in DME (10 mL) a solution of quinolin-6-ol (**1**: 2.23 g, 15.4 mmol) in DME (200 mL) was added dropwise and the resultant mixture was stirred at 0 °C for 1.5 h. To this mixture the appropriate bromide (**2**: 23 mmol) was then added and stirring was continued at r.t. for 24 h. The reaction mixture was quenched with sat. aq. NaCl (400 mL) and extracted with EtOAc (3 x 100 mL). The combined organic layers were washed with 1N NaOH solution and H₂O and then dried over MgSO₄. The solvent was removed under reduced pressure and the residue was purified by column chromatography on silica gel (CH₂Cl₂ – acetone 4:1).

6-Allyloxyquinoline (3a). Yield: 69%; brown oil; *R_f* = 0.77 (CH₂Cl₂ – acetone 2:1). ¹H NMR (CDCl₃): δ= 4.66 (d, *J*= 5.4 Hz, 2H, CH₂), 5.34 (dd, *J*= 10.6 and 1.1 Hz, 1H, =CH), 5.47 (dd, *J*= 17.2 and 1.2 Hz, 1H, =CH), 6.12 (m, 1H, =CH), 7.08 (d, *J*= 2.7 Hz, 1H, C₅-H), 7.34 (dd, *J*= 8.2 and 1.2 Hz, 1H, C₃-H), 7.40 (dd, *J*= 9.1 and 2.7 Hz, 1H, C₇-H), 7.99 (d, *J*= 9.1 Hz, 1H, C₈-H), 8.02 (m, 1H, C₄-H), 8.77 (dd, *J*= 4.2 and 1.6 Hz, 1H, C₂-H).

¹³C NMR (CDCl₃): δ= 69.07 (C-2'), 106.38 (C-5), 118.04 (C-4'), 121.37 (C-3), 122.51 (C-6), 129.25 (C-4a), 130.94 (C-7), 132.84 (C-3'), 134.81 (C-4), 144.48 (C-7a), 148.06 (C-2), 156.67 (C-6).

6-(3-Methylbut-2-en-1-yloxy)quinoline (3b). Yield: 83%; brown oil; *R_f* = 0.56 (CHCl₃ – acetone 10:1). ¹H NMR (CDCl₃): δ= 1.78 (s, 3H, CH₃), 1.82 (s, 3H, CH₃), 4.62 (d, *J*= 6.6 Hz, 2H, CH₂), 5.56 (t, *J*= 6.6 Hz, 1H, C₂'-H), 7.08 (d, *J*= 2.5 Hz, 1H, C₅-H), 7.34 (dd, *J*= 8.2 and 4.2 Hz, 1H, C₃-H), 7.39 (dd, *J*= 9.2 and 2.5 Hz, 1H, C₇-H), 8.01 (d, *J*= 9.6 Hz, 1H, C₈-H), 8.03 (d, *J*= 9.8 Hz, 1H, C₄-H). ¹³C NMR (CDCl₃): δ= 18.21 (CH₃), 25.69 (CH₃), 64.90 (C-1'), 105.74 (C-5), 118.86 (C-2'), 120.93 (C-3), 122.41 (C-7), 128.93 (C-4a), 130.28 (C-8), 134.52 (C-4), 138.29 (C-3'), 143.75 (C-8a), 147.25 (C-2), 156.54 (C-6). Anal. Calcd for C₁₄H₁₅NO: C, 78.84; H, 7.09; N, 6.57. Found: C, 78.57; H, 7.12; N, 6.32.

6-(Cyclohex-2-en-1-yloxy)quinoline (3c). Yield: 46%; yellow solid mp 65-69 °C; *R_f* = 0.28 (hexane – EtOAc 4:1). ¹H NMR (CDCl₃): δ= 1.78 (m, 1H, C₅'-H), 1.87 (m, 1H, C₅'-H), 1.95 (m, 1H, C₆'-H), 2.03 (m, 1H, C₆'-H), 2.06 (m, 1H, C₄'-H), 2.17 (m, 1H, C₄'-H), 4.95 (m, 1H, C₁'-H),

