

An expedient approach to the 2,3,5,6-tetrasubstituted pyridine core of nosiheptide using oxidative cleavage of 2,3,5,8-tetrasubstituted quinolines

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Abstract

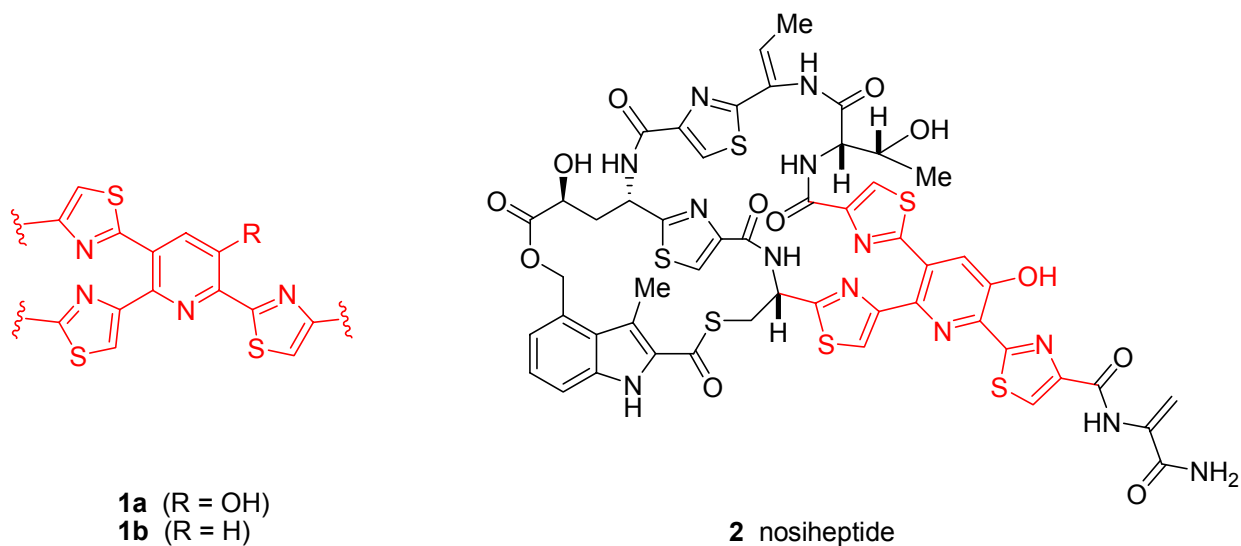
Pyridine cores **1a** and **1b** bearing thiazole rings feature in several natural products such as nosiheptide **2** and are often accessed through tedious synthetic routes. We have developed a new approach to the tetrasubstituted pyridine fragment of nosiheptide, which is based on the ozonolysis of tetrasubstituted quinolines, readily obtained by the Meth-Cohn “double Vilsmeier” process.

Keywords: Pyridine, quinoline, ozonolysis

Introduction

The thiopeptide antibiotics¹ constitute a class of highly modified cyclic peptides with reported biological activity against MRSA² and malaria,³ and which inhibit protein synthesis by interfering with the elongation protein factors in bacteria.⁴ Despite their interesting biological activity, only the total syntheses of micrococcin P1,⁵ thiostrepton,⁶ promothiocin A⁷ and amythiamycin D⁸ have been reported to date. The total synthesis of nosiheptide **2** has not been completed yet, although the preparation of some key fragments has been described in several papers.⁹⁻¹⁵

[‡] Chris Moody was Chairman of the RSC Heterocyclic Group during the period 1993-1995.



Much of the structural complexity of the thiopeptides is centred on a conserved pyridine core **1a-b**, substituted by thiazole rings, and such pyridines have attracted considerable interest from synthetic chemists. Indeed, previous strategies towards this key fragment include the stepwise functionalization of an existing pyridine ring as described by the Shin group,¹⁴ the cyclization of a suitable 1,5-diketone precursor with ammonia prior to aromatization by dehydrogenation as demonstrated by Ciufolini,¹⁶ or the sequential regioselective arylation of 2,5,6-tribromopyridine using palladium-catalyzed Negishi and Stille cross-couplings as shown recently by Bach and co-workers.¹⁷

