

Synthesis, X-ray crystal structure and antimycobacterial activity of enantiomerically pure 1-ethyl-2,3-dicyano-5-(het)aryl-6-hetaryl-1,6-dihydropyrazines

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Dedicated to Professor Oleg N. Chupakhin on the occasion of his 80th anniversary

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Abstract

The Petasis reaction of 6-alkoxy adducts of 1-alkyl-2,3-dicyano-5-arylpyrazinium salts with aromatic boronic acids, such as 2-thienylboronic, 2-furanylboronic and 3-thienylboronic acids, or their benzo analogs in dichloromethane proceeds smoothly at room temperature with the formation of the corresponding 5-aryl-6-hetaryl substituted 1,6-dihydropyrazine derivatives. All dihydropyrazines were separated as pure enantiomers by chiral HPLC, and their absolute configurations for each pair of enantiomers have been determined by X-ray analysis. Individual enantiomers were screened *in vitro* for their antimycobacterial activities against *Mycobacterium tuberculosis* H₃₇Rv, *avium*, *terrae* and extensively drug-resistant and multi-drug-resistant strains isolated from tuberculosis patients in Ural region (Russia). It has been shown that several compounds exhibit a good level of antituberculosis activity compared to the reference drugs.

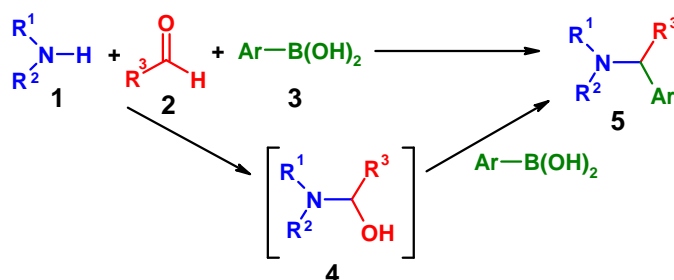
Keywords: Pyrazines, Petasis reaction, antituberculosis activity, extensively drug-resistant and multi-drug-resistant tuberculosis

Introduction

Tuberculosis is an illness that results from infection with *Mycobacterium tuberculosis* (MTB). Tuberculosis is responsible for millions of human deaths annually, claiming more lives than any

other single infectious agent.¹⁻³ Almost one-third to one-half of the world's population is infected with *M. tuberculosis* and approximately 10% of infected individuals will experience active disease at some time in their life.⁴ Currently, treatment for active Tuberculosis includes at least 6 months of therapy with the first line drugs, such as isoniazid, rifampicin, pyrazinamide and ethambutol.⁵⁻⁷ Failure of patients to complete a full treatment protocol and the outbreak of acquired immune deficiency syndrome (AIDS) has resulted in the emergence of multi-drug resistance (MDR). Recent estimates show that 10% of all new tuberculosis infections are resistant to at least one antituberculosis drug. Subsequently, the World Health Organization (WHO) has declared tuberculosis as global public health emergency.⁸ For multi-drug resistance (MDR) and extensively drug resistance (XDR) combinations of the first line drugs and the second line drugs, such as aminoglycosides, fluoroquinolones, thioamides, cycloserine and *p*-amino salicylic acid are used.⁹ Since the discovery of rifampicin in 1960, there is no more drugs developed to treat tuberculosis. The third line drugs include rifabutin, clarithromycin, linezolid, thiacetone, arginine and vitamin D, which are still being developed, have less or unproven efficacy and are very expensive.¹⁰ MTB have the cell with a high lipid content, thus resulting in a high degree of lipophilicity and resistance to alcohol, acids, alkali and some disinfectants.¹¹ It is worth noting that practically all drugs used to treat tuberculosis are capable of causing toxic side effects.¹¹ Therefore, it is obvious that new original drugs that are more active and effective than the existing ones are supposed to be discovered. In addition, pyrazinecarbonitrile analogs (pyrazinecarboxamide and pyrazine-2-carboxylic acid derivatives) are important and famous as the first-line antituberculosis drugs.¹¹

It is known that the addition reaction of 1,4-diazines and their benzo derivatives is of great importance for the synthesis of biologically active compounds, including alkaloids.¹²⁻¹⁷ In order to cause these transformations azaaromatic compounds are usually activated by quaternization to obtain *N*-alkyl-1,4-diazinium salts.^{17,18} Indeed, it has recently been shown that 1-alkyl-1,4-diazinium cations are prone to add carbo- and heteroatomic nucleophiles to give mono- and diadducts, or to be transformed into condensed tetrahydropyrazines through the tandem addition reactions with bifunctional nucleophiles.¹⁷⁻²⁶ On the other hand, the Petasis reaction is one of the multicomponent condensation reactions, which provides a convenient method for the preparation of α -amino acids by reacting an amine **1** with a carbonyl compound **2** and a boronic acid **3**.²⁷⁻³⁴ The reaction proceeds via the formation of aminal intermediate **4** (Scheme 1).

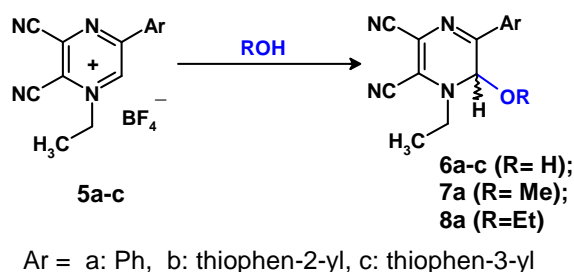


Scheme 1. The Petasis reaction with aryl boronic acid.

The reaction has also been applied to the synthesis of heterocyclic systems³⁵⁻⁴⁰ and solid phase synthesis of peptides.⁴¹⁻⁴³ The reaction is a very efficient with alkenyl and electron-rich aromatic boronic acids. In addition to primary amines, the secondary amines, sterically hindered primary amines, and tertiary aromatic amines can also participate in these transformations. A mild, practical, and novel method for the C–C bond formation in the series of pyrazine (quinoxaline) derivatives by using the Petasis reaction has been developed through intermediacy of 6-alkoxy-1-ethyl-1,6-dihydropyrazines. All obtained dihydropyrazines were screened *in vitro* for their antimycobacterial activities against *Mycobacterium tuberculosis* H₃₇Rv, *avium*, *terrae* and extensively drug-resistant and multi-drug-resistant strains isolated from tuberculosis patients in Ural region (Russia). It has been found that some of these compounds exhibit a high tuberculostatic activity comparable with that of pyrazinamide.

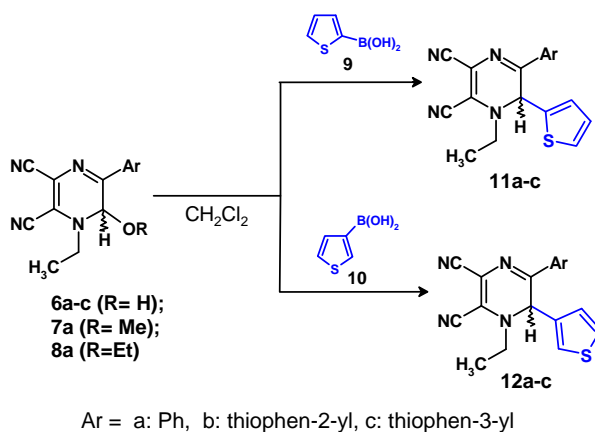
Results and Discussion

We have earlier carried out a series of studies on the properties of substituted pyrazinium salts **5a-c**.^{22,23,44-50} It has been shown that they are capable of reacting with various types of nucleophiles under mild conditions to form products of nucleophilic addition to the unsubstituted positions of the pyrazine ring. The aim of the present study is to examine the further modifications of 6-alkoxy adducts **6-8** derived from the reaction of 5-(het)aryl-2,3-dicyano-1-ethylpyrazinium tetrafluoroborates **5a-c** with O-nucleophiles (water and alcohols) (Scheme 2).²³



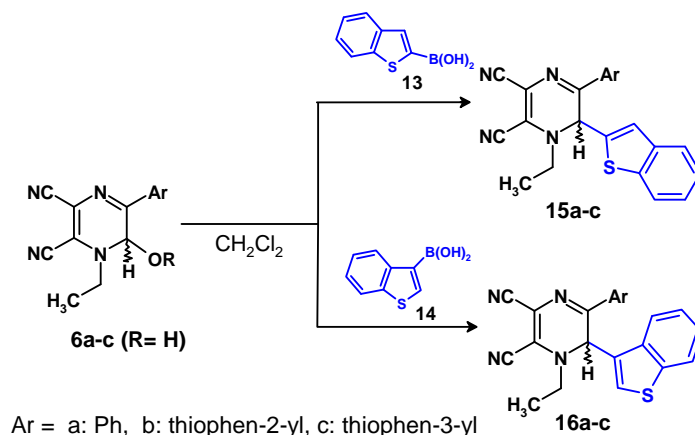
Scheme 2

Taking into account a structural analogy between O-adducts **6-8** and amino alcohols **4** derived from the Petasis reaction, we have investigated the reaction of **6-8** with thienylboronic acids. It has been established that the alkoxy adducts **6-8** react readily with 2-thienylboronic **9** and 3-thienylboronic **10** acids under mild conditions (CH₂Cl₂, 20-25 °C, 1 hour) to give 1-ethyl-5-(het)aryl-6-thiophen-2-yl-1,6-dihydropyrazine-2,3-dicarbonitriles **11a-c** and 1-ethyl-5-(het)aryl-6-thiophen-3-yl-1,6-dihydropyrazine-2,3-dicarbonitriles **12a-c** (Scheme 3, Table 1).