5.94 (m, 1H, C₂-H), 6.02 (m, 1H, C₃-H), 7.12 (d, *J* = 2.6 Hz, 1H, C₅-H), 7.33 (dd, *J* = 8.3 and 4.3 Hz, 1H, C₃-H), 7.38 (dd, *J* = 9.2 and 2.6 Hz, 1H, C₇-H), 8.01 (d, *J* = 9.2 Hz, 1H, C₈-H), 8.03 (d, *J* = 9.2 Hz, 1H, C₄-H), 8.75 (dd, *J* = 4.3 and 1 Hz, 1H, C₂-H). ¹³C NMR (CDCl₃): δ = 19.11 (C-5'), 25.19 (C-4'), 28.25 (C-6'), 71.13 (C-1'), 107.26 (C-5), 121.08 (C-3), 123.14 (C-7), 125.55 (C-2'), 129.11 (C-4a), 130.59 (C-8), 132.39 (C-3'), 134.60 (C-4), 143.83 (C-8a), 147.45 (C-2'), 155.61 (C-6). Anal. Calcd for C₁₅H₁₅NO: C, 79.97; H, 6.71; N, 6.22. Found: C, 79.71; H, 6.53; N, 6.34.

6-(Cyclohept-2-en-1-yloxy)quinoline (3d). Yield: 38%; brown oil; *R_f* = 0.71 (CH₂Cl₂ – acetone 2:1). ¹H NMR (CDCl₃): δ = 1.45 (m, 1H, C₅-H), 1.72 (m, 1H, C₆-H), 1.76 (m, 1H, C₅-H), 1.83 (m, 1H, C₇-H), 2.07 (m, 1H, C₆-H), 2.10 (m, 1H, C₇-H), 2.20 (m, 1H, C₄-H), 2.27 (m, 1H, C₄-H), 5.02 (m, 1H, C₁-H), 5.84 (m, 1H, C₂-H), 5.92 (m, 1H, C₃-H), 7.01 (d, *J* = 2.5 Hz, 1H, C₅-H), 7.32 (dd, *J* = 8.3 and 4.2 Hz, 1H, C₃-H), 7.37 (dd, *J* = 9.2 and 2.5 Hz, 1H, C₇-H), 8.00 (d, *J* = 9.0 Hz, 1H, C₈-H), 8.02 (d, *J* = 7.0 Hz, 1H, C₄-H), 8.75 (d, *J* = 3.8 Hz, 1H, C₂-H). ¹³C NMR (CDCl₃): δ = 26.48 (C-5'), 27.49 (C-6'), 28.58 (C-4'), 77.65 (C-1'), 107.44 (C-5), 121.25 (C-3), 123.18 (C-7), 129.30 (C-4a), 130.85 (C-8), 131.44 (C-3'), 134.84 (C-4), 135.16 (C-2'), 144.17 (C-8a), 147.76 (C-2), 155.73 (C-6). Anal. Calcd for C₁₆H₁₇NO: C, 80.30; H, 7.16; N, 5.85. Found: C, 80.02; H, 7.12; N, 6.01.

5-Allylquinolin-6-ol (4a)

3a (1.2 g, 6.4 mmol) was heated at 175 °C for 8 h in a microwave oven. After cooling, the light brown solid was treated with CH₂Cl₂ and the precipitated crystals were collected by filtration. Yield: 0.59 g (49.5%); mp 158-162 °C [lit.,² 155°C]; *R_f* = 0.44 (CH₂Cl₂ – acetone 2:1).

¹H NMR (CDCl₃): δ = 3.73 (d, *J* = 5.7 Hz, 2H, CH₂), 4.93 (m, 2H, =CH), 5.95 (m, 1H, HC=), 7.41 (dd, *J* = 9 and 4 Hz, 1H, C₃-H), 7.42 (d, *J* = 9 Hz, 1H, C₇-H), 7.78 (d, *J* = 9 Hz, 1H, C₈-H), 8.26 (d, *J* = 8.5 Hz, 1H, C₄-H), 8.66 (d, *J* = 3 Hz, 1H, C₂-H), 9.88 (s, 1H, OH). ¹³C NMR (CDCl₃): δ = 28.46 (CH₂), 115.16 (=C), 116.80 (C-5), 121.20 (C-3), 121.49 (C-7), 128.23 (C-4a), 128.75 (C-8), 131.33 (C-4), 136.79 (C=), 143.56 (C-C-8a), 146.72 (C-2), 152.59 (C-6).

5-(2-Methylbut-3-en-2-yl)quinolin-6-ol (4b) and 5-(3-methylbut-3-en-2-yl)quinolin-6-ol (4c)

3b (1.0 g, 4.7 mmol) was heated at 175 °C for 10 h in a microwave oven. After cooling, the reaction mixture was dissolved in a mixture of CH₂Cl₂ – acetone (5:2) and purified by column chromatography to yield 0.44 g of **4b** (44%) and 0.12 g of **4c** (12%).

