

Enantioselectivity for catalytic cyclopropanation with diazomalonates

Michael P. Doyle* and Wenhao Hu

Department of Chemistry, University of Arizona, Tucson, Arizona, 85721, USA

E-mail: mdoyle@u.arizona.edu

This paper is dedicated to Professor M. A. McKervey on the occasion of his retirement from Queen's University, Belfast, Ireland

(received 30 Jan 03; accepted 06 Mar 03; published on the web 24 Mar 03)

Abstract

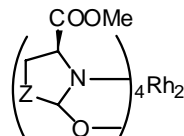
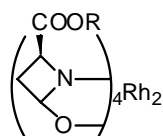
The use of chiral azetidinone-ligated dirhodium(II) catalysts activates dinitrogen extrusion from diazomalonates and provides access to cyclopropanation products with selectivities as high as 40-50% ee.

Keywords: Cyclopropanation, diazomalonates, chiral catalysts, asymmetric synthesis

Introduction

There are few reports of successful intermolecular cyclopropanation reactions of diazomalonates that occur with enantiocontrol,^{1,2} and only a limited number have documented enantioselective intramolecular cyclopropanation reactions of allyl esters of β -ketoacetates.^{3,4} There are two reasons for this. One is the relative unreactivity of diazomalonates toward diazo decomposition.⁵ The other is the placement of two like (identical) substituents on the diazo carbon that minimizes (eliminates) carbene substituent differentiation in the cyclopropanation reaction.

We have recently developed a set of chiral azetidinone-carboxylate ligands for placement on the dirhodium(II) core.⁶ Because the amide OCN angle is greater in these compounds than in their five-membered ring counterparts,⁷ the Rh-Rh bond length is increased and with it the electronic reactivity of the carboxamidate-ligated catalyst for diazo decomposition.⁸ These catalysts have the structure represented by **1**, and they differ in reactivity towards diazo esters from their five-membered ring pyrrolidinone (**2**),⁹ oxazolidinone (**3**),¹⁰ or imidazolidinone (**4**) analogues.¹¹ Their reactivity does in fact approach that of proline catalysts such as **5**,¹² often referred to as "the McKervey catalyst," or the *tert*-lucinate-based catalyst **6**.¹³ This communication describes their selectivities in cyclopropanation in reactions of representative vinyl compounds.



1a R = Me, Rh₂(4S-MEAZ)₄

2 Z = CH₂, Rh₂(5S-MEPY)₄

1b R = ⁱBu, Rh₂(4S-IBAZ)₄

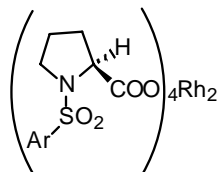
3 Z = O, Rh₂(4S-MEOX)₄

1c R = CH₂CM₂, Rh₂(4S-NEPAZ)₄

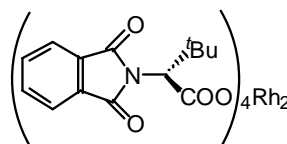
4 Z = NCOCH₂CH₂Ph, Rh₂(4S-MPPIM)₄

1d R = ^cC₆H₁₁, Rh₂(4S-CHAZ)₄

1e R = PhCH₂, Rh₂(4S-BNAZ)₄



5 Ar = *p*-*tert*-BuC₆H₄



6

Results and Discussion

The common reaction with styrene was first examined under standard conditions (1.0 % mol catalyst, refluxing CH₂Cl₂, 10 equiv olefin). Dimethyl diazomalonate did not undergo decomposition during 2.5h with **2-4**, but complete reaction was achieved with **1a-e**, **5**, and **6** under the same conditions. Product yields and measured enantioselectivities from these reactions (eq 1) are presented in Table 1. High product yields are obtained in each case, and enantioselectivities from the use of **1**, especially **1a** and **1b**, are the highest achieved.