Scheme 3

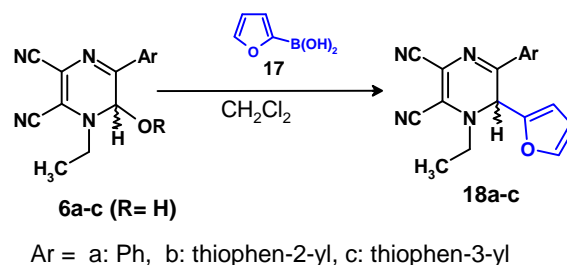
In a similar way, the reaction of the hydroxy adducts **6a-c** with benzo[*b*]thien-2-ylboronic **13** and benzo[*b*]thien-3-ylboronic **14** acids leads to the corresponding 6-benzo[*b*]thiophen-2-yl-1-ethyl-5-(het)aryl-1,6-dihydropyrazine-2,3-dicarbonitriles **15a-c** and 6-benzo[*b*]thiophen-3-yl-1-ethyl-5-(het)aryl-1,6-dihydropyrazine-2,3-dicarbonitriles **16a-c** in good yields 55-98% (Scheme 4, Table 1).



Scheme 4

On treatment of the adducts **6a-c** with 2-furanylboronic acid (**17**) the corresponding 1-ethyl-6-furan-2-yl-5-(het)aryl-1,6-dihydropyrazine-2,3-dicarbonitriles **18a-c** were obtained in yields 55-98% (Scheme 5, Table 1).

The evidence for the structures of all di(hetero)aryl-substituted dihydropyrazines has been obtained by means of ^1H and ^{13}C NMR spectroscopy, and X-ray crystallography analysis performed for racemic 1-ethyl-5-phenyl-6-thiophen-2-yl-1,6-dihydropyrazine-2,3-dicarbonitrile (**11a**), 1-ethyl-5-thiophen-3-yl-6-thiophen-2-yl-1,6-dihydro-pyrazine-2,3-dicarbonitrile (**11c**) and 1-ethyl-5-phenyl-6-thiophen-3-yl-1,6-dihydropyrazine-2,3-dicarbonitriles (**12a**) (Figs. 1-3).



Scheme 5

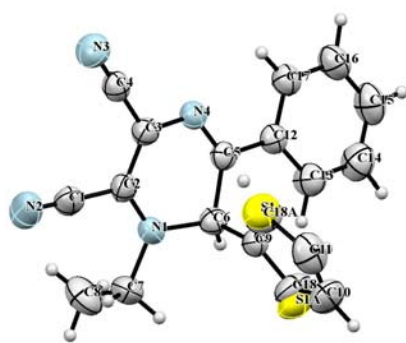


Figure 1. X-ray structure of (±)-**11a**.

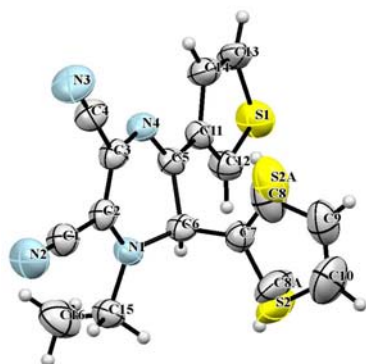


Figure 2. X-ray structure of (±)-**11c**.

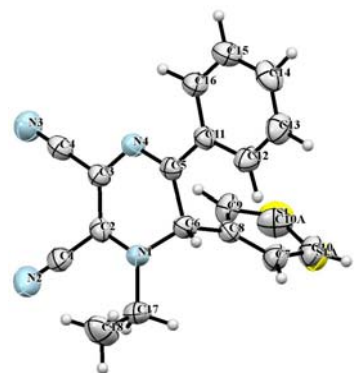


Figure 3. X-ray structure of (±)-**12a**.

In case of highly reactive alkoxy adducts **6a**, **7a** and **8a** yields of dihydropyrazine **11a** are nearly independent of the substituents R in the alkoxy-group (see entries 1-3).

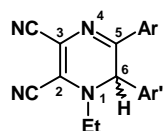
The reactions of alkoxy adducts **6a-c**, **7a**, **8a** with π -rich heteroaromatic boronic acids proved to proceed in good yields (entries 1-17), however phenylboronic acid did not enter the reaction at all due to its low reactivity.⁵¹ It should be noted that the Petasis type reaction for 1,4-diazines is observed for the first time, although similar transformations have been described earlier for derivatives of quinoline and isoquinoline.⁵¹

According to the X-ray data, all racemic compounds **11a,c** and **12a** are crystallized in centrosymmetric $P2_1/n$ space group of monoclinic crystal system, and are characterized by the similarity of magnitudes in unit cell parameters (Tables 2, 4).

Table 1. Reaction conditions and yields of compounds **11**, **12**, **15**, **16** and **18**

Entry	Reaction	Time, h	Product	Isolated yields (%)	Yield on GC-MS (%)
1	6a+9	1	11a	76	97
2	7a+9	1	11a	n.d.	98
3	8a+9	1	11a	n.d.	97
4	6a+10	20	12a	57	n.d.
5	6b+9	20	11b	68	n.d.
6	6b+10	20	12b	64	n.d.
7	6c+9	20	11c	65	n.d.
8	6c+10	20	12c	79	n.d.
9	6a+13	48	15a	55	n.d.
10	6b+13	17	15b	58	n.d.
11	6c+13	17	15c	98	n.d.
12	6a+14	48	16a	59	n.d.
13	6b+14	17	16b	76	n.d.
14	6c+14	17	16c	80	n.d.
15	6a+17	24	18a	75	n.d.
16	6b+17	24	18b	86	n.d.
17	6c+17	24	18c	71	n.d.

n.d. – not determined

Table 2. Some stereometric characteristics of dihydropyrazines according to the X-Ray data. A numeration of ring's atoms is given according to the figure below

Compound	(±)-11a	(±)-11c	(±)-12a
d(C6-N1C2C3C4C5), Å.	0.408	0.499	0.394
Dihedral angle	21.47	26.75	20.37
C6N1C2C3/C3N4C5C6, deg.			
Dihedral angle	27.59	33.96	26.45
N1C2C3C4C5/C5C6N1, deg.			
C6-C5, Å	1.518(2)	1.517(3)	1.533(3)
C6-C(Ar'), Å	1.5066(19)	1.490(3)	1.499(3)
C5-C(Ar), Å	1.4806(19)	1.455(3)	1.472(3)
N1-C2, Å	1.3456(18)	1.348(3)	1.342(2)

Table 2. Continued

C6-N1, Å	1.4769(18)	1.470(2)	1.472(2)
Angle C5C6N1, deg.	109.89(12)	108.82(16)	109.79(18)
C2-C(N), Å	1.446(2)	1.430(3)	1.442(3)
C3-C(N), Å	1.429(2)	1.431(3)	1.432(3)
(C2)C≡(N), Å	1.1401(18)	1.139(3)	1.133(3)
(C3)C≡(N), Å	1.1431(18)	1.136(3)	1.141(2)

Low variations in unit cell parameters for these structures are rationalised by the very similar packing requirements of the thiophene and benzene ring substituents in compounds **11a/12a** and **11c**. The dihydropyrazine ring in compounds is non-planar, the sp^3 -hybridized atom is deviated from the least-squared plane of the ring on 0.4-0.5 Å, and dihedral angles between the planes C6N1C2C3 and C3N4C5C6 are in limits 20-30° (numeration of atoms is given according to the figure for Table 2). However, atoms C3, N4, C5 are placed approximately in the plane of (het)aryl substituent at C5 and the double C=N bond of the azadienic fragment is obviously conjugated with the π -electronic system of the substituent. Measured sterical parameters of the dihydropyrazinic system, in particular, deviation of the sp^3 -atom from a plane of the pyrazine ring d(C6-N1C2C3C4C5), dihedral angles between planes C6N1C2C3/C3N4C5C6 and N1C2C3C4C5/C5C6N1, are depending not from the type of substituents at the sp^3 -hybridised C-atom (Ar'), but from effects of the dihydropyrazine azadiene system conjugation with Ar substituent at C5. For pair **11a/12a** a good structural correspondence is observed, however at the same time the structural parameters for the pair **11a/11c** show significant differences (see Table 2). The next structural differences of dihydropyrazines are connected with their conformations and molecular packing. Dihedral angle between the plane of substituent Ar' at the sp^3 -hybridized C-atom and the plane C5C6C(Ar') for the pair **11a/12a** is ~50°, the same angle for compound **11c** is ~20° (exact definition of magnitudes is not possible due to disordering of substituent Ar'). Despite on proximity of unit cell parameters, molecular packing for the pair **11a/12a** and compound **11c** show some variations. In particular, molecular packing for the pair **11a/12a** is characterized by the presence of π -contacts between C≡N-groups and the azadiene system of the dihydropyrazine ring with distance ~3.3 Å. This type of interaction is absent in packing of compound **11c**, and their molecules form layers parallel 0b axe by means of axial N...S type contacts with distances ~3.25 Å between C≡N-groups and S-atoms of thienyl substituent at C5. All dihydropyrazines (\pm)-**11**, **12**, **15**, **16** and **18** were separated as pure enantiomers [enantiomeric excess approximately 95-99.9%] by semi-preparative chiral HPLC using Kromasil 5-CelluCoat (150 × 4.6 mm, 5 μ m) column. Chiral HPLC and polarimetric analysis indicated that for all racemates, the first eluted enantiomers rotated polarized light in the negative direction, accordingly to a similar nature of substituents at the stereogenic center C(6). Crystals of enantiomeric pairs **R(-)-11b/S(+)-11b** and **R(-)-11c/S(+)-11c** suitable for X-ray experiments were grown by crystallization from acetonitrile solution. Enantiomeres **R(-)-11b/S(+)-11b** and **R(-)-11c/S(+)-11c** were crystallized in very typical chiral P₂₁2₁2₁ space group of

orthorhombic crystal system with the nearest magnitudes in unit cell parameters (Tables 5, Figs. 4 and 5). All deviations in magnitudes in unit cell parameters for the pair of enantiomers can be explained by variation in temperatures of experiments and experimental errors. The presence of sulfur atoms in their structures permits to establish an absolute configuration of compounds. The absolute configuration of compounds was established on the basis of anomalous scattering effect (by Flack parameters), however, only for the ***R*-(-)-11b** and ***S*-(+)-11c** isomers their parameters are significant (Table 5). For the isomers ***R*-(+)-11b** and ***S*-(-)-11c** their parameters are less than the measurement mistakes, but are in accordance with measurements for enantiopodes. General stereochemistry of the enantiomers is in good agreement with the stereochemistry of racemates. Some variations in dihedral angles due to rotation of a substituent around the single C-C bond's axis can be explained by changing in the molecular packing environment. Any shortened intermolecular contacts in the molecular packing are absent.