Compound 4b. Brown crystalline solid; mp 132-136 °C; *R_f* = 0.18 (CHCl₃ – acetone 10:1). ¹H NMR (CDCl₃): δ = 1.80 (s, 6H, 2 CH₃), 5.20 (m, 2H, =CH), 6.44 (m, 1H, HC=), 7.28 (m, 2H, C₃-H and C₇-H), 7.84 (d, *J* = 9 Hz, 1H, C₈-H), 8.67 (d, *J* = 3.7 Hz, 1H, C₂-H), 8.79 (d, *J* = 8.9 Hz, 1H, C₄-H). ¹³C NMR (CDCl₃): δ = 29.84 (2 CH₃), 42.65 (C-2'), 110.95 (C-4'), 119.13 (C-3), 123.23 (C-5), 124.33 (C-7), 128.89 (C-4a), 129.28 (C-8), 135.36 (C-4), 144.61 (C-8a), 145.59 (C-2), 151.22 (C-3'), 153.33 (C-6). Anal. Calcd for C₁₄H₁₅NO: C, 78.84; H, 7.09; N, 6.57. Found: C, 78.60; H, 7.17; N, 6.35.

Compound 4c. Light brown crystals; mp 172-176 °C; *R_f* = 0.26 (CHCl₃ – acetone 10:1). ¹H NMR (CDCl₃): δ = 1.55 (d, *J* = 6.2 Hz, 3H, CH₃), 1.71 (s, 3H, CH₃), 4.32 (q, *J* = 6.5 Hz, 1H, C₂-H), 5.22 (d, *J* = 0.9 Hz, 1H, C₄-H), 5.29 (s, 1H, C₄-H), 7.34 (d, *J* = 9.1 Hz, 1H, C₇-H), 7.37 (dd,

$J = 8.7$ and 4.2 Hz, 1H, C₃-H), 7.89 (d, $J = 9.1$ Hz, 1H, C₈-H), 8.47 (d, $J = 8.6$ Hz, 1H, C₄-H), 8.74 (dd, $J = 4.2$ and 1.2 Hz, 1H, C₂-H). ¹³C NMR (CDCl₃): $\delta = 17.34$ (C-1'), 22.77 (CH₃), 37.48 (C-2'), 111.43 (C-4'), 120.67 (C-5), 120.81 (C-3), 122.86 (C-7), 128.37 (C-4a), 129.08 (C-8), 131.35 (C-4), 144.09 (C-8a), 146.53 (C-2), 150.06 (C-3'), 153.36 (C-6). Anal. Calcd for C₁₄H₁₅NO: C, 78.84; H, 7.09; N, 6.57. Found: C, 78.66; H, 6.92; N, 6.38.

5-(3-Methylbut-2-en-1-yl)quinolin-6-ol (7) and 5,5-bis(3-methylbut-2-en-1-yl)quinolin-6(5H)-one (8)

To a stirred mixture of **1** (1.0 g, 6.9 mmol), KOH (24.4 mL 12.5% solution), and triethylbenzylammonium chloride (0.73 g, 4 mmol) in toluene (70 mL) was added **2b** (1.87 g, 12.6 mmol), and stirring was continued at r.t. for 16 h. The reaction mixture was extracted with EtOAc (3 x 50 mL), the combined extracts were washed with sat. aq NaCl, and dried over MgSO₄. Evaporation of the solvent under reduced pressure provided a mixture of three compounds which was separated by column chromatography on silica gel (CHCl₃ – acetone 10:1) to give **3b** (0.25 g, 16.8%), **7** (0.38 g, 26.2%), and **8** (0.47 g, 32.4%).

Compound 7. Brown crystals; mp 154-158 °C (EtOH - H₂O 3:2); $R_f = 0.21$ (CHCl₃ – acetone 10:1). IR (KBr): 3450 (s, OH) cm⁻¹. ¹H NMR (CDCl₃): $\delta = 1.69$ (s, 3H, CH₃), 1.88 (s, 3H, CH₃), 3.76 (d, $J = 6$ Hz, 2H, C₁'-H), 3.83 (br s, 1H, OH), 5.19 (t, $J = 6.5$ Hz, 1H, C₂'-H), 7.36 (t, $J = 6.5$ Hz, 1H, C₃-H), 7.38 (d, $J = 9$ Hz, 1H, C₇-H), 7.80 (d, $J = 9$ Hz, 1H, C₈-H), 8.27 (d, $J = 9$ Hz, 1H, C₄-H), 8.63 (d, $J = 5$ Hz, 1H, C₂-H). ¹³C NMR (CDCl₃): $\delta = 17.82$ (CH₃), 23.67 (C₁'-H), 25.38 (CH₃), 119.64 (C-5), 120.63 (C-3), 121.79 (C-7), 122.65 (C-2'), 127.16 (C-8), 128.62 (C-4a), 131.98 (C-3'), 132.42 (C-4), 143.14 (C-8a), 145.88 (C-2), 152.12 (C-6). Anal. Calcd for C₁₄H₁₅NO: C, 78.84; H, 7.09; N, 6.57. Found: C, 78.60; H, 6.87; N, 6.33.

