

# Utilisation of chiral enamines and azomethine imines in the synthesis of functionalised pyrazoles

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## Abstract

Chiral enamines, derived from commercially available enantiopure starting materials, such as (+)-camphor and  $\alpha$ -amino acids, were employed in cyclocondensation reactions with hydrazine derivatives to afford the corresponding pyrazoles, functionalised with terpene, alanine, 2-phenylethylamine, and  $\beta$ -amino alcohol moiety. On the other hand, recent study on stereocontrol in cycloadditions of racemic (1*Z*,4*R*\*,5*R*\*)-1-arylmethylidene-4-benzoylamino-5-phenylpyrazolidin-3-one-1-azomethine imines, available in three steps from hippuric acid, showed, that stereodirecting phenyl group, as well as *ortho*-substituents at the aromatic ring, control the selectivity of these cycloadditions. In extension, these results are now applied in a study, which is oriented towards combinatorial synthesis of pyrazolo[1,2-*a*]pyrazolone type of peptidomimetics with variable, yet predictable configuration.

**Keywords:** Enaminones, heterocycles, azomethine imines, cyclisations, cycloadditions, chiral pool

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## 1. Introduction

Pyrazoles belong among the most representative five-membered heterocyclic systems.<sup>1</sup> Despite the fact, that pyrazole ring is rarely a constituent of natural products, numerous synthetic pyrazole derivatives found use in various pharmaceutical, agrochemical, photographic, and other applications. Such examples of important pyrazole derivatives are natural products (*S*)-pyrazolylalanine,<sup>2</sup> pyrazomycin,<sup>3</sup> and withasomine<sup>4</sup> and synthetic compounds sildenafil (Viagra®),<sup>5</sup> lonazolac,<sup>6</sup> difenamizole,<sup>7</sup> mepirizole,<sup>8</sup> phenidone,<sup>9</sup> and bicyclic pyrazolidinone LY 186826.<sup>10</sup>

On the other hand, synthesis and transformations of heterocyclic compounds represent the major topics of our research interest, which is primarily focused on development of synthetic methodologies for the preparation of various heterocyclic systems. In extension, these methodologies are then used for preparation of various types of heterocyclic compounds, which are functionalised with an amino acid, hydroxy acid, amino alcohol, polyol, nucleoside, terpene, dipeptide, and related structural motifs.<sup>11</sup> Within this context, a part of our studies was also focused on the synthesis of functionalised pyrazoles. For this purpose, we used two 3+2 heterocyclisation approaches, which are, also in general, the most frequently employed methods for the formation of the pyrazole ring:

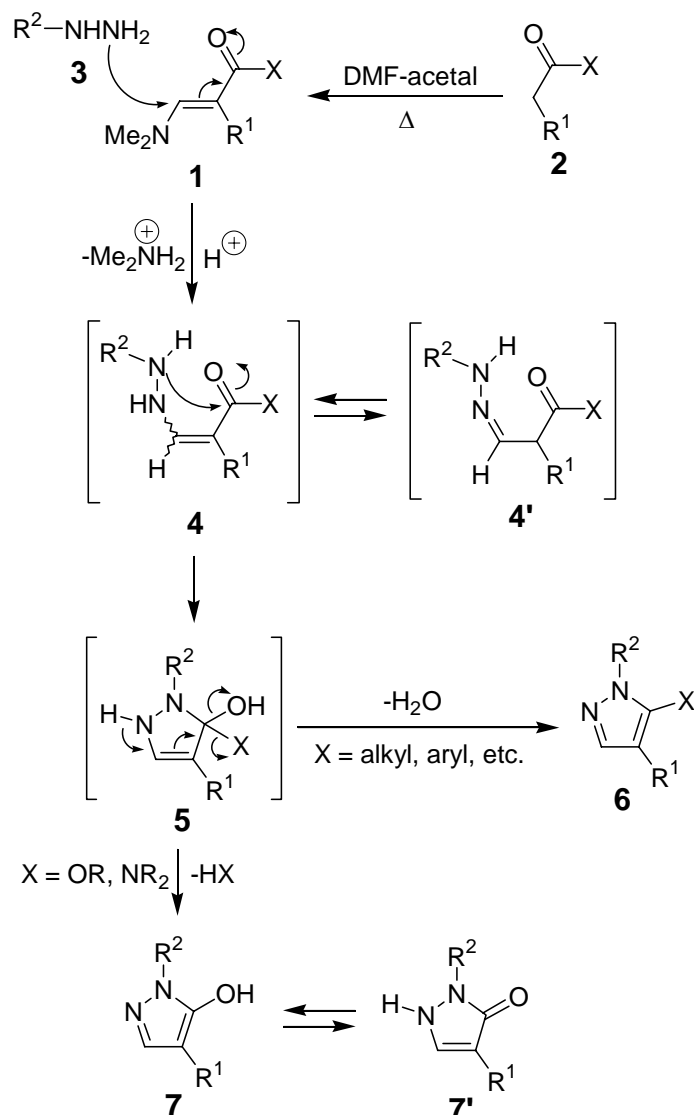
- cyclocondensation between a hydrazine derivative and a suitably functionalised chiral enaminone as enamino masked 1,3-dicarbonyl compound analogue and
- 1,3-dipolar cycloaddition of chiral 3-pyrazolidinone-1-azomethine imine to a suitable dipolarophile.

The present review represents a summary of our most recent results in the synthesis of functionalised pyrazoles.

## 2. Syntheses of functionalised pyrazoles from chiral enaminones

In the last two decades, a series of 2-substituted alkyl 3-(dimethylamino)propenoates **1** and related enaminones was synthesized and used for the preparation of various heterocyclic

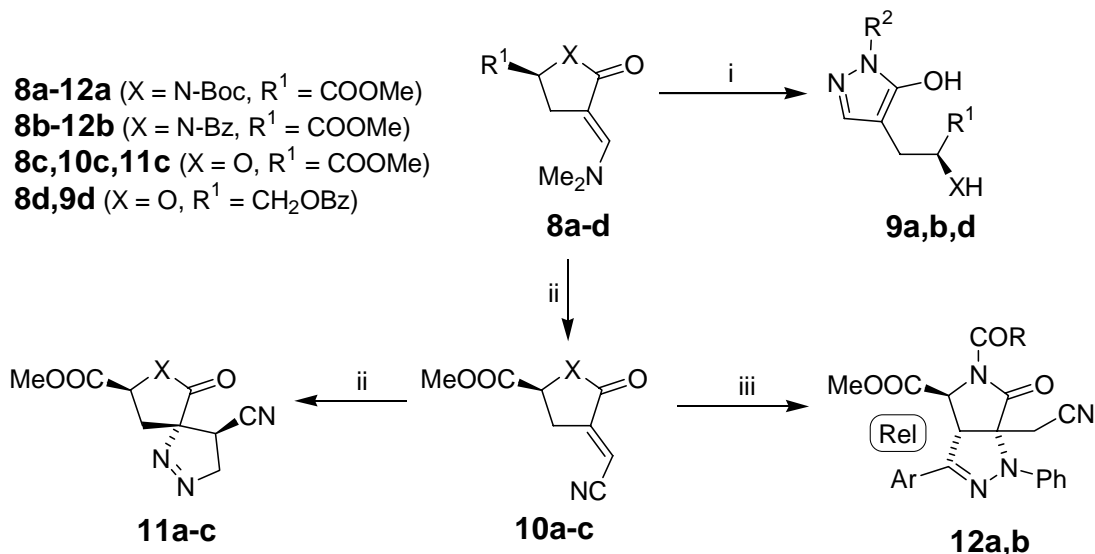
systems, functionalised heterocycles, and natural product analogues.<sup>12,13</sup> Recently, 3-(dimethylamino)propenoates **1** and analogs also found use in combinatorial synthesis.<sup>14</sup> The most usual synthesis of alkyl 3-(dimethylamino)propenoates **1** consists of a treatment of an active methylene compound **2** with a formamide acetal. Propenoates **1** exhibit similar reactivity as their parent 1,3-dicarbonyl compounds: a) they react with electrophiles at position 2 and b) two electrophilic sites at positions 3 and 1 enable cyclisations with various 1,2- and 1,3-dinucleophiles leading to five- and six-membered heterocyclic systems. Reactions of **1** with nucleophiles are acid-catalysed and proceed by initial substitution of the dimethylamino group, followed condensation to the carbonyl group.<sup>12</sup>



Scheme 1

For example, in the reactions of **1** with hydrazine derivatives **3**, substitution of the dimethylamino group takes place first to give the enehydrazines **4** (or tautomeric hydrazones **4'**), followed by cyclisation to the carbonyl group to give the intermediate **5**. In the case of enamino ketones ( $X = \text{alkyl, aryl, etc.}$ ), elimination of water affords the 4,5-disubstituted pyrazole **6**, while in the case of enamino esters and enamino amides, 4-substituted 5-hydroxypyrazoles **7** and/or their tautomers **7'** are usually formed.<sup>12</sup> In some cases, the intermediates **4/4'**,<sup>15,16</sup> and **5**<sup>17</sup> were isolated under mild reaction conditions and their structures were confirmed by X-Ray diffraction (Scheme 1).

