

Bicyclo[2.2.2]octane analogues of patchouli alcohol by Sakurai reaction and Nagata cyclization. Synthesis and olfactory properties of novel isopropyl derivatives[†]

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Dedicated to Prof. Csaba Szántay on his 80th birthday

Abstract

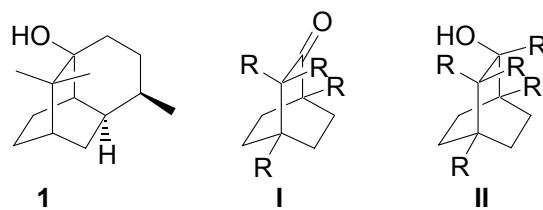
The synthesis of bicyclo[2.2.2]octane *patchouli alcohol* analogues by the Sakurai conjugate addition and Nagata cyclization is described. By this approach, complementary to those so far adopted and based on the Diels-Alder addition, known analogues **2**, **3** and **20** and new analogues **8-11**, with 1-isopropylbicyclo[2.2.2]octane structure, could be obtained. The olfactory properties of **8** and **10** were also evaluated.

Keywords: 1-isopropylbicyclo[2.2.2]octane derivatives, synthesis, Sakurai allylation, Nagata cyclization, *patchouli alcohol* analogues, olfactory properties

Introduction

The olfactory properties of *patchouli alcohol* **1**, a sesquiterpenoid largely available from natural sources, are well known. Since total synthesis² has proven uneconomical, a systematic search for synthetic analogues with simpler structures **I** (R=H, alkyl, alkenyl) and **II** (R=H, alkyl, alkenyl) has been carried out by Spreitzer³ and Weyerstahl.⁴

[†] The work described in this paper constitutes part of the Ph.D. Thesis of A.L.B.¹

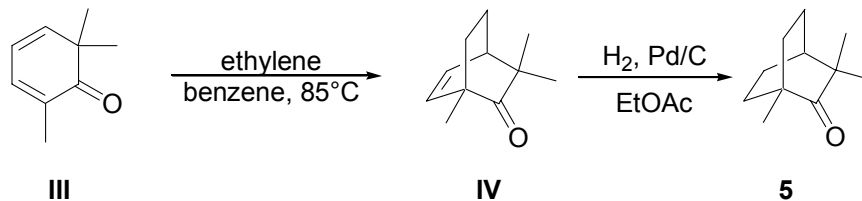


Some compounds of type **I** and **II** display olfactory properties similar to those of **1**.⁴ A general requisite for patchouli alcohol-like olfactory properties is a 13-15 C-atoms skeleton.⁵ In the case of analogues of type **II**, another requisite is that the “hydroxyl group should be sterically shielded by a methyl or another group to a large extent but not completely”.⁶



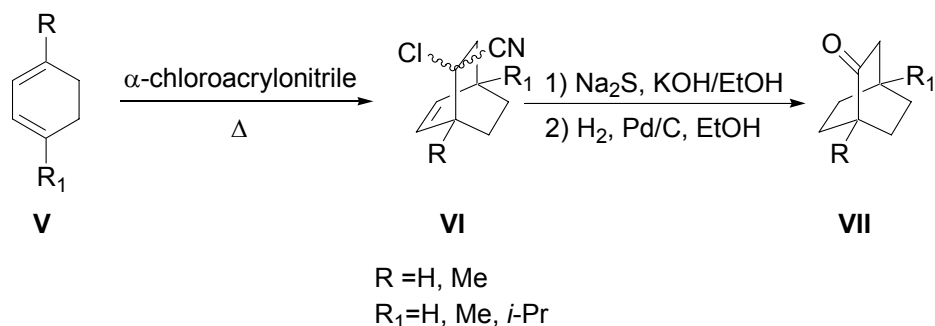
- | | |
|---|--|
| 2 R ₁ =H, R ₂ =H, R ₃ =H, R ₄ =H | 6 R ₁ =H, R ₂ =H, R ₃ =Me, R ₄ =Me |
| 3 R ₁ =Me, R ₂ =Me, R ₃ =H, R ₄ =H | 7 R ₁ =Me, R ₂ =Me, R ₃ =Me, R ₄ =Me |
| 4 R ₁ =Me, R ₂ = <i>i</i> -Pr, R ₃ =H, R ₄ =H | 10 R ₁ = <i>i</i> -Pr, R ₂ =H, R ₃ =Me, R ₄ =Me |
| 5 R ₁ =Me, R ₂ =H, R ₃ =Me, R ₄ =Me | 11 R ₁ = <i>i</i> -Pr, R ₂ =H, R ₃ =H, R ₄ =H |
| 8 R ₁ = <i>i</i> -Pr, R ₂ =H, R ₃ =H, R ₄ =H | |
| 9 R ₁ = <i>i</i> -Pr, R ₂ =H, R ₃ =Me, R ₄ =Me | |