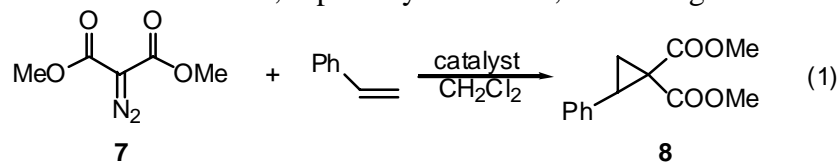


Table 1. Product yields and enantioselectivities from catalyzed reactions of dimethyl diazomalonate with styrene

Catalyst	Yield 8 , %	ee, %
1a	97	44
1b	87	38
1c	91	36
1d	92	19
1e	88	25
5	63	7 ^a
6	88	23

^a From ref. 2; reaction performed in pentane.

We reasoned that placement of electron-withdrawing groups on the benzene ring of styrene might decrease the nucleophilicity of the carbon-carbon double bond towards the intermediate electrophilic metal carbene. Using *p*-trifluoromethylstyrene (eq 2), selectivities did, in fact increase, but only modestly (Table 2). We then investigated possible steric enhancement of enantiocontrol through

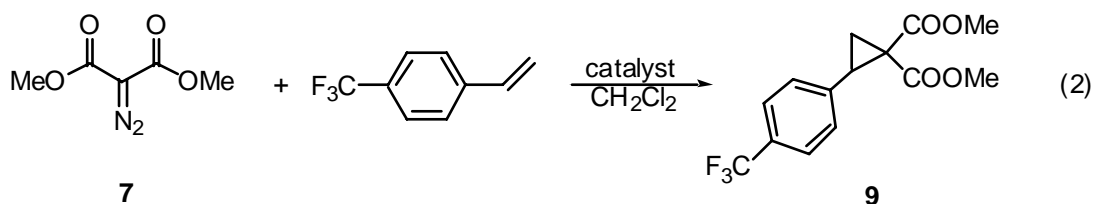


Table 2. Product yields and enantioselectivities from catalyzed reactions of dimethyl diazomalonate with *p*-trifluoromethylstyrene^a

Catalyst	Yield 9 , %	ee, %
1a	73	50
1b	52	42

^a Same conditions as reactions in Table 1.

the use of di-*tert*-butyl diazomalonate. However, these reactions were complicated by competing insertion into the ester primary carbon-hydrogen bond (eq 3) – a rare observation in reactions of this type.^{5,14} Results are described in Table 3. Notable is the influence of the ligand ester group in the catalyst on the extent of C–H insertion and on enantiocontrol in cyclopropanation.

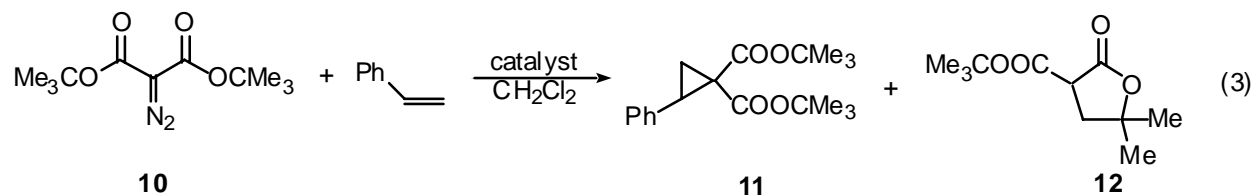


Table 3. Product yields and enantioselectivities from catalyzed reactions of di-*tert*-butyl diazomalonate with styrene^a

Catalyst	Yield (11 + 12), %	11 : 12	ee 11 , %
1a	60	25 : 75	28
1c	55	36 : 64	0
1d	65	59 : 41	0

^a Same conditions as reactions in Table 1.

Two other alkenes were evaluated for enantioselective cyclopropanation with dimethyl diazomalonate. Vinyl acetate gave **13** in good yield and expected modest enantiocontrol (Table 4), but vinylcyclohexane underwent cyclopropanation to **14** with virtually no enantioselectivity (Table 4), and the reason for this is unknown. A complex reaction mixture was obtained in attempted cyclopropanation of *n*-butyl vinyl ether, and the mixture was not further analyzed.

