

Synthesis and antiviral evaluation of 2-amino-1,6-dihydro-6-oxo-9-[4-bis(hydroxymethyl)-2-cyclopenten-1-yl]-9H-purine: an analog of the anti-HIV compound Carbovir

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Abstract

A novel carbovir analog, (\pm)-2-amino-1,6-dihydro-6-oxo-9-[4-bis(hydroxymethyl)-2-cyclopenten-1-yl]-9H-purine, was synthesized starting from the γ -lactam, 2-azabicyclo[2.2.1]hept-5-en-3-one (**7**). The key intermediate, 1-acetamido-4-bis(hydroxymethyl)cyclopent-2-ene (**10**), was prepared by the Pfitzner-Moffatt oxidation of the hydroxymethyl group of **8**, aldol condensation of the resulting aldehyde, followed by a Cannizzaro reaction. The guanine base was constructed on the amino group of **10**. The structures of the target molecule and the intermediates were confirmed by NMR, UV and HRMS data. Anti-HIV evaluation results are reported.

Keywords: Anti HIV, carbovir, Pfitzner-Moffatt oxidation, Cannizzaro reaction

Introduction

Modified analogs of natural nucleosides have been widely studied as potential antitumor, antiviral and fungicidal agents.^{1,2} Search for nucleoside analogs which function as nontoxic, selective inhibitors of polymerases and other enzymes for the control of viral diseases and cancer has been the subject of intense research.³⁻⁶ However, as some of these nucleosides also undergo enzymatic degradations, a number of modifications have been carried out on both the sugar portion and the heterocycle to circumvent or to retard these deactivating processes. Nucleoside analogs that are good substrates for cellular kinases, but are resistant to other host enzymes such as phosphorylases (glycosidic bond cleavage) and deaminases (hydrolytic deamination) are essential for the development of useful therapeutic antiviral agents.

One important discovery in this area has been the replacement of the oxygen in the sugar portion of the nucleoside with a methylene unit, which results in carbocyclic nucleoside analogs which are highly resistant to phosphorylases.⁷ Carbocyclic analogs of nucleosides (CANs) have

recently become the object of increased interest owing to their demonstrated antineoplastic⁶ and antiviral⁸ properties. While the carbocyclic analog of adenosine was first described by Shealy and Clayton⁹ in 1966, it was the discovery that the natural carbocyclic nucleosides, aristeromycin (**1**)¹⁰ and neplanocin A (**2**),¹¹ display antibiotic and antitumor activity that sparked the search for other CANs with biological activity (Figure 1). Among the antiviral CANs discovered in the search for agents active against human immunodeficiency virus (HIV), the most promising are carbovir (**3**),¹² the structurally related abacavir (**4**)¹³ as well as BCA (**5**).¹⁴