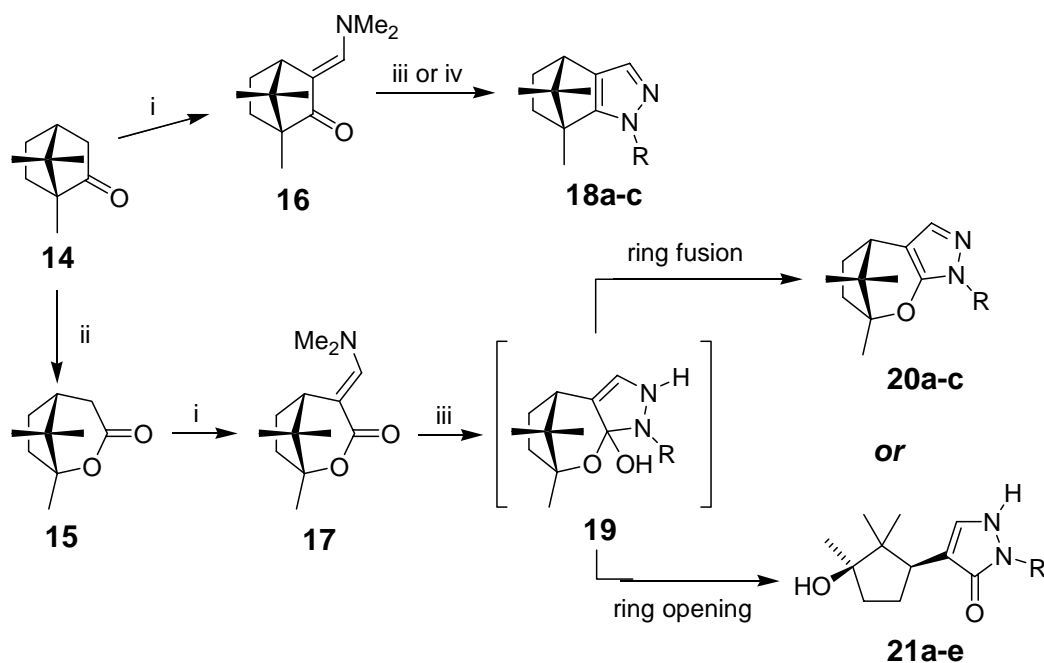
In the chiral enamino series, our previous studies were based on transformations of 5-substituted lactams **8a,b** and lactones **8c,d**. Acid-catalysed reactions of **8a,b,d** with various hydrazine derivatives **3**, resulted in 'ring switching' formation<sup>18</sup> of methyl (*S*)-*N*-acyl-3-(1-substituted-5-hydroxy-1*H*-pyrazol-4-yl)alanines **9a,b**<sup>19</sup> and (*S*)-1-*O*-benzoyl-3-(1-substituted-5-hydroxy-1*H*-pyrazol-4-yl)propane-1,2-diols **9c**.<sup>20</sup> In a recent extension, the 'ring switching' methodology was also applied in a parallel solution-phase synthesis of 3-pyrazolylalanines.<sup>21</sup> On the other hand, when the dimethylamino group in enaminoes **8a–c** was substituted by the cyano group, nitriles **10a–c** were obtained and used as chiral dipolarophiles in 1,3-dipolar cycloadditions to diazomethane and nitrile imines to afford spiro pyrazoles **11a–c** and fused pyrazoles **12a,b** with a dipeptide or closely related structural unit (Scheme 2).<sup>22</sup>



**Scheme 2.** (i)  $\text{R}^2\text{NHNH}_2$  **3**, AcOH, 80–120 °C; (ii) KCN, AcOH, r.t.; (iii)  $\text{CH}_2\text{N}_2\text{-Et}_2\text{O}$ , 0 °C; (iii)  $\text{ArC}(\text{Cl})=\text{NNHPh}$ ,  $\text{Et}_3\text{N}$ ,  $\text{CH}_2\text{Cl}_2$ , reflux.

Similarly, 1-acyl-3-methyl-5-[(*Z*)-cyanomethylidene]imidazolidin-2,4-diones **13**, available in three steps from hydantoin, reacted with diazomethane, azomethine imines, and nitrile imines to give, depending on the reaction conditions, the spiro pyrazole hydantoin and the pyrazole-5-carboxamide derivatives.<sup>23</sup>

## 2.1 Synthesis of terpene-functionalised pyrazoles



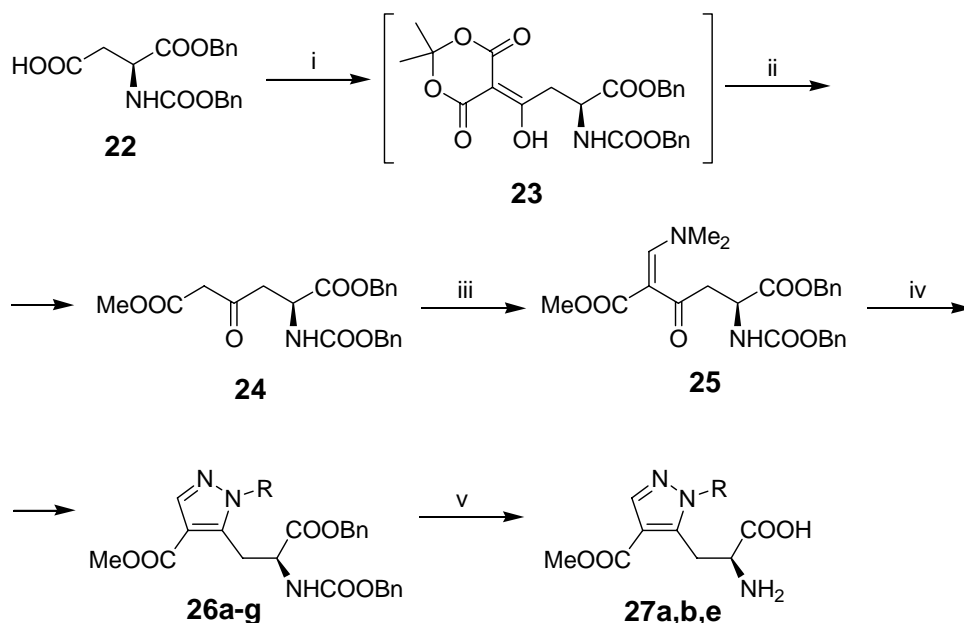
Compound	R	Yield (%)
<b>3a, 18a</b>	H	81
<b>3b, 18b</b>	Benzyl	63
<b>3c</b>	6-chloropyridazin-3-yl	-
<b>18c</b>	6-oxo-1,6-dihydropyridazin-3-yl	83
<b>3d, 20a</b>	Ph	91
<b>3e, 20b</b>	3-methylphenyl	74
<b>3f, 20c</b>	4-methylphenyl	85
<b>3a, 21a</b>	H	83
<b>3g, 21b</b>	2-methylphenyl	70
<b>3h, 21c</b>	2-chlorophenyl	61
<b>3i, 21d</b>	2-bromophenyl	63
<b>3j, 21e</b>	Pentafluorophenyl	56

**Scheme 3.** (i) *t*-BuOCH(NMe<sub>2</sub>)<sub>2</sub>, DMF, reflux; (ii) AcOOH, AcOH, AcONa, r.t.; (iii) R<sup>2</sup>-NHNH<sub>2</sub> (**3a,d-j**), *n*-PrOH, 37% HCl (1 equiv.), reflux; (iv) R<sup>2</sup>-NHNH<sub>2</sub> **3b,c**, AcOH, reflux.

Recently, (1*R*,3*E*,4*R*)-3-[(dimethylamino)methylidene]-1,7,7-trimethylbicyclo [2.2.1]-heptan-2-one (**16**)<sup>16</sup> and (1*R*,4*E*,5*R*)-4-[(dimethylamino)methylidene]-1,8,8-trimethyl-2-oxabicyclo[3.2.1]octan-3-one (**17**),<sup>24</sup> were prepared from (+)-camphor (**14**). Reactions of **16** and **17** with hydrazines **3a-j** afforded terpene-functionalised pyrazoles **18**, **20**, and **21**. Reactions of **16** were selective and gave the corresponding pyrazolo fused camphors **18a-c**. In the reaction of

**16** with 6-chloro-3-hydrazinopyridazine (**3c**), substitution of the chloro by the hydroxy group took place. On the other hand, reactions of the lactone analogue **17** proceeded in two ways. Treatment of **17** with *ortho*-unsubstituted phenylhydrazines **3d–f** furnished the pyrazolo fused lactones **20a–c** as products of elimination of water from the intermediate **19**, while with *ortho*-substituted hydrazines **3g–j** and with hydrazine hydrochloride (**3a**), opening of the lactone ring took place to give the ‘ring switched’ products **21a–e**. It has to be emphasized, that also these reactions were highly selective and led to a single type of product, depending on the type of hydrazine derivative employed. Selectivity of these transformations might be attributed to steric, as well as to electronic effects (Scheme 3).<sup>25</sup>

## 2.2 Synthesis of (*S*)-3-(1-substituted-4-methoxycarbonyl-1*H*-pyrazol-5-yl)alanines from *L*-aspartic acid



Compound	R	Yield (%)	
		<b>26</b>	<b>27</b>
<b>3a, 26a, 27a</b>	H	57	82
<b>3d, 26b, 27b</b>	Ph	94	75
<b>3j, 26c</b>	pentafluorophenyl	73	
<b>3k, 26d</b>	4-methoxyphenyl	82	
<b>3l, 26e, 27e</b>	pyridin-2-yl	85	82
<b>3m, 26f</b>	6-phenylpyridazin-3-yl	56	
<b>3n, 26g</b>	pyrimidin-2-yl	80	

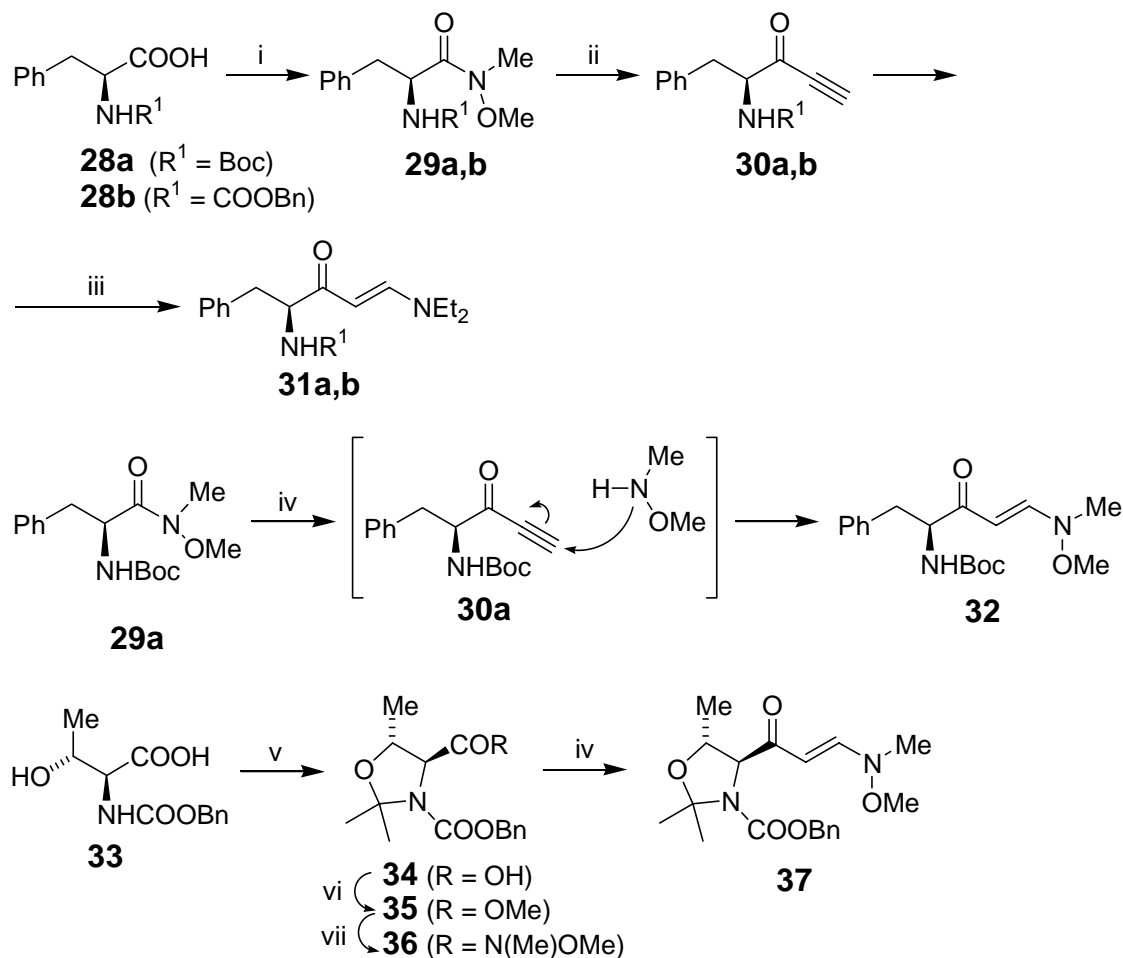
**Scheme 4.** (i) Meldrum's acid, DCC, CH<sub>2</sub>Cl<sub>2</sub>, DMAP, –5 °C (Ref. ); (ii) MeOH, reflux; (iii) DMFDMA, CH<sub>2</sub>Cl<sub>2</sub>, r.t.; (iv) R–NHNH<sub>2</sub> × HCl **3a,d,j–n**, MeOH or EtOH, r.t.→reflux; (v) H<sub>2</sub> (1 bar), Pd–C, MeOH, r.t.