The key intermediate in the Spreitzer approach was the bicyclo[2.2.2]octan-2-one **5**, obtained by catalytic hydrogenation of the Diels-Alder addition product **IV** of ethylene to the unsymmetrical activated diene **III** (2,6,6-trimethylcyclohexadienone). By standard steps **5** was converted into analogues of type **II** (Scheme 1).^{3b}



Scheme 1

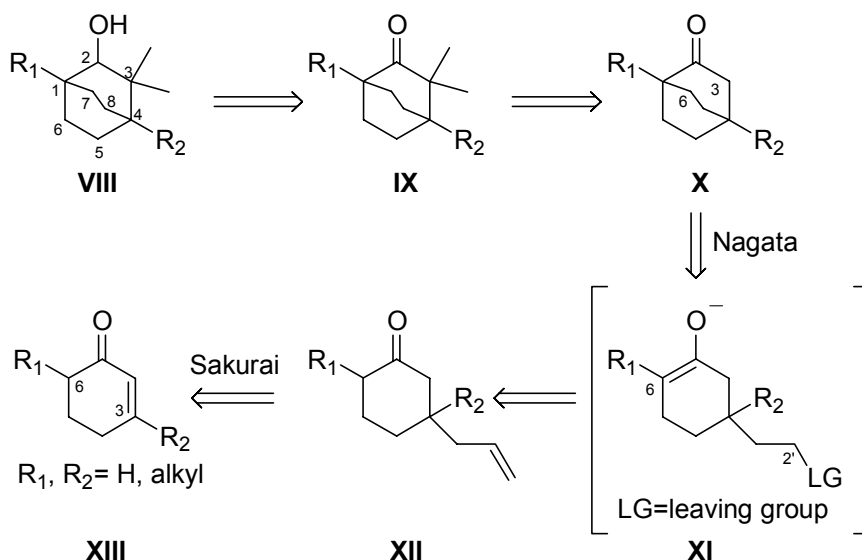
Weyerstahl obtained intermediates **I** *via* catalytic hydrogenation of the Diels-Alder addition product **VI** of an activated unsymmetrical dienophile (α -chloroacrylonitrile) to symmetrically 1,4-disubstituted unactivated dienes (cyclohexadiene or 1,4-dimethylcyclohexadiene) or to readily available α -terpinene (Scheme 2).⁴



Scheme 2

Steric, regiochemical and electronic restrictions of the Diels–Alder reaction as well as the availability of suitable dienes limit the versatility of this approach and the number of analogues **I** and **II** of *patchouli alcohol* obtainable.

In our studies for the synthesis of natural products containing bicyclo[2.2.2]octane systems or *via* intermediates of this type,⁷ we have developed a synthetic approach to *patchouli alcohol* analogues, complementary to those so far adopted,^{3,4} and based on the Nagata 3-sulfonyloxyethylcyclohexanone cyclization⁸ and the Sakurai cyclohex-2-en-1-one conjugate addition⁹ (Scheme 3).