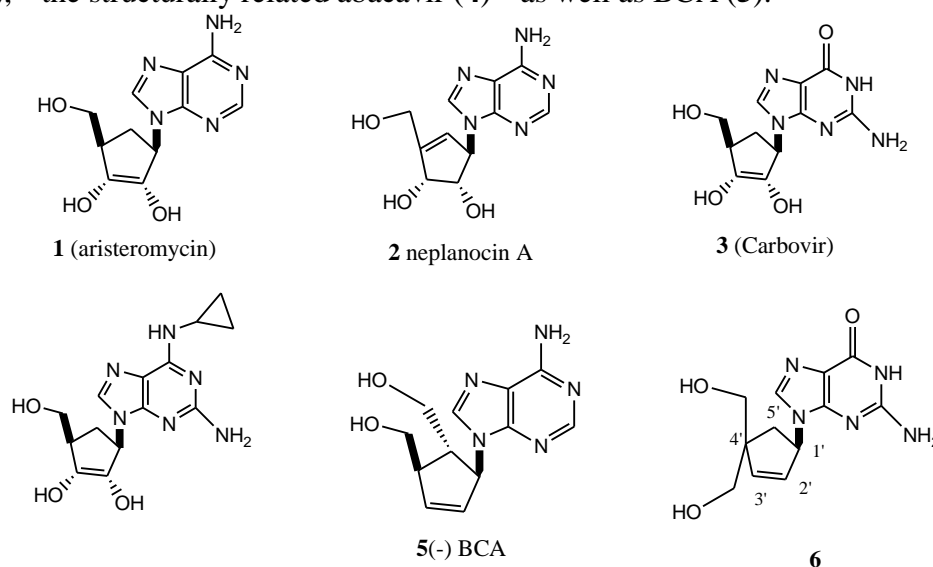


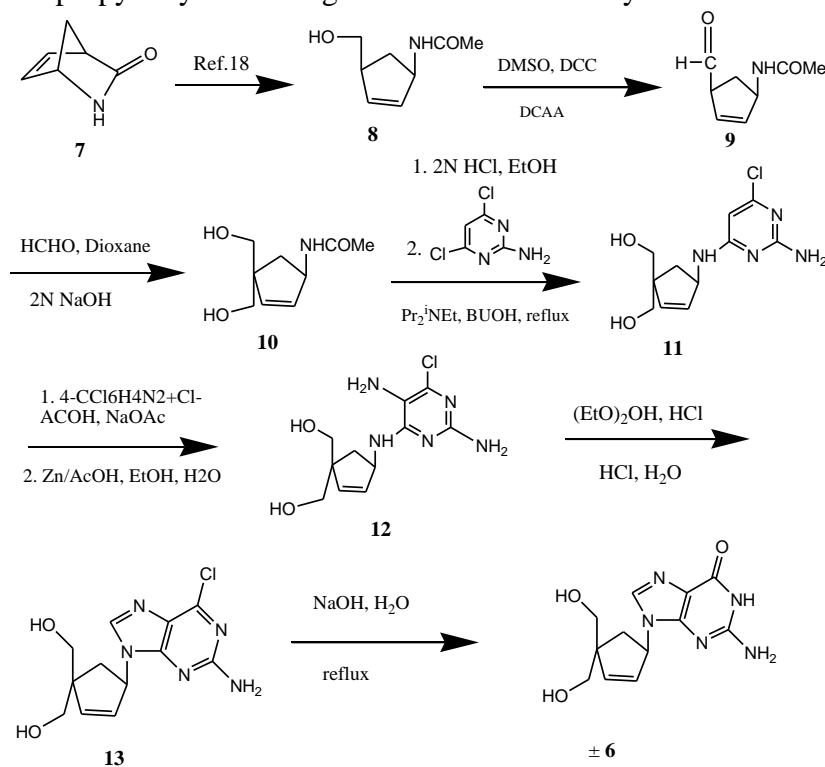
Figure 1

Although promising new antiviral agents have been discovered, their long-term usefulness is somewhat limited by their toxicities¹⁵ and development of resistant strains on prolonged clinical use. Therefore, the search for new inhibitors of HIV continues. Our interest in the title compound **6** was stimulated by the observation of the *in vitro* anti-HIV activity of 4'-substituted nucleosides.^{16,17} For example, 4'-hydroxymethylthymidine exhibits good activity against HIV-1.¹⁷ In addition, compound **5** also shows good activity against HIV-1 in MT-4 cells (IC₅₀ 0.20 $\mu\text{g/mL}$).¹⁴ However, to the best of our knowledge, there is no report on the synthesis and anti-HIV evaluation of 4'-substituted CANs. We describe herein the synthesis and antiviral evaluation of a 4'-substituted carbovir analog, 2-amino-1,6-dihydro-6-oxo-9-[4-bis(hydroxymethyl)-2-cyclopenten-1-yl]-9H-purine (**6**).

Results and Discussion

The route adopted for the synthesis of **6** utilizes the γ -lactam, 2-azabicyclo[2.2.1]hept-5-en-3-one (**7**), which is commercially available (Scheme 1). Hydrolysis of the lactam **7** with aqueous hydrochloric acid followed by esterification, acetylation and reduction by calcium borohydride provided the alcohol **8**.¹⁸ We thought of introducing the hydroxymethyl group at the 4'-position

of the cyclopentene ring through the aldehyde **9**, using in tandem, an aldol and a Cannizzaro reaction.¹⁹ Oxidation of the hydroxymethyl group of **8** under Swern conditions²⁰ resulted in a mixture of products which could not be identified. The oxidation of alcohol **8** using Moffatt conditions (DCC, DMSO, DCAA) was then investigated with good results.^{21,22} The resulting aldehyde **9**, without isolation, on aldol condensation with formaldehyde, followed by a Cannizzaro reaction provided the required 1-acetamido-4-bis(hydroxymethyl)cyclopent-2-ene (**10**) in 51% overall yield. Bishydroxymethyl compound **10** was deacetylated under acidic conditions and the resulting amino alcohol was coupled with 2-amino-4,6-dichloropyrimidine in the presence of diisopropylethylamine to give **11** in 50% overall yield.



Scheme 1

Diazotization of the diamine **11** using 4-chlorophenyldiazonium chloride followed by reduction of the yellow diazo compound with zinc-acetic acid afforded **12**. Cyclization of **12** with triethyl orthoformate under acidic conditions and subsequent treatment of the resulting compound **13** with aqueous sodium hydroxide under reflux gave the target compound (\pm) **6**. Interestingly, compound **13** is a very poor substrate for mammalian adenosine deaminase. Finally, compound **6** did not show significant anti-HIV activity *in vitro* in infected CEM-SS cells ($IC_{50} > 200 \mu M$).