Compound 8. Brown oil; $R_f = 0.82$ (CHCl₃ – acetone 10:1). IR (film): 1680 (s, CO) cm⁻¹. ¹H NMR (CDCl₃): $\delta = 1.43$ (s, 6H, 2 CH₃), 1.47 (s, 6H, 2 CH₃), 2.49 (dd, $J = 14$ and 7 Hz, 2H, C₁'-H), 2.88 (dd, $J = 14$ and 7 Hz, 2H, C₁'-H), 4.58 (t, $J = 7$ Hz, 2H, C₂'-H), 6.36 (d, $J = 10$ Hz, 1H, C₇-H), 7.29 (dd, $J = 7$ and 7.5 Hz, 1H, C₃-H), 7.57 (d, $J = 10$ Hz, 1H, C₈-H), 7.69 (d, $J = 8$ Hz, 1H, C₄-H), 8.56 (d, $J = 5$ Hz, 1H, C₂-H). ¹³C NMR (CDCl₃): $\delta = 17.83$ (CH₃), 25.58 (CH₃), 56.02 (C-5), 117.84 (C-2'), 123.19 (C-3), 129.81 (C-7), 134.24 (C-4), 134.90 (C-3'), 140.18 (C-4a), 145.97 (C-8), 147.98 (C-2), 149.85 (C-8a), 202.69 (C-6). Anal. Calcd for C₁₉H₂₃NO: C, 81.10; H, 8.24; N, 4.98. Found: C, 80.82; H, 7.95; N, 4.69.

5-(Cyclohex-2-en-1-yl)quinolin-6-ol (4d)

A solution of **3c** (1.0 g, 4.4 mmol) in chlorobenzene (45 mL) was stirred under reflux for 100 h. The solvent was evaporated under reduced pressure and the residue was purified by column chromatography on silica gel (CH₂Cl₂ – acetone 10:1) to yield **4d** (0.7 g, 70%) as colorless crystals. Mp 148-152°C. $R_f = 0.27$ (CHCl₃ – acetone 10:1).

IR (KBr): 3400 (s, OH) cm⁻¹. ¹H NMR (CDCl₃): $\delta = 1.84$ (m, 1H, C₅'-H), 1.97 (m, 2H, C₅'-H and C₆'-H), 2.05 (m, 1H, C₆'-H), 2.20 (m, 1H, C₄'-H), 2.29 (m, 1H, C₄'-H), 4.51 (m, 1H, C₁'-H), 5.82 (m, 1H, C₂'-H), 5.88 (m, 1H, C₃'-H), 7.34 (dd, $J = 8.5$ and 4.0 Hz, 1H, C₃-H), 7.42 (d, $J = 9.0$ Hz, 1H, C₇-H), 7.80 (d, $J = 9.0$ Hz, 1H, C₈-H), 8.65 (d, $J = 8.5$ Hz, 1H, C₄-H), 8.66 (d, $J = 4.0$ Hz, 1H, C₂-H), 8.8 (br. s, 1H, OH). ¹³C NMR (CDCl₃): $\delta = 24.12$ (C-5'), 25.57 (C-4'), 29.8 (C-6'), 34.99

(C-1'), 121.07 (C-3), 122.19 (C-7), 123.46 (C-5), 127.71 (C-3'), 129.25 (C-4a), 130.29 (C-8), 133.00 (C-4), 133.02 (C-2'), 145.85 (C-8a), 147.59 (C-2), 153.18 (C-6). Anal. Calcd for C₁₅H₁₅NO: C, 79.97; H, 6.71; N, 6.22. Found: C, 80.22; H, 6.95; N, 6.01.

5-(Cyclohept-2-en-1-yl)quinolin-6-ol (4e)

A solution of **3d** (0.5 g, 2.1 mmol) in chlorobenzene (20 mL) was stirred under reflux for 100 h. The solvent was evaporated under reduced pressure and the residue was purified by column chromatography on silica gel (CH₂Cl₂ – acetone 10:1) to yield **4e** (0.2 g, 38%) as colorless crystals. Mp 85-90°C. *R_f* = 0.42 (CH₂Cl₂ – acetone 4:1). IR (KBr): 3420 (s, OH) cm⁻¹.