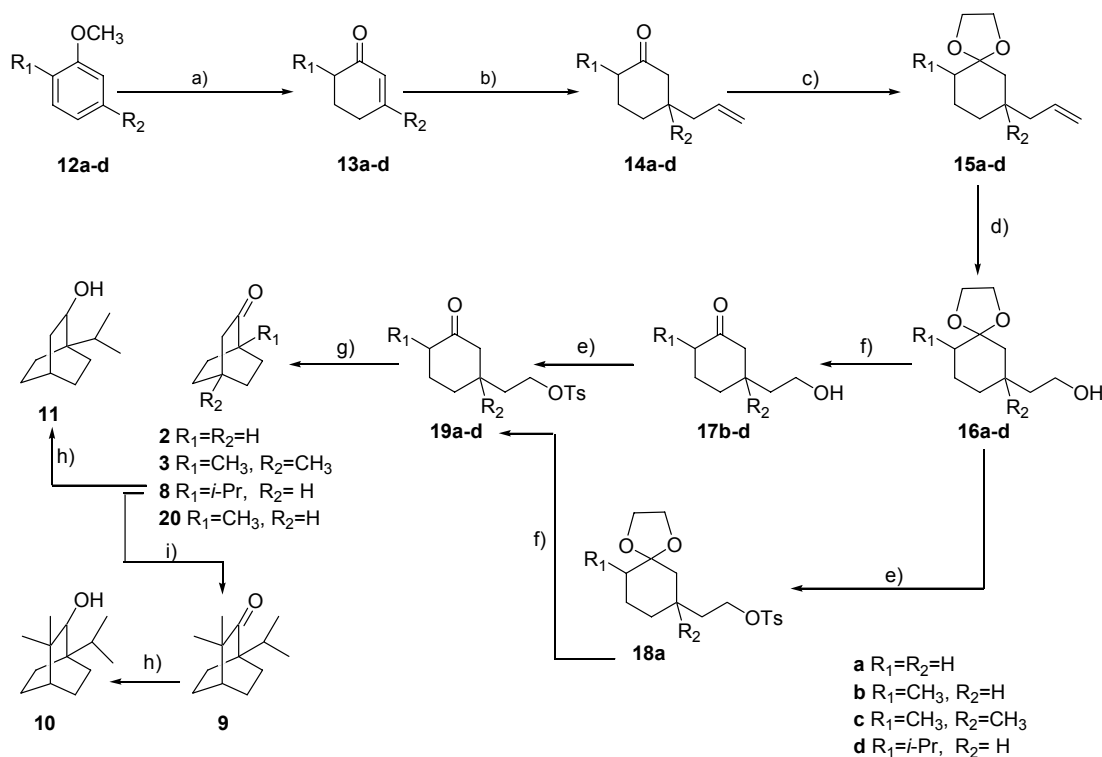
Scheme 3

The targets we selected were the known **2**⁴, **3**⁴, **20**¹⁰ and the novel 1-isopropylbicyclo[2.2.2]octan-2-one **8**. Compound **8** was selected since its C(4) homologue **4** could be obtained only in trace amounts by the Diels–Alder approach, owing to the “*strong steric influence of the bulky isopropyl group*”.⁴ In addition 1-isopropylbicyclo[2.2.2]octan-2-one **8** can

be transformed into **9**, a new analogue of type **I**, and into **10** and **11**, new analogues of type **II**. Thus information on the effect of a bulky alkyl group at C(1) on the olfactory properties of analogues of type **I** and **II** could be obtained.

Results and Discussion

The starting materials for this study (Scheme 4) were commercially available anisoles **12** which were converted into α,β -unsaturated ketones **13** by Birch reduction followed by acidic hydrolysis.

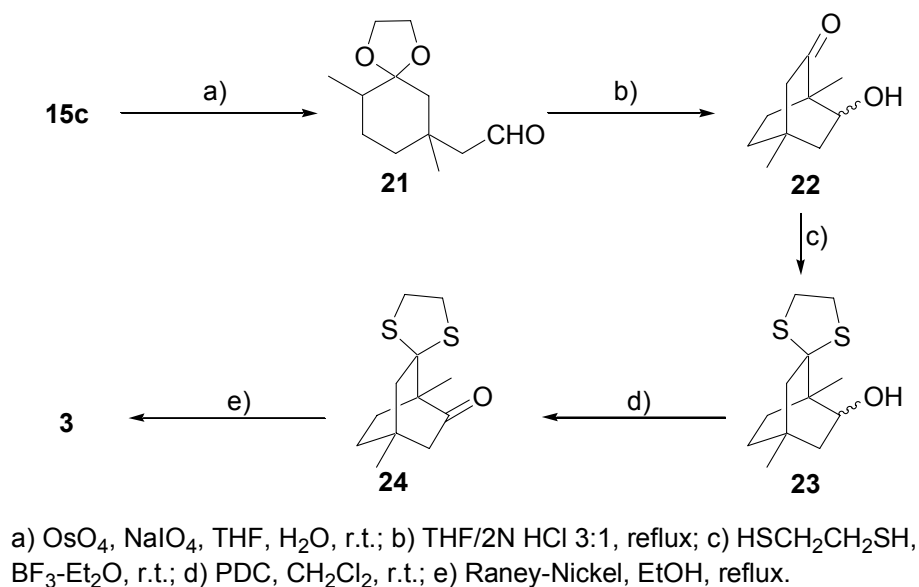


a) i) Li, NH₃, *t*-BuOH, THF, -50°C; ii) HCl; b) TiCl₄, CH₂Cl₂, Me₃SiCH₂CH=CH₂, -78°C, Ar; c) HO(CH₂)₂OH, benzene, TsOH, reflux; d) i) O₃, CH₂Cl₂, -78°C; ii) NaBH₄, MeOH, r.t.; e) TsCl, Py, r.t.; f) THF/1N HCl 4/1, r.t.; g) *t*-BuOH/*t*-BuOK, 0°C; h) LiAlH₄, THF, r.t.; i) CH₃I, NaH, THF, reflux.

Scheme 4

The latter were allowed to react according to Sakurai⁹ with allyltrimethylsilane in the presence of TiCl₄ to give **14**. Protection of the carbonyl function of **14** as ethylene glycol acetal gave then **15**. The side chain double bond was cleaved with O₃/NaBH₄ to give **16**, which on treatment with 1N HCl/THF gave **17**. The latter were converted into tosylates **19**. In the case of **16a** the transformation into **19a** was also achieved by tosylation of **16a** to **18a**, which was then

deprotected giving **19a**. Exposure of tosylates **19** to *t*-BuOK in *t*-BuOH gave **2**, **3**, **8** and **20**. Previously¹ compound **3** had been obtained from **15c** as reported in Scheme 5.