In search for alternative synthetic routes towards 3-pyrazolylalanines and other 3-heteroarylalanines, we have recently developed an enaminone-based methodology for the synthesis of (*S*)-3-(1*H*-pyrazol-5-yl)alanines. Starting from L-aspartic acid, the (*S*)-*N*-benzyloxycarbonylaspartic acid-1-benzyl ester (**22**) was prepared according to the literature procedure.<sup>26</sup> Following closely related literature examples,<sup>27</sup> compound **22** was then coupled with Meldrum's acid to give the intermediate **23**, which was transformed with methanol into 1-benzyl-6-methyl (*S*)-2-benzyloxycarbonylamino-4-oxohexanedioate (**24**) in 66% yield over two steps. Further treatment of **24** with *N,N*-dimethylformamide dimethyl acetal (DMFDMA) in dichloromethane at r.t. furnished the desired enaminone **25** in 98% yield. Cyclocondensations of **25** with substituted hydrazine hydrochlorides **3a,d,j-n** gave the protected (*S*)-pyrazolylalanines **26a-g** in 56–94% yields. Deprotection of **26a,b,e** by catalytic hydrogenation furnished the free pyrazolylalanines **27a,b,e** (Scheme 4).<sup>28</sup>

## 2.3 Synthesis of pyrazolyl and pyrazolo[1,5-*a*]pyrimidinyl substituted 2-phenylethylamines and $\beta$ -amino alcohols

### 2.3.1 Synthesis of L-3-phenylalanine and L-threonine derived enamino ketones

According to the literature known methodology,<sup>29</sup> the *N*-protected L-3-phenylalanines **28a,b** were transformed into the Weinreb amides **29a,b**, which were then treated with excess ethynylmagnesium bromide. Upon quenching excess Grignard reagent with aqueous NaHSO<sub>4</sub>, the corresponding ethynyl ketones **30a,b** were obtained in 90 and 61% yield, respectively. Addition of diethylamine to the triple C $\equiv$ C bond then afforded enamino ketones **31a,b** in 92% and 86% yield, respectively. On the other hand, upon treatment of **29a** with excess ethynylmagnesium bromide followed by quenching with aqueous NH<sub>4</sub>Cl, the *N*-methyl-*N*-methoxy substituted enaminone **32** was obtained in 50% yield. Formation of **32** could be explained by initial formation of the ethynyl ketone **30a** followed by addition of *N,O*-dimethylhydroxylamine to the triple C $\equiv$ C bond. Similarly, the enaminone **37** was prepared in four steps from *N*-benzyloxycarbonyl-L-threonine (**33**) (Scheme 5).<sup>30</sup>

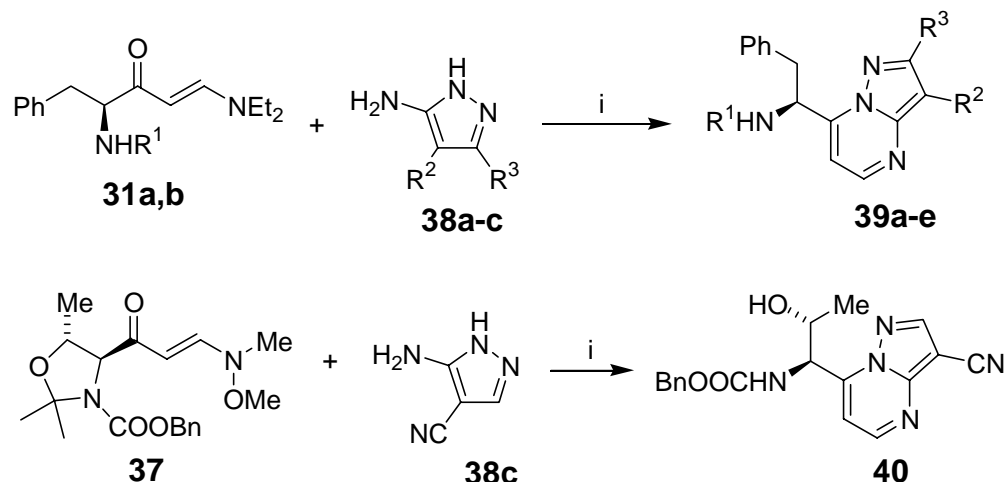


**Scheme 5.** (i)  $\text{ClCOOBu}$ , *N*-methylmorpholine,  $\text{EtOAc}$ ,  $0\text{ }^\circ\text{C}$ , then  $\text{MeNHOMe}$ ,  $0\text{ }^\circ\text{C} \rightarrow \text{r.t.}$ ; (ii)  $\text{HC}\equiv\text{CMgBr}$ , THF,  $-78\text{ }^\circ\text{C} \rightarrow \text{r.t.}$ , then aq.  $\text{NaHSO}_4$ ; (iii)  $\text{Et}_2\text{NH}$ ,  $\text{CH}_2\text{Cl}_2$ ,  $0\text{ }^\circ\text{C} \rightarrow \text{r.t.}$ ; (iv)  $\text{HC}\equiv\text{CMgBr}$ , THF,  $-78\text{ }^\circ\text{C} \rightarrow \text{r.t.}$ , then aq.  $\text{NH}_4\text{Cl}$ ; (v)  $\text{Me}_2\text{C}(\text{OMe})_2$ ,  $\text{BF}_3 \cdot \text{Et}_2\text{O}$ , r.t.; (vi)  $\text{MeI}$ ,  $\text{K}_2\text{CO}_3$ , acetone,  $0\text{ }^\circ\text{C} \rightarrow \text{r.t.}$ ; (vii) *i*-PrMgBr,  $\text{MeNHOMe}$ , THF,  $-78\text{ }^\circ\text{C} \rightarrow -20\text{ }^\circ\text{C}$ , then aq.  $\text{NH}_4\text{Cl}$ .

### 2.3.2 Synthesis of pyrazolo[1,5-*a*]pyrimidinyl substituted 2-phenylethylamines and $\beta$ -amino alcohols

Enaminones **31a,b** and **37** reacted with 3-aminopyrazole derivatives **38a-c** as 1,3-dinucleophiles, to afford the corresponding (*S*)-1-(pyrazolo[1,5-*a*]pyrimidin-7-yl)-2-phenylethylamines **39a-e** and (*S*)-1-amino-1-(pyrazolo[1,5-*a*]pyrimidin-7-yl)propan-2-ol (**40**) in 20–77% yields. In the reaction of **37** with 5-amino-1*H*-pyrazole-4-carbonitrile (**38c**), simultaneous removal of the ketal protecting group also took place (Scheme 6).<sup>30</sup>



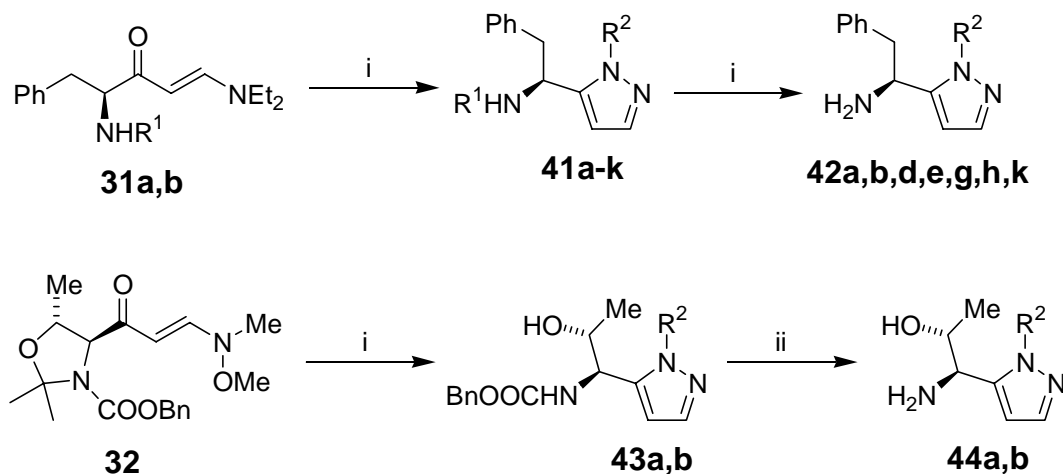


Compound	R <sup>1</sup>	R <sup>2</sup>	R <sup>3</sup>	Yield (%)
<b>38a, 39a</b>	COOBn	H	H	76
<b>38a, 39b</b>	Boc	H	H	44
<b>38b, 39c</b>	COOBn	H	Me	32
<b>38c, 39d</b>	COOBn	CN	H	73
<b>38c, 39e</b>	Boc	CN	H	20
<b>40</b>				77