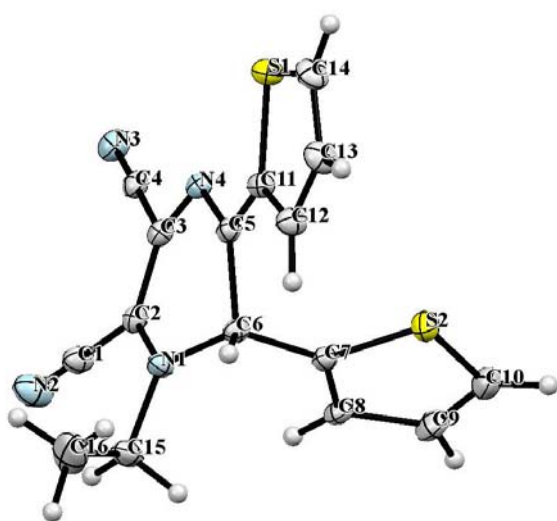


Figure 4. X-ray structure of enantiomer ***R*-(-)-11b**.

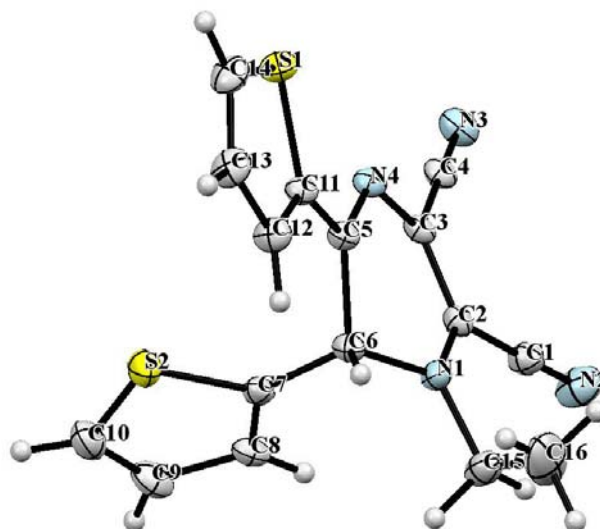


Figure 5. X-ray structure of enantiomer ***S*-(+)-11b**.

On the basis of combined chromatographic and X-ray studies, the absolute configuration assignment was unambiguously completed, and configurations were assigned to the fifteen enantiomeric pairs: ***R*-(-)-11a-c**, **12a-c**, **15a-c**, **16a-c**, **18a-c**/***S*-(+)-11a-c**, **12a-c**, **15a-c**, **16a-c**, **18a-c**. Due to comparable chiral centers in this group of molecules, the results clearly indicate at the (*S*)-absolute configuration for all dextrorotatory enantiomers of the series and at the (*R*)-absolute configuration for the corresponding levorotatory isomers.

The antimycobacterial activity of compounds **11**, **12**, **15**, **16** and **18** was tested against *M. tuberculosis* *H*₃₇*Rv*, *avium*, *terrae* and extensively drug-resistant and multi-drug-resistant strains. All new dihydropyrazines were compared with the commercially available drugs Isoniazid and Pyrazineamide under the same experimental conditions (Table 3).

Table 3. Antimycobacterial activity of 1,6-dihydropyrazines

Entry	Compound	Antimycobacterial activity against different strains of <i>Mycobacterium tuberculosis</i> (MIC in µg/mL)				
		H ₃₇ Rv	<i>M. avium</i>	<i>M. terrae</i>	MDR	XDR
1	R(-)-11a	6.2	6.2	6.2	6.2	6.2
2	S(+)-11a	6.2	6.2	6.2	6.2	6.2
3	R(-)-11b	6.2	6.2	6.2	6.2	6.2
4	S(+)-11b	6.2	6.2	6.2	6.2	6.2
5	R(-)-11c	6.2	3.1	3.1	6.2	6.2
6	S(+)-11c	6.2	6.2	6.2	6.2	6.2
7	R(-)-12a	1.5	0.75	0.75	1.5	1.5
8	S(+)-12a	3.1	3.1	1.5	3.1	3.1
9	R(-)-12b	0.75	0.3	0.3	0.75	0.75
10	S(+)-12b	1.5	0.75	0.75	1.5	1.5
11	R(-)-12c	6.2	6.2	6.2	6.2	6.2
12	S(+)-12c	6.2	6.2	6.2	6.2	6.2
13	R(-)-15a	12.5	12.5	12.5	12.5	12.5
14	S(+)-15a	12.5	12.5	12.5	12.5	12.5
15	R(-)-15b	12.5	12.5	12.5	12.5	12.5
16	S(+)-15b	12.5	12.5	12.5	12.5	12.5
17	R(-)-15c	12.5	12.5	12.5	12.5	12.5
18	S(+)-15c	12.5	12.5	12.5	12.5	12.5
19	R(-)-16a	12.5	12.5	12.5	12.5	12.5
20	S(+)-16a	12.5	12.5	12.5	12.5	12.5
21	R(-)-16b	1.5	0.75	0.75	1.5	1.5
22	S(+)-16b	12.5	12.5	12.5	12.5	12.5
23	R(-)-16c	1.5	0.75	0.75	1.5	1.5
24	S(+)-16c	6.2	6.2	6.2	6.2	6.2
25	R(-)-18a	12.5	12.5	12.5	12.5	12.5
26	S(+)-18a	12.5	12.5	12.5	12.5	12.5
27	R(-)-18b	0.75	0.3	0.3	0.75	0.75
28	S(+)-18b	12.5	12.5	12.5	12.5	12.5
29	R(-)-18c	12.5	12.5	12.5	12.5	12.5
30	S(+)-18c	12.5	12.5	12.5	12.5	12.5
	Pyrazinamide	12.5	n.d.	n.d.	-	-
	Isoniazid	0.1	0.1	0.1	-	-

n.d. – not determined; MDR (multi-drug-resistant tuberculosis strain) – Rifampin and isoniazid resistant *Mycobacterium tuberculosis* strain having Beijing genotype with a combination of

mutations *Ser 531 - Leu 315* and *Ser-Thr* in *rpoB* and *katG* genes, respectively; XDR – extensively drug-resistant tuberculosis strain.

It has been established that the most active compounds of the series contain thiophenyl groups, while a low activity was found for the phenyl-substituted dihydropyrazines. Compounds, bearing 3-thiophenyl **12a,b** and 3-benzo[*b*]thiophenyl **16b,c** fragments at C(6) position of the dihydropyrazine ring, were highly antimicrobial and inhibited *M. tuberculosis* growth at concentration of 0.75-1.5 µg/mL. Incorporation of 2-thiophenyl **11a-c** group at C(6) position of the dihydropyrazine ring resulted in a slightly less activity (MIC 6.2 µg/mL). In contrast, introduction of other groups, such as 2-benzo[*b*]thiophenyl (**15a-c**) or 2-furyl (**18a,c**), resulted in rather poor activity (MIC 12.5 µg/mL). Furthermore, the most active compounds exhibited enantiomer selectivity, generally preferring the *R*-(-)-forms over the *S*-(+)-enantiomers, as demonstrated by the data for compounds *R*-(-)-**12a**, *R*-(-)-**12b**, *R*-(-)-**16b**, *R*-(-)-**16c** and *R*-(-)-**18b** (see Table 3).

The results obtained show that antimycobacterial activity of several derivatives of the series of 1-alkyl-2,3-dicyano-5-aryl-6-hetaryl-1,6-dihydropyrazines is similar to the standard drugs. In addition, the compounds of this series are active against extensively drug-resistant and multi-drug-resistant strains of *Mycobacterium tuberculosis*.

Conclusions

It has been shown that the Petasis reaction provides a mild, practical, and novel method for noncatalytic synthesis of dihydropyrazine derivatives. Novel 1-alkyl-2,3-dicyano-5-(het)aryl-6-hetaryl-1,6-dihydropyrazines have been obtained and characterized. The structures of these compounds have been established by X-ray crystallography analysis. Several compounds of this series proved to possess a better antimycobacterial activity, than those from the family of classic antituberculosis drugs, for example Pyrazinamide. Indeed, 6-thienyl and 6-furyl substituted 1-alkyl-2,3-dicyano-5-(het)aryl-6-hetaryl-1,6-dihydropyrazines can be regarded as promising candidates for the further development of antitubercular agents.