Scheme 5

Compound **8** was also converted with MeI and NaH into the highly volatile gem-dimethylated compound **9** which could not be isolated. It was therefore reduced with LiAlH₄ to **10**. LiAlH₄ reduction of **8** gave then **11**.

Evaluation of olfactory properties.

The evaluation of olfactory properties requires a rather large amount of material. Thus only compounds **8** (type **I**) and **10** (type **II**) were subjected to olfactory evaluation.

Evaluation of analogue **8** revealed a scent reminiscent of eucalyptol and camphor, with the earthy-fruity part of *patchouli* oil. Thus the presence of the isopropyl group at C(1) appears to be sufficient for maintaining the earthy-fruity note of *patchouli* fragrance. In contrast, previously prepared analogues of type **I** having such olfactory properties were substituted at C(1), C(3) and C(4).⁴

The 13 C skeleton analogue **10** gave an earthy, mouldy and harsh odour with a technical and solvent-like note in olfactory evaluation. The HO-C(2) shielding by the isopropyl group at C(1) and by the two methyl groups at C(3) seems therefore to be responsible for the lack of the *patchouli alcohol* note in **10** (see prerequisites noted in the Introduction).

Conclusions

In conclusion, by preparing known analogues **2**, **3** and **20** and new analogues **8-11**, we have shown that the approach based on the Sakurai cyclohex-2-en-1-one conjugate addition and on the Nagata 3-sulfonyloxyethylcyclohexanone cyclization is quite convenient for the preparation of *patchouli alcohol* analogues of type **I** and **II**. Thus optically active *patchouli alcohol* analogues of type **II** can be prepared by performing the last reductive step with an asymmetric reducing reagent.

This approach could be useful for preparing a number of compounds of type **I** and **II** and in evaluating the influence on the olfactory properties of C(1)-substituents different than H and methyl, thus contributing to the knowledge of structure/odour relationships in this class of compounds, a target which deserves considerable attention and efforts.¹¹

Acknowledgements

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Experimental Section

General Procedure: All solvents were anal. grade. TLC: Merck silica gel 60 F₂₅₄. Column Chromatography (CC): silica gel 60, 70-230 mesh ASTM. IR Spectra: *Shimadzu-470* scanning infrared spectrophotometer; in cm⁻¹. ¹H- and ¹³C NMR: *Varian-Gemini-200*, at 200 and 50 MHz respectively; chemical shifts are on the δ scale and were referenced to residual CDCl₃ (at 7.26 for ¹H and the center line of the triplet at 77.0 for ¹³C NMR); δ in ppm; *J* in Hz. Compounds **12** and **13a** are commercially available; compounds **2**,^{4,12} **3**,^{4,13} **13b**,¹⁴ **13c**,¹⁵ **13d**,¹⁶ **14a**,^{9,17} **14b**,¹⁸ **15a**,¹⁹ **16a**,²⁰ **17b**,²¹ **17d**,²² **18a**,^{20a,23} **19a**,^{8b} **20**,¹⁰ were already described in the literature. The ¹³C-NMR spectra of compounds obtained as not easily separable diastereoisomeric mixtures (**14c**, **15c**, **15d**, **16c**, **16d**, **19b**, **19c**, **19d**) are not reported. Olfactory properties of compounds **8** and **10** were evaluated at *Givaudan Schweiz* AG in a 10% dipropylene glycol (DPG) solution.

5-Allyl-2,5-dimethylcyclohexanone (14c). To a solution of enone **13c** (7.9 g, 63 mmol) in anhydrous CH_2Cl_2 (40 mL), cooled to -78°C , a solution of TiCl_4 (6.8 mL, 63 mmol) in anhydrous CH_2Cl_2 (13 mL) was added dropwise. To the well stirred mixture a solution of allyltrimethylsilane (11 mL, 69 mmol) in anhydrous CH_2Cl_2 (60 mL) was added dropwise. After 1 h the mixture was allowed to warm slowly to -30°C and stirred for 45 min. The reaction was then quenched at 0°C with H_2O and the whole poured into a separatory funnel. The layers were separated, the aqueous was extracted with CH_2Cl_2 (2x50 mL). The combined organic layers were repeatedly washed with sat. NaHCO_3 solution, brine, dried with anhydrous Na_2SO_4 and concentrated at atmospheric pressure distilling off the solvent through a *Vigreux* column. The crude product was then purified by CC (SiO_2 : petroleum ether (40-70 $^\circ$)/ Et_2O : 8.5/1.5) to afford **14c** as an oil (7.8 g, 50 mmol, 75%). Data of **14c**: IR (CCl_4): 1711 ($\nu_{\text{C=O}}$); $^1\text{H-NMR}$ (CDCl_3): 5.86-5.58 (*m*, 1H), 5.07-4.91 (*m*, 2H), 2.34-1.34 (*m*, 9H), 1.01-0.78 (*m*, 6H). $\text{C}_{11}\text{H}_{18}\text{O}$ (166.26); Calc. C: 79.46; H: 10.91%. Found C: 79.28; H: 11.18%.