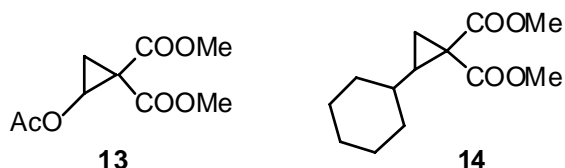


Table 4. Product yields and selectivities from catalyzed reactions of dimethyl diazomalonate with vinyl acetate and vinylcyclohexane^a

RCH=CH ₂	Catalyst	Yield, %	ee, %
R= AcO	1a	65 (13)	33
	1b	72 (13)	34
^c C ₆ H ₁₁	1b	60 (14)	6

^a Same conditions as reactions in Table 1.

Efforts were also undertaken to effect intramolecular cyclopropanation with allyl diazomalonate **15** (eq 4). Here reaction conditions were identical to those typically performed for intermolecular reactions without, of course, added alkene. Analyses were the same as those reported by Koskinen and Tamm.^{4,15} We were gratified to find that enantiocontrol for intramolecular cyclopropanation reached new high levels, but product yields were unexpectedly low (Table 5). The reason(s) for the low yields in these reactions are not evident as yet. Further investigations are underway.

Coinciding with the initial reports of chiral semicorrin and bis-oxazoline ligands for copper,⁵ were the first reports of chiral (homochiral) catalysts of dirhodium(II).^{12a, 16, 17} M. A. McKervery contributed substantially and creatively to this development, and his efforts continue to influence the field.^{5, 12b, 18}

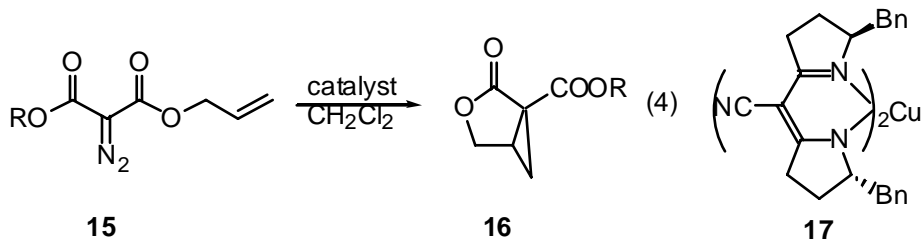


Table 5. Product yields and selectivities from catalyzed reactions of **15**^a

R =	Catalyst	Yield 16 , %	ee, %
Me	1a	30	57
	1b	21	46
	1d	14	42
	1c	13	43
^t Bu	1a	5	30
	17^b	72	32

^a Same conditions as reactions in Table 1. ^b Ref. 4.

Experimental Section

Cyclopropanation of styrene with dimethyl diazomalonate.⁵ General procedure

A solution of dimethyl diazomalonate **13** (63 mg, 0.39 mmol) in dichloromethane (CH₂Cl₂) (2 mL) was added *via* syringe pump (1.0 mL/hr) over 2 hours to a refluxing solution of Rh₂(4*S*-BNAZ)₄ (4.3 mg, 1.0 mol %) and styrene (0.40 ml, 3.9 mmol) in CH₂Cl₂ (4 mL). After complete addition the reaction mixture was stirred at reflux for a further 30 minutes to ensure complete reaction. The mixture was then cooled to room temperature, then passed through a short silica plug, which was subsequently washed with CH₂Cl₂ (40 mL). The solvent was removed at reduced pressure, to furnish the desired cyclopropane **8** (80 mg, 0.34 mmol, 88%) as a colorless oil, 25% ee (determined by GC on a 30-m Chiraldex β-DM column 100°C for 5 min, then 1°C/min to 160°C. *t*_{R minor} = 63.4 min, *t*_{R major} = 64.7 min). ¹H NMR (300 MHz, CDCl₃) δ 7.38-7.15 (comp, 5H), 3.78 (s, 3H), 3.23 (dd, *J* = 9.3, 7.8 Hz, 1H), 2.20 (dd, *J* = 7.8, 5.4 Hz, 1H), 1.74 (dd, *J* = 9.3, 5.4 Hz, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 170.1, 166.9, 134.5, 128.3, 128.0, 127.2, 52.6, 52.0, 37.1, 32.4, 18.9.