Experimental Section

General. All solvents and reagents were purified and prepared according to the literature data.⁵² Compounds **6-8** were synthesized according to the procedures described in the literature.²³ ¹H and ¹³C NMR spectra were recorded on a Bruker AVANCE-500 instrument using Me₄Si as an internal standard. All signals in the ¹H and ¹³C NMR spectra were assigned on the basis of 2D ¹H-¹H COSY, ¹H-¹³C HSQC and HMBC experiments. Elemental analysis was carried on a Eurovector EA 3000 automated analyzer. Melting points were determined on Boetius combined

heating stages and were not corrected. The GC-MS analysis of all samples was carried out using an Agilent GC 7890A MS 5975C Inert XL EI/CI GC-MS spectrometer with a quadrupole mass-spectrometric detector with electron ionization (70 eV) and scan over the total ionic current in the range m/z 20–1000 and a quartz capillary column HP-5MS (30 m \times 0.25 mm, film thickness 0.25 mm). Helium was used as a carrier gas, the split ratio of the flow was 1 : 50, and the consumption through the column was 1.0 mL min^{-1} ; the initial temperature of the column was 40 $^{\circ}\text{C}$ (storage 3 min), programming rate was 10 $^{\circ}\text{C min}^{-1}$ to 290 $^{\circ}\text{C}$ (storage 20 min), the temperature of the evaporator was 250 $^{\circ}\text{C}$, the temperature of the source was 230 $^{\circ}\text{C}$, the temperature of the quadrupole was 150 $^{\circ}\text{C}$, and the temperature of the transition chamber was 280 $^{\circ}\text{C}$. Solutions of the samples with a concentration of 3–4 mg mL^{-1} were prepared in acetonitrile. Samples of 1 mL of the obtained solutions were analyzed. Column chromatography was carried out using Lancaster silica gel 0.040–0.063 mm (230–400 mesh), eluting with ethyl acetate–hexane, 1 : 2. The progress of reactions and the purity of compounds were checked by TLC on Sorbfil plates (Russia), in which the spots were visualized with UV light (λ 254 or 365 nm).

Microwave experiments were carried out in a Discover unimodal microwave system (CEM, USA) with a working frequency of 2.45 GHz and the power of microwave radiation ranged from 0 to 300 W. The reactions were carried out in a 10 mL reaction tube with the hermetic Teflon cork. The temperature of the reaction was monitored using an inserted IR sensor by the external surface of the reaction vessel.

Semi-preparative HPLC enantioseparations were performed on an Agilent 1100 instrument using stainless-steel Kromasil 5-CelluCoat (150 \times 4.6 mm, 5 μm) column. The analytical HPLC to determine the enantiomeric excess of the optically pure compounds was performed on the same instrument using chiral Kromasil CelluCoat (150 \times 4.6 mm, 5 μm) column and mobile phase of $\text{CH}_3\text{CN-H}_2\text{O}$ at the different ratio. The values of retention time (t_R) were given in minutes. Enantiomeric excess values were calculated from peak areas using the formula $(A_1 - A_2)/(A_1 + A_2)$, where A_1 is the major enantiomer. Specific rotations of enantiomers were measured at 589 nm with a Perkin-Elmer polarimeter model 341 equipped with a Na lamp. The volume of the cell was 1 mL, and the optical path was 10 cm. A standard solution was prepared by dissolving 5 mg of compound into 2 mL of CH_3CN ($c = 0.25$).

X-ray diffraction analysis of single crystals of **11a,c**, **12a** and **R(-)-11c** was performed on an automated X-ray diffractometer “Xcalibur 3” equipped with CCD detector. Compounds **R(-)-11b**, **S-(+)-11b**, **S-(+)-11c** were analyzed on an automated X-ray diffractometer “Xcalibur E” on a standard procedure (λMoK_α , graphite monochromator, $\omega/2\theta$ scanning with step 1°). Empirical absorption correction was performed. Structures of compounds **11a,c**, **12a** and **R(-)-11c** were solved and refined by full-matrix least-squared method on F^2 with using SHELXTL program package.⁵³ Compounds **R(-)-11b**, **S-(+)-11b**, **S-(+)-11c** were solved and refined by using Olex2 program.⁵⁴ Non hydrogen atoms were refined in anisotropic approximation; H-atoms were refined in isotropic approximation in riding model. General X-Ray data are summarized in the Table 4. The X-ray crystallography data for structures **11a-c** and **12a** reported in this paper have

been deposited with Cambridge Crystallography Data Centre as supplementary publications no CCDC 991536 for (\pm)-**11a**, CCDC 991537 for (\pm)-**11c**, CCDC 991538 for (\pm)-**12a**, CCDC 991532 for (+)-**11b**, CCDC 991533 for (-)-**11b**, CCDC 991534 for (+)-**11c** and CCDC 991535 for (-)-**11c**. These data can be obtained free of charge from the Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

Table 4. General X-ray structural data for racemic compounds **11a,c** and **12a**

Compound	11a	11c	12a
Crystal size, mm	0.25×0.20×0.15	0.25×0.20×0.15	0.25×0.20×0.15
Crystal color	Orange	orange	orange
Empirical formula	C ₁₈ H ₁₄ N ₄ S	C ₁₆ H ₁₂ N ₄ S ₂	C ₁₈ H ₁₄ N ₄ S
Temperature, K	295(2)	295(2)	295(2)
Crystal system	Monoclinic	Monoclinic	Monoclinic
Space group	P2 ₁ /n	P2 ₁ /n	P2 ₁ /n
a, Å	10.1483(9)	8.1995(7)	10.1126(17)
b, Å	16.0775(16)	16.8758(15)	15.9301(14)
c, Å	10.3350(10)	12.2379(9)	10.2752(13)
α	90	90	90
β	107.086(8)	108.675(7)	107.097(13)
γ	90	90	90
Volume (Å ³), Z	1611.8(3), 4	1604.2(2), 4	1582.1(4), 4
μ , mm ⁻¹	0.205	0.333	0.209
Reflections collected	10204	8088	5918
Independent reflections(R _{int})	3958 (0.0291)	3862 (0.0219)	3157 (0.0311)
Reflections with I>2 σ (I)	1615	1860	1354
GooF	1.002	0.990	1.002
R ₁ [I>2 σ (I)]	0.0348	0.0455	0.0362
wR ₂ [I>2 σ (I)]	0.0710	0.1231	0.0492
R ₁ (all data)	0.0995	0.0946	0.0951
wR ₂ (all data)	0.0748	0.1334	0.0512
Largest diff. peak and hole, e ⁻ /Å ³	0.145/-0.220	0.258/-0.300	0.167/-0.237
Completeness to θ (deg.)	98.9 % (28.28)	98.8 % (26.00)	97.5 % (26.38)

Table 5. General X-ray structural data for enantiomeric compounds **R-(-)-11b** /**S-(+)-11b** and **R-(-)-11c** /**S-(+)-11c**

Compound	R-(-)-11b	S-(+)-11b	R-(-)-11c	S-(+)-11c
Crystal size, mm	0.26×0.21×0.17	0.23×0.17×0.08	0.30x0.25x0.20	0.25×0.2×0.15
Crystal color	orange	orange	orange	orange
Empirical formula	C ₁₆ H ₁₂ N ₄ S ₂	C ₁₆ H ₁₂ N ₄ S ₂	C ₁₆ H ₁₂ N ₄ S ₂	C ₁₆ H ₁₂ N ₄ S ₂
Temperature, K	120.00(10)	150.0(1)	295(2)	149.91(18)
Crystal system	orthorhombic	orthorhombic	orthorhombic	orthorhombic
Space group	P2 ₁ 2 ₁ 2 ₁	P2 ₁ 2 ₁ 2 ₁	P2 ₁ 2 ₁ 2 ₁	P2 ₁ 2 ₁ 2 ₁
a, Å	7.2966(3)	7.30865(17)	7.3139(10)	7.24218(12)
b, Å	12.5764(5)	12.6253(4)	12.8680(6)	12.7486(2)
c, Å	17.2086(7)	17.2173(4)	17.0494(18)	17.0675(3)
α	90	90.00	90.00	90.00
β	90	90.00	90.00	90.00
γ	90	90.00	90.00	90.00
Volume (Å ³), Z	1579.15(11)	1588.71(7), 4	1604.6(3), 4	1575.80(4), 4
μ, mm ⁻¹	0.338	0.336	0.333	0.339
Reflections collected	9187	7269	11930	10783
Independent reflections(R _{int})	5200 (0.0295)	4330 (0.0237)	3981 (0.0376)	6291 (0.0196)
Reflections with I>2σ(I)	4745	3980	1748	5655
GooF	1.002	1.219	1.000	1.001
R ₁ [I>2σ(I)]	0.0392	0.0369	0.0472	0.0386
wR ₂ [I>2σ(I)]	-	0.0905	0.0906	0.1084
R ₁ (all data)	0.0458	0.0425	0.1053	0.0457
wR ₂ (all data)	0.1021	0.1009	0.0945	0.1173
Largest diff. peak and hole, e ⁻ /Å ³	0.59/-0.28	0.46/-0.33	0.342/-206	0.47/-0.35
Flack parameter			0.06(9)	0.07(6)
Completeness to θ (deg.)			99.9 % (28.30)	99.80 % (33.07)

General procedure for the synthesis of 5,6-disubstituted 1,6-dihydropyrazine derivatives (11,12,15,16 and 18). A mixture of compound **6** (**7** or **8**) (0.5 mmol) and hetero(aryl)boronic (2-thienylboronic, 3-thienylboronic, benzo[*b*]thien-2-ylboronic, benzo[*b*]thien-3-ylboronic or 2-furanylboronic) acid (0.6 mmol) was dissolved in 3 mL CH₂Cl₂. The resulting solution was stirred for an appropriate time (see Table 1) at room temperature, the solvent was distilled off *in vacuo*, and the residue was purified by flash column chromatography (hexane/ethyl acetate) to afford the desired di(hetero)aryl-substituted dihydropyrazine derivatives **11**, **12**, **15**, **16** and **18**.