**Scheme 6.** (i) (i) EtOH, 37% HCl (1 equiv.), r.t.→reflux.

### 2.3.4 Synthesis of pyrazolyl substituted 2-phenylethylamines and $\beta$ -amino alcohols

Acid-catalysed treatment of enaminones **31a,b** with substituted hydrazines **3a,c,d,i,h-o**, afforded the corresponding *N*-protected (*S*)-1-pyrazolyl-2-phenylethylamines **41a-k** in 54–98% yields. Deprotection of compounds **41a,b,d,e,g,h,k** by catalytic hydrogenation furnished free (*S*)-1-pyrazolyl-2-phenylethylamines **42a,b,d,e,g,h,k** in 69–98% yields. In the same manner, L-threonine derived enamino ketone **32** was transformed with phenylhydrazine (**3d**) and 3-hydrazino-6-phenylpyridazine (**3m**) into compounds **34a** and **34b**, respectively. Deprotection by catalytic hydrogenation afforded (2*S*,3*R*)-1-amino-1-(1-phenyl-1*H*-pyrazol-5-yl)propan-2-ol (**44a**) and (2*S*,3*R*)-1-amino-1-[1-(6-phenylpyridazin-3-yl)-1*H*-pyrazol-5-yl]propan-2-ol (**44b**) (Scheme 7).<sup>30</sup>



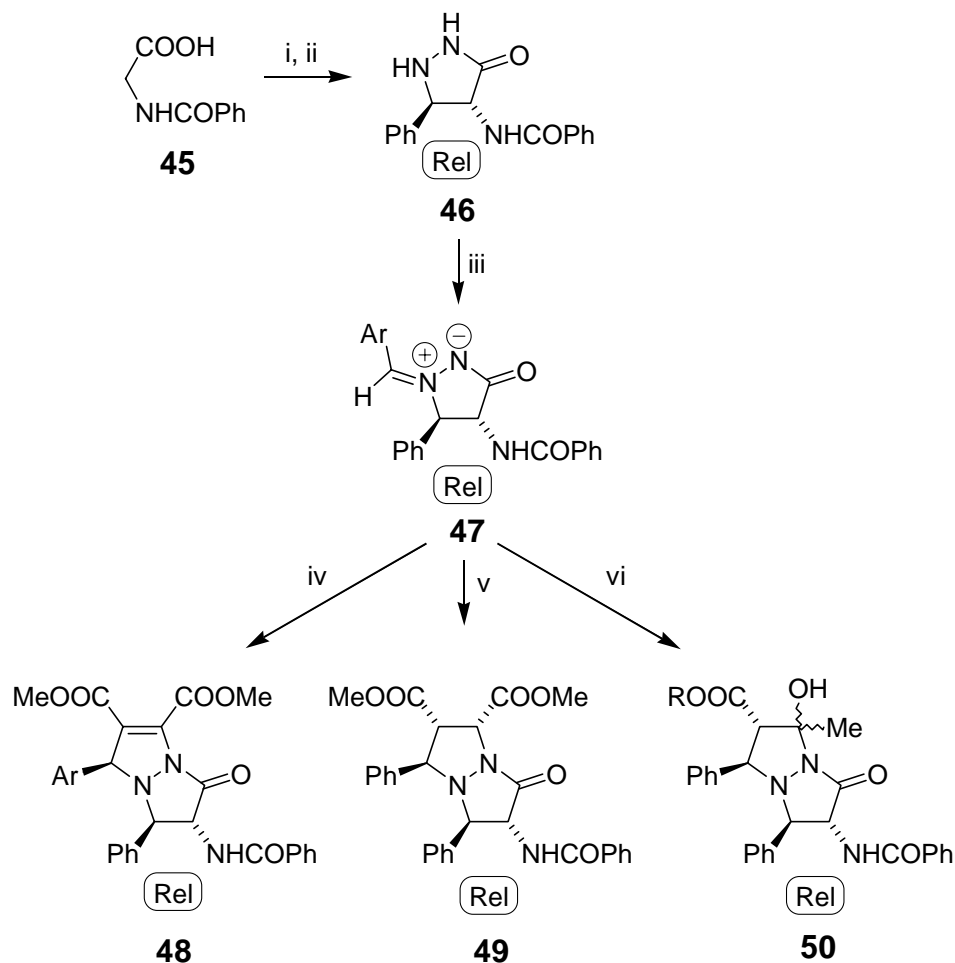
Compound	R <sup>1</sup>	R <sup>2</sup>	Yield (%)	
			41 or 43	42 or 44
3a, 41a, 42a	COOBn	H	98	92
3a, 41b, 42b	Boc	H	86	82
3d, 41c	COOBn	Ph	95	
3o, 41d, 42d	COOBn	4-nitrophenyl	80	95
3k, 41e, 42e	COOBn	4-methoxyphenyl	70	98
3i, 41f	Boc	2-bromophenyl	90	
3l, 41g, 42g	COOBn	pyridin-2-yl	83	85
3n, 41h, 42h	COOBn	pyrimidin-2-yl	89	69
3c, 41i	COOBn	6-chloropyridazin-3-yl	91	
3c, 41j	Boc	6-chloropyridazin-3-yl	54	
3m, 41k, 42k	COOBn	6-phenylpyridazin-3-yl	78	70
3d, 43a, 44a		Ph	92	83
3m, 43b, 44b		6-phenylpyridazin-3-yl	63	64

**Scheme 7.** (i) R<sup>2</sup>-NHNH<sub>2</sub> **3a,c,d,i,k-o**, EtOH, 37% HCl (1 equiv.), r.t.→reflux; (ii) H<sub>2</sub> (1 bar), EtOH-THF, 10% Pd-C, r.t.

### 3. Syntheses of functionalised pyrazoles from (1*Z*,4*R*\*,5*R*\*)-1-arylmethylidene-4-benzoylamino-5-phenylpyrazolidin-3-one-1-azomethine imines

1,3-Dipolar cycloaddition reactions are useful methods for preparation of five-membered heterocycles. They enable access to polyfunctional compounds with multiple asymmetric centers, often with excellent stereocontrol.<sup>31</sup> In contrast to well elaborated asymmetric cycloadditions in chiral nitron, nitrile oxide, and azomethine ylide series,<sup>32</sup> much less examples of asymmetric cycloadditions to chiral azomethine imines have been reported.<sup>33-37</sup>

The importance of pyrazolidin-3-ones significantly rose in the last two decades, since several pyrazolidin-3-one derivatives exhibit biological activities and due to their applicability in industrial processes.<sup>9,10,38,39</sup> For example, 2-acylamino-1-oxo-1*H*,5*H*-pyrazolo[1,2-*a*]pyrazole-7-carboxylates are useful scaffolds for preparation of conformationally constrained peptidomimetics.<sup>10,40</sup> Since the first reports of *Dorn*<sup>41</sup> and *Oppolzer*,<sup>42</sup> 1,3-dipolar cycloaddition of stable, pyrazolidin-3-one derived, azomethine imines represent a simple and efficient method for preparation of 1*H*,5*H*-pyrazolo[1,2-*a*]pyrazol-1-ones. However, most of these studies were performed on achiral dipoles and on poorly substituted chiral azomethine imines.<sup>10,38-40,43,44</sup> In connection with our studies in the field of 3-pyrazolidinones,<sup>34,45-50</sup> we have previously reported regio- and stereoselective 1,3-dipolar cycloadditions to polysubstituted racemic (1*Z*,4*R*\*,5*R*\*)-1-arylmethylidene-4-benzoylamino-5-phenylpyrazolidin-3-one-1-azomethine imines **47** leading to polysubstituted 1*H*,5*H*-pyrazolo[1,2-*a*]pyrazol-1-ones **48-50** (Scheme 8).<sup>34</sup>



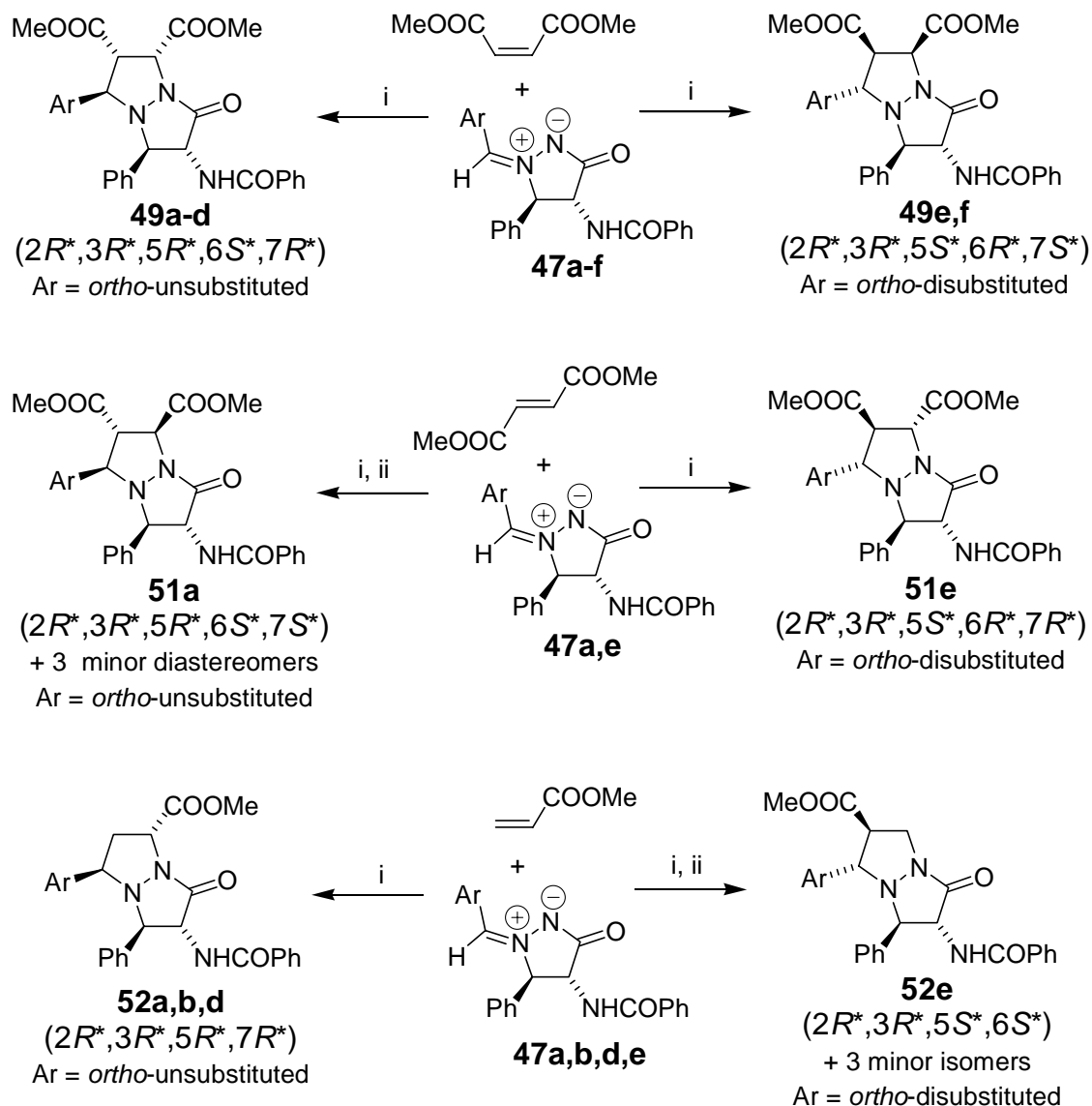
**Scheme 8.** (i) PhCHO, Ac<sub>2</sub>O, AcONa, 100 °C; (ii) N<sub>2</sub>H<sub>4</sub>×H<sub>2</sub>O (80%), reflux; (iii) ArCHO, EtOH, TFA (cat.), reflux; (iv) dimethyl acetylenedicarboxylate, anisole, reflux; (v) dimethyl maleate, anisole reflux; (vi) methyl or ethyl acetoacetate, MeOH, Et<sub>3</sub>N (1 equiv.), r.t.