Cyclopropanation of *p*-trifluoromethylstyrene with dimethyl diazomalonate.⁵ General procedure

A solution of dimethyl diazomalonate **9** (32 mg, 0.20 mmol) in dichloromethane (CH₂Cl₂) (2 mL) was added *via* syringe pump (0.5 mL/hr) over 4 hours to a refluxing solution of Rh₂(4*S*-MEAZ)₄ (1.4 mg, 1.0 mol %) and *p*-trifluoromethylstyrene (0.34 ml, 2.0 mmol) in CH₂Cl₂ (4 mL). After complete addition the reaction mixture was stirred at reflux for a further 30 minutes to ensure complete reaction. The mixture was then cooled to room temperature, then passed through a short silica plug, which was subsequently washed with CH₂Cl₂ (40 mL). The crude product was purified by flash column chromatography on silica gel eluting with 40% ethyl acetate in hexanes to furnish the desired cyclopropane **9** (44 mg, 0.14 mmol, 73%) as a colorless oil, 50% ee (determined by GC on a 30-m Chiraldex β-DM column 100°C for 5 min, then 0.5°C/min to 170°C. *t*_{R minor} = 88.9 min, *t*_{R major} = 91.2 min). ¹H NMR (300 MHz, CDCl₃) δ 7.53 (d, *J* = 8.4 Hz, 2H), 7.30 (d, *J* = 8.4 Hz, 2H), 7.30 (d, *J* = 8.4 Hz, 2H), 3.80 (s, 3H), 3.38 (s, 3H), 3.25 (dd, *J* = 9.0, 7.8 Hz, 1H), 2.20 (dd, *J* = 7.8, 5.1 Hz, 1H), 1.78 (dd, *J* = 9.0, 5.1 Hz, 1H); ¹³C

NMR (125 MHz, CDCl₃) δ 169.8, 166.7, 138.8, 128.8, 128.0, 125.1, 52.9, 52.4, 37.3, 31.8, 29.7, 19.1.

Cyclopropanation of styrene with di-*tert*-butyl diazomalonate. General procedure

A solution of di-*tert*-butyl diazomalonate **10** (97 mg, 0.4 mmol) in dichloromethane (CH₂Cl₂) (2 mL) was added *via* syringe pump (1.0 mL/hr) over 2 hours to a refluxing solution of Rh₂(4*S*-CHAZ)₄ (3.9 mg, 1.0 mol %) and styrene (0.40 mL, 3.9 mmol) in CH₂Cl₂ (4 mL). After complete addition the reaction mixture was stirred at reflux for a further 30 minutes to ensure complete reaction. The mixture was then cooled to room temperature, then passed through a short silica plug, which was subsequently washed with CH₂Cl₂ (40 mL). The solvent was removed under reduced pressure, to furnish a 59:41 mixture (76 mg, 0.26 mmol, 65%) of cyclopropane **11** (ee 0%, determined by GC on a 30-m ChiralDex β -DM column 140°C isotherm, $t_{R \text{ major}} = 52.9$ min, $t_{R \text{ minor}} = 53.7$ min), and insertion product **12**. Data for **11**: ¹H NMR (500 MHz, CDCl₃) δ 7.28 – 7.17 (comp, 5H), 3.09 (dd, $J = 9.1, 7.8$ Hz, 1H), 2.03 (dd, $J = 7.8, 5.1$ Hz, 1H), 1.52 (dd, $J = 9.1, 5.1$ Hz, 1H), 1.50 (s, 9H), 1.09 (s, 9H); ¹³C NMR (125 MHz, CDCl₃) δ 169.3, 65.9, 135.0, 128.8, 127.9, 81.7, 80.8, 39.3, 30.9, 28.0, 27.4, 17.6. Data for **12**: ¹H NMR (300 MHz, CDCl₃) δ 3.65 (t, $J = 9.9$ Hz, 1H), 2.46 (dd, $J = 12.9, 9.9$ Hz, 1H), 2.29 (dd, $J = 12.9, 9.9$ Hz, 1H), 1.51 (s, 3H), 1.50 (s, 9H), 1.41 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 171.6, 167.4, 83.4, 82.7, 48.6, 38.3, 28.3, 27.8, 27.7.