5-Allyl-2-isopropylcyclohexanone (14d). Compound **14d** was prepared from known **13d** (2.7 g, 20 mmol) as described for **14c** from **13c**. The crude product was then purified by CC (SiO_2 : petroleum ether (40-70 $^\circ$)/ Et_2O : 8.5/1.5) to afford two oily diastereomers. Data of **14d**_{RF<} (0.9 g, 5 mmol, 25%): IR (CCl_4): 1711 ($\nu_{\text{C=O}}$); $^1\text{H-NMR}$ (CDCl_3): 5.84-5.58 (*m*, 1H), 5.08-4.91 (*m*, 2H), 2.35-1.17 (*m*, 11H), 0.92-0.73 (*m*, 6H); $^{13}\text{C-NMR}$ (CDCl_3): 214.0, 135.8, 116.6, 57.3, 45.7, 40.1, 38.9, 27.0, 26.9, 26.7, 20.8, 19.8. $\text{C}_{12}\text{H}_{20}\text{O}$ (180.29); Calc. C: 79.94; H: 11.18 %. Found C: 79.70; H: 11.35%. Data of **14d**_{RF>} (0.9 g, 5 mmol, 25%): IR (CCl_4): 1710 ($\nu_{\text{C=O}}$); $^1\text{H-NMR}$ (CDCl_3): 5.81-5.55 (*m*, 1H), 5.03-4.90 (*m*, 2H), 2.43-1.20 (*m*, 11H), 0.94-0.78 (*m*, 6H); $^{13}\text{C-NMR}$ (CDCl_3): 211.9, 135.6, 116.5, 56.1, 48.4, 41.0, 39.9, 31.3, 27.6, 25.8, 21.0, 18.6. $\text{C}_{12}\text{H}_{20}\text{O}$ (180.29); Calc. C: 79.94; H: 11.18 %. Found C: 79.75; H: 11.50%.

9-Allyl-6,9-dimethyl-1,4-dioxaspiro[4.5]decane (15c). To a solution of ketone **14c** (7.8 g, 50 mmol) in anhydrous benzene (50 mL) an excess of ethylene glycol (0.3 mol) and a catalytic amount of TsOH were added. The mixture was refluxed under Ar with azeotropic removal of H_2O (*Dean-Stark* trap), until the TLC (petroleum ether (40-70 $^\circ\text{C}$)/ Et_2O : 8.5/1.5, $R_f(\mathbf{14c}) < R_f(\mathbf{15c})$) indicated the complete disappearance of the starting material. The reaction mixture was then cooled to r.t., diluted with Et_2O , and washed with sat. NaHCO_3 solution till neutral, brine, dried with anhydrous Na_2SO_4 and concentrated at atmospheric pressure distilling off the solvent through a *Vigreux* column. The crude product was then purified by CC (SiO_2 : petroleum ether (40-70 $^\circ$)/ Et_2O : 9/1) affording **15c** as an oil (7.7 g, 36 mmol, 73%). Data of **15c**: $^1\text{H-NMR}$ (CDCl_3): 5.90-5.67 (*m*, 1H), 5.04-4.92 (*m*, 2H), 3.97-3.80 (*m*, 4H), 2.31-1.10 (*m*, 9H), 0.97-0.81 (*m*, 6H). $\text{C}_{13}\text{H}_{22}\text{O}_2$ (210.31); Calc. C: 74.24; H: 10.54%. Found C: 74.03; H: 10.89%.

9-Allyl-6-isopropyl-1,4-dioxaspiro[4.5]decane (15d). Compound **15d** was prepared from **14d** (1.8 g, 10 mmol), as described for **15c** from **14c**. The crude product was then purified by CC (SiO_2 : petroleum ether (40-70 $^\circ$)/ Et_2O : 9/1) affording **15d** as an oil (1.9 g, 8.5 mmol, 85%). Data of **15d**: $^1\text{H-NMR}$ (CDCl_3): 5.88-5.64 (*m*, 1 H), 5.08-4.90 (*m*, 2 H), 4.08-3.77 (*m*, 4H), 2.20-1.19 (*m*, 11 H), 0.95-0.81 (*m*, 6H). $\text{C}_{14}\text{H}_{24}\text{O}_2$ (224.34); Calc. C: 74.95; H: 10.78%. Found C: 75.18; H: 11.13%.