Similarly, *Chuang* and *Sharpless*,<sup>36</sup> as well as *Husson*, *Bonin*, *Micouin*, and coworkers,<sup>35,37</sup> reported high facial selectivity of 1,3-dipolar cycloadditions to chiral azomethine imines, derived from pyrazolidinones and related 1,3,4-oxadiazinones.

On the other hand, our previous study on reactions of 1-arylmethylidene-5,5-dimethylpyrazolidin-3-on-1-azomethine imines with methyl propiolate showed, that the regioselectivity was strongly dependent on the *ortho*-substituents at the aromatic ring.<sup>49</sup> This results prompted us to investigate also the influence of *ortho*-substituents in chiral azomethine imines **47** on stereoselectivity and regioselectivity of cycloadditions.

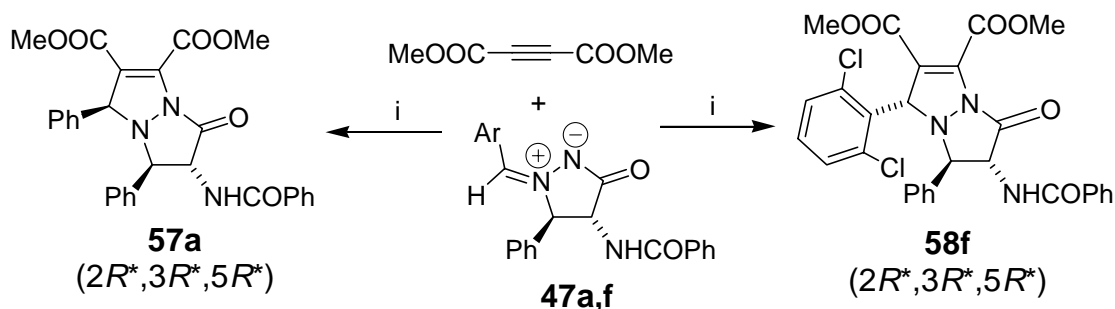
### 3.1 Stereocontrol in cycloadditions to dimethyl maleate, dimethyl fumarate, methyl acrylate, and dimethyl acetylenedicarboxylate

Azomethine imines **47a–f** with the following aryl substituents attached to the exocyclic C=N double bond were chosen as the model 1,3-dipoles: phenyl (**47a**), 4-nitrophenyl (**47b**), 4-methoxyphenyl (**47c**), 3,4,5-trimethoxyphenyl (**47d**), 2,4,6-trimethylphenyl (**47e**), and 2,6-dichlorophenyl (**47f**). Cycloadditions were carried out with dimethyl maleate, dimethyl fumarate, and methyl acrylate as the model dipolarophiles. Cycloadditions of dipoles **47a–f** to dimethyl maleate were all stereoselective, however, two diastereomeric types of cycloadducts were formed, depending on *ortho*-substituents at the aromatic ring. *ortho*-Unsubstituted dipoles **47a–d** afforded the cycloadducts **49a–d** with (2*R*\*,3*R*\*,5*R*\*,6*S*\*,7*R*\*)-configuration, while *ortho*-disubstituted azomethine imines **47e,f** gave, stereoselectively, the major (2*R*\*,3*R*\*,5*S*\*,6*R*\*,7*S*\*)-isomers **49e,f**. In contrast, cycloaddition of *ortho*-unsubstituted **47a** to dimethyl fumarate (**5**) was not stereoselective and gave a mixture of **51a** and three isomeric cycloadducts in a ratio of 33:30:22:15, which were separated by chromatography. On the other hand, reaction of dimethyl fumarate with *ortho*-disubstituted **47e** afforded the major (2*R*\*,3*R*\*,5*S*\*,6*R*\*,7*R*\*)-isomer **51e** in 68% d.e.. Reactions of *ortho*-unsubstituted **47a,b,d** with methyl acrylate proceeded regioselectively and stereoselectively to give the (2*R*\*,3*R*\*,5*R*\*,7*R*\*)-isomers **55a,b,d**, while cycloaddition of *ortho*-disubstituted **47e** was not selective and furnished a mixture of the major (2*R*\*,3*R*\*,5*S*\*,6*S*\*)-isomer **52e** and three minor isomers in a ratio of 34:30:21:15, which were separated by chromatography (Scheme 9).<sup>51</sup>



**Scheme 9.** (i) Anisole, reflux; (ii) chromatographic separation (CC followed by MPLC).

These results prompted us to reinvestigate the configuration of compounds, formed upon cycloaddition of dipoles **47a** and **47f** to dimethylacetylene dicarboxylate.<sup>34</sup> Also in this case, stereocontrol was dependent on *ortho*-substituents. Thus, **47a** gave cycloadduct **48a** with  $(2R^*,3R^*,5R^*)$ -configuration, while the *ortho*-disubstituted **48f** gave cycloadduct **58f** with the opposite sense of configuration at position 5 (Scheme 10).<sup>51</sup>

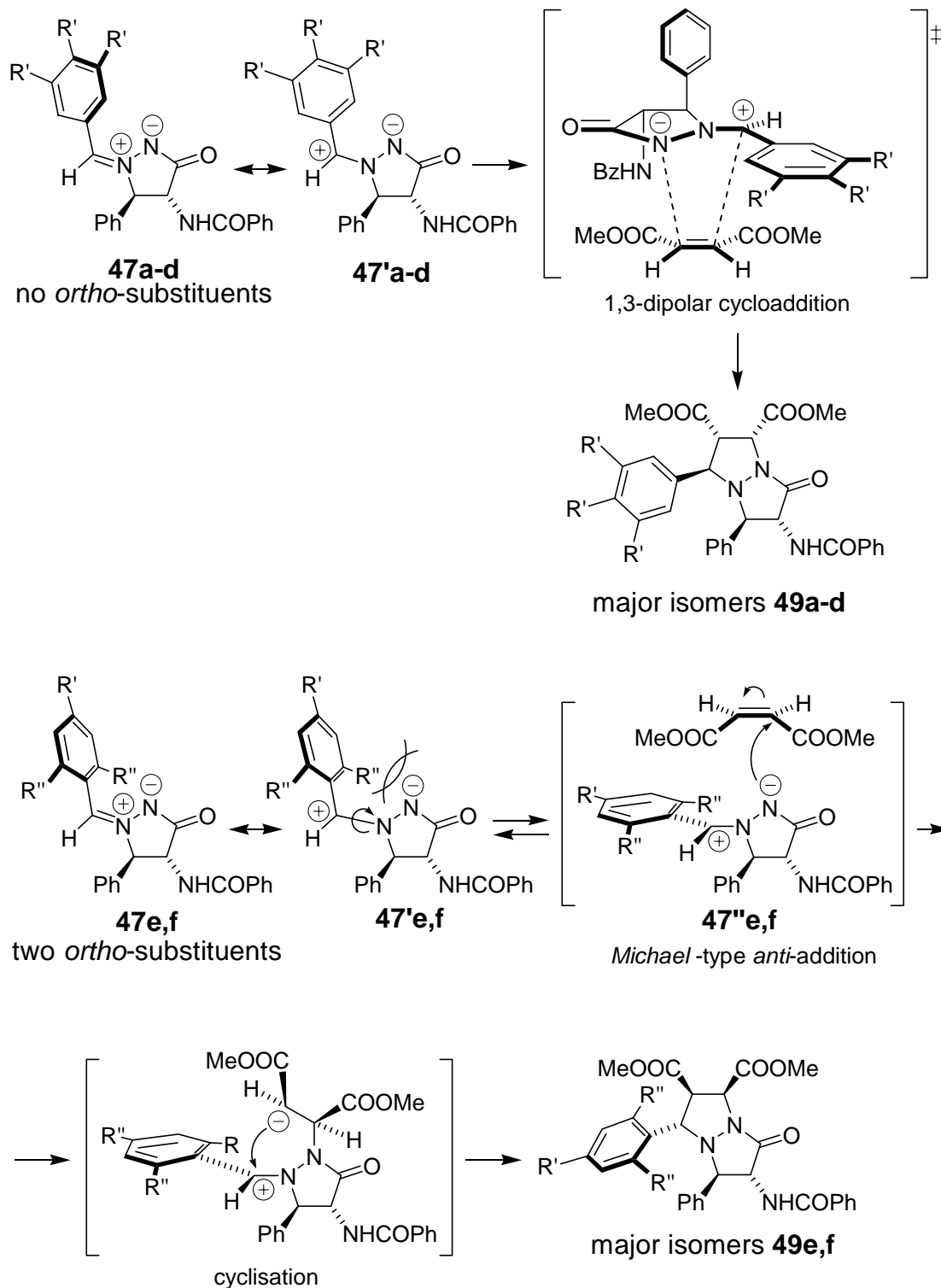


**Scheme 10.** (i) Anisole, reflux.

Stereochemistry of these cycloadditions, which is apparently controlled by the stereodirecting phenyl group at position 3, as well as by the *ortho*-substituents at the aromatic ring, might be summarized in the following way:

- ortho*-unsubstituted dipoles favoured formation of the major isomers with *syn*-oriented *H*-C(3) and *H*-C(5), while *ortho*-disubstituted dipoles favoured formation of the major isomers with *anti*-oriented *H*-C(3) and *H*-C(5),
- in all major isomers with a stereocenter at position 6, the *H*-C(5) and *H*-C(6) were always *trans*-oriented,
- cycloadditions to dimethyl maleate and dimethyl acetylenedicarboxylate were always stereoselective,
- cycloaddition to dimethyl fumarate was stereoselective only in the case of two *ortho*-substituents, and
- cycloadditions to methyl acrylate were selective only in the case of no *ortho*-substituents.