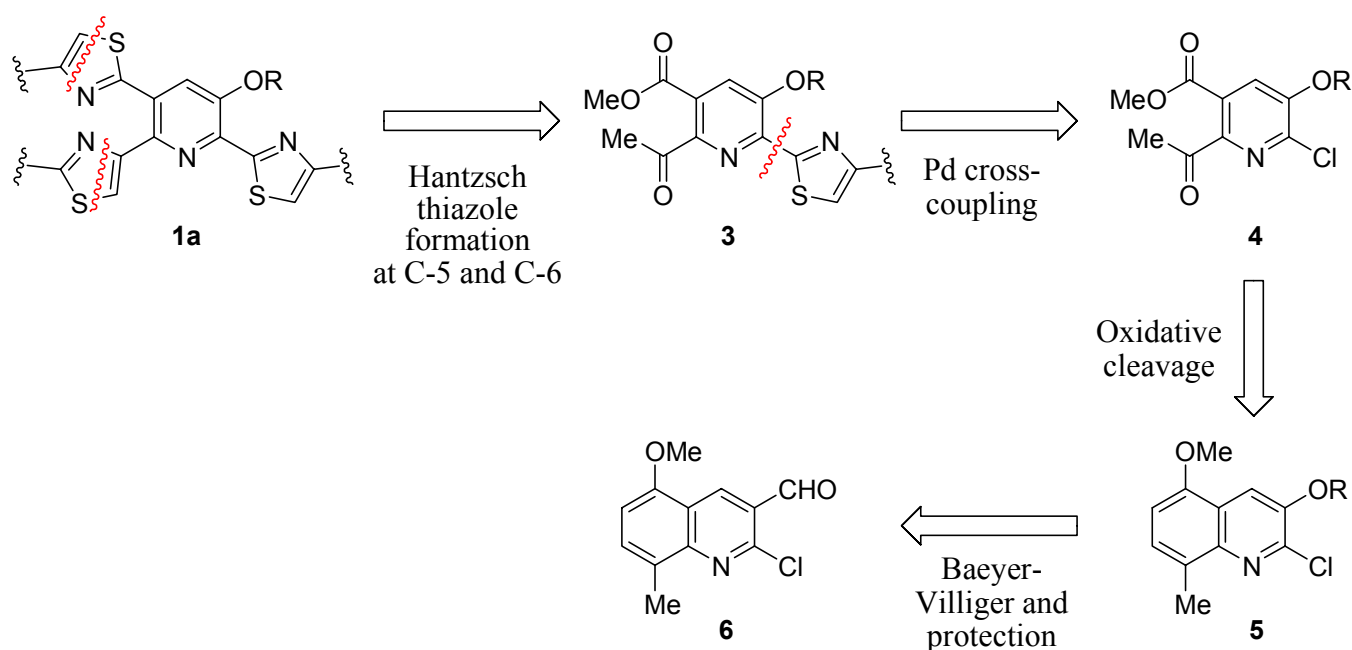
In contrast, our group has previously focused on a 'biomimetic' hetero-Diels-Alder-aromatization sequence under microwave irradiation,¹⁸ and the Bohlmann-Rahtz reaction between an enamine and an alkynone¹⁹ to generate the pyridine cores of amythiamycin D⁸ and promothiocin A⁷ respectively. Despite these advances there remains a need for a new synthetic approach to access the crucial pyridine core of nosiheptide and related thiopeptide natural products.

Results and Discussion

Pyridine-5,6-dicarboxylates have recently been prepared by Barré and Perrio by oxidative cleavage of quinolines using ozonolysis or ruthenium tetroxide in biphasic carbon tetrachloride and water.²⁰ This was interesting to us since direct precursors to the pyridine core of nosiheptide could be obtained in one key step from readily accessible quinolines. However, the sequential functionalization of these carboxylates into two different thiazoles as in pyridines **1a-b** would be difficult to achieve, and this clearly constituted the likely downfall of this methodology.

Therefore, our approach required a pyridine intermediate with two different substituents at the 5- and 6-positions that could be later differentiated into the desired thiazole rings. We

postulated that the nosiheptide pyridine core **1a** could be accessed through classical Hantzsch cyclizations following derivatization of a methyl ester at C-5 and an acetyl moiety at C-6, whilst the third thiazole ring would be introduced through a palladium-catalyzed cross-coupling at C-2 of the 2-chloropyridine intermediate **3**. It was next anticipated that pyridine **4** could be obtained by ozonolysis of the tetrasubstituted quinoline **5** and that a Baeyer-Villiger oxidation/hydroxyl protection of quinoline **6** would afford the required protected substituent at C-3 (Scheme 1).