¹H NMR (CDCl₃): δ = 1.50 (m, 1H, C₅'-H), 1.75 (m, 1H, C₆'-H), 1.88 (m, 1H, C₇'-H), 1.93 (m, 1H, C₅'-H), 2.08 (m, 1H, C₆'-H), 2.11 (m, 1H, C₇'-H), 2.27 (m, 1H, C₄'-H), 2.41 (m, 1H, C₄'-H), 4.62 (m, 1H, C₁'-H), 5.95 (m, 2 H, C₂'-H and C₃'-H), 7.34 (d, *J* = 9.0 Hz, 1H, C₇-H), 7.36 (dd, *J* = 9.0 and 4.2 Hz, 1H, C₃-H), 7.74 (d, *J* = 9.0 Hz, 1H, C₈-H), 8.56 (d, *J* = 8.4 Hz, 1H, C₄-H), 8.70 (dd, *J* = 4.2 Hz, 1H, C₂-H), 11.7 (br. s, 1H, OH). ¹³C NMR (CDCl₃): δ = 27.59 (C-5'), 29.58 (C-4'), 31.24 (C-6'), 34.86 (C-7'), 38.36 (C-1'), 120.23 (C-3), 122.73 (C-7), 125.48 (C-5), 127.36 (C-8), 128.14 (C-4a), 131.65 (C-3'), 133.90 (C-4), 136.82 (C-2'), 143.14 (C-8a), 145.48 (C-2), 152.65 (C-6). Anal. Calcd for C₁₆H₁₇NO: C, 80.30; H, 7.16; N, 5.85. Found: C, 80.12; H, 7.32; N, 6.03.

Furo[3,2-*f*]quinolines (5a-e). General procedure

A mixture of **4** (2.5 mmol) and concd H₂SO₄ (0.25 g, 5.3 mmol) was heated in a water bath for 1.5 h. After cooling, the reaction mixture was poured onto ice (10g), basified with 1N NaOH (15 ml) and extracted with CHCl₃ (3 x 15 mL). The combined organic layers were washed with H₂O, dried over MgSO₄ and the solvent was evaporated in vacuo. The residue was purified by column chromatography on silica gel (using CH₂Cl₂ – acetone 5:1 as eluent).

1,2-Dihydro-2-methylfuro[3,2-*f*]quinoline (5a). Yield: 35%; brown oil, [lit.,² 26.5°C], *R_f* = 0.87 (CH₂Cl₂ – acetone 2:1). ¹H NMR (CDCl₃): δ = 1.56 (d, *J* = 6.3 Hz, 3H, CH₃), 3.07 (dd, *J* = 15.2 and 7.5 Hz, 1H, C₁-H), 3.60 (dd, *J* = 15.2 and 9.3 Hz, 1H, C₁-H), 5.17 (m, 1H, C₂-H), 7.29 (d, *J* = 9 Hz, 1H, C₄-H), 7.35 (dd, *J* = 8.4 and 4.1 Hz, 1H, C₈-H), 7.90 (d, *J* = 8.3 Hz, 1H, C₉-H), 7.93 (d, *J* = 9 Hz, 1H, C₅-H), 8.74 (d, *J* = 2.9, 1H, C₇-H). ¹³C NMR (CDCl₃): δ = 22.09 (CH₃), 35.56 (C-1), 80.88 (C-2), 115.39 (C-4), 118.20 (C-9b), 121.31 (C-8), 126.02 (C-9a), 130.33 (C-5), 130.88 (C-9), 144.54 (C-5a), 147.16 (C-7), 157.20 (C-3a).

1,2-Dihydro-1,1,2-trimethylfuro[3,2-*f*]quinoline (5b). Yield: 42.2%; light yellow oil; *R_f* = 0.44 (CHCl₃ – acetone 10:1). ¹H NMR (CDCl₃): δ = 1.32 (s, 3H, C₁-CH₃), 1.46 (d, *J* = 6.6 Hz, 3H, C₂-CH₃), 1.62 (s, 3H, C₁-CH₃), 4.53 (q, *J* = 6.6 Hz, 1H, C₂-H), 7.29 (d, *J* = 8.9 Hz, 1H, C₄-H), 7.33 (dd, *J* = 8.5 and 4.0 Hz, 1H, C₈-H), 7.93 (d, *J* = 8.9 Hz, 1H, C₅-H), 8.29 (d, *J* = 8.5 Hz, 1H, C₉-H), 8.73 (d, *J* = 3.0 Hz, 1H, C₇-H). ¹³C NMR (CDCl₃): δ = 13.89 (CH₃), 22.07 (CH₃), 26.56 (CH₃), 45.31 (C-1), 89.45 (C-2), 115.77 (C-4), 120.97 (C-8), 125.68 (C-9a), 126.89 (C-9b), 129.77 (C-9), 130.60 (C-5), 145.15 (C-5a), 146.88 (C-7), 156.10 (C-3a). Anal. Calcd for C₁₄H₁₅NO: C, 78.84; H, 7.09; N, 6.57. Found: C, 79.07; H, 6.92; N, 6.33