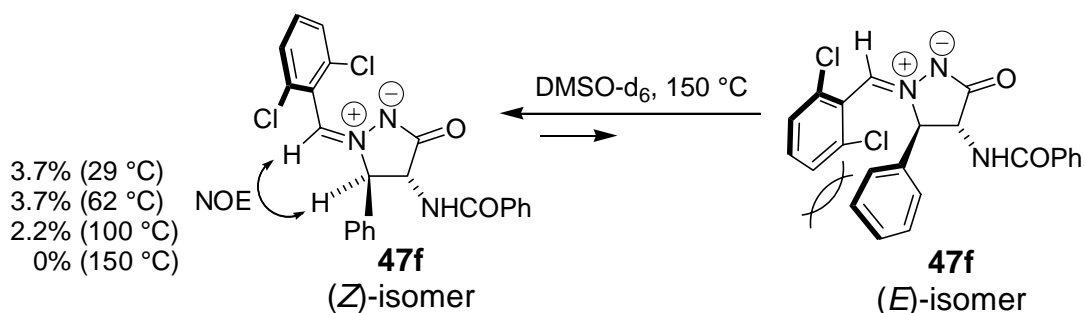
Possible explanation for different selectivity might be exemplified at best by cycloadditions of **47a-f** to dimethyl maleate. Dipoles **47a-d** with free *ortho*-positions in the aromatic ring can adopt the planar conformation **47'a-d** allowing transition state for the concerted 1,3-dipolar cycloaddition. Formation of the compounds **49a-d** could be explained by preferential *endo*-approach of dipolarophile from the less hindered face of the (1*Z*,4*R*\*,5*R*\*)-dipole. On the other hand, such planar conformation is not possible in the case of dipoles **47e,f** with two *ortho*-substituents. Alternatively, stereoselective formation of **49e,f** might be explained by a two-step 1,4-addition-cyclization mechanism.<sup>52</sup> In the mesomeric structures **47'e,f**, rotation around the N(1)-C(1') single bond gives the rotamers **47''e,f** with the bulky aryl group twisted away from the phenyl ring at position 3. Conformers **47''e,f** can undergo *Michael*-type *anti*-addition to the dipolarophile to form the intermediate zwitterions (or a biradicals),<sup>52</sup> which cyclise into the final products **49e,f** (Scheme 11).<sup>51</sup>



Scheme 11

Stereoselective formation of compounds **49e,f** could also be in agreement with the *exo*-approach of the dipolarophile from the less hindered face of the (1*E*,4*R*<sup>\*</sup>,5*R*<sup>\*</sup>)-dipoles **47e,f**.

However, this explanation does not seem suitable, since both, (*Z*)- and (*E*)-planar conformation of dipoles **47e,f** would be sterically unfavourable due to two *ortho*-substituents and because *Z/E*-isomerization of dipoles **47e,f** at 150 °C would consequently lead to a mixture of isomeric cycloadducts. In order to determine the possible *Z/E*-isomerisation, <sup>1</sup>H NMR and NOESY spectra of azomethine imine **47f** were recorded at 29, 62, 100, and 150 °C. Only one set of signals, observed in <sup>1</sup>H NMR spectra even at 150 °C, was in agreement with retention of the (*Z*)-configuration. On the other hand, decreasing NOE between 1'-H and 5-H did not exclude the possibility of *Z/E*-isomerisation (Scheme 12).<sup>51</sup>

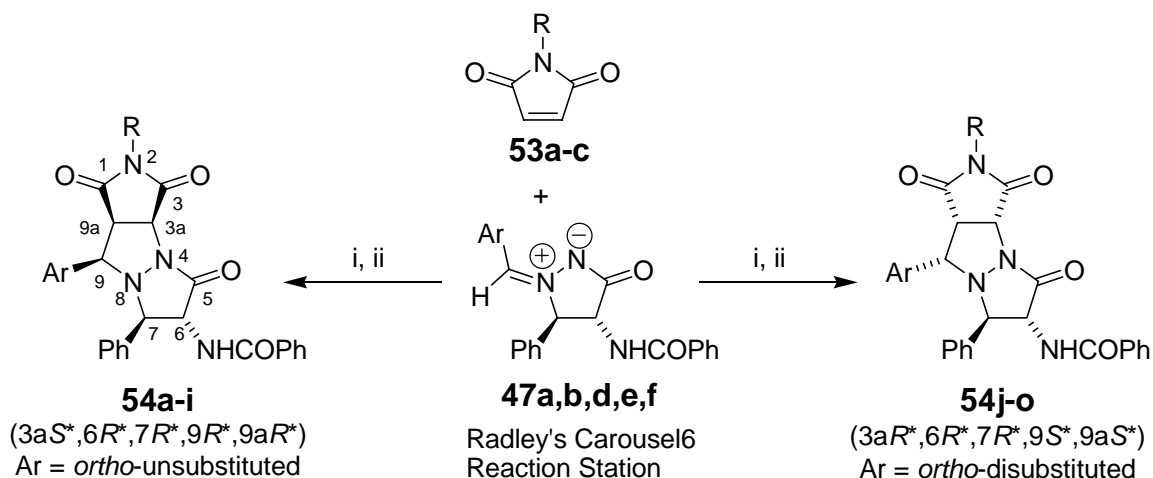


**Scheme 12**

### 3.2 Combinatorial synthesis of 5*H*-pyrazolo[1,2-*a*]pyrrolo[3,4-*c*]pyrazole-1,3,5(2*H*,3*aH*)-triones

In continuation, we focused our attention also on combinatorial studies. Since the reaction and isolation conditions for azomethine imines **47** are always the same (the dipoles precipitate from the reaction mixture), a series of azomethine imines **47** has recently been prepared by the parallel solution-phase approach and isolated simply by filtration, washing, and drying. In extension, the solution-phase approach was applied for the combinatorial synthesis of 15 tetrahydro-5*H*-pyrazolo[1,2-*a*]pyrrolo[3,4-*c*]pyrazole-1,3,5(2*H*,3*aH*)-triones **54a–o** by reacting five azomethine imines **47a,b,d,e,f** with three maleimides **53a–c**. Maleimides were chosen, since they are also, like dimethyl maleate, the *cisoid*-dipolarophiles and cycloadditions were expected to proceed stereoselectively regardless to the *ortho*-substituents at the aromatic ring (*c.f.* Scheme 9). Upon heating in anisole followed by cooling, evaporation, trituration with ether, filtration, washing, and thorough drying, all products **54a–i** and **54j–o** were isolated in analytically pure form in 18–89% yields.<sup>53</sup> According to expectations, two stereochemical types of cycloadducts, **54a–i** and **54j–o**, were formed, depending on *ortho*-substituents at the aromatic ring. Surprisingly, recent NMR and X-ray structural determinations showed, that the configurations at positions 6, 7 and 9 were in agreement with the previously established stereocontrol, while the configurations at positions 3*a* and 9*a* were not.<sup>51</sup> This might be due to possible isomerisation at positions 3*a* and 9*a* in cycloadduct **54** or/and due to different steric demand of the dipolarophile **53** in comparison to dimethyl maleate (Scheme 13).<sup>53</sup>





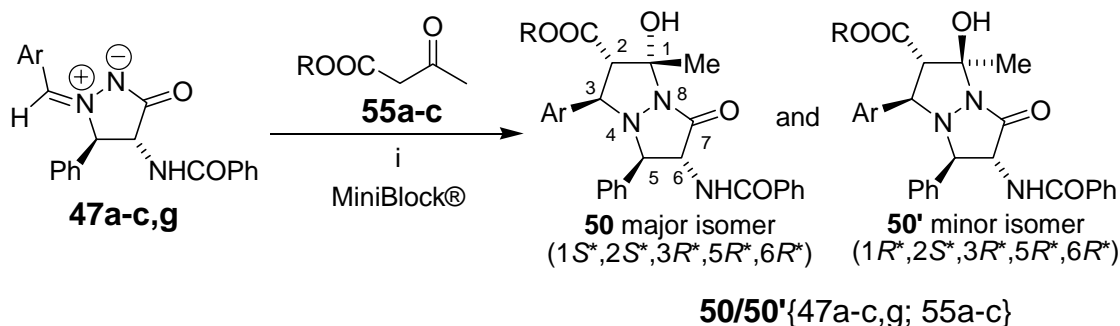
Compound	Ar	R	Yield (%)
(3aS*,6R*,7R*,9R*,9aR*)-isomers			
<b>54a</b>	Ph	Ph	74
<b>54b</b>	4-nitrophenyl	Ph	84
<b>54c</b>	3,4,5-trimethoxyphenyl	Ph	75
<b>54d</b>	Ph	4-methylphenyl	48
<b>54e</b>	4-nitrophenyl	4-methylphenyl	57
<b>54f</b>	3,4,5-trimethoxyphenyl	4-methylphenyl	76
<b>54g</b>	Ph	Me	55
<b>54h</b>	4-nitrophenyl	Me	73
<b>54i</b>	3,4,5-trimethoxyphenyl	Me	85
(3aR*,6R*,7R*,9S*,9aS*)-isomers			
<b>54j</b>	2,4,6-trimethylphenyl	Ph	63
<b>54k</b>	2,6-dichlorophenyl	Ph	82
<b>54l</b>	2,4,6-trimethylphenyl	4-methylphenyl	55
<b>54m</b>	2,6-dichlorophenyl	4-methylphenyl	89
<b>54n</b>	2,4,6-trimethylphenyl	Me	18
<b>54o</b>	2,6-dichlorophenyl	Me	84

**Scheme 13.** (i) maleimide **53a-c**, anisole, reflux; (ii) evaporation, trituration with Et<sub>2</sub>O, filtration, washing with Et<sub>2</sub>O, drying *in vacuo*.