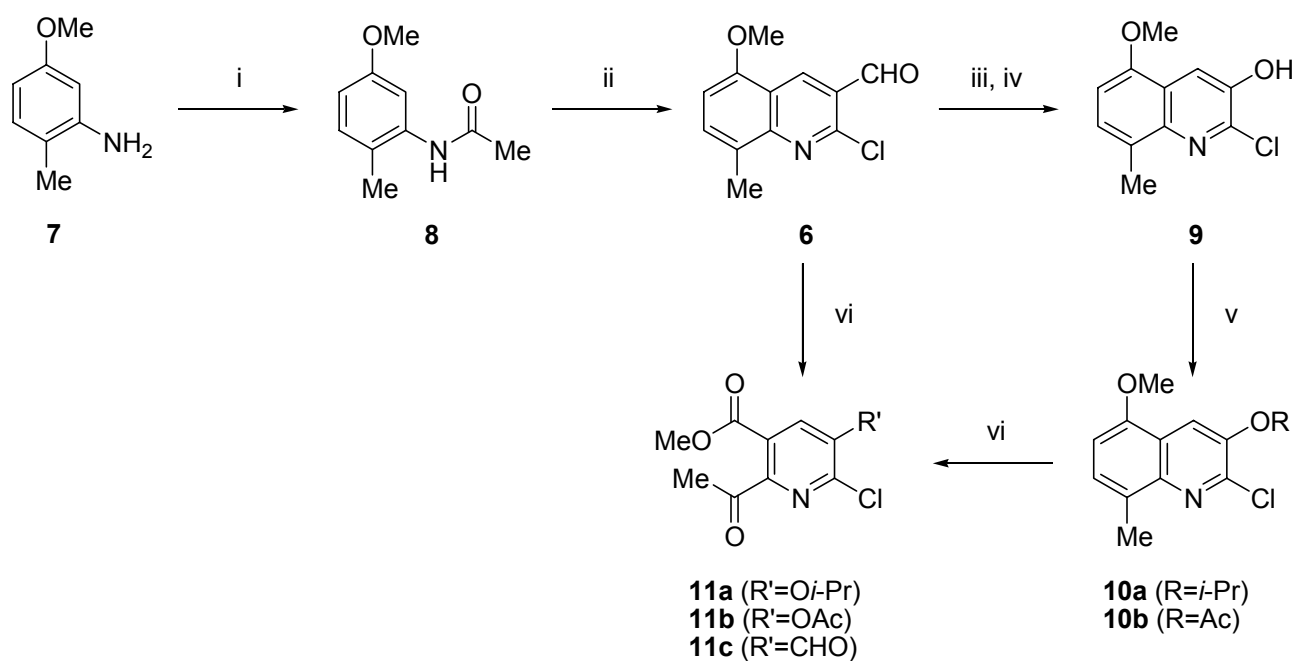


Scheme 1. Retrosynthetic analysis of tetrasubstituted pyridine **1a**

Access to the 2-chloroquinoline-3-carbaldehyde **6** was readily accomplished using the Meth-Cohn double-Vilsmeier protocol.²¹ Thus, reacting the commercially available aniline **7** with acetic anhydride in acetonitrile and treating the acetanilide **8** under Vilsmeier-Haack conditions with excess phosphorus oxychloride gave the required tetrasubstituted quinoline **6**. Oxidation of the 3-formyl moiety of quinoline **6** with *m*-CPBA in dichloromethane was unsatisfactory as the reaction was either incomplete or afforded the quinoline *N*-oxide side-product predominantly. However, when an aqueous solution of peracetic acid was slowly added to quinoline **6** in chloroform and the intermediate formate ester hydrolyzed with potassium hydrogen carbonate in aqueous methanol, 3-hydroxyquinoline **9** was isolated in 46% yield over the two steps with recovery of the unreacted starting material. Subsequent protection of the 3-hydroxyl group as its isopropyl ether or acetyl ester was performed using standard conditions in 97% and 93% yield respectively.

However, when quinolines **10a-b** were treated with ruthenium tetroxide in water/chloroform as described by Barré and Perrio, no reaction occurred over 24 h and the starting materials were recovered. Our attention then turned to the ozonolysis of the same

substrates below 0 °C. While no product could be isolated when treating the intermediate under oxidative conditions (hydrogen peroxide then sodium hydroxide), we were delighted to find that the desired tetrasubstituted pyridines could be obtained in 39-45% yield following quenching with dimethyl sulfide and purification of the crude product by flash chromatography on silica gel. The same procedure was also implemented on quinoline **6** and gave the desired product **11c** in 45% yield.