2-(10-Methyl-1,4-dioxaspiro[4.5]dec-7-yl)-ethanol (16b). Compound **15b** (5.5 g, 28 mmol) was dissolved in CH₂Cl₂ (20 mL) and cooled to -78°C; a stream of O₃ was then slowly passed through the solution until a faint blue color persisted. NaBH₄ (2 g, 54 mmol) was then added portionwise, and the mixture stirred for 4 h at -78°C. After evaporation of the solvent under reduced pressure, the residue was taken up with water, neutralized with 5% HCl solution and extracted with CH₂Cl₂. Combined extracts were washed with water, brine, dried with anhydrous Na₂SO₄ and evaporated under reduced pressure. The crude product was then purified by CC (SiO₂: petroleum ether (40-70°)/Et₂O: 6/4) to afford two oily diastereomers. Data of **16b**_{RF>} (0.8 g, 4.2 mmol, 15%): IR (CCl₄): 3635 (ν_{OH}); ¹H-NMR (CDCl₃): 3.97-3.86 (*m*, 4H), 3.73-3.52 (*m*, 2H), 2.08-1.07 (*m*, 11H), 0.89 (*d*, *J*=6.04, 3H); ¹³C-NMR (CDCl₃): 110.8, 64.9, 64.7, 60.9, 44.5, 38.3, 36.7, 34.8, 31.1, 22.7, 10.7. C₁₁H₂₀O₃ (200.27); Calc. C: 65.97; H: 10.07%. Found C: 65.85; H: 10.34%. Data of **16b**_{RF<} (3.4 g, 17 mmol, 60%): IR (CCl₄): 3642 (ν_{OH}); ¹H-NMR (CDCl₃): 3.98-3.82 (*m*, 4H), 3.73 (*t*, *J*=6.87, 2H), 2.09 (*s*, 1H), 1.85-0.87 (*m*, 10H), 0.83 (*d*, *J*=6.41, 3H); ¹³C-NMR (CDCl₃): 110.6, 65.2, 64.8, 60.5, 42.1, 39.7, 39.6, 32.3, 32.1, 31.8, 13.8. C₁₁H₂₀O₃ (200.27); Calc. C: 65.97; H: 10.07%. Found C: 65.78; H: 10.42%.

2-(7,10-Dimethyl-1,4-dioxaspiro[4.5]dec-7-yl)-ethanol (16c). Compound **16c** was prepared from **15c** (7.7 g, 36 mmol), as described for **16b** from **15b**. The crude product was then purified by CC (SiO₂: petroleum ether (40-70°)/Et₂O: 6/4) to afford **16c** as an oil (5.4 g, 25 mmol, 70%). Data of **16c**: IR (CCl₄): 3475 (ν_{OH}); ¹H-NMR (CDCl₃): 3.96-3.75 (*m*, 4H), 3.67-3.54 (*m*, 2H), 2.16 (*s*, 1H), 1.91-1.03 (*m*, 9H), 0.96-0.79 (*m*, 6H). C₁₂H₂₂O₃ (214.30); Calc. C: 67.26; H: 10.35%. Found C: 66.96; H: 10.72%.

2-(10-Isopropyl-1,4-dioxaspiro[4.5]dec-7-yl)-ethanol (16d). Compound **16d** was prepared from **15d** (1.9 g, 8.5 mmol), as described for **16b** from **15b**. The crude product was then purified by CC (SiO₂: petroleum ether (40-70°)/Et₂O: 6/4) to afford **16d** as an oil (1.5 g, 6.5 mmol, 77%). Data of **16d**: IR (CCl₄): 3422 (ν_{OH}); ¹H-NMR (CDCl₃): 4.03-3.84 (*m*, 4H), 3.68-3.60 (*m*, 2H), 2.17-0.96 (*m*, 12H), 0.92-0.79 (*m*, 6H). C₁₃H₂₄O₃ (228.33); Calc. C: 68.38; H: 10.59%. Found C: 68.22; H: 10.81%.

5-(2-Hydroxyethyl)-2,5-dimethylcyclohexanone (17c). A 4:1 THF/1N HCl solution (10 mL) of **16c** (5.4 g, 25 mmol) was stirred at r.t. until TLC analysis (SiO₂; petroleum ether (40-70°)/Et₂O: 1/1; *R_f*(**16c**)>*R_f*(**17c**)) showed the disappearance of the starting material (about 72 h). The reaction mixture was neutralized with a sat. NaHCO₃ solution and diluted with Et₂O; after separation, the aqueous phase was thoroughly extracted with Et₂O and the combined organic extracts were washed with H₂O and brine, dried with anhydrous Na₂SO₄ and evaporated under reduced pressure. The crude product was purified by CC (SiO₂: petroleum ether (40-70°)/Et₂O: 7/3) to afford two oily diastereomers. Data of **17c**_{RF<} (1.5 g, 8.8 mmol, 35 %): IR (CCl₄): 1715 (ν_{C=O}); ¹H-NMR (CDCl₃): 3.71 (*t*, *J*=7.23, 2H), 2.45-1.33 (*m*, 10H), 1.08-0.77 (*m*, 6H); ¹³C-NMR (CDCl₃): 213.0, 58.7, 53.3, 46.7, 44.5, 38.8, 36.6, 31.3, 23.1, 14.3. C₁₀H₁₈O₂ (170.25); Calc. C: 70.55; H: 10.66%. Found C: 70.31; H: 10.90%. Data of **17c**_{RF>} (2.3 g, 14 mmol, 55%): IR (CCl₄): 1713 (ν_{C=O}); ¹H-NMR (CDCl₃): 3.74-3.57 (*m*, 2H), 2.45-1.37 (*m*, 10H), 1.09-0.92 (*m*,

6H); ^{13}C -NMR (CDCl_3): 213.3, 58.9, 53.6, 44.2, 39.9, 38.8, 36.6, 31.2, 28.2, 14.3. $\text{C}_{10}\text{H}_{18}\text{O}_2$ (170.25); Calc. C: 70.55; H: 10.66%. Found C: 70.24; H: 11.03%.