1,2-Dihydro-1,2,2-trimethylfuro[3,2-*f*]quinoline (5c). Yield: 40%; light yellow oil; R_f = 0.59 (CHCl₃ – acetone 10:1). ¹H NMR (CDCl₃): δ = 1.31 (d, J = 7.1 Hz, 3H, C₁-CH₃), 1.44 (s, 3H, C₂-CH₃), 1.53 (s, 3H, C₂-CH₃), 3.44 (q, J = 7.1 Hz, 1H, C₁-H), 7.27 (d, J = 8.9 Hz, 1H, C₄-H), 7.34 (dd, J = 8.4 and 4.1 Hz, 1H, C₈-H), 7.93 (d, J = 8.9 Hz, 1H, C₅-H), 8.04 (d, J = 8.4 Hz, 1H, C₉-H), 8.73 (d, J = 3.3 Hz, 1H, C₇-H). ¹³C NMR (CDCl₃): δ = 16.37 (C₁-CH₃), 22.41 (C₂-CH₃), 28.52 (C₂-CH₃), 44.66 (C-1), 90.23 (C-2), 115.96 (C-4), 121.13 (C-8), 123.90 (C-9b), 126.05 (C-9a), 130.36 (C-9), 130.44 (C-5), 144.80 (C-5a), 146.91 (C-7), 155.32 (C-3a). Anal. Calcd for C₁₄H₁₅NO: C, 78.84; H, 7.09; N, 6.57. Found: C, 79.01; H, 6.92; N, 6.30.

2,3-Dihydro-3,3-dimethyl-1*H*-pyrano[3,2-*f*]quinoline (9). Yield: 54%; brown oil; R_f = 0.66 (CHCl₃ – acetone 10:1). ¹H NMR (CDCl₃): δ = 1.39 (s, 6H, 2 CH₃), 1.95 (m, 2H, C₂-H), 3.00 (m, 2H, C₁-H), 7.25 (d, J = 9.6 Hz, 1H, C₅-H), 7.40 (m, 1H, C₉-H), 7.91 (d, J = 9.0 Hz, 1H, C₆-H), 8.19 (d, J = 8.2 Hz, 1H, C₁₀-H), 8.74 (m, 1H, C₈-H). ¹³C NMR (CDCl₃): δ = 18.84 (C-1), 26.48 (CH₃), 32.23 (C-2), 74.53 (C-3), 112.08 (C-10b), 120.88 (C-9), 123.80 (C-5), 128.12 (C-10a), 128.45 (C-6), 130.77 (C-10), 143.51 (C-6a), 146.38 (C-8), 151.75 (C-4a).

7a,8,9,10,11,11a-Hexahydrobenzofuro[3,2-*f*]quinoline (5d). Yield: 75%; yellow oil; ; R_f = 0.72 (CH₂Cl₂ – acetone 2:1). **cis-Isomer.** t_R = 15.72 min. ¹H NMR (CDCl₃): δ = 1.25 (m, 1H, C₁₁-H), 1.32 (m, 1H, C₁₀-H), 1.57 (m, 1H, C₉-H), 1.67 (m, 1H, C₁₀-H), 1.86 (m, 1H, C₈-H), 2.12 (m, 1H, C₁₁-H), 2.35 (m, 1H, C₈-H), 3.45 (m, 1H, C_{11a}-H), 4.82 (m, 1H, C_{7a}-H), 7.32 (dd, J = 8.5 and 4.2 Hz, 1H, C₂-H), 7.34 (d, J = 9 Hz, 1H, C₆-H), 7.93 (d, J = 9 Hz, 1H, C₅-H), 7.99 (d, J = 8.2 Hz, 1H, C₁-H), 8.73 (d, J = 3 Hz, 1H, C₃-H). ¹³C NMR (CDCl₃): δ = 20.18 (C-9), 22.37 (C-10), 27.24 (C-8), 29.19 (C-11), 39.59 (C-11a), 84.00 (C-7a), 115.85 (C-6), 121.20 (C-2), 125.35 (C-11c), 126.48 (C-11b), 129.89 (C-5), 130.88 (C-1), 144.66 (C-4a), 147.19 (C-3), 156.93 (C-6a). Anal. Calcd for C₁₅H₁₅NO: C, 79.97; H, 6.71; N, 6.22. Found: C, 80.25; H, 6.98; N, 6.00.