### 3.3 Combinatorial studies on cycloadditions to $\beta$ -keto esters.

Next, we investigated cycloadditions of azomethine imines **47** to  $\beta$ -keto esters **55**,<sup>34</sup> which result in the formation of two epimeric alkyl 3-aryl-6-benzoylamino-1-hydroxy-1-methyl-7-oxo-5-phenyltetrahydro-1*H*,5*H*-pyrazolo[1,2-*a*]pyrazole-2-carboxylates, the major (1*S*\*,2*S*\*,3*R*\*,5*R*\*,6*R*\*)-isomers **50'** and the minor (1*R*\*,2*S*\*,3*R*\*,5*R*\*,6*R*\*)-isomers **50**. During the preliminary studies it turned out, that *ortho*-disubstituted dipoles **47e,f** do not react.

Therefore, we have chosen three model *ortho*-unsubstituted azomethine imines **47a–c** and one model *ortho*-monosubstituted azomethine imine **47g** and nine model  $\beta$ -keto esters **55a–i** for a combinatorial study on cycloaddition reactions. First, 12 reactions were carried out with ethyl (**55a**), benzyl (**55b**), and *tert*-butyl acetoacetate (**55c**). With exception of **50**{**47b**; **55c**}, the products were isolated as mixtures of epimers **50** and **50'**. Phenyl (**47a**) and 4-nitrophenyl (**47b**) substituted dipoles reacted with all three acetoacetates **55a–c**, while 4-methoxyphenyl (**47c**) and 2,4-dichlorophenyl substituted azomethine imine (**47g**) did not react in all cases. On the other hand, *tert*-butyl acetoacetate (**55c**) was the most reactive, since it underwent cycloadditions with all four azomethine imines **47a–c,g**. (Scheme 14).<sup>53</sup>



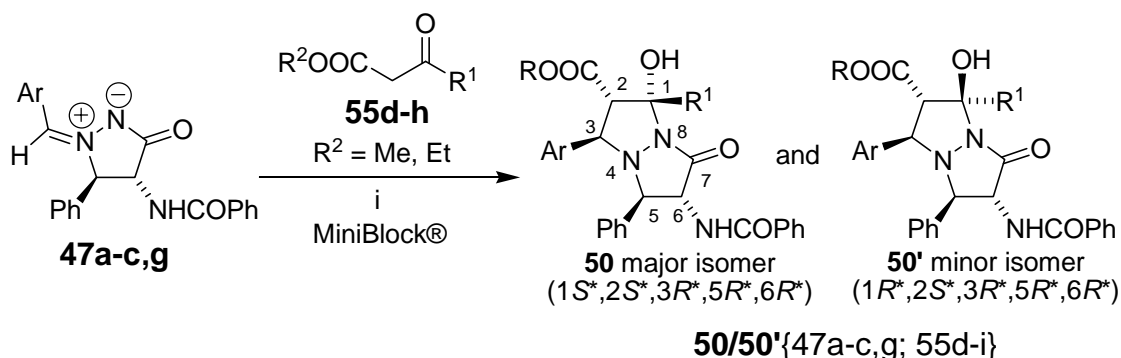
R ( <b>55a–c</b> ) → Ar ( <b>47</b> ) ↓	Yield/D.e. (%)		
	<b>55a</b> Et	<b>55b</b> Bn	<b>55c</b> <i>t</i> -Bu
<b>47a</b> phenyl	76/48	60/14	78/46
<b>47b</b> 4-nitrophenyl	75/62	89/60	86/100
<b>47c</b> 4-methoxyphenyl	<i>n.c.</i>	34/16	80/68
<b>47g</b> 2,4-dichlorophenyl	<i>n.c.</i>	<i>n.c.</i>	66/64

*n.c.*) No conversion detected by TLC.

**Scheme 14.** (i) MeOH, Et<sub>3</sub>N (1 equiv.), r.t..

Finally, 24 cycloadditions were performed with dipoles **47a–c,g** and  $\beta$ -keto esters **55d–i** with variable substituents at the  $\beta$ -position (**55d** (R = Et), **55e** (R = *n*-Pr), **55f** (R = *i*-Rr), **55g** (R = *t*-Bu), **55h** (R = Ph), and **55i** (R = CH<sub>2</sub>COOMe). In contrast to cycloadditions to alkyl acetoacetates **55a–c**, isomerically pure cycloadducts were isolated upon reactions of **47a–c,g** with  $\beta$ -keto esters **55d–i**. An exception was compound **50/50'**{**47a**; **55d**}, which was obtained in 72% d.e. 4-Nitrophenyl substituted azomethine imine **47b** was again the most reactive and gave cycloadducts also with sterically more demanding keto esters **55g–i** (R = *t*-Bu, Ph, CH<sub>2</sub>COOMe), while the other three dipoles **47a,c,g** reacted only with sterically less demanding keto esters **55d–f** (R = Et, *n*-Pr, *i*-Rr). These results also support the indication, that, besides steric factors, the

electronic effects should also be taken into account by planning a combinatorial synthesis of pyrazolo[1,2-*a*]pyrazolone type of peptidomimetics (Scheme 15).<sup>53</sup>



R <sup>1</sup> (55d-i)→ Ar (47a-c,g)↓	Yield/D.e. <sup>a</sup> (%)					
	55d	55e	55f	55g	55h	55i
	Et	<i>n</i> -Pr	<i>i</i> -Pr	<i>t</i> -Bu	Ph	CH <sub>2</sub> COOMe
<b>47a</b> phenyl	60/72	19	32	<i>n.c.</i>	<i>n.c.</i>	76
<b>47b</b> 4-nitrophenyl	85	85	49	41	14	86
<b>47c</b> 4-methoxyphenyl	29	22	43	<i>n.c.</i>	<i>n.c.</i>	<i>n.c.</i>
<b>47g</b> 2,4-dichlorophenyl	45	54	23	<i>n.c.</i>	6	<i>n.c.</i>

*n.c.*) No conversion detected by TLC.

<sup>a</sup>) Unless otherwise stated, the d.e. was 100%.

**Scheme 15.** (i) MeOH, Et<sub>3</sub>N (1 equiv.), r.t..

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## References and Notes

1. For recent reviews see: (a) Elguero, J. *Pyrazoles in Comprehensive Heterocyclic Chemistry II*; A. R. Katritzky, C. W. Rees, E. F. V. Scriven; Eds.; Vol 3; Elsevier Science Ltd., Oxford, 1996, pp. 1–75. (b) Varvounis, G.; Fiamegos, Y.; Pilidis, G. *Adv. Heterocycl. Chem.* **2001**, *80*, 75–165. (c) Stanovnik, B.; Svete, J. *Pyrazoles in Science of Synthesis, Houben-Weyl Methods of Molecular Transformations, Vol. 12*; Neier, R.; Ed.; Thieme Verlag: Stuttgart, 2002, pp 15–225.
2. (a) Sugimoto, N.; Watanabe, H.; Ide, A. *Tetrahedron* **1960**, *11*, 231. (b) Finar, I. L.; Utting, K. *J. Chem. Soc.* **1960**, 5272.
3. Buchanan, J. G.; Stobie, A.; Wightman, R. H. *J. Chem. Soc.* **1981**, 2374.
4. Morimoto, A.; Noda, K.; Watanabe, T.; Takasugi, H. *Tetrahedron Lett.* **1968**, *9*, 5707.
5. (a) Mulhall, J. *British journal of urology* **1997**, *79*, 363. (b) Osterloh, I. H. *The discovery and development of Viagra (sildenafil citrate) in Sildenafil*; Dunzendorfer, U.; Ed.; Birkhaeuser Verlag: Basel, 2004, pp 1–13.
6. (a) Rainer, G.; Kruger, U.; Klemm, K. *Arzneimittel-Forschung* **1981**, *31*, 649. (b) Goertz, R.; Appelboom, T. *Int. Tissue React.i* **1985**, *7*, 263. (c) Vinge, E.; Bjorkman, S. B. *Acta Pharmacol. Toxicol.* **1986**, *59*, 165.
7. (a) Kameyama T.; Nabeshima T. *Neuropharmacology* **1978**, *17*, 249. (b) Kameyama, T.; Ukai, M.; Nabeshima, T. *Chem. Pharm. Bull.* **1978**, *26*, 3265. (c) Yoshida, N.; Nabeshima, T.; Yamaguchi, K. *Neurosciences* **1979**, *5*, 54.
8. (a) Naito, T.; Yoshikawa, T.; Kitahara, S.; Aoki, N. *Chem. Pharm. Bull.* **1969**, *17*, 1467. (b) Oshima, Y.; Akimoto, T.; Tsukada, W.; Yamasaki, T.; Yamaguchi, K.; Kojima, H. *Chem. Pharm. Bull.* **1969**, *17*, 1492. (c) Takabatake, E.; Kodama, R.; Tanaka, Y.; Dohmori, R.; Tachizawa, H.; Naito, T. *Chem. Pharm. Bull.* **1970**, *18*, 1900.
9. (a) Heijnen, H.; Geuze, H. *Histochemistry* **1977**, *54*, 39. (b) Blackwell, G. J.; Flower, R. J. *Prostaglandins* **1978**, *16*, 417. (c) Bragt, P. C.; Bansberg, J. I.; Bonta, I. L. *Inflammation* **1980**, *4*, 289.
10. (a) Jungheim, L. N.; Sigmund, S. K.; Fisher, J. W. *Tetrahedron Lett.* **1987**, *28*, 285. (b) Jungheim, L. N.; Sigmund, S. K. *J. Org. Chem.* **1987**, *52*, 4007. (c) Indelicato, J. M.; Pasini, C. E. *J. Med. Chem.* **1988**, *31*, 1227. (d) Ternansky, R. J.; Draheim, S. E. *Tetrahedron Lett.* **1990**, *31*, 2805. (e) Holmes, R. E.; Neel, D. A. *Tetrahedron Lett.* **1990**, *31*, 5567. (f) Allen, N. E.; Hobbs, J. N., Jr.; Preston, D. A.; Turner, J. R.; Wu, C. Y. E. *J. Antibiot.* **1990**, *43*, 92.
11. For recent reviews see: (a) Svete, J. *J. Heterocycl. Chem.* **2005**, *42*, 361. (b) Svete, J. *Monatsh. Chem.* **2004**, *135*, 629. (c) Svete, J. *J. Heterocycl. Chem.* **2002**, *39*, 437.
12. For recent reviews see: (a) Stanovnik, B.; Svete, J. *Chem. Rev.* **2004**, *104*, 2433–2480. (b) Stanovnik, B.; Svete, J. *Synlett* **2000**, 1077–1091. (c) Stanovnik, B.; Svete, J. *Targets in Heterocyclic Systems* **2000**, *4*, 105–137; (d) Stanovnik, B. *J. Heterocycl. Chem.* **1999**, *36*, 1581–1593.
13. (a) Stanovnik, B.; Svete, J. *Mini-Reviews in Organic Chemistry* **2005**, *2*, 211–224; (b) Časar, Z.; Bevk, D.; Svete, J.; Stanovnik, B. *Tetrahedron* **2005**, *61*, 7508.