Experimental Section

General Procedures. Melting points reported are uncorrected and were determined on a Thomas

Hoover apparatus fitted with a microscope. NMR spectra were recorded on Bruker AC-300 pulse Fourier transform spectrometer. Mass spectra were determined on a VG ZAB-HF instrument. UV spectra were recorded on a Cary 3 UV spectrometer. Preparative layer chromatography used plates prepared with E. Merck PF254 silica gel. Flash chromatography was carried out on columns packed with 240-400 mesh silica gel.

1-Acetamido-4-bis(hydroxymethyl)cyclopent-2-ene (10). A solution of **8**¹⁸ (880 mg, 5.68 mmol), dicyclohexylcarbodiimide (4.69 g, 22.71 mmol) and dichloroacetic acid (366 mg, 2.84 mmol) in DMSO (50 mL) was stirred under nitrogen at room temperature for 48 h. Solvent was evaporated under reduced pressure and the residue, presumably containing **9**, was suspended in petroleum ether (400 mL) and extracted with water (3 x 150 mL). The water later was filtered and concentrated under reduced pressure. 37% solution of formaldehyde (1.84 mL, 22.71 mmol) and 2N sodium hydroxide (11.36 mL, 22.71 mmol) were added to a solution of the above residue in 1,4-dioxane (50 mL). The resulting mixture was stirred at room temperature for 46 h, neutralized with 1N hydrochloric acid and the solvents were evaporated under reduced pressure. The residue was purified on silica gel (15% MeOH-CHCl₃) to give **10** as a colorless viscous liquid (540 mg, 51.4%). ¹H NMR (CD₃OD): δ 5.93-5.85 (m, 2H, 2-*H* and 3-*H*), 4.83 (m, 1H, 1-*H*), 3.60-3.40 (m, 4H, 2 × CH₂OH), 2.20-2.10 (dd, 1H, 5-*H*), 1.91 (s, 3H, CH₃), 1.48-1.38 (dd, 1H, 5' -*H*); ¹³C NMR (CD₃OD): δ 170.5 (CON), 136.1 and 132.1 (2-C and 3-C), 65.11 and 65.08 (2 × CH₂O), 56.0 (1-C), 54.3 (4-C), 36.2 (5-C), 20.9 (CH₃); HRMS (FAB) calculated for C₉H₁₆NO₃: (M + H)⁺ 186.1131, found 186.1122.

4-Bis(hydroxymethyl)-1-[(2-amino-6-chloropyrimidin-4-yl)amino]cyclopent-2-ene (11). Bis hydroxymethyl compound **10** (575 mg, 3.11 mmol) was refluxed under nitrogen in a mixture of ethanol (11 mL) and 2N hydrochloric acid (16.1 mL) for 70 h. The solvent was removed under reduced pressure. Butanol (17 mL), 2-amino-4,6-dichloropyrimidine (1.02 g, 6.22 mmol) and diisopropylethylamine (7.73 mL) were added and the mixture refluxed under nitrogen for 28 h, then poured into water (21 mL) and extracted with dichloromethane (4 × 80 mL). The combined organic layers were dried (MgSO₄) and concentrated. Column chromatography (11% MeOH-CHCl₃) gave compound **11** (425 mg, 50.6%) as a white foam, mp 108-110 °C. ¹H NMR (CD₃OD): δ 5.80-5.65 (m, 3H, 2' -*H*, 3' -*H* and pyrimidine-5-*H*), 4.78 (m, 1H, 1' -*H*), 3.54-3.38 (m, 4H, 2 × CH₂OH), 2.24-2.08 (dd, 1H, 5' -*H*), 1.48-1.33 (dd, 1H, 5'' -*H*); ¹³C NMR (CD₃OD): δ 163.0 (2-C), 162.3 (4-C), 135.6 and 132.3 (2' -C and 3' -C), 108.7 (6-C), 107.4 (5-C), 65.03 (2 × CH₂O), 55.82 and 55.80 (4' -C and 1' -C), 36.32 (5' -C).

4-Bis(hydroxymethyl)-1-[(2,5-diamino-6-chloropyrimidin-4-yl)amino]cyclopent-2-ene (12). A solution of sodium nitrite (129 mg, 1.85 mmol) in water (1.5 mL) was added dropwise to a solution of 4-chloroaniline (217 mg, 1.70 mmol) in 2.7 N hydrochloric acid (3.7 mL) at 9 °C. The resultant solution was added quickly dropwise to a solution of pyrimidine **11** (400 mg, 1.48 mmol) and sodium acetate (1.78 g) in water (7.4 mL) and glacial acetic acid (7.4 mL) at room temperature. The mixture was stirred vigorously for 44 h and the bright orange solid was filtered, washed with water until neutral and crystallized from ethyl acetate to afford 4-

bis(hydroxymethyl)-1-[[2-amino-6-chloro-5-[(4-chlorophenyl)azo]-4-pyrimidinyl]amino]-cyclopent-2-ene (431 mg, 71.3%) as a bright yellow solid, mp 252.8-253.4 °C (decomp.). ¹H NMR (CD₃OD): δ 10.41 (d, NH), 7.8-7.7 (d, 2H, Ar-H), 7.48-7.38 (d, 2H, Ar-H), 5.96-5.84 (m, 2H, 2' -H and 3' -H), 5.46-5.30 (m, 1H, 1' -H), 3.64-3.48 (m, 4H, 2 × CH₂OH), 2.40-2.33 (dd, 1H, 5' -H), 1.75-1.58 (dd, 1H, 5'' -H).