2-(4-Methyl-3-oxocyclohexyl)ethyl-4-methylbenzenesulfonate (19b). To a stirred solution of **17b** (3.6 g, 23 mmol) in pyridine (5 mL) TsCl (4.4 g, 23 mmol) was added. After stirring for 18 h at r.t. H_2O (5 ml) was added, followed, after additional 10 min, by Et_2O (20 mL). The aqueous layer was separated and the organic one washed with 2N HCl, H_2O , sat. NaHCO_3 solution till neutral, brine, dried with anhydrous Na_2SO_4 and evaporated under reduced pressure. The crude product was then purified by CC (SiO_2 : petroleum ether (40-70 $^\circ$)/ Et_2O : 6/4, $R_f(\mathbf{17b}) < R_f(\mathbf{19b})$) to afford **19b** as an oil (6.8 g, 22 mmol, 95%). Data of **19b**: IR (CCl_4): 1715 ($\nu_{\text{C=O}}$); ^1H -NMR (CDCl_3): 8.10-7.56 (*m*, 4H), 4.37-4.26 (*m*, 2H), 2.72 (*s*, 3H), 2.66-1.42 (*m*, 10H), 1.32-1.20 (*m*, 3H).

$\text{C}_{16}\text{H}_{22}\text{SO}_4$ (310.41); Calc. C: 61.91; H: 7.14; S: 10.33%. Found C: 61.72; H: 7.39; S: 10.64%.

2-(1,4-Methyl-3-oxocyclohexyl)ethyl-4-methylbenzenesulfonate (19c). Compound **19c** was prepared from **17c** (3.8 g, 22 mmol) as described for **19b** from **17b**. The crude product was then purified by CC (SiO_2 : petroleum ether (40-70 $^\circ$)/ Et_2O : 6/4) to afford **19c** as an oil (5.2 g, 16 mmol, 73%). Data of **19c**: IR (CCl_4): 1713 ($\nu_{\text{C=O}}$); ^1H -NMR (CDCl_3): 8.09-7.58 (*m*, 4H), 4.43-4.32 (*m*, 2H), 2.74 (*s*, 3H), 2.65-1.53 (*m*, 9H), 1.32-1.07 (*m*, 6H).

$\text{C}_{17}\text{H}_{24}\text{SO}_4$ (324.44); Calc. C: 62.93; H: 7.46; S: 9.88%. Found C: 63.23; H: 7.61; S: 10.11%.

2-(4-Isopropyl-3-oxocyclohexyl)ethyl-4-methylbenzenesulfonate (19d). Compound **19d** was prepared from **17d** (2.1 g, 11 mmol) as described for **19b** from **17b**. The crude product was purified by CC (SiO_2 : petroleum ether (40-70 $^\circ$)/ Et_2O : 6/4) to afford **19d** as an oil (3.3 g, 9.9 mmol, 90%). Data of **19d**: IR (CCl_4): 1712 ($\nu_{\text{C=O}}$); ^1H -NMR (CDCl_3): 7.78-7.32 (*m*, 4H), 4.06-3.99 (*m*, 2H), 2.43 (*s*, 3H), 2.31-1.13 (*m*, 11H), 1.00-0.76 (*m*, 6H).

$\text{C}_{18}\text{H}_{26}\text{SO}_4$ (338.46); Calc. C: 63.87; H: 7.74; S: 9.47%. Found C: 64.01; H: 8.07; S: 9.82 %.

1-Isopropylbicyclo[2.2.2]octan-2-one (8). To a solution of **19d** (3.3 g, 9.9 mmol) in *t*-BuOH (8 mL), *t*-BuO $^-\text{K}^+$ (1.4 g, 12.5 mmol) was added. The mixture was stirred at r.t. until TLC (petroleum ether (40-70 $^\circ$)/ Et_2O : 1/1, $R_f(\mathbf{19d}) < R_f(\mathbf{8})$) showed the complete disappearance of the starting material (1 h). After careful neutralization with 0.1N HCl, Et_2O (10 mL) was added, the aqueous layer separated, extracted with Et_2O . The combined organic phases were washed with H_2O , brine, dried with anhydrous Na_2SO_4 and evaporated at atmospheric pressure. The crude product was purified by CC (SiO_2 : petroleum ether (40-70 $^\circ$)/ Et_2O : 8/2, $R_f(\mathbf{19d}) < R_f(\mathbf{8})$) to afford **8** as an oil (1.4 g, 8.6 mmol, 87%). Data of **8**: IR (CCl_4): 1715 ($\nu_{\text{C=O}}$); ^1H -NMR (CDCl_3): 2.20-2.19 (*m*, 2H), 2.10 (*ps*, 1H), 2.01 (*sept*, $J=6.87$, 1H), 1.77-1.41 (*m*, 8H), 0.80 (*d*, $J=6.87$, 6H); ^{13}C -NMR (CDCl_3): 217.7, 47.8, 45.3, 28.9, 27.7, 25.2, 24.6, 17.6.