Scheme 2. Reagents and conditions: i) Ac₂O, MeCN, 89%; ii) POCl₃, DMF, 0 °C, 30 min then 120 °C, 3 h, 79%; iii) MeCO₃H, CHCl₃, 72 h; iv) KHCO₃, aqueous MeOH, 46% (over 2 steps); v) *i*-PrI, K₂CO₃, DMF, 75 °C, 97% (R = *i*-Pr, **10a**); or Ac₂O, K₂CO₃, DMF, 50 °C, 93% (R = Ac, **10b**); vi) O₃, dichloromethane, -20 °C, 30 min then Me₂S, -20 °C, 12 h, 39-45%.

To conclude, the ozonolysis of 2,3,5,8-tetrasubstituted quinolines, readily obtained using the Meth-Cohn reaction, gives access to 2,3,5,6-tetrasubstituted pyridines. This approach provides a short and elegant alternative to the traditionally used strategies towards the tetrasubstituted pyridines of thiopeptides. We are currently investigating the palladium-catalyzed cross-coupling of these 2-chloropyridines with various metallated thiazoles.

Experimental Section

General Procedures. Commercially available reagents were used throughout without further purification unless otherwise stated; solvents were dried by standard procedures. Light petroleum refers to the fraction with bp 40-60 °C. Reactions were routinely carried out under a nitrogen

atmosphere. Fully characterized compounds were chromatographically homogeneous. IR spectra were recorded in the range 4000-600 cm^{-1} . ^1H and ^{13}C NMR spectra were recorded at 400 MHz and 100 MHz respectively; J values were recorded in Hz. In the ^{13}C NMR spectra, signals corresponding to CH, CH_2 , or CH_3 groups, as assigned from DEPT, are noted; all others are C.

***N*-(5-Methoxy-2-methylphenyl)acetamide (8).** Acetic anhydride (3.6 ml, 32.6 mmol) was added dropwise to a stirring solution of 5-methoxy-2-methylaniline **7** (3 g, 21.9 mmol) in acetonitrile (50 ml) under a nitrogen atmosphere. The solution was stirred for 12 h, evaporated *in vacuo*, partitioned between ethyl acetate (100 ml) and saturated aqueous sodium sodium bicarbonate (200 ml). The aqueous layer was further extracted with ethyl acetate and the combined organic extracts were washed with water, dried (MgSO_4) and evaporated *in vacuo*. Recrystallization from petroleum ether gave the *title compound 8* (3.47 g, 89%) as colourless needles, mp 95-96 $^\circ\text{C}$, (lit.,²² mp 95-95.7 $^\circ\text{C}$); ν_{max} (CHCl_3)/ cm^{-1} 3452, 2938, 2837, 1688, 1618, 1588, 1486, 1456, 1370, 1241, 1164, 1126, 1110, 1041, 1009, 873; δ_{H} (400 MHz; CDCl_3) 7.43 (1 H, br s, NH), 7.34 (1 H, d, J 2.0, ArH), 7.01 (1 H, d, J 8.2, ArH), 6.60 (1 H, dd, J 8.2 and J 2.0, ArH), 3.72 (3 H, s, OCH_3), 2.12 (2 x 3 H, s, 2 x CH_3); δ_{C} (100 MHz; CDCl_3) 168.8 (C), 158.2 (C), 136.5 (C), 130.9 (CH), 121.3 (C), 111.2 (CH), 109.0 (CH), 55.4 (Me), 24.2 (Me), 16.9 (Me).

