

Preparation of aryl benzyl ketones by [1,2]-Wittig rearrangement

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Dedicated to Professor S. V. Kessar on the occasion of his 70th birthday

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Abstract

α -(Benzotriazol-1-yl)benzyl ethers **4a–f** were readily prepared by condensation of benzotriazole, aryl aldehydes and benzyl alcohols. A one-pot reaction involving deprotonation of **4** followed by [1,2]-Wittig rearrangement and departure of the benzotriazolyl group resulted in aryl benzyl ketones **6a–f** in good yields.

Keywords: α -(Benzotriazol-1-yl)benzyl ethers, [1,2]-Wittig rearrangement, aryl benzyl ketones

Introduction

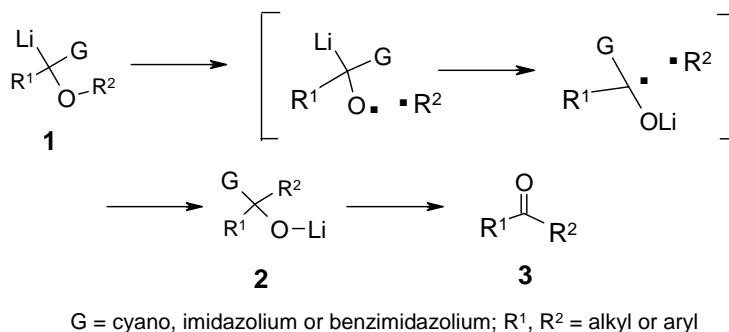
The [1,2]-Wittig rearrangement¹ has been reviewed extensively.² It is now widely accepted that the rearrangement proceeds *via* a radical dissociation-recombination mechanism,^{3,4} with configuration retention at the migrating carbon and inversion at the Li-bearing terminus.⁵

Substituted ethers **1**, in which G can function as a leaving group as well as an electron-withdrawing group, eliminate group G after [1,2]-Wittig rearrangement to form the carbonyl compounds **3** (Scheme 1). Such reactions have been reported for G = cyano,⁶ imidazolium and benzimidazolium⁷ groups. α -(Benzotriazol-1-yl)alkyl allyl ethers have been successfully utilized in the synthesis of homoallylic ketones and alcohols *via* [2,3]-Wittig rearrangements.⁸ We now report an extension of this methodology: [1,2]-Wittig rearrangements of α -(benzotriazol-1-yl)arylmethyl ethers to give aromatic ketones.

Results and Discussion

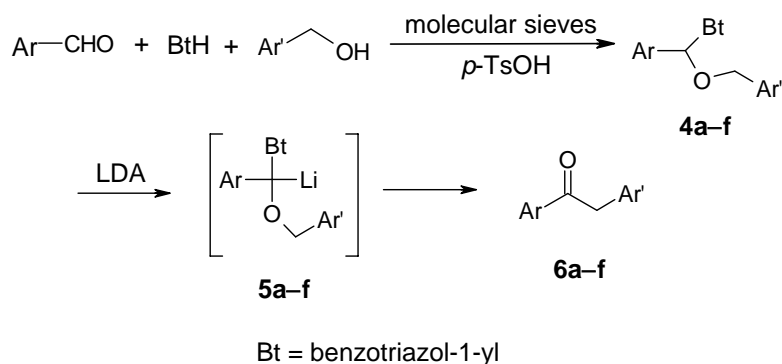
Condensation of an aromatic aldehyde, an aliphatic alcohol and benzotriazole in the presence of a catalytic amount of *p*-TsOH and 4Å molecular sieves in methylene chloride at room temperature as expected,^{8,9} gave the α -(benzotriazol-1-yl)alkyl ethers **4a–f** in 60–78% yield.

Novel ethers **4b–f** were fully characterized by their ^1H , ^{13}C NMR spectra and elemental analyses or HRMS data.