1-Ethyl-5-phenyl-6-thiophen-2-yl-1,6-dihydropyrazine-2,3-dicarbonitrile (11a). Yield of the racemate (see Table 1, entries 1-3), yellow-orange powder, mp 135-137 °C. Racemate was separated by chiral HPLC with a Kromasil 5-CelluCoat (150 × 4.6 mm, 5 μm) column. The eluent mixture CH₃CN–H₂O 65:35 was used in isocratic mode (0.6 mL/min at 25 °C), and the UV signal was monitored at λ 268 nm to obtain the resolved enantiomeric pair **R(-)-11a/S(-)-11a** (*t_R* 7.37, *t_R* 11.00). **R(-)-11a**: [α]_D²⁰ -968 (c 0.25, CH₃CN), ee 98.7% (Kromasil 5-CelluCoat, CH₃CN–H₂O 70:30, *t_R* 3.93); **S(+)-11a**: [α]_D²⁰ +877 (c 0.25, CH₃CN), ee 97.4% (Kromasil 5-CelluCoat, CH₃CN–H₂O 70:30, *t_R* 5.48 min). δ_H (500 MHz, CDCl₃) 1.34 (t, 3H, CH₃, *J* 7.2 Hz), 3.71 (dq, 1H, NCH^B, *J* 14.5, 7.2 Hz), 3.76 (dq, 1H, NCH^A, *J* 14.5, 7.2 Hz), 5.96 (s, 1H, H6), 6.92 (dt, 1H, H3', *J* 3.6, 1.0 Hz), 6.96 (dd, 1H, H4', *J* 5.1, 3.6 Hz), 7.33 (dd, 1H, H5', *J* 5.1, 1.2 Hz), 7.44 (m, 2H, Hm;), 7.49 (m, 1H, Hp), 7.90 (m, 2H, Ho). δ_C (126 MHz, CDCl₃) 14.97 (CH₃), 48.33 (NCH₂), 52.36 (C6), 110.47 (CN), 111.31 (CN), 115.99 (C3), 118.24 (C2), 127.39 (C5'), 127.45 (C4'), 127.59 (C3'), 127.97 (Co), 129.01 (Cm), 131.93 (Cp), 133.70 (Ci), 136.68 (C2'), 147.37 (C5). GC *t_R* 28.85 min; MS *m/z* (rel intensity) 318 (M⁺, 100). Calcd. for C₁₈H₁₄N₄S (318.40): C 67.90, H 4.43, N 17.60. Found C 68.30, H 4.50, N 17.85.

1-Ethyl-5,6-dithiophen-2-yl-1,6-dihydropyrazine-2,3-dicarbonitrile (11b). Yield of the racemate (see Table 1, entry 5), orange powder, mp 124-126 °C. Racemate was separated by chiral HPLC with a Kromasil 5-CelluCoat (150 × 4.6 mm, 5 μm) column. The eluent mixture CH₃CN–H₂O 75:25 was used in isocratic mode (20 mL/min at 25 °C), and the UV signal was monitored at λ 280 nm to obtain the resolved enantiomeric pair **R(-)-11b/S(-)-11b** (*t_R* 2.92, *t_R* 3.46). **R(-)-11b**: [α]_D²⁰ -1673 (c 0.25, CH₃CN), ee 99.9% (Kromasil 5-CelluCoat, CH₃CN–H₂O 70:30, *t_R* 5.10); **S(+)-11b**: [α]_D²⁰ +1603 (c 0.25, CH₃CN), ee 93.5% (Kromasil 5-CelluCoat, CH₃CN–H₂O 70:30, *t_R* 6.32). δ_H (500 MHz, CDCl₃) 1.34 (t, 3H, CH₃, *J* 7.2 Hz), 3.68 (dq, 1H, NCH^B, *J* 14.5, 7.2 Hz), 3.73 (dq, 1H, NCH^A, *J* 14.5, 7.2 Hz), 5.88 (s, 1H, H6), 6.95-6.97 (m, 2H, H3' and H4'), 7.09 (dd, 1H, H4'', *J* 5.1, 3.8 Hz), 7.31 (m, 1H, H5'), 7.47 (dd, 1H, H3'', *J* 3.8, 1.1 Hz), 7.56 (dd, 1H, H5'', *J* 5.1, 1.1 Hz). δ_C (126 MHz, CDCl₃) 14.72 (CH₃), 48.27 (NCH₂), 53.39 (C6), 110.18 (CN), 111.34 (CN), 115.84 (C3), 118.21 (C2), 127.21, 127.28 and 127.41 (C3', C4' and C5'), 128.25 (C4''), 129.88 (C3''), 132.99 (C5''), 136.97 (C2'), 140.40 (C2''), 143.26 (C5). GC *t_R* 29.38 min; MS *m/z* (rel intensity) 324 (M⁺, 100). Calcd. for C₁₆H₁₂N₄S₂ (324.43): C 59.24, H 3.73, N 17.27, S 19.77. Found C 59.42, H 3.65, N 16.97, S 19.96.

1-Ethyl-5-thophen-3-yl-6-thophen-2-yl-1,6-dihydropyrazine-2,3-dicarbonitrile (11c). Yield of the racemate (see Table 1, entry 7), bright yellow powder, mp 113-114 °C. Racemate was separated by chiral HPLC with a Kromasil 5-CelluCoat (150 × 4.6 mm, 5 μm) column. The

eluent mixture CH₃CN–H₂O 65:35 was used in isocratic mode (0.8 mL/min at 25 °C), and the UV signal was monitored at λ 268 nm to obtain the resolved enantiomeric pair **R-(-)-11c/S-(-)-11c** (t_R 4.67, t_R 6.18). **R-(-)-11c**: $[\alpha]_D^{20}$ -1001 (c 0.25, CH₃CN), ee >99.9% (Kromasil 5-CelluCoat, CH₃CN–H₂O 70:30, t_R 4.67); **S-(+)-11c**: $[\alpha]_D^{20}$ +907 (c 0.25, CH₃CN), ee 94.2% (Kromasil 5-CelluCoat, CH₃CN–H₂O 70:30, t_R 6.33). δ_H (500 MHz, CDCl₃) 1.34 (t, 3H, CH₃, J 7.2 Hz), 3.68 (dq, 1H, NCH^B, J 14.3, 7.2 Hz), 3.73 (dq, 1H, NCH^A, J 14.3, 7.2), 5.82 (s, 1H, H6), 6.93 (dm, 1H, H3', J 3.6 Hz), 6.97 (dd, 1H, H4', J 5.0, 3.6 Hz), 7.32 (dd, 1H, H5', J 5.0, 1.2 Hz), 7.37 (dd, 1H, H4'', J 5.1, 2.8 Hz), 7.64 (dd, 1H, H5'', J 5.1, 1.2 Hz), 7.78 (dd, 1H, H2'', J 2.8, 1.2 Hz). δ_C (126 MHz, CDCl₃) 14.80 (CH₃), 48.18 (CH₂), 53.44 (C6), 110.18 (CN), 111.34 (CN), 116.05 (C3), 118.14 (C2), 127.09 (C5''), 127.21 (C4''), 127.29 (C3'), 127.32 (C5'), 127.47 (C4'), 128.76 (C2''), 136.98 (C2'), 137.89 (C3''), 143.87 (C5). GC t_R 29.28 min; MS m/z (rel intensity) 324 (M⁺, 100). Calcd. for C₁₆H₁₂N₄S₂ (324.43): C 59.24, H 3.73, N 17.27, S 19.77. Found C 59.45, H 3.62, N 16.95, S 19.98.

1-Ethyl-5-phenyl-6-thiophen-3-yl-1,6-dihydropyrazine-2,3-dicarbonitrile (12a). Yield of the racemate (see Table 1, entry 4), dark yellow powder, mp 154–155 °C. Racemate was separated by chiral HPLC with a Kromasil 5-CelluCoat (150 × 4.6 mm, 5 μ m) column. The eluent mixture CH₃CN–H₂O 60:40 was used in isocratic mode (0.6 mL/min at 25 °C), and the UV signal was monitored at λ 268 nm to obtain the resolved enantiomeric pair **R-(-)-12a/S-(-)-12a** (t_R 9.44, t_R 13.97). **R-(-)-12a**: $[\alpha]_D^{20}$ -927 (c 0.25, CH₃CN), ee 99.7% (Kromasil 5-CelluCoat, CH₃CN–H₂O 60:40, t_R 6.05); **S-(+)-12a**: $[\alpha]_D^{20}$ +859 (c 0.25, CH₃CN), ee 97.6% (Kromasil 5-CelluCoat, CH₃CN–H₂O 60:40, t_R 9.00). δ_H (500 MHz, CDCl₃) 1.31 (t, 3H, CH₃, J 7.2 Hz), 3.68 (dq, 1H, NCH^B, J 14.5, 7.2 Hz), 3.76 (dq, 1H, NCH^A, J 14.5, 7.2 Hz), 5.82 (s, 1H, H6), 6.98 (dd, 1H, H5', J 5.1, 1.4 Hz), 7.14 (ddd, 1H, H2', J 2.9, 1.4, 0.8 Hz), 7.36 (dd, 1H, H4', J 5.1, 2.9), 7.44 (m, 2H, Hm), 7.49 (m, 1H, Hp), 7.89 (m, 2H, Ho). δ_C (126 MHz, CDCl₃) 15.11 (CH₃), 48.60 (NCH₂), 52.83 (C6), 110.21 (CN), 111.41 (CN), 116.12 (C3), 118.37 (C2), 124.71 (C2'), 125.70 (C5'), 127.92 (Co), 128.06 (C4'), 129.01 (Cm), 131.85 (Cp), 133.95 (Ci), 135.44 (C3'), 147.78 (C5). GC t_R 29.10 min; MS m/z (rel intensity) 318 (M⁺, 100). Calcd. for C₁₈H₁₄N₄S (318.40): C 67.90, H 4.43, N 17.60. Found C 68.05, H 4.65, N 17.55.