2-Chloro-5-methoxy-8-methylquinoline-3-carbaldehyde (6). Phosphorus oxychloride (7 ml, 72 mmol) was added dropwise to a stirring solution of the *N*-arylacetamide **8** (1.79 g, 10 mmol) in DMF (2 ml, 26 mmol) at 0 $^\circ\text{C}$ under a nitrogen atmosphere. The resulting mixture was heated at 130 $^\circ\text{C}$ for 2 h, cooled and poured on to ice-water (150 ml) to precipitate a yellow solid. The mixture was carefully treated with concentrated ammonia to pH 9, extracted with ethyl acetate (3 x 80 ml) and the combined organic extracts were dried (K_2CO_3), evaporated *in vacuo* and purified by flash chromatography on silica, eluting with ethyl acetate/light petroleum (1:9 then 1:4) to give the *title compound 6* (1.86 g, 79%) as fluorescent yellow needles, mp 132-133 $^\circ\text{C}$; (Found: MH^+ , 236.0489, $\text{C}_{12}\text{H}_{10}^{35}\text{ClNO}_2 + \text{H}$ requires 236.0478); ν_{max} (CHCl_3)/ cm^{-1} 2964, 2938, 2872, 2842, 1692, 1613, 1587, 1575, 1484, 1462, 1438, 1382, 1349, 1322, 1276, 1174, 1151, 1106, 1036, 1004, 977, 951, 854, 823, 636; δ_{H} (400 MHz; CDCl_3) 10.48 (1 H, s, CHO), 9.00 (1 H, s, 4-H), 7.54 (1 H, d, J 8.0, ArH), 6.76 (1 H, d, J 8.0, ArH), 3.98 (3 H, s, OCH_3), 2.62 (3 H, s, CH_3); δ_{C} (100 MHz; CDCl_3) 189.3 (CH), 155.1 (C), 149.6 (C), 148.8 (C), 135.8 (C), 134.0 (CH), 128.0 (CH), 124.9 (CH), 119.2 (C), 105.4 (CH), 55.9 (Me), 17.1 (Me); m/z (ESI) 270/268 (MNa^+ , 30/100%), 238/236 (MH^+ , 7/20%).

2-Chloro-3-hydroxy-5-methoxy-8-methylquinoline (9). Peracetic acid (8.75 ml, 40% in acetic acid/water, 50 mmol) in chloroform (125 ml) was added dropwise to a yellow solution of the quinoline-3-carbaldehyde **6** (2.5 g, 10.6 mmol) in chloroform (125 ml) under a nitrogen atmosphere with vigorous stirring. The mixture was stirred for 3 days, extracted with water (2 x 150 ml), and evaporated *in vacuo*. Methanol/THF (2:1) (60 ml), water (10 ml) and potassium hydrogen carbonate (2 g) were added sequentially to the yellow residue and the darkening mixture was stirred for 2 h, evaporated *in vacuo*, and partitioned between ethyl acetate (100 ml) and water (50 ml). The aqueous layer was further extracted with ethyl acetate, dried (MgSO_4), evaporated *in vacuo* and purified by flash chromatography on silica, eluting with ethyl

acetate/light petroleum (4:96 then 15:85) to give the *title compound 9* (485 mg, 46%) as a pale yellow solid, mp 139-140 °C; (Found: MH^+ , 224.0476, $\text{C}_{11}\text{H}_{10}^{35}\text{ClNO}_2 + \text{H}$ requires 224.0473); $\nu_{\text{max}}(\text{CHCl}_3)/\text{cm}^{-1}$ 3562, 2940, 2841, 1608, 1479, 1464, 1402, 1372, 1347, 1329, 1272, 1150, 1100, 950, 900; δ_{H} (400 MHz; CDCl_3) 8.01 (1 H, s, 4-H), 7.29 (1 H, d, J 8.0, ArH), 6.72 (1 H, d, J 8.0, ArH), 5.75 (1 H, s, OH), 3.95 (3 H, s, OCH_3), 2.63 (3 H, s, CH_3); δ_{C} (100 MHz; CDCl_3) 152.7 (C), 144.3 (C), 142.6 (C), 140.2 (C), 127.9 (C), 127.3 (CH), 121.8 (C), 114.7 (CH), 104.9 (CH), 55.7 (Me), 17.3 (Me); m/z (ESI) 226/224 (MH^+ , 30/100%).