Scheme 1

The acidic benzotriazole-activated α -proton in the α -(benzotriazol-1-yl)alkyl benzyl ethers **4a–f** on treatment with LDA at $-78\text{ }^\circ\text{C}$ readily gives the α -lithiated intermediates **5a–f**, which eliminate the benzotriazole anion to give, by [1,2]-Wittig rearrangement, the carbonyl products **6a–f** in 53–63% yield. The structures of **6a–f** were confirmed by ^1H and ^{13}C NMR spectroscopy. ^1H NMR spectra of all of the ketones **6a–f** displayed a similar pattern including the disappearance of the α -proton and benzotriazolyl signals from the aromatic region. The benzylic protons in **6a–f** resonated around 4.2 ppm and the ^{13}C NMR signal for this carbon appeared around 45 ppm. The aromatic signals in all of the ketones **6a–f** experienced a negligible change in the chemical shift values from those in the ^1H and ^{13}C NMR spectra of starting ethers **4a–f**. The disappearance of the α -carbon signal around 88 ppm and the appearance of a carbonyl signal near 197 ppm in the ^{13}C NMR spectra further confirmed the formation of ketones **6a–f** (Scheme 2), (Table 1).



Scheme 2

Table 1. Synthesis of α -(benzotriazol-1-yl) ethers **4a–f** and aryl benzyl ketones **6a–f**

Entry	Ar	Ar'	4 (%yield) ^a	6 (%yield) ^a
a	Ph	Ph	78	63
b	4-F-C ₆ H ₄	Ph	63	54
c	4-Cl-C ₆ H ₄	Ph	60	62
d	3-MeO-C ₆ H ₄	Ph	72	53
e	Ph	4-MeO-C ₆ H ₄	73	61
f	4-Cl-C ₆ H ₄	4-MeO-C ₆ H ₄	63	63

^a Isolated yields.

General methods for the preparation of ketones from aldehydes include (i) nucleophilic addition of an organometallic reagent followed by oxidation of the secondary alcohol;¹⁰ (ii) transformation into a carbonyl anion equivalent such as an (α -benzotriazol-1-ylalkyl)methyl ether followed by lithiation, alkylation and deprotection.⁹ The present two-step [1,2]-Wittig rearrangement procedure provides access to aryl benzyl ketones from an aromatic aldehyde and a benzyl alcohol.

Experimental Section

General Procedures. Melting points were determined using a Bristoline hot-stage microscope and are uncorrected. ¹H NMR (300 MHz) and ¹³C NMR (75 MHz) spectra were recorded on a Gemini 300 NMR spectrometer in CDCl₃ (with TMS for ¹H and chloroform-*d* for ¹³C as the internal reference). HRMS were measured on an AEI-30 mass spectrometer. Elemental analyses were performed on a Carlo Erba-1106 instrument. THF was distilled from sodium/benzophenone and dichloromethane from calcium hydride prior to use. All of the reactions were carried out under N₂.

General procedure for the preparation of α -(benzotriazol-1-yl)alkyl ethers **4**

A mixture of an aldehyde (20 mmol), an alcohol (20 mmol), benzotriazole (25 mmol), *p*-TsOH (10 mmol) and 4Å molecular sieves (10 g) in dry methylene chloride (60 mL) was stirred at 25 °C for 2 days. After filtration, the undissolved solid was washed with methylene chloride (3×30 mL). The solvent of the combined filtrate was removed in vacuo and the residue was purified by column chromatography using hexanes/EtOAc (9:1) as eluent to afford the pure products **4a–f**.

1-[(Benzyloxy)(phenyl)methyl]-1*H*-1,2,3-benzotriazole (4a). Colorless oil;¹¹ yield, 78%; ¹H NMR δ 8.10–8.04 (m, 1H), 7.44–7.42 (m, 2H), 7.34–7.25 (m, 12H), 4.58 (s, 2H); ¹³C NMR δ 147.0, 136.0, 131.1, 129.0, 128.6, 128.5, 128.2, 128.1, 127.5, 126.0, 124.2, 119.9, 111.7, 88.3, 70.7.

1-[(Benzyloxy)(4-fluorophenyl)methyl]-1*H*-1,2,3-benzotriazole (4b). Colorless oil; yield, 63%; ¹H NMR δ 8.12–8.05 (m, 1H), 7.46–7.22 (m, 11H), 7.04 (t, *J* = 8.5 Hz, 2H), 4.57 (s, 2H); ¹³C NMR δ 163.1 (d, *J* = 248.5 Hz), 147.3, 136.1, 132.2 (d, *J* = 3.4 Hz), 131.2, 128.9, 128.3 (d, *J*

= 23.5 Hz), 128.4, 128.1, 127.9, 124.6, 120.3, 115.8, 111.7, 88.0, 71.0. HRMS calcd for $C_{20}H_{17}FN_3O_2$: 334.1356 (M+1), found: 334.1346.

1-[(Benzyloxy)(4-chlorophenyl)methyl]-1H-1,2,3-benzotriazole (4c). Colorless oil; yield, 60%; 1H NMR δ 8.12–8.05 (m, 1H), 7.40–7.24 (m, 13H), 4.57 (s, 2H); ^{13}C NMR δ 147.3, 136.0, 135.3, 134.9, 131.2, 129.0, 128.8, 128.6, 128.4, 127.9, 127.7, 124.6, 120.3, 111.7, 87.9, 71.1. Anal. Calcd for $C_{20}H_{16}ClN_3O$: C, 68.67; H, 4.61; N, 12.01. Found: C, 68.54; H, 4.83; N, 12.16.

1-[(Benzyloxy)(3-methoxyphenyl)methyl]-1H-1,2,3-benzotriazole (4d). Colorless oil; yield, 72%; 1H NMR δ 8.10–8.04 (m, 1H), 7.40–7.20 (m, 10H), 7.04 (s, 1H), 6.98 (d, $J = 7.7$ Hz, 1H), 6.86 (dd, $J = 8.2, 2.2$ Hz, 1H), 4.57 (s, 2H), 3.74 (s, 3H); ^{13}C NMR δ 159.9, 147.2, 137.8, 136.2, 131.3, 129.9, 128.7, 128.4, 128.3, 127.7, 124.5, 120.1, 118.4, 114.5, 112.0, 111.9, 88.3, 71.0, 55.5. Anal. Calcd for $C_{21}H_{19}N_3O_2$: C, 73.03; H, 5.54; N, 12.17. Found: C, 72.79; H, 5.69; N, 11.96.

1-[(4-Methoxybenzyl)oxy](phenyl)methyl)-1H-1,2,3-benzotriazole (4e). Colorless oil; yield, 73%; 1H NMR δ 8.20–8.00 (m, 1H), 7.45–7.37 (m, 2H), 7.35–7.22 (m, 9H), 6.86 (d, $J = 7.5$ Hz, 2H), 4.53 (d, $J = 11.7$ Hz, 1H), 4.47 (d, $J = 11.7$ Hz, 1H), 3.77 (s, 3H); ^{13}C NMR δ 159.8, 147.1, 136.4, 131.3, 130.2, 129.1, 128.7, 128.2, 127.7, 126.1, 124.4, 120.1, 114.1, 111.9, 88.1, 70.6, 55.4. Anal. Calcd for $C_{21}H_{19}N_3O_2$: C, 73.03; H, 5.54; N, 12.17. Found: C, 72.84; H, 5.75; N, 12.17.

1-[(4-Chlorophenyl)[(4-methoxybenzyl)oxy]methyl]-1H-1,2,3-benzotriazole (4f). Colorless oil; yield, 63%; 1H NMR δ 8.12–8.04 (m, 1H), 7.40–7.20 (m, 10H), 6.86 (d, $J = 8.5$ Hz, 2H), 4.53 (d, $J = 11.7$ Hz, 1H), 4.46 (d, $J = 11.7$ Hz, 1H), 3.79 (s, 3H); ^{13}C NMR δ 159.9, 147.2, 135.2, 135.0, 131.2, 130.2, 129.0, 127.9, 127.8, 127.6, 124.6, 120.2, 114.1, 111.7, 87.4, 70.7, 55.4. Anal. Calcd for $C_{21}H_{18}ClN_3O_2$: C, 66.40; H, 4.78; N, 11.06. Found: C, 66.04; H, 4.66; N, 10.84.

General procedure for the preparation of ketones (6a–f) via [1,2]-Wittig rearrangement

To a solution of **4a–f** (2 mmol) in dry THF (10 mL) at -78 °C was added dropwise LDA (2.2 mmol). The solution was stirred at -78 °C for 3 hours and then allowed to warm up to room temperature overnight. The reaction was quenched using water (5 mL) and then extracted with ethyl acetate (3 \times 20 mL). The combined organic extracts were dried over Na_2SO_4 . After removal of the solvent *in vacuo* the residue was purified by column chromatography using hexanes/EtOAc (20:1) as eluent to afford the pure ketones **6a–f**.

1,2-Diphenyl-1-ethanone (6a). White prisms (pentane); mp 53.0–55.0 °C (Lit.¹² mp 55.0–56.0 °C); yield, 63%; 1H NMR δ 8.01 (d, $J = 7.7$ Hz, 2H), 7.57–7.42 (m, 3H), 7.35–7.24 (m, 5H), 4.28 (s, 2H); ^{13}C NMR δ 197.4, 136.4, 133.0, 129.3, 128.5, 128.5, 128.4, 126.7, 45.3.

1-(4-Fluorophenyl)-2-phenyl-1-ethanone (6b). White microcrystals (hexane); mp 77.5–78.5 °C (Lit.¹³ mp 83.0 °C); yield, 54%; 1H NMR δ 8.03 (dd, $J = 8.8, 5.5$ Hz, 2H), 7.38–7.22 (m, 5H), 7.11 (t, $J = 8.5$ Hz, 2H), 4.25 (s, 2H); ^{13}C NMR δ 196.2, 165.9 (d, $J = 254.8$ Hz), 134.5, 133.1, 131.5, 131.4, 129.2 (d, $J = 48.1$ Hz), 127.2, 115.9 (d, $J = 11.8$ Hz), 45.7.

1-(4-chlorophenyl)-2-phenyl-1-ethanone (6c). White needles (hexane); mp 102.0–103.0 °C (Lit.¹⁴ mp 107.5 °C); yield, 62%; 1H NMR δ 7.94 (d, $J = 8.4$ Hz, 2H), 7.42 (d, $J = 8.5$ Hz, 2H),

7.38–7.20 (m, 5H), 4.25 (s, 2H); ^{13}C NMR δ 196.6, 139.8, 135.0, 134.4, 130.2, 129.6, 129.2, 129.0, 127.2, 45.8.

1-(3-Methoxyphenyl)-2-phenyl-1-ethanone (6d). Colorless oil;¹⁵ yield, 53%; ^1H NMR δ 7.60 (d, $J = 7.7$ Hz, 1H), 7.55–7.51 (m, 1H), 7.40–7.22 (m, 6H), 7.10 (dd, $J = 8.2, 2.2$ Hz, 1H), 4.27 (s, 2H), 3.84 (m, 3H); ^{13}C NMR δ 197.7, 160.0, 138.2, 134.8, 129.8, 129.6, 128.9, 127.1, 121.5, 119.9, 113.0, 55.6, 45.8.

2-(4-Methoxyphenyl)-1-phenyl-1-ethanone (6e). White powder (hexane); mp 92.0 °C (Lit.¹⁶ mp 99.5–100.5 °C); yield, 61%; ^1H NMR δ 8.01 (d, $J = 7.1$ Hz, 2H), 7.55 (t, $J = 7.4$ Hz, 1H), 7.45 (t, $J = 7.7$ Hz, 2H), 7.20 (d, $J = 8.5$ Hz, 2H), 6.86 (d, $J = 8.5$ Hz, 2H), 4.22 (s, 2H), 3.78 (s, 3H); ^{13}C NMR δ 198.1, 158.7, 136.8, 133.3, 130.7, 128.8, 128.7, 126.7, 114.3, 55.4, 44.8.

1-(4-Chlorophenyl)-2-(4-methoxyphenyl)-1-ethanone (6f). White prisms (hexane); mp 107.5 °C (Lit.¹⁷ mp 111.0 °C); yield, 63%; ^1H NMR δ 7.93 (d, $J = 8.5$ Hz, 2H), 7.42 (d, $J = 8.5$ Hz, 2H), 7.16 (d, $J = 8.5$ Hz, 2H), 6.86 (d, $J = 8.6$ Hz, 2H), 4.19 (s, 2H), 3.78 (s, 3H); ^{13}C NMR δ 196.7, 158.6, 139.6, 134.9, 130.4, 130.1, 129.0, 126.1, 114.2, 55.3, 44.7.

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