1-Ethyl-5-thiophen-2-yl-6-thiophen-3-yl-1,6-dihydropyrazine-2,3-dicarbonitrile (12b). Yield of the racemate (see Table 1, entry 6), bright yellow powder, mp 141–143 °C. Racemate was separated by chiral HPLC with a Kromasil 5-CelluCoat (150 × 4.6 mm, 5 μ m) column. The eluent mixture CH₃CN–H₂O 65:35 was used in isocratic mode (20 mL/min at 25 °C), and the UV signal was monitored at λ 280 nm to obtain the resolved enantiomeric pair **R-(-)-12b/S-(-)-12b** (t_R 3.42, t_R 4.36). **R-(-)-12b**: $[\alpha]_D^{20}$ -1475 (c 0.25, CH₃CN), ee >99.9% (Kromasil 5-CelluCoat, CH₃CN–H₂O 60:40, t_R 5.68); **S-(+)-12b**: $[\alpha]_D^{20}$ +907 (c 0.25, CH₃CN), ee 93.5% (Kromasil 5-CelluCoat, CH₃CN–H₂O 60:40, t_R 7.32). δ_H (500 MHz, CDCl₃) 1.31 (t, 3H, CH₃, J 7.2 Hz), 3.65 (dq, 1H, NCH^B, J 14.5, 7.2 Hz), 3.73 (dq, 1H, NCH^A, J 14.5, 7.2 Hz), 5.72 (s, 1H, H6), 6.98 (dd, 1H, H5', J 5.1, 1.4 Hz), 7.09 (dd, 1H, H4'', J 5.1, 3.8 Hz), 7.18 (ddd, 1H, H2', J 2.9, 1.4, 0.7 Hz), 7.36 (dd, 1H, H4', J 5.1, 2.9 Hz), 7.45 (dd, 1H, H3'', J 3.8, 1.1 Hz), 7.56 (dd, 1H, H5'', J 5.1, 1.1 Hz). δ_C (126 MHz, CDCl₃): 14.82 (CH₃), 48.49 (CH₂), 53.88 (C6), 109.89 (CN), 111.43 (CN),

115.96 (C3), 118.37 (C2), 124.32 (C2'), 125.65 (C5'), 128.09 (C4'), 128.21 (C4''), 129.69 (C3''), 132.85 (C5''), 135.70 (C3'), 140.75 (C2''), 143.50 (C5). GC t_R 29.63 min; MS m/z (rel intensity) 324 (M^+ , 100). Calcd. For $C_{16}H_{12}N_4S_2$ (324.43): C 59.24, H 3.73, N 17.27, S 19.77. Found C 59.36, H 3.54, N 17.18, S 19.92.

1-Ethyl-5,6-dithiophen-3-yl-1,6-dihydropyrazine-2,3-dicarbonitrile (12c). Yield of the racemate (see Table 1, entry 8), bright yellow powder, mp 113-114 °C. Racemate was separated by chiral HPLC with a Kromasil 5-CelluCoat (150 × 4.6 mm, 5 μm) column. The eluent mixture CH_3CN-H_2O 65:35 was used in isocratic mode (0.8 mL/min at 25 °C), and the UV signal was monitored at λ 270 nm to obtain the resolved enantiomeric pair **R-(-)-12c/S-(-)-12c** (t_R 5.08, t_R 6.60). **R-(-)-12c:** $[\alpha]_D^{20}$ -992 (c 0.25, CH_3CN), ee >99.9% (Kromasil 5-CelluCoat, CH_3CN-H_2O 65:35, t_R 4.61); **S-(+)-12c:** $[\alpha]_D^{20}$ +967 (c 0.25, CH_3CN), ee 95.2% (Kromasil 5-CelluCoat, CH_3CN-H_2O 65:35, t_R 6.23). δ_H (500 MHz, $CDCl_3$) 1.31 (t, 3H, CH_3 , J 7.3 Hz), 3.65 (dq, 1H, NCH^B , J 14.5, 7.3 Hz), 3.73 (dq, 1H, NCH^A , J 14.5, 7.3 Hz), 5.67 (s, 1H, H6), 6.97 (dd, 1H, H5', J 5.1, 1.4 Hz), 7.16 (ddd, 1H, H2', J 2.9, 1.4, 0.7 Hz), 7.36 (dd, 1H, H4', J 5.1, 2.9 Hz), 7.37 (dd, 1H, H4'', J 5.1, 2.9 Hz), 7.64 (dd, 1H, H5'', J 5.1, 1.3 Hz), 7.75 (dd, 1H, H2'', J 2.9, 1.3 Hz). δ_C (126 MHz, $CDCl_3$) 14.90 (CH_3), 48.42 (CH_2), 53.95 (C6), 109.90 (CN), 111.44 (CN), 116.17 (C3), 118.28 (C2), 124.38 (C2'), 125.68 (C5'), 127.12 (C5''), 127.19 (C4''), 128.13 (C4'), 128.52 (C2''), 135.74 (C3'), 138.21 (C3''), 144.19 (C5). GC t_R 29.59 min; MS m/z (rel intensity) 324 (M^+ , 100). Calcd. for $C_{16}H_{12}N_4S_2$ (324.43): C 59.24, H 3.73, N 17.27, S 19.77. Found C 58.98, H 3.40, N 17.39, S 20.23.

6-Benzo[b]thiophen-2-yl-1-ethyl-5-phenyl-1,6-dihydropyrazine-2,3-dicarbonitrile (15a). Yield of the racemate (see Table 1, entry 9), bright yellow powder, mp 68-70 °C. Racemate was separated by chiral HPLC with a Kromasil 5-CelluCoat (150 × 4.6 mm, 5 μm) column. The eluent mixture CH_3CN-H_2O 75:25 was used in isocratic mode (0.6 mL/min at 25 °C), and the UV signal was monitored at λ 265 nm to obtain the resolved enantiomeric pair **R-(-)-15a/S-(-)-15a** (t_R 6.94, t_R 10.51). **R-(-)-15a:** $[\alpha]_D^{20}$ -955 (c 0.25, CH_3CN), ee >99.9% (Kromasil 5-CelluCoat, CH_3CN-H_2O 70:30, t_R 4.67); **S-(+)-15a:** $[\alpha]_D^{20}$ +918 (c 0.25, CH_3CN), ee 94.2% (Kromasil 5-CelluCoat, CH_3CN-H_2O 70:30, t_R 6.33). δ_H (500 MHz, $CDCl_3$) 1.38 (t, 3H, CH_3 , J 7.2 Hz), 3.75 (dq, 1H, NCH^B , J 14.5, 7.2 Hz), 3.82 (dq, 1H, NCH^A , J 14.5, 7.2 Hz), 6.02 (d, 1H, H6, J 0.8 Hz), 7.11 (s, 1H, H3'), 7.34-7.38 (m, 2H, H5',6'), 7.46 (m, 2H, Hm), 7.51 (t, 1H, H p , J 7.2, 1.4 Hz), 7.70 (m, 1H, H4'), 7.79 (m, 1H, H7'), 7.95 (dd, 2H, H o , J 8.4, 1.3 Hz). δ_C (126 MHz, $CDCl_3$) 15.10 (CH_3), 48.72 (NCH_2), 53.18 (C6), 110.88 (CN), 111.24 (CN), 115.87 (C3), 118.21 (C2), 122.41 (C7'), 124.20 (C4'), 124.56 (C3'), 125.04 and 125.62 (C5' and C6'), 128.04 (C o), 129.10 (C m), 132.09 (C p), 133.66 (C i), 136.94 (C2'), 138.68 (C3'a), 139.79 (C7'a), 146.99 (C5). GC t_R 36.22 min; MS m/z (rel intensity) 368 (M^+ , 100). Calcd. for $C_{22}H_{16}N_4S$ (368.46): C 71.72, H 4.38, N 15.21. Found C 71.68, H 4.44, N 15.48.

6-Benzo[b]thiophen-2-yl-1-ethyl-5-thiophen-2-yl-1,6-dihydropyrazine-2,3-dicarbonitrile (15b). Yield of the racemate (see Table 1, entry 10), dark yellow powder, mp 77-79 °C. Racemate was separated by chiral HPLC with a Kromasil 5-CelluCoat (150 × 4.6 mm, 5 μm) column. The eluent mixture CH_3CN-H_2O 80:20 was used in isocratic mode (20 mL/min at 25

°C), and the UV signal was monitored at λ 280 nm to obtain the resolved enantiomeric pair **R-(-)-15b/S-(+)-15b** (t_R 3.41, t_R 4.21). **R-(-)-15b**: $[\alpha]_D^{20}$ -1615 (c 0.25, CH₃CN), ee >99.9% (Kromasil 5-CelluCoat, CH₃CN–H₂O 70:30, t_R 4.67); **S-(+)-15b**: $[\alpha]_D^{20}$ +1578 (c 0.25, CH₃CN), ee 94.2% (Kromasil 5-CelluCoat, CH₃CN–H₂O 70:30, t_R 6.33). δ_H (500 MHz, CDCl₃) 1.37 (t, 3H, CH₃, J 7.2 Hz), 3.72 (dq, 1H, NCH^B, J 14.5, 7.2 Hz), 3.78 (dq, 1H, NCH^A, J 14.5, 7.2 Hz), 5.92 (s, 1H, H₆), 7.11 (dd, 1H, H^{4''}, J 5.0, 3.9 Hz), 7.15 (s, 1H, H^{3'}), 7.34-7.39 (m, 2H, H^{5',6'}), 7.52 (dd, 1H, H^{3''}, J 3.9, 1.0 Hz), 7.59 (dd, 1H, H^{5''}, J 5.0, 1.0 Hz), 7.72 (m, 1H, H^{4'}), 7.78 (m, 1H, H^{7'}). δ_C (126 MHz, CDCl₃) 14.84 (CH₃), 48.63 (NCH₂), 110.68 (CN), 111.28 (CN), 115.67 (C₃), 118.17 (C₂), 122.43 (C^{7'}), 124.09 (C^{3'}), 124.19 (C^{4'}), 125.07 (C^{6'}), 125.63 (C^{5'}), 128.30 (C^{4''}), 129.95 (C^{3''}), 133.24 (C^{5''}), 137.17 (C^{2'}), 138.66 (C^{3'a}), 139.74 (C^{7'a}), 140.44 (C^{2''}), 142.74 (C₅). GC t_R 37.35 min; MS m/z (rel intensity) 374 (M⁺, 100). Calcd. for C₂₀H₁₄N₄S₂ (374.49): C 64.15, H 3.77, N 14.96. Found C 63.91, H 3.56, N 15.02.