Cyclopropanation of vinyl acetate with dimethyl diazomalonate. General procedure

A solution of dimethyl diazomalonate (63 mg, 0.39 mmol) in dichloromethane (CH₂Cl₂) (2 mL) was added *via* syringe pump (1.0 mL/hr) over 2 hours to a refluxing solution of Rh₂(4*S*-MEAZ)₄ (2.8 mg, 1.0 mol %) and vinyl acetate (0.34 mL, 3.9 mmol) in CH₂Cl₂ (4 mL). After complete addition the reaction mixture was stirred at reflux for a further 30 minutes to ensure complete reaction. The mixture was then cooled to room temperature, then passed through a short silica plug, which was subsequently washed with CH₂Cl₂ (40 mL). The crude product was purified by flash column chromatography on silica gel eluting with 10% ethyl acetate in hexanes to furnish the desired cyclopropane **13** (56 mg, 0.26 mmol, 65%) as a colorless oil, 33% ee (determined by GC on a 30-m ChiralDex β -DM column 100°C for 5 min, then 1°C/min to 160°C, $t_{R \text{ minor}} = 33.8$ min, $t_{R \text{ major}} = 34.2$ min). ¹H NMR (500 MHz, CDCl₃) δ 4.77 (dd, $J = 7.0, 5.2$ Hz, 1H), 3.78 (s, 3H), 3.75 (s, 3H), 2.04 (s, 3H), 1.95 (dd, $J = 7.0, 5.2$ Hz, 1H), 1.72 (dd, $J = 7.0$ Hz, 1H); ¹³C NMR (125 MHz, CDCl₃) δ 170.2, 168.4, 165.8, 56.8, 52.8, 33.9, 20.4, 19.3; HRMS (FAB⁺) Calcd for C₉H₁₆O₆: 217.0712. Found: 217.0714.

Cyclopropanation of vinylcyclohexane with dimethyl diazomalonate. General procedure

A solution of dimethyl diazomalonate (63 mg, 0.39 mmol) in dichloromethane (CH₂Cl₂) (2 mL) was added *via* syringe pump (1.0 mL/hr) over 2 hours to a refluxing solution of Rh₂(4*S*-IBAZ)₄ (3.5 mg, 1.0 mol %) and vinylcyclohexane (0.54 mL, 3.9 mmol) in CH₂Cl₂ (4 mL). After complete addition the reaction mixture was stirred at reflux for a further 30 minutes to ensure

complete reaction. The mixture was then cooled to room temperature, then passed through a short silica plug, which was subsequently washed with CH₂Cl₂ (40 mL). The solvent was removed at reduced pressure, to furnish the desired cyclopropane **14** (56 mg, 0.23 mmol, 60%) as a colorless oil, 6% ee (determined by GC on a 30-m Chiraldex β-DM column 140°C isotherm, t_{R major} = 13.0 min, t_{R minor} = 13.5 min). ¹H NMR (300 MHz, CDCl₃) δ 3.75 (s, 3H), 3.71 (s, 3H), 1.89 – 1.60 (comp, 6H), 1.41 (dd, *J* = 7.9, 4.5 Hz, 1H), 1.34 (dd, *J* = 7.9, 4.5 Hz, 1H), 1.25 – 1.08 (comp, 5H), 0.79 (m, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 170.8, 168.8, 52.4, 52.2, 37.6, 35.0, 33.8, 32.7, 32.3, 26.1, 25.9, 25.7, 20.1; HRMS (FAB⁺) Calcd for C₁₃H₂₁O₄ : 241.1440. Found: 241.1448.