2-Chloro-3-isopropoxy-5-methoxy-8-methylquinoline (10a). 2-Iodopropane (68 μl , 0.68 mmol) was added to a stirring solution of the 3-hydroxyquinoline **9** (100 mg, 0.45 mmol) and potassium carbonate (200 mg, 1.45 mmol) at room temperature under a nitrogen atmosphere. The mixture was stirred for 6 h at 75 °C, cooled, poured on to ice-water (50 ml), and extracted with ethyl acetate (2 x 15 ml). The combined organic extracts were dried (MgSO_4), evaporated *in vacuo* and purified by flash chromatography on silica, eluting with ethyl acetate/light petroleum (2:98) to give the *title compound 10a* (115 mg, 97%) as a colourless crystalline solid, mp 140-141 °C; (Found: MH^+ , 266.0959, $\text{C}_{14}\text{H}_{16}^{35}\text{ClNO}_2 + \text{H}$ requires 266.0942); $\nu_{\text{max}}(\text{CHCl}_3)/\text{cm}^{-1}$ 2981, 2938, 1839, 1593, 1478, 1463, 1455, 1368, 1321, 1270, 1153, 1139, 1108, 1048, 980, 954, 884, 844; δ_{H} (400 MHz; CDCl_3) 7.83 (1 H, s, 4-H), 7.28 (1 H, d, J 8.0, ArH), 6.72 (1 H, d, J 8.0, ArH), 4.74 (1 H, sept, J 6.1, OCHMe_2), 3.96 (3 H, s, OCH_3), 2.65 (3 H, s, CH_3), 1.47-1.45 (6 H, d, J 6.1, $\text{OCH}(\text{CH}_3)_2$); δ_{C} (100 MHz; CDCl_3) 152.4 (C), 146.7 (C), 143.7 (C), 141.9 (C), 127.9 (C), 127.0 (CH), 120.8 (C), 112.7 (CH), 104.8 (CH), 72.3 (CH), 55.7 (Me), 21.8 (Me), 17.3 (Me); m/z (ESI) 268/266 (MH^+ , 30/100%).

3-Acetoxy-2-chloro-5-methoxy-8-methylquinoline (10b). Acetic anhydride (83 μl , 0.88 mmol) was added to a solution of the 3-hydroxyquinoline **9** (100 mg, 0.45 mmol) and potassium carbonate (340 mg, 2.50 mmol) in anhydrous DMF at room temperature under a nitrogen atmosphere. The resulting yellowish mixture was stirred at 50 °C for 6 h, cooled, poured on to ice-water (15 ml) and extracted with ethyl acetate (2 x 10 ml). The combined organic extracts were dried (MgSO_4), evaporated *in vacuo* and purified by flash chromatography on silica, eluting with ethyl acetate/light petroleum (5:95) to give the *title compound 10b* as a colourless solid (105 mg, 93%), mp 175-176 °C; (Found: MH^+ , 266.0558, $\text{C}_{13}\text{H}_{12}^{35}\text{ClNO}_3 + \text{H}$ requires 266.0578); $\nu_{\text{max}}(\text{CHCl}_3)/\text{cm}^{-1}$ 2960, 2938, 2840, 1774, 1616, 1598, 1463, 1439, 1393, 1359, 1324, 1272, 1150, 1107, 1080, 1041, 1008, 951, 919, 875, 824, 645; δ_{H} (400 MHz; CDCl_3) 8.28 (1 H, s, 4-H), 7.42 (1 H, d, J 8.0, ArH), 6.77 (1 H, d, J 8.0, ArH), 3.95 (3 H, s, OCH_3), 2.65 (3 H, s, CH_3), 2.42 (3 H, s, COCH_3); δ_{C} (100 MHz; CDCl_3) 168.6 (C), 153.2 (C), 145.3 (C), 144.1 (C), 139.7 (C), 129.8 (CH), 128.0 (CH), 125.3 (CH), 120.4 (C), 105.2 (CH), 55.7 (Me), 20.7 (Me), 17.3 (Me); m/z (ESI) 268/266 (MH^+ , 30/100%).

General procedure for the ozonolysis of quinolines **6**, **10a** and **10b**

A stream of ozone in oxygen was bubbled into a slowly stirring solution of the quinoline (1 mmol) in dichloromethane at -20 °C for 20-30 min until the solution was saturated. The gas flow was stopped and stirring was continued for a further 30 min. The solution was flushed with oxygen for 10 min, then nitrogen for 5 min and treated with dimethyl sulfide (15 mmol) at -20

°C, stirred for 12 h at room temperature, evaporated *in vacuo* and purified by flash chromatography, eluting with ethyl acetate/light petroleum (2:98 to 10:90) to give the compounds **11a-c**.