6-Benzo[*b*]thiophen-2-yl-1-ethyl-5-thiophen-3-yl-1,6-dihydropyrazine-2,3-dicarbonitrile (15c). Yield of the racemate (see Table 1, entry 11), dark yellow powder, mp 75-77 °C. Racemate was separated by chiral HPLC with a Kromasil 5-CelluCoat (150 × 4.6 mm, 5 μ m) column. The eluent mixture CH₃CN–H₂O 65:35 was used in isocratic mode (0.8 mL/min at 25 °C), and the UV signal was monitored at λ 265 nm to obtain the resolved enantiomeric pair **R-(-)-15c/S-(+)-15c** (t_R 10.36, t_R 13.55). **R-(-)-15c**: $[\alpha]_D^{20}$ -1026 (c 0.25, CH₃CN), ee >99.9% (Kromasil 5-CelluCoat, CH₃CN–H₂O 70:30, t_R 4.67); **S-(+)-15c**: $[\alpha]_D^{20}$ +830 (c 0.25, CH₃CN), ee 94.2% (Kromasil 5-CelluCoat, CH₃CN–H₂O 70:30, t_R 6.33). δ_H (500 MHz, CDCl₃) 1.37 (t, 3H, CH₃, J 7.2 Hz), 3.72 (dq, 1H, NCH^B, J 14.5, 7.2 Hz), 3.79 (dq, 1H, NCH^A, J 14.5, 7.2 Hz), 5.87 (s, 1H, H₆), 7.13 (s, 1H, H^{3'}), 7.35-7.39 (m, 2H, H^{5',6'}), 7.39 (dd, 1H, H^{5''}, J 5.1, 2.8 Hz), 7.68 (dd, 1H, H^{4''}, J 5.1, 1.4 Hz), 7.72 (m, 1H, H^{4'}), 7.78 (m, 1H, H^{7'}), 7.84 (dd, 1H, H^{2''}, J 2.8, 1.4 Hz). δ_C (126 MHz, CDCl₃) 14.92 (CH₃), 48.56 (NCH₂), 54.27 (C₆), 110.60 (CN), 111.28 (CN), 115.92 (C₃), 118.10 (C₂), 122.43 (C^{7'}), 124.17 and 124.19 (C^{3'} and C^{4'}), 125.08 and 125.62 (C^{5'} and C^{6'}), 127.12 (C^{4''}), 127.39 (C^{5''}), 128.91 (C^{2''}), 137.20 (C^{2'}), 137.91 (C^{3''}), 138.70 (C^{3'a}), 139.75 (C^{7'a}), 143.40 (C₅). GC t_R 37.36 min; MS m/z (rel intensity) 374 (M⁺, 100). Calcd. for C₂₀H₁₄N₄S₂ (374.49): C 64.15, H 3.77, N 14.96. Found C 64.08, H 3.77, N 14.70.

6-Benzo[*b*]thiophen-3-yl-1-ethyl-5-phenyl-1,6-dihydropyrazine-2,3-dicarbonitrile (16a). Yield of the racemate (see Table 1, entry 12), yellow powder, mp 174-176 °C. Racemate was separated by chiral HPLC with a Kromasil 5-CelluCoat (150 × 4.6 mm, 5 μ m) column. The eluent mixture CH₃CN–H₂O 75:25 was used in isocratic mode (0.6 mL/min at 25 °C), and the UV signal was monitored at λ 265 nm to obtain the resolved enantiomeric pair **R-(-)-16a/S-(+)-16a** (t_R 6.18, t_R 8.79). **R-(-)-16a**: $[\alpha]_D^{20}$ -751 (c 0.25, CH₃CN), ee >99.9% (Kromasil 5-CelluCoat, CH₃CN–H₂O 70:30, t_R 4.67); **S-(+)-16a**: $[\alpha]_D^{20}$ +776 (c 0.25, CH₃CN), ee 94.2% (Kromasil 5-CelluCoat, CH₃CN–H₂O 70:30, t_R 6.33). δ_H (500 MHz, CDCl₃) 1.28 (t, 3H, CH₃, J 7.2 Hz), 3.70 (dq, 1H, NCH^B, J 14.4, 7.2 Hz), 3.83 (dq, 1H, NCH^A, J 14.4, 7.2 Hz), 6.08 (s, 1H, H₆), 7.27 (s, 1H, H^{2'}), 7.38 (dd, 1H, H_m, J 8.3, 7.2 Hz), 7.45 (tt, 1H, H_p, J 7.2, 1.2 Hz), 7.48 (ddd, 1H, H_{6'}, J 8.2, 7.1, 1.0 Hz), 7.56 (ddd, 1H, H^{5'}, J 8.1, 7.2, 1.0 Hz), 7.81 (m, 2H, H_o). δ_C

(126 MHz, CDCl₃) 15.05 (CH₃), 49.14 (NCH₂), 51.02 (C6), 110.70 (CN), 111.35 (CN), 116.14 (C3), 118.06 (C2), 120.39 (C4'), 123.66 (C7'), 125.20 (C6'), 125.28 (C5'), 127.80 (C_o), 128.99 (C_m), 129.16 (C₂)^δ 129.20 (C3'), 131.95 (C_p), 133.72 (C_i), 136.25 (C3'a), 140.99 (C7'a), 148.03 (C5). GC *t_R* 35.29 min; MS *m/z* (rel intensity) 368 (M⁺, 100). Calcd. for C₂₂H₁₆N₄S (368.46): C 71.72, H 4.38, N 15.21. Found C 71.63, H 4.29, N 15.39.

6-Benzo[*b*]thiophen-3-yl-1-ethyl-5-thiophen-2-yl-1,6-dihydropyrazine-2,3-dicarbonitrile

(16b). Yield of the racemate (see Table 1, entry 13), yellow powder, mp 224-226 °C (decomp.). Racemate was separated by chiral HPLC with a Kromasil 5-CelluCoat (150 × 4.6 mm, 5 μm) column. The eluent mixture CH₃CN–H₂O 80:20 was used in isocratic mode (20 mL/min at 25 °C), and the UV signal was monitored at λ 227 nm to obtain the resolved enantiomeric pair **R(-)-16b/S(-)-16b** (*t_R* 3.12, *t_R* 4.18). **R(-)-16b**: [α]_D²⁰ -1035 (c 0.25, CH₃CN), ee >99.9% (Kromasil 5-CelluCoat, CH₃CN–H₂O 75:25, *t_R* 6.32); **S(+)-16b**: [α]_D²⁰ +941 (c 0.25, CH₃CN), ee 98.1% (Kromasil 5-CelluCoat, CH₃CN–H₂O 75:25, *t_R* 7.58). δ_H (500 MHz, CDCl₃) 1.28 (t, 3H, CH₃, *J* 7.2 Hz), 3.65 (dq, 1H, NCH^B, *J* 14.4, 7.2 Hz), 3.77 (dq, 1H, NCH^A, *J* 14.4, 7.2 Hz), 6.00 (s, 1H, H₆), 6.99 (dd, 1H, H^{4''}, *J* 5.0, 3.9 Hz), 7.29 (dd, 1H, H^{3''}, *J* 3.9, 1.0 Hz), 7.35 (s, 1H, H^{2'}), 7.47 (ddd, 1H, H^{6'}, *J* 8.2, 7.1, 1.0 Hz), 7.56 (ddd, 1H, H^{5'}, *J* 8.1, 7.1, 1.0 Hz), 7.91-7.94 (m, 2H, H^{4',7'}). δ_C (126 MHz, CDCl₃) 14.64 (CH₃), 48.85 (NCH₂), 51.73 (C6), 110.05 (CN), 111.37 (CN), 115.99 (C3), 118.19 (C2), 120.24 (C4'), 123.65 (C7'), 125.22 (C6'), 125.36 (C5'), 128.19 (C^{4''}), 128.67 (C^{2'}), 129.48 (C^{3''}), 130.07 (C^{2'}), 132.81 (C^{5''}), 136.41 (C3'a), 140.48 (C^{2''}), 140.79 (C7'a), 143.98 (C5). GC *t_R* 36.61 min; MS *m/z* (rel intensity) 374 (M⁺, 100). Calcd. for C₂₀H₁₄N₄S₂ (374.49): C 64.15, H 3.77, N 14.96. Found C 63.91, H 3.56, N 15.02.

6-Benzo[*b*]thiophen-3-yl-1-ethyl-5-thiophen-3-yl-1,6-dihydropyrazine-2,3-dicarbonitrile

(16c). Yield of the racemate (see Table 1, entry 14), yellow powder, mp 227-228 °C. Racemate was separated by chiral HPLC with a Kromasil 5-CelluCoat (150 × 4.6 mm, 5 μm) column. The eluent mixture CH₃CN–H₂O 80:20 was used in isocratic mode (20 mL/min at 25 °C), and the UV signal was monitored at λ 266 nm to obtain the resolved enantiomeric pair **R(-)-16c/S(-)-16c** (*t_R* 3.34, *t_R* 5.20). **R(-)-16c**: [α]_D²⁰ -684 (c 0.25, CH₃CN), ee 99.3% (Kromasil 5-CelluCoat, CH₃CN–H₂O 75:25, *t_R* 5.68); **S(+)-16c**: [α]_D²⁰ +679 (c 0.25, CH₃CN), ee 98.4% (Kromasil 5-CelluCoat, CH₃CN–H₂O 75:25, *t_R* 9.31). δ_H (500 MHz, CDCl₃) 1.28 (t, 3H, CH₃, *J* 7.2 Hz), 3.65 (dq, 1H, NCH^B, *J* 14.4, 7.2 Hz), 3.77 (dq, 1H, NCH^A, *J* 14.4, 7.2 Hz), 5.94 (s, 1H, H₆), 7.32 (s, 1H, H^{2'}), 7.33 (dd, 1H, H^{5''}, *J* 5.0, 2.9 Hz), 7.47 (ddd, 1H, H^{6'}, *J* 8.2, 7.1, 1.0 Hz), 7.56 (ddd, 1H, H^{5'}, *J* 8.1, 7.1, 1.0 Hz), 7.60-7.62 (m, 2H, H^{4'',2''}), 7.91-7.95 (m, 2H, H^{4',7'}). δ_C (126 MHz, CDCl₃) 14.75 (CH₃), 48.82 (NCH₂), 51.80 (C6), 110.05 (CN), 111.37 (CN), 116.22 (C3), 118.14 (C2), 120.16 (C4'), 123.70 (C7'), 125.23 (C6'), 125.38 (C5'), 126.95 (C^{4''}), 127.14 (C^{5''}), 128.44 (C^{2''}), 128.86 (C^{2'}), 130.02 (C3'), 136.43 (C3'a), 137.97 (C^{3''}), 140.88 (C7'a), 144.63 (C5). GC *t_R* 36.40 min; MS *m/z* (rel intensity) 374 (M⁺, 100). Calcd. for C₂₀H₁₄N₄S₂ (374.49): C 64.15, H 3.77, N 14.96. Found C 64.12, H 3.83, N 14.96.