trans-Isomer. t_R = 13.37 min. ¹H NMR (CDCl₃): δ = 1.32 (m, 1H, C₉-H), 1.50 (m, 1H, C₉-H), 1.67 (m, 1H, C₈-H), 1.88 (m, 2H, C₁₀-H), 2.03 (m, 2H, C₁₁-H), 2.15 (m, 1H, C₈-H), 3.62 (br. s, 1H, C_{11a}-H), 4.68 (br. s, 1H, C_{7a}-H), 7.30 (d, J = 9.1 Hz, 1H, C₆-H), 7.34 ((dd, J = 8.5 and 4.0 Hz, 1H, C₂-H), 7.88 (d, J = 9.1 Hz, 1H, C₅-H), 8.15 (d, J = 8.5 Hz, 1H, C₁-H), 8.72 (d, J = 3 Hz, 1H, C₃-H). ¹³C NMR (CDCl₃): δ = 17.86 (C-9), 26.33 (C-11a), 29.20 (C-11), 31.06 (C-10), 33.79 (C-8), 70.84 (C-7a), 116.83 (C-11b), 120.94 (C-2), 121.67 (C-6), 126.92 (C-11c), 128.97 (C-5), 129.56 (C-1), 144.22 (C-4a), 146.83 (C-3), 153.90 (C-6a). Anal. Calcd for C₁₅H₁₅NO: C, 79.97; H, 6.71; N, 6.22. Found: C, 80.17; H, 6.99; N, 5.98.

8,9,10,11,12,12a-hexahydro-7a*H*-cycloheptafuro[3,2-*f*]quinoline (5e)

Yield: 74%; R_f = 0.71 (CH₂Cl₂ – acetone 4:1).

cis-Isomer. t_R = 7.78 min. ¹H NMR (CDCl₃): δ = 1.45 (m, 1H, C₉-H), 1.50 (m, 2H, C₁₀-H and C₁₁-H), 1.71 (m, 1H, C₁₁-H), 1.75 (m, 1H, C₁₀-H), 1.89 (m, 2H, C₉-H and C₁₂-H), 1.94 (m, 1H, C₁₂-H), 2.08 (m, 1H, C₈-H), 2.27 (m, 1H, C₈-H), 3.87 (m, 1H, C_{12a}-H), 5.14 (m, 1H, C_{7a}-H), 7.27 (d, J = 8.0 Hz, 1H, C₆-H), 7.32 (dd, J = 8.5 and 4.2 Hz, 1H, C₂-H), 7.92 (d, J = 8.0 Hz, 1H, C₅-H), 8.03 (d, J = 8.3 Hz, 1H, C₁-H), 8.72 (dd, J = 3.7 and 1.0 Hz, 1H, C₃-H). ¹³C NMR (CDCl₃): δ = 23.67 (C-9), 28.84 (C-11), 30.01 (C-12), 31.21 (C-10), 31.45 (C-8), 46.47 (C-12a), 88.19 (C-7a), 115.23 (C-6), 121.12 (C-2), 122.59 (C-5), 125.69 (C-12b), 130.32 (C-1), 130.67 (C-5), 144.89

(C-4a), 146.95 (C-3), 156.63 (C-6a). Anal. Calcd for C₁₆H₁₇NO: C, 80.30; H, 7.16; N, 5.85. Found: C, 80.10; H, 7.38; N, 6.04.

trans-Isomer. $t_R = 7.08$ min. ¹H NMR (CDCl₃): $\delta = 1.34$ (m, 1H, C₁₁-H), 1.48 (m, 1H, C₉-H), 1.57 (m, 1H, C₁₁-H), 1.93 (m, 1H, C₈-H), 2.10 (m, 3H, C₈-H, C₁₀-H and C₁₂-H), 2.36 (m, 1H, C₁₀-H), 3.68 (br. s, 1H, C_{12a}-H), 4.82 (br. s, 1H, C_{7a}-H), 7.24 (d, $J = 9.2$ Hz, 1H, C₆-H), 7.36 ((dd, $J = 8.5$ and 4.2 Hz, 1H, C₂-H), 7.86 (d, $J = 9.2$ Hz, 1H, C₅-H), 8.17 (d, $J = 8.5$ Hz, 1H, C₁-H), 8.73 (d, $J = 3.3$ Hz, 1H, C₃-H). ¹³C NMR (CDCl₃): $\delta = 23.77$ (C-9), 25.67 (C-11), 27.50 (C-12a), 28.43 (C-10), 34.63 (C-12), 36.89 (C-8), 73.03 (C-7a), 117.05 (C-12b), 120.78 (C-2), 122.64 (C-6), 127.14 (C-12c), 129.20 (C-5), 130.30 (C-1), 144.61 (C-4a), 146.71 (C-3), 151.84 (C-6a). Anal. Calcd for C₁₆H₁₇NO: C, 80.30; H, 7.16; N, 5.85. Found: C, 80.06; H, 7.28; N, 6.07.