14. (a) Pirc, S.; Bevk, D.; Golič Grdadolnik, S.; Svete, J. *ARKIVOC*, **2003** (xiv), 37. (b) Westman, J.; Lundin, R. *Synthesis* **2003**, 1025. (c) Čebašek, P.; Wagger, J.; Bevk, D.; Jakše, R.; Svete, J.; Stanovnik, B. *J. Comb. Chem.* **2004**, 6, 356.
15. (a) Bratušek, U.; Hvala, A.; Stanovnik, B. *J. Heterocycl. Chem.* **1998**, 35, 971. (b) Bevk, D.; Kmetič, M.; Rečnik, S.; Svete, J.; Golič, L.; Golobič, A.; Stanovnik, B. *Chem. Heterocycl. Compd.* **2001**, 1651.
16. Grošelj, U.; Rečnik, S.; Svete, J.; Meden, A.; Stanovnik, B.; *Tetrahedron: Asymmetry* **2002**, 13, 821.
17. Hanzlowsky, A.; Jeleničič, B.; Rečnik, S.; Svete, J.; Golobič, A.; Stanovnik, B. *J. Heterocycl. Chem.* **2003**, 40, 487.
18. (a) Bowler, A. N.; Doyle, P. M.; Young, D. W. *J. Chem. Soc., Chem. Commun.* **1991**, 314. (b) Dinsmore, A.; Doyle, P. M.; Young, D. W. *Tetrahedron Lett.* **1995**, 36, 7503. (c) Dinsmore, A.; Doyle, P. M.; Young, D. W. *J. Chem. Soc., Perkin Trans. 1*, **1997**, 1297.
19. Škof, M.; Svete, J.; Stanovnik, B. *Heterocycles* **2000**, 53, 339.
20. Mihelič, D.; Jakše, R.; Svete, J.; Stanovnik, B. *J. Heterocycl. Chem.* **2001**, 38, 1307.
21. Čebašek, P.; Stanovnik, B.; Svete, J.; to be published.
22. (a) Škof, M.; Svete, J.; Stanovnik, B.; Golič, L.; Golič Grdadolnik, S.; Selič, L. *Helv. Chim. Acta* **1998**, 81, 2332. (b) Pirc, S.; Rečnik, S.; Škof, M.; Svete, J.; Golič, L.; Meden, A.; Stanovnik, B. *J. Heterocycl. Chem.* **2002**, 39, 411. (c) Škof, M.; Pirc, S.; Rečnik, S.; Svete, J.; Stanovnik, B.; Golič, L.; Selič, L. *J. Heterocycl. Chem.* **2002**, 39, 957.
23. Grošelj, U.; Drobnič, A.; Rečnik, S.; Svete, J.; Stanovnik, B.; Golobič, A.; Lah, N.; Leban, I.; Golič Grdadolnik, S. *Helv. Chim. Acta* **2001**, 84, 3403.
24. Grošelj, U.; Bevk, D.; Jakše, R.; Meden, A.; Pirc, S.; Rečnik, S.; Stanovnik, B.; Svete, J. *Tetrahedron: Asymmetry* **2004**, 15, 2367.
25. Grošelj, U.; Bevk, D.; Jakše, R.; Rečnik, S.; Meden, A.; Stanovnik, B.; Svete, J. *Tetrahedron* **2005**, 61, 3991.
26. Bergmann, M.; Zervas, L. *Chem. Ber.* **1932**, 65, 1192.
27. Li, B.; Franck, R. W. *Bioorg. Med. Chem. Lett.* **1999**, 9, 2629.
28. Uršič, U.; Stanovnik, B.; Svete, J.; submitted for publication.
29. (a) Saari, W. S.; Fisher, T. E. *Synthesis*, **1990**, 453. (b) Sibi, M. P. *Org. Prep. Proc. Int.* **1993**, 25, 15. (c) De Luca, L.; Falorini, M.; Giacomelli, L.; Porcheddu, A.; *Tetrahedron Lett.* **1999**, 40, 8701. (d) De Luca, L.; Giacomelli, G.; Porcheddu, A.; Spannedda, A. M.; Falorini, M. *Synthesis*, **2000**, 1295. (e) Evans, D. A.; Hu, E.; Tedrow, J. S. *Org. Lett.*, **2001**, 3, 3133.
30. Pirc, S.; Bevk, D.; Golobič, A.; Stanovnik, B.; Svete, J. *Helv. Chim. Acta*; in press.
31. (a) *Synthetic Applications of 1,3-Dipolar Cycloaddition Chemistry toward Heterocycles and Natural Products*; Padwa, A.; Pearson, W. H.; Eds.; John Wiley & Sons, Inc.: Hoboken, New Jersey, 2003. (b) *1,3-Dipolar Cycloaddition Chemistry*; Padwa, A.; Ed.; Vol 1; John Wiley & Sons, Inc.: New York, 1984.
32. Gothelf, V. K.; Jørgensen, K. A. *Asymmetric Reactions in Synthetic Applications of 1,3-Dipolar Cycloaddition Chemistry toward Heterocycles and Natural Products*; Padwa, A.; Pearson, W. H.; Eds.; John Wiley & Sons, Inc.: Hoboken, New Jersey, 2003, pp 817–899.

33. (a) Stanovnik, B. *Tetrahedron* **1991**, *47*, 2925. (b) Žličar, M.; Stanovnik, B.; Tišler, M. *Tetrahedron* **1992**, *48*, 7965. (c) Žličar, M.; Stanovnik, B.; Tišler, M. *J. Heterocycl. Chem.* **1993**, *30*, 1209. (d) Stanovnik, B.; Jelen, B.; Žličar, M. *Il Farmaco* **1993**, *48*, 231. (e) Stanovnik, B.; Jelen, B.; Turk, C.; Žličar, M.; Svete, J. *J. Heterocycl. Chem.* **1998**, *35*, 1187.
34. Svete, J.; Prešeren, A.; Stanovnik, B.; Golič, L.; Golič Grdadolnik, S. *J. Heterocycl. Chem.* **1997**, *34*, 1323.
35. Roussi, F.; Chauveau, A.; Bonin, M.; Micouin, L.; Husson, H.-P. *Synthesis*, **2000**, 1170.
36. Chuang, T.-H.; Sharpless, K. B. *Helv. Chim. Acta* **2000**, *83*, 1734.
37. Chauveau, A.; Martens, T.; Bonin, M.; Micouin, L.; Husson, H.-P. *Synthesis* **2002**, 1885.
38. Dorn, H. *Chem. Heterocycl. Compd. USSR* **1981**, 3.
39. Claramunt, R. M.; Elguero, J. *Org. Proc. Prep. Int.* **1991**, *23*, 273.
40. Hanessian, S.; McNaughton-Smith, G.; Lombart, H.-G.; Lubell, W. D. *Tetrahedron*, **1997**, *53*, 12789. and references cited therein.
41. (a) Dorn, H.; Otto, A. *Chem. Ber.* **1968**, *101*, 3287. (b) Dorn, H.; Ozegowski, R.; Gründemann, E. *J. Prakt. Chem.* **1979**, *321*, 565. (c) Dorn, H. *Tetrahedron Lett.* **1985**, *26*, 5123.
42. (a) Oppolzer, W. *Tetrahedron Lett.* **1970**, *15*, 2199. (b) Oppolzer, W. *Tetrahedron Lett.* **1972**, *17*, 1707.
43. Kim, K. S.; Ryan, P. *Heterocycles*, **1990**, *31*, 79.
44. (a) Schantl, J. *Azomethine Imines In Science of Synthesis, Houben-Weyl Methods of Molecular Transformations, Vol. 27*; Padwa, A.; Ed.; Thieme Verlag: Stuttgart, 2004, pp 731–824; (b) Grashey, R. *Azomethine Imines In 1,3-Dipolar Cycloaddition Chemistry*; Padwa, A.; Ed.; Vol 1; John Wiley & Sons, Inc.: New York, 1984, pp 733–814.
45. Prešeren, A.; Svete, J.; Stanovnik, B. *J. Heterocycl. Chem.* **1999**, *36*, 799.
46. Svete, J.; Stanovnik, B. *Heterocycles* **1999**, *51*, 2073.
47. Zupančič, S.; Svete, J.; Stanovnik, B. *J. Heterocycl. Chem.* **1999**, *36*, 607.
48. Turk, C.; Svete, J.; Stanovnik, B.; Golič, L.; Golobič, A.; Selič, L. *Org. Lett.* **2000**, *2*, 423.
49. Turk, C.; Svete, J.; Stanovnik, B.; Golič, L.; Golič Grdadolnik, S.; Golobič, A.; Selič, L. *Helv. Chim. Acta* **2001**, *84*, 146.
50. Turk, C.; Golič, L.; Selič, L.; Svete, J.; Stanovnik, B. *ARKIVOC* **2001** (v), 87.
51. Pezdirc, L.; Jovanovski, V.; Bevk, D.; Jakše, R.; Pirc, S.; Meden, A.; Stanovnik, B.; Svete, J. *Tetrahedron* **2005**, *61*, 3977.
52. (a) Huisgen, R. *1,3-Dipolar Cycloadditions – Introduction, Survey, Mechanism in 1,3-Dipolar Cycloaddition Chemistry*; Padwa, A.; Ed.; Vol 1; John Wiley & Sons, Inc.: New York, 1984, pp 1–176. (b) Houk, K. N.; Yamaguchi, K. *Theory of 1,3-Dipolar Cycloadditions in 1,3-Dipolar Cycloaddition Chemistry*; Padwa, A.; Ed.; Vol 2; John Wiley & Sons, Inc.: New York, 1984, pp 407–450.
53. Pezdirc, L.; Bevk, D.; Stanovnik, B.; Svete, J.; to be published.