2-Acetyl-6-chloro-5-isopropoxynicotinic acid methyl ester (11a). (111 mg, 39%) as a yellow oil; (Found: MH^+ , 272.0689. $\text{C}_{12}\text{H}_{14}^{35}\text{ClNO}_4 + \text{H}$ requires 272.0690); $\nu_{\text{max}}(\text{CHCl}_3)/\text{cm}^{-1}$ 2981, 2938, 1839, 1593, 1478, 1463, 1455, 1368, 1321, 1270, 1153, 1139, 1108, 1048, 980, 954, 884, 844; δ_{H} (400 MHz; CDCl_3) 7.27 (1 H, s, 4-H), 4.67 (1 H, sept, J 6.1, $\text{OCH}(\text{CH}_3)_2$), 3.91 (3 H, s, OCH_3), 2.64 (3 H, s, COCH_3), 1.41 (6 H, d, J 6.1, $\text{CH}(\text{CH}_3)_2$); δ_{C} (100 MHz; CDCl_3) 197.7 (C), 167.4 (C), 152.2 (C), 143.5 (C), 141.1 (C), 128.8 (C), 119.8 (CH), 73.1 (Me), 53.2 (Me), 26.5 (Me), 21.7 (Me); m/z (ESI) 274/272 (MH^+ , 30/100%).

5-Acetoxy-2-acetyl-6-chloronicotinic acid methyl ester (11b). (117 mg, 45%) as a yellow oil; (Found: MH^+ , 294.0142, $\text{C}_{11}\text{H}_{10}^{35}\text{ClNO}_5 + \text{Na}$ requires 294.0140); $\nu_{\text{max}}(\text{CHCl}_3)/\text{cm}^{-1}$ 2955, 2399, 1780, 1733, 1586, 1557, 1370, 1359, 1292, 1174, 1131, 1099, 1026, 963, 857; δ_{H} (400 MHz; CDCl_3) 7.82 (1 H, s, 4-H), 3.88 (3 H, s, OCH_3), 2.64 (3 H, s, OCOCH_3), 2.38 (3 H, s, COCH_3); δ_{C} (100 MHz; CDCl_3) 209.9 (C), 166.7 (C), 150.8 (C), 144.6 (C), 143.8 (C), 132.0 (CH), 52.2 (Me), 26.2 (Me), 19.6 (Me); m/z (ESI) 296/294 (MNa^+ , 30/100%).

2-Acetyl-6-chloro-5-formylnicotinic acid methyl ester (11c). (108 mg, 45%) as a yellow oil; (Found: MH^+ , 242.0222. $\text{C}_{10}\text{H}_8^{35}\text{ClNO}_4 + \text{H}$ requires 242.0220); $\nu_{\text{max}}(\text{CHCl}_3)/\text{cm}^{-1}$ 2956, 2881, 1733, 1703, 1585, 1548, 1465, 1380, 1358, 1278, 1132, 1098, 966, 914; δ_{H} (400 MHz; CDCl_3) 10.77 (1 H, s, CHO), 8.67 (1 H, s, 4-H), 3.54 (3 H, s, OCH_3), 2.15 (3 H, s, COCH_3); δ_{C} (100 MHz; CDCl_3) 198.2 (C), 187.4 (C), 164.8 (C), 159.4 (C), 154.4 (C), 139.5 (CH), 129.1 (C), 125.4 (C), 53.3 (Me), 27.5 (Me); m/z (ESI) 244/242 (MH^+ , 30/100%).

Acknowledgements

We thank the EPSRC for funding.

References and Footnotes

1. Bagley, M. C.; Dale, J. W.; Merritt, E. A.; Xiong, X.; *Chem. Rev.* **2005**, *105*, 685.
2. Shimanaka, K.; Kinoshita, N.; Inuma, H.; Hamada, M.; Takeuchi, T.; *J. Antibiotics* **1994**, *47*, 668.
3. Clough, B.; Strath, M.; Preiser, P.; Denny, P.; Wilson, R. J. M.; *FEBS Lett.* **1997**, *406*, 123.
4. (a) Xing, Y. Y.; Draper, D. E.; *Biochemistry* **1996**, *35*, 1581. (b) Heffron, S. E.; Journak, F.; *Biochemistry* **2000**, *39*, 37.
5. Synthetic efforts have focused on the Bycroft-Gowland proposed structure of micrococcin P1, which has since been revised by Bagley and Merritt. Hence, the NMR spectra of synthetic micrococcin samples have not been found identical to those of the natural product;