$\text{C}_{11}\text{H}_{18}\text{O}$ (166.26); Calc. C: 79.46; H: 10.91%. Found C: 79.58; H: 11.12%.

1-Isopropyl-3,3-dimethylbicyclo[2.2.2]octan-2-ol (10). To a stirred solution of **8** (380 mg, 2.3 mmol) in THF (3 mL) NaH (0.8 g, 3.5 mmol) was added portionwise under Ar and the mixture was stirred at r.t. for 40 min. CH_3I (4 mL, 0.07 mol) was then added dropwise and the mixture refluxed under Ar until TLC monitoring (SiO_2 : petroleum ether (40-70 $^\circ$)/ Et_2O : 9/1, $R_f(\mathbf{8}) < R_f(\mathbf{9})$) showed the disappearance of the starting material. The reaction mixture was neutralized with

0.5N HCl, washed with H₂O, brine, dried with anhydrous Na₂SO₄ and evaporated at atmospheric pressure. The residue constituted by 1-isopropyl-3,3-dimethyl-bicyclo[2.2.2]octan-2-one (**9**) was used as such in the following step.

A solution of compound **9** in anhydrous THF (10 mL) was treated with LiAlH₄ (130 mg, 3.3 mmol). The reaction mixture was stirred at r.t. until TLC analysis (petroleum ether (40-70°)/Et₂O: 9/1, *R_f*(**10**)<*R_f*(**9**)) showed the disappearance of the starting material (1h). Excess LiAlH₄ was quenched by dropwise addition of H₂O and neutralized with 0.1N HCl. The layers were separated and the aqueous one extracted three times with Et₂O. The combined organic layers were washed with brine, dried with anhydrous Na₂SO₄ and concentrated at atmospheric pressure. The crude residue was purified by CC (SiO₂; petroleum ether (40-70°)/Et₂O: 9.5/0.5) to afford **10** as an oil (350 mg, 1.8 mmol, 77%). Data of **10**: IR (CCl₄): 3516 (ν_{OH}); ¹H-NMR (CDCl₃): 3.36 (s, 1H), 1.89-1.12 (m, 11H), 1.01 (s, 3H), 0.99 (s, 3H), 0.81 (d, *J*=6.85, 3H), 0.76 (d, *J*=6.92, 3H); ¹³C-NMR (CDCl₃): 78.0, 38.6, 36.4, 36.3, 30.8, 30.0, 23.2, 22.9, 22.2, 21.9, 21.6, 17.2, 16.9.

C₁₃H₂₄O (196.33); Calc. C: 79.53; H: 12.32%. Found C: 79.83; H: 12.56%.

1-Isopropyl-bicyclo[2.2.2]octan-2-ol (11). To a solution of compound **8** (150 mg, 0.9 mmol) in anhydrous THF (5 mL) LiAlH₄ (50 mg, 1.3 mmol) was added. The reaction mixture was stirred at r.t. until TLC analysis (SiO₂; petroleum ether (40-70°)/Et₂O: 9/1, *R_f*(**8**)>*R_f*(**11**)) showed the disappearance of the starting material (1h). Excess LiAlH₄ was quenched by dropwise addition of H₂O and neutralized with 0.1N HCl. The layers were separated, and the aqueous one extracted with Et₂O, washed with brine, dried with anhydrous Na₂SO₄ and concentrated at atmospheric pressure. The crude residue was purified by CC (SiO₂; petroleum ether (40-70°)/Et₂O: 9.5/0.5) to afford **11** as an oil (116 mg, 0.7 mmol, 77%). Data of **11**: IR (CCl₄): 3543 (ν_{OH}); ¹H-NMR (CDCl₃): 3.91-3.85 (m, 1H), 2.05-1.92 (m, 1H), 1.71-1.03 (m, 12H), 0.83 (d, *J*=6.32, 3H), 0.80 (d, *J*=6.68, 3H); ¹³C-NMR (CDCl₃): 69.7, 38.3, 36.7, 30.8, 26.1, 25.0, 24.8, 22.9, 21.5, 17.1, 17.0.

C₁₁H₂₀O (168.28); Calc. C: 78.51; H: 11.98%. Found C: 78.68; H: 12.28 %.

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