1-Ethyl-6-furan-2-yl-5-phenyl-1,6-dihydropyrazine-2,3-dicarbonitrile (18a). Yield of the racemate (see Table 1, entry 15), yellow-orange powder, mp 137-140 °C. Racemate was separated by chiral HPLC with a Kromasil 5-CelluCoat (150 × 4.6 mm, 5 μm) column. The

eluent mixture CH₃CN–H₂O 65:35 was used in isocratic mode (0.6 mL/min at 25 °C), and the UV signal was monitored at λ 270 nm to obtain the resolved enantiomeric pair **R-(-)-18a/S-(-)-18a** (t_R 6.32, t_R 9.35). **R-(-)-18a**: $[\alpha]_D^{20}$ -1032 (c 0.25, CH₃CN), ee 96.7% (Kromasil 5-CelluCoat, CH₃CN–H₂O 65:35, t_R 5.53); **S-(+)-18a**: $[\alpha]_D^{20}$ +945 (c 0.25, CH₃CN), ee 96.6% (Kromasil 5-CelluCoat, CH₃CN–H₂O 65:35, t_R 7.96 min). δ_H (500 MHz, CDCl₃) 1.36 (t, 3H, CH₃, J 7.2 Hz), 3.69 (dq, 1H, NCH^A, J 14.5, 7.2 Hz), 3.79 (dq, 1H, NCH^B, J 14.5, 7.2 Hz), 5.82 (s, 1H, H₆), 6.19 (d, 1H, H_{4'}, J 3.2 Hz), 6.32 (dd, 1H, H_{5'}, J 3.2, 1.8 Hz), 7.43-7.51 (m, 4H, H_{3'} and Ph), 7.86-7.87 (m, 2H, Ph). δ_C (126 MHz, CDCl₃) 14.71 (CH₃), 48.82 (NCH₂), 50.84 (C₆), 110.07, 111.03, 111.07, 111.30, 116.10, 118.23, 127.83, 128.96, 131.87, 133.72, 144.20, 145.29, 147.35. GC t_R 27.04 min; MS m/z (rel intensity) 302 (M⁺, 100). Calcd. for C₁₈H₁₄N₄O (302.34): C 71.51, H 4.67, N 18.53. Found C 71.30, H 4.48, N 18.75.

1-Ethyl-6-furan-2-yl-5-thiophen-2-yl-1,6-dihydropyrazine-2,3-dicarbonitrile(18b). Yield of the racemate (see Table 1, entry 16), yellow powder, mp 102-105 °C. Racemate was separated by chiral HPLC with a Kromasil 5-CelluCoat (150 × 4.6 mm, 5 μ m) column. The eluent mixture CH₃CN–H₂O 65:35 was used in isocratic mode (20 mL/min at 25 °C), and the UV signal was monitored at λ 280 nm to obtain the resolved enantiomeric pair **R-(-)-18b/S-(-)-18b** (t_R 5.75, t_R 7.10). **R-(-)-18b**: $[\alpha]_D^{20}$ -906 (c 0.25, CH₃CN), ee 88.5% (Kromasil 5-CelluCoat, CH₃CN–H₂O 65:35, t_R 5.27); **S-(+)-18b**: $[\alpha]_D^{20}$ +1721 (c 0.25, CH₃CN), ee 95.2% (Kromasil 5-CelluCoat, CH₃CN–H₂O 65:35, t_R 6.42). δ_H (500 MHz, CDCl₃) 1.34 (t, 3H, CH₃, J 7.2 Hz), 3.67 (dq, 1H, NCH^A, J 14.5, 7.2 Hz), 3.74 (dq, 1H, NCH^B, J 14.5, 7.2 Hz), 5.72 (s, 1H, H₆), 6.23 (d, 1H, H_{4'}, J 3.4 Hz), 6.33 (dd, 1H, H_{5'}, J 3.4, 1.8 Hz), 7.09 (dd, 1H, H_{4''}, J 5.0, 3.8 Hz), 7.42-7.43 (m, 2H, H_{3'} and H_{3''}), 7.56 (dd, 1H, H_{5''}, J 5.0, 1.0). δ_C (126 MHz, CDCl₃) 14.46 (CH₃), 48.65 (NCH₂), 51.74 (C₆), 109.85, 110.47, 111.03, 111.34, 115.92, 118.26, 128.19, 129.71, 132.82, 140.47, 141.29, 144.09, 147.35. GC t_R 27.42 min; MS m/z (rel intensity) 308 (M⁺, 100). Calcd. for C₁₆H₁₂N₄OS (308.36): C 62.32, H 3.92, N 18.17. Found C 62.22, H 3.77, N 18.45.

1-Ethyl-6-furan-2-yl-5-thiophen-3-yl-1,6-dihydropyrazine-2,3-dicarbonitrile (18c). Yield of the racemate (see Table 1, entry 17), yellow powder, mp 103-106 °C. Racemate was separated by chiral HPLC with a Kromasil 5-CelluCoat (150 × 4.6 mm, 5 μ m) column. The eluent mixture CH₃CN–H₂O 60:40 was used in isocratic mode (0.8 mL/min at 25 °C), and the UV signal was monitored at λ 270 nm to obtain the resolved enantiomeric pair **R-(-)-18c/S-(-)-18c** (t_R 5.58, t_R 7.62). **R-(-)-18c**: $[\alpha]_D^{20}$ -1112 (c 0.25, CH₃CN), ee 99.6% (Kromasil 5-CelluCoat, CH₃CN–H₂O 60:40, t_R 6.60); **S-(+)-18c**: $[\alpha]_D^{20}$ +982 (c 0.25, CH₃CN), ee 93.0% (Kromasil 5-CelluCoat, CH₃CN–H₂O 60:40, t_R 8.93). δ_H (500 MHz, CDCl₃) 1.34 (t, 3H, CH₃, J 7.2 Hz), 3.67 (dq, 1H, NCH^A, J 14.5, 7.2 Hz), 3.74 (dq, 1H, NCH^B, J 14.5, 7.2), 5.67 (s, 1H, H₆), 6.21 (d, 1H, H_{3'}, J 3.3), 6.33 (dd, 1H, H_{4'}, J 3.3, 1.8 Hz), 7.37 (dd, 1H, H_{5''}, J 5.2, 2.8 Hz), 7.43 (dd, 1H, H_{5'}, J 1.8, 0.7 Hz), 7.63 (dd, 1H, H_{4''}, J 5.2, 1.3 Hz), 7.75 (dd, 1H, H_{2''}, J 2.8, 1.3 Hz). δ_C (126 MHz, CDCl₃) 14.54 (CH₃), 48.61 (CH₂), 51.85 (C₆), 109.79, 110.59, 111.06, 111.35, 116.16, 118.16, 126.96, 127.13, 128.59, 137.93, 141.93, 144.08, 147.44. GC t_R 27.43 min; MS m/z (rel intensity) 308 (M⁺, 100). Calcd. for C₁₆H₁₂N₄OS (308.36): C 62.32, H 3.92, N 18.17. Found C 62.12, H 3.76, N 18.44.

Antimycobacterial assay. To evaluate the inhibitory efficiency of molecules on *Mycobacterium tuberculosis*, *M. tuberculosis H₃₇Rv*, which is susceptible to all classical antituberculosis drugs, was used. The minimal inhibitory concentration (MIC) for *M. tuberculosis H₃₇Rv* for each compound was determined by a micro broth dilution method. All molecules tested were dissolved in dimethylsulfoxide and their 1/2 dilutions were prepared in 5 mL tubes using Löwenstein-Jensen medium. A few colonies from freshly grown *M. tuberculosis H₃₇Rv* were suspended in Löwenstein-Jensen medium to obtain 1.0 McFarland turbidity and diluted ten times using the same medium and the tubes were incubated at 37 °C medium with a different concentration of the tested molecule and to a positive control tube containing only clear growth medium. After 24 hours the tubes were placed in a vertical position and the free edge of the buried 0.3 mL of the substance in the test compounds concentrations: 12.5, 6.2, 3.1, 1.5, 0.75 µg/mL. The tubes were then placed in an thermostat at a temperature of 37 °C and incubated for 10 days. Growth estimate for the MTB were determined by standard methods, where the appearance of zones of growth retardation MTB (over 10 mm) indicated the presence of tuberculostatic properties in concentration of the compounds under study. Penetration size stunting MTB (in mm) is proportional to the degree of tuberculostatic activity. Growth delay of 100 mm or more is considered as a complete growth inhibition MTB. The extensively drug-resistant (XDR) and multi-drug-resistant (MDR) tuberculosis strains have been isolated from tuberculosis patients in Ural Research Institute for Phthisiopulmonology (Russia). The minimal inhibitory concentrations against *Mycobacterium tuberculosis avium*, *Mycobacterium tuberculosis terrae*, XDR and MDR tuberculosis strains were evaluated similarly.

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