- Shin, C.-G.; Okumura, K.; Shigekuni, M.; Nakamura, Y.; *Chem. Lett.* **1998**, 139. Okumura, K.; Nakamura, Y.; Shin, C.; *Bull. Chem. Soc. Jpn.* **1999**, 72, 1561. Okumura, K.; Ito, A.; Yoshioka, D.; Shin, C.-G.; *Heterocycles* **1998**, 48, 1319. Okumura, K.; Suzuki, T.; Nakamura, Y.; Shin, C.-G.; *Bull. Chem. Soc. Jpn.* **1999**, 72, 2483. Ciufolini, M. A.; Shen, Y.-C.; *Org. Lett.* **1999**, 1, 1843. Bagley, M.C.; Merritt, E. A.; *J. Antibiot.* **2004**, 57, 829.
- (a) Nicolaou, K. C.; Safina, B. S.; Zak, M.; Lee, S. H.; Nevalainen, M.; Bella, M.; Estrada, A. A.; Funke, C.; Zecri, F. J.; Bulat, S.; *J. Am. Chem. Soc.* **2005**, 127, 11159. (b) Nicolaou, K. C.; Zak, M.; Safina, B. F.; Estrada, A. A.; Lee, S. H.; Nevalainen, M.; *J. Am. Chem. Soc.* **2005**, 127, 11176.
 - Bagley, M. C.; Bashford, K. E.; Hesketh, C. L.; Moody, C. J.; *J. Am. Chem. Soc.* **2000**, 122, 3301.
 - Hughes, R. A.; Thompson, S. P.; Alcaraz, L.; Moody, C. J.; *J. Am. Chem. Soc.* **2005**, 127, 15644.
 - Koerber-Plé, K.; Massiot, G.; *Synlett* **1994**, 759.
 - Shin, C.; Nakamura, Y.; Yamada, Y.; Yonezawa, Y.; Umemura, K.; Yoshimura, J.; *Bull. Chem. Soc. Jpn.* **1995**, 68, 3151.
 - Koerber-Plé, K.; Massiot, G.; *J. Heterocycl. Chem.* **1995**, 32, 1309.
 - Umemura, K.; Tate, T.; Yamaura, M.; Yoshimura, J.; Yonezawa, Y.; Shin, C.; *Synthesis* **1995**, 1423.
 - Shin, C.; Yamada, Y.; Hayashi, K.; Yonezawa, Y.; Umemura, K.; Tanji, T.; Yoshimura, J.; *Heterocycles* **1996**, 43, 891.
 - Umemura, K.; Noda, H.; Yoshimura, J.; Konn, A.; Yonezawa, Y.; Shin, C. G.; *Tetrahedron Lett.* **1997**, 38, 3539.
 - Umemura, K.; Noda, H.; Yoshimura, J.; Konn, A.; Yonezawa, Y.; Shin, C. G.; *Bull. Chem. Soc. Jpn.* **1998**, 71, 1391.
 - Ciufolini, M. A.; Shen, Y. C.; *J. Org. Chem.* **1997**, 62, 3804.
 - Heckmann, G.; Bach, T.; *Angew. Chem., Int. Ed.* **2005**, 44, 1199.
 - Moody, C. J.; Hughes, R. A.; Thompson, S. P.; Alcaraz, L.; *Chem. Commun.* **2002**, 1760.
 - (a) Bohlmann, F.; Rahtz, D.; *Chem. Ber.* **1957**, 90, 2265. (b) Bagley, M. C.; Chapaneri, K.; Dale, J. W.; Xiong, X.; Bower, J.; *J. Org. Chem.* **2005**, 70, 1389. (c) Bagley, M. C.; Dale, J. W.; Ohnesorge, M.; Xiong, X.; Bower, J.; *J. Comb. Chem.* **2003**, 5, 41.
 - Le Bas, M.-D.; Guéret, C; Perrio, C ; Lasne, M.-C. ; Barré, L ; *Synthesis* **2001**, 2495.
 - (a) Meth-Cohn, O.; Narine, B.; Tarnowski, B.; *Tetrahedron Lett.* **1979**, 33, 3111. (b) Meth-Cohn, O.; Narine, B.; Tarnowski, B.; *J. Chem. Soc., Perkin Trans. 1* **1981**, 1520.
 - Akhavan-Tafti, H.; DeSilva, R.; Arghavani, Z.; Eickholt, R. A.; Handley, R. S.; Schoenfelner, B. A.; Sugioka, K.; Sugioka, Y.; Schaapet, A. P.; *J. Org. Chem.* **1998**, 63, 930.