This azo compound (430 mg, 1.051 mmol), zinc dust (740 mg, 11.32 mmol), glacial acetic acid (0.36 mL), ethanol (17.5 mL) and water (17.5 mL) was refluxed under nitrogen for 3 h. The mixture was filtered and concentrated, azeotroping several times with ethanol. The residue was dissolved in methanol and absorbed onto silica. Column chromatography (10% MeOH-CHCl₃) gave the triamine **12** as a white solid (185 mg, 61.6%), mp 81.2-82.8 °C. ¹H NMR (CD₃OD): δ 6.35(d, NH), 5.85-5.75 (m, 2H, 2' -H and 3' -H), 5.15 (m, 1H, 1' -H), 3.62-3.49 (m, 4H, 2 × CH₂OH), 2.23 (dd, 1H, 5' -H), 1.53 (dd, 1H, 5'' -H); HRMS (FAB) calculated for C₁₁H₁₇ClN₅O₂: (M + H)⁺ 286.1072, found 286.1065.

2-Amino-6-chloro-9-[4-bis(hydroxymethyl)-2-cyclopenten-1-yl]-9H-purine (13). Pyrimidine **12** (177 mg, 0.62 mmol) was added to conc. hydrochloric acid (0.15 mL) in freshly distilled triethyl orthoformate (3.8 mL) and the mixture stirred for 31 h at room temperature. The mixture was concentrated and the resultant solid stirred in 0.5 N hydrochloric acid (6 mL) for 1 h when sufficient 1N sodium hydroxide solution was added to adjust the mixture to pH 8. The mixture was concentrated, dissolved in methanol and absorbed onto silica. Column chromatography (12% MeOH-CHCl₃) gave **13** (157 mg, 85.7%) as a white foam-like solid, mp 179.1-179.5 °C. ¹H NMR (CD₃OD): δ 8.08 (s, 1H, 8-H), 6.10-5.90 (m, 2H, 2' -H and 3' -H), 5.62 (m, 1H, 1' -H), 3.65-3.45 (m, 4H, 2 × CH₂OH), 2.47 (dd, 1H, 5' -H), 1.89 (dd, 1H, 5'' -H); ¹³C NMR (CD₃OD): δ 159.6 (6-C), 152.9 (2-C), 149.5 (4-C), 141.0 (5-C), 140.0 (8-C), 129.3 (2' -C), 123.2 (3' -C), 65.2 and 64.5 (2 × CH₂O), 59.36 (1' -C), 57.3 (4' -C), 36.3 (5' -C).

2-Amino-1,6-dihydro-6-oxo-9-[4-bis(hydroxymethyl)-2-cyclopenten-1-yl]-9H-purine (6).

Chloropurine **13** (146 mg, 0.49 mmol) was gently refluxed under nitrogen in 0.33 N sodium hydroxide (6.8 mL) for 5 h. The mixture was concentrated in vacuo, dissolved in methanol and absorbed onto silica. Column chromatography (30% MeOH-CHCl₃) gave **6** contaminated with silica. Further purification by ion-exchange chromatography using DOWEX 50 W × 8-200 ion exchange resin gave **6** (90 mg, 65.8%) as white solid which was crystallized from methanol, mp 273.8-274.1 °C. UV (0.1 N HCl) λ_{max} = 253 and 278 nm (ε = 14700 and 11300), ¹H NMR (DMSO-d₆): δ 10.32 (brs, 1H, NH), 7.60 (s, 1H, 8H), 6.42 (s, 2H, NH₂), 5.94-5.81 (m, 2H, 2' -H and 3' -H), 5.39 (m, 1H, 1' -H), 4.68-4.62 (m, 2H, OH), 3.49-3.28 (m, 4H, 2 × CH₂OH), 2.30-2.25 (dd, 1H, 5' -H), 1.70-1.60 (dd, 1H, 5'' -H); ¹³C NMR (DMSO-d₆): δ 156.7, 153.3, 150.7, 141.2, 139.2, 136.0, 131.0, 116.5, 116.4, 64.87, 64.55, 57.34; HRMS (FAB) calculated for C₁₂H₁₆N₅O₃: (M + H)⁺ 278.1254, found 278.1248.

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