Acknowledgements

This work was supported by the Hungarian OTKA Foundation (T 42515) and NKFP program contract MediChem II (Hungarian Ministry of Education). We would like to thank Csilla Hegyi and Sarolta Pilbak for the HPLC separations.

References and Notes

1. Dey, B. B.; Seshadri, T. R. *Quart. J. Indian Chem. Soc.* **1926**, *3*, 166.
2. Péne, C.; Demerseman, P.; Cheutin, A.; Royer, R. *Bull. Soc. Chim. Fr.* **1966**, 586.
3. Muratake, H.; Hayakawa, A.; Natsume, M. *Tetrahedron Lett.* **1997**, *38*, 7577.
4. Törincsi, A.; Kolonits, P.; Pálosi, E.; Novák, L. *Synthesis* **2007**, 284.
5. Microwave –assisted organic reactions have gained considerable attention over the past decade, and the method also found application in preparing pharmaceutically active agents. See: Farand, J. A.; Denissova, I.; Barriault, L. *Heterocycles* **2004**, *62*, 735. Kotha, S.; Mandal, K.; Deb, A. Ch.; Banerjee, S. *Tetrahedron Lett.* **2004**, *45*, 9603. Rao, V. V.; Reddy, G. V.; Yadla, R.; Narsaiah, B.; Rao, P. S. *Arkivoc* **2005**, *3*, 211. Jacob, A. M.; Moody, Ch. J. *Tetrahedron Lett.* **2005**, *46*, 8823. Kremsner, J. M.; Kappe, C. O. *J. Org. Chem.* **2006**, *71*, 4651, and references cited therein.
6. For excellent monograph on Claisen rearrangement, see: Tarbell, D. S. *Org. React.* **1944**, *2*, 2. Zigler, F. E. *Acc. Chem. Res.* **1977**, *10*, 227. Murray, A. W. *Org. React. Mech.* **1980**, 517. Lutz, R. P. *Chem. Rev.* **1984**, *84*, 205. Zigler, F. E. *Chem. Rev.* **1988**, *88*, 1423. Nubbemeyer, U. *Synthesis* **2003**, 961. Castro, A. M. M. *Chem. Rev.* **2004**, *104*, 2939.
7. This type of rearrangement (**3b**→**4c**) was first reported by W.M. Lauer and W.F. Filbert: *J. Am. Chem. Soc.* **1936**, *58*, 1388. The mechanism of the reactions has been identified as the result of three consecutive processes: [3,3] rearrangement followed by 1,5-hydrogen shifts. Hurd, C. D. ; Pollack, M. A. *J. Org. Chem.* **1939**, *3*, 550. Marwell, E. N. *J. Org. Chem.*

- 1960**, 25, 676. Lauer, W. M.; Douldoras, G. A.; Hileman, R. E.; Lupino, R. *J. Org. Chem.* **1961**, 26, 4785. Marwell, E. N.; Anderson, D. R.; Ong, J. *J. Org. Chem.* **1962**, 27, 1109. Roberts, R. M.; Landolt, R. G.; Greene, R. N.; Heyer, E. W. *J. Am. Chem. Soc.* **1967**, 89, 1404. Scheinmann, F.; Barner, R.; Schmid, H. *Helv. Chim. Acta* **1968**, 51, 1603. Rhoads, S. J.; Raulins, N. R. *Org. React.* **1974**, 22, 1. Schobert, R.; Siegfried, S.; Gordon, G.; Mulholland, D.; Nieuwenhuyzen, M. *Tetrahedron Lett.* **2001**, 42, 4561.
8. Cox, R. A.; Swallow, A. J. *J. Chem. Soc.* **1958**, 3727.
 9. Dondas, H. A.; Grigg, R.; Thibault, S. *Tetrahedron* **2001**, 57, 7035.