Acknowledgements

Support for this research from the National Science Foundation and the National Institutes of Health (GM-46503) is gratefully acknowledged. Thanks especially to M. A. (Tony) McKervey for his inspiration, insight, and intensity in the pursuit of excellence in the chemistry of diazo compounds and in asymmetric catalysis.

References and Notes

1. Doyle, M. P.; Davies, S. B.; Hu, W. *Org. Lett.* **2000**, *2*, 1145.
2. Davies, H. M. L.; Bruzinski, P.; Fall, M. J. *Tetrahedron Lett.* **1996**, *37*, 4133.
3. Pique, C.; Fahndrich, B.; Pfaltz, A. *Synlett* **1995**, 491.
4. Koskinen, A. M. P.; Hassila, H. *J. Org. Chem.* **1993**, *58*, 4479.
5. Doyle, M. P.; McKervey, M. A.; Ye, T. *Modern Catalytic Methods for Organic Synthesis with Diazo Compounds: From Cyclopropanes to Ylides*; Wiley: New York, 1998.
6. Doyle, M. P.; Zhou, Q.-L.; Simonsen, S. H.; Lynch, V. *Synlett* **1996**, 697.
7. Doyle, M. P.; Ren, T. *Prog. Inorg. Chem.* **2001**, *49*, 113.
8. Doyle, M. P.; Hu, W. *Adv. Synth. & Cat.* **2001**, *343*, 299.
9. Doyle, M. P.; Winchester, W. R.; Hoorn, J. A. A.; Lynch, V.; Simonsen, S. H.; Ghosh, R. *J. Am. Chem. Soc.* **1993**, *115*, 9968.
10. Doyle, M. P.; Dyatkin, A. B.; Protopova, M. N.; Yang, C. I.; Miertschin, C. S.; Winchester, W. R.; Simonsen, S. H.; Lynch, V.; Ghosh, R. *Rec. Trav. Chim. Pays-Bas* **1995**, *114*, 163.
11. Doyle, M. P.; Zhou, Q.-L.; Raab, C. E.; Roos, G. H. P.; Simonsen, S. H.; Lynch, V. *Inorg. Chem.* **1996**, *35*, 6064.
12. (a) Kennedy, M.; McKervey, M. A.; Maguire, A. R.; Roos, G. H. P. *J. Chem. Soc., Chem. Commun.* **1990**, 361. (b) Davies, H. M. L.; Antoulinakis, E. G. *Org. Rxns.* **2001**, *57*, 1.
13. Watanabe, N.; Ogawa, T.; Ohtake, Y.; Ikegami, S.; Hashimoto, S. *Synlett* **1996**, 85.

14. Doyle, M. P. In *Catalytic Asymmetric Synthesis, Second Edition*; Ojima, I., Ed.; Wiley: New York, 2000; Chapter 5.
15. (a) Koskinen, A. M. P.; Muñoz, L. *J. Org. Chem.* **1993**, *58*, 879. (b) Wick, L.; Tamm, C.; Boller, T. *Helv. Chim. Acta* **1995**, *78*, 403.
16. Brunner, H.; Kluschanzoff, H.; Wutz, K. *Bull. Soc. Chem. Belg.* **1989**, *98*, 63.
17. Doyle, M. P.; Brandes, B. D.; Kazala, A. P.; Pieters, R. J.; Jarstfer, M. B.; Watkins, L. M.; Eagle, C. T. *Tetrahedron Lett.* **1990**, *31*, 6613.
18. Doyle, M. P.; McKervey, M. A. *J. Chem. Soc., Chem. Commun.* **